



HIGH-RESOLUTION REMOTE SENSING APPLIED TO MINERAL EXPLORATION IN
AUSTRALIA

VOLUME II

ANDREW R. GABELL

DEPARTMENT OF GEOLOGY AND GEOPHYSICS, UNIVERSITY OF ADELAIDE

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features.
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Figure 1.1 Location of test sites.

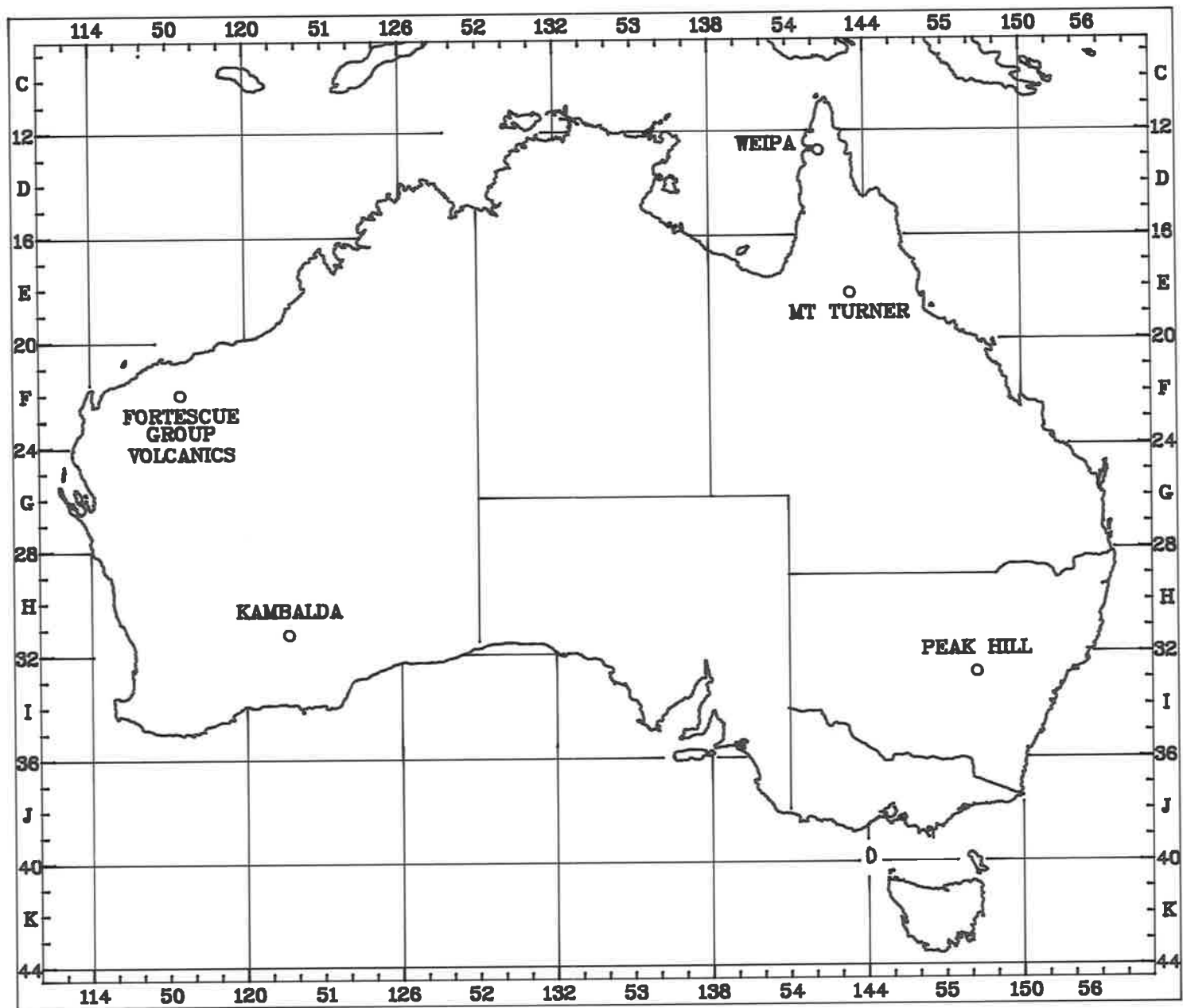


FIGURE 1.1

Figure 2.1 The electromagnetic spectrum and atmospheric transmission.

The wavelength regions defined in section 1.5 are shown, with their percentage atmospheric transmission. Landsat MSS and TM bandpasses are also shown for reference. The regions most utilised for remote sensing are the VNIR (.4 to 1.1 μ m), SWIR (1.1 to 2.5 μ m) and MIR (8 to 14 μ m).

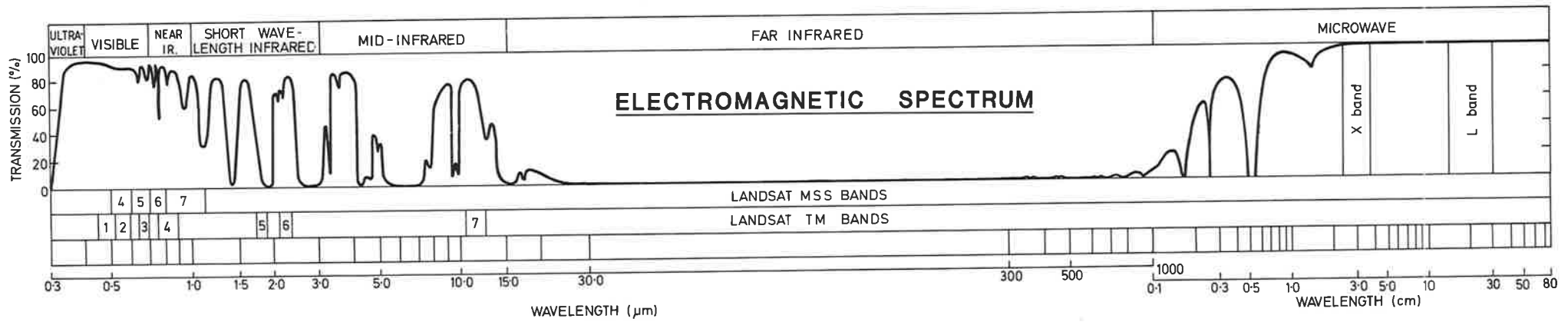
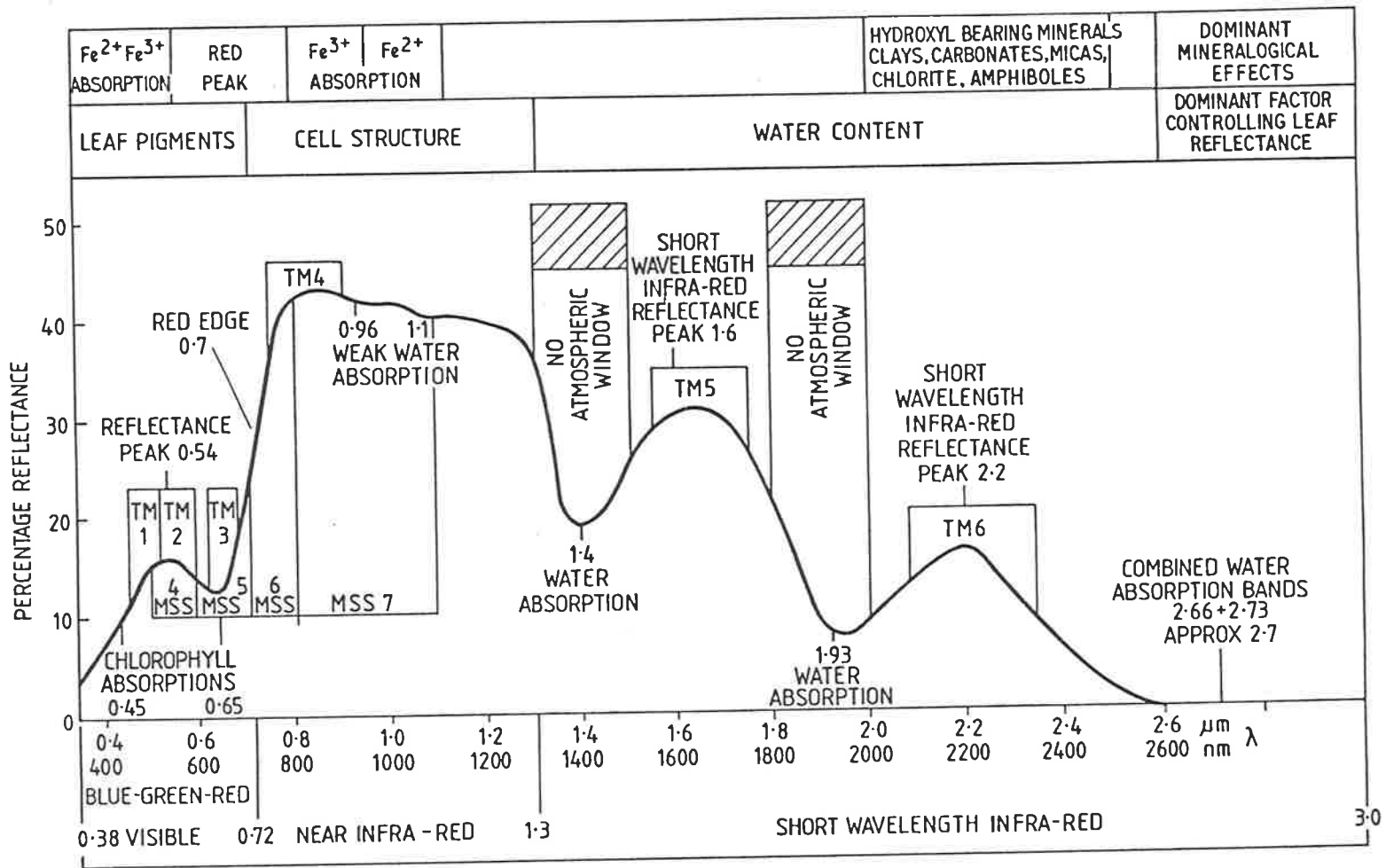


Figure 2.2 The spectral response of green vegetation in the VNIR and SWIR.

In addition to the reflectance of green vegetation, the cause of the spectral features in each region is shown. The spectral regions where mineralogical effects occur are also depicted, as are Landsat MSS and TM bandpasses.

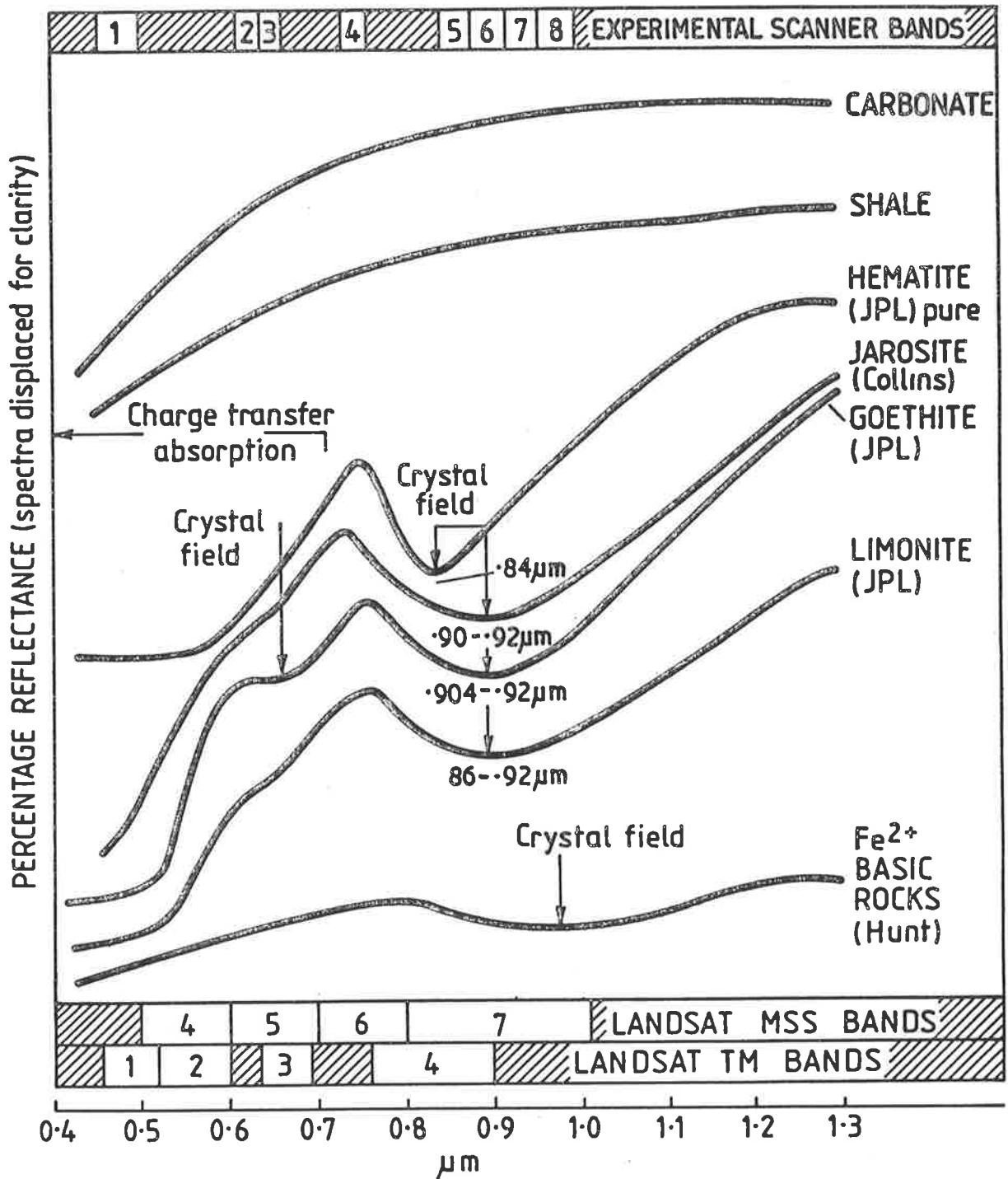


SPECTRAL RESPONSE OF GREEN VEGETATION AND LANDSAT BANDS



Figure 2.3 Typical VNIR spectra of the iron oxides.

The absorption features in iron-bearing rocks occur at wavelengths that vary slightly because of crystal field or charge transfer effects. Typical absorption wavelengths are shown for some iron oxides and the sulphate, jarosite. Note the absence of diagnostic features in the spectra of the iron-free shale and carbonate rocks. Landsat MSS and TM bandpasses are included for reference.



TYPICAL VISIBLE AND NEAR INFRARED SPECTRA OF THE IRON OXIDES

Figures 2.4a&b MIR kaolinite transmission spectra.

Figure 2.4a depicts the fundamental O-H stretching absorption features (ν_{OH}).

Figure 2.4b shows the fundamental O-H bending absorption doublet, near 900 cm^{-1} . The doublet structure is due to the occupation by hydroxyls of two non-equivalent lattice sites.

KAOLINITE INFRARED TRANSMISSION SPECTRA

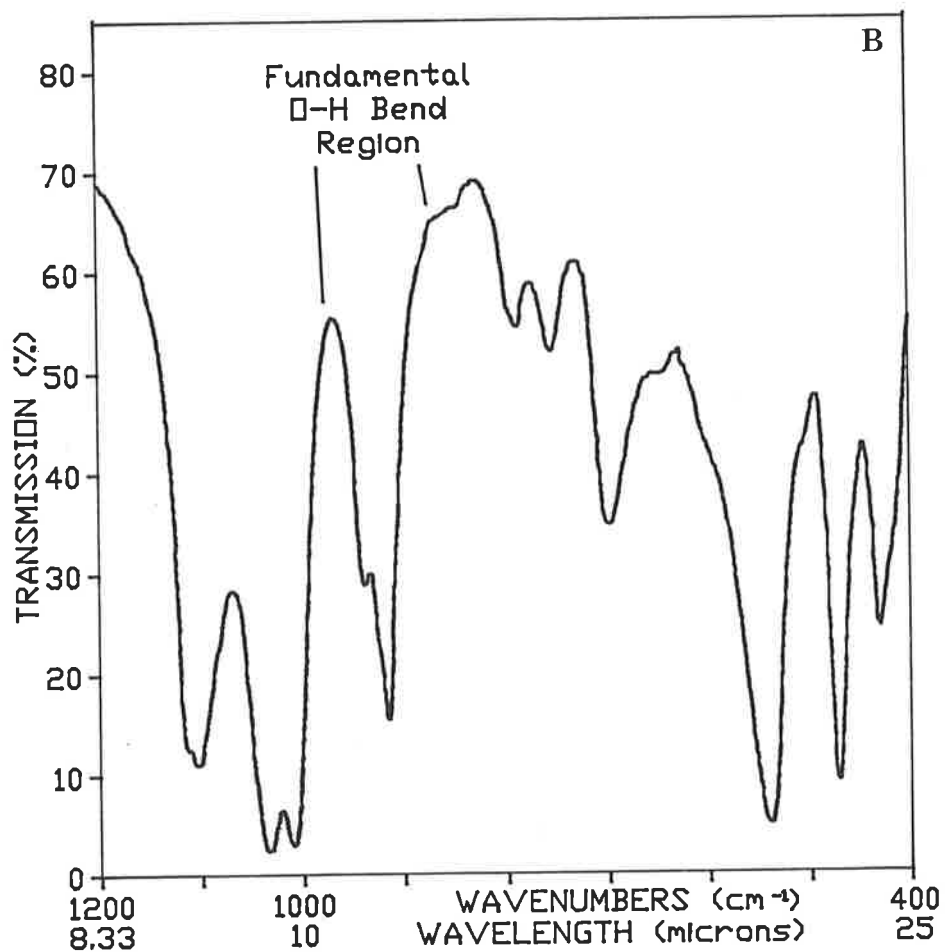
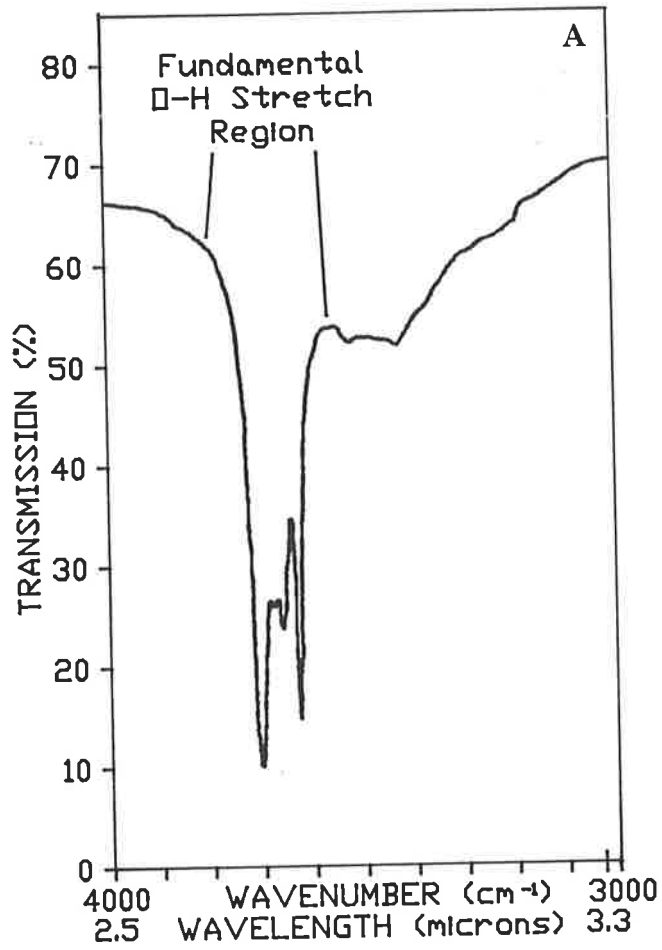


Figure 2.5 SWIR reflectance spectra - Al-OH minerals.

The SWIR reflectance spectra of some important dioctahedral (Al-OH) phyllosilicates are shown. In a remote sensing mode, the features at 1.4 and 1.9 μm are largely obscured by atmospheric absorption. The major absorption feature between 2.0 and 2.5 μm is caused by the combination of fundamental bending and stretching vibration frequencies. It is this region that is most useful for remote sensing. Note that the doublet structure exhibited by kaolinite is very similar to that of its fundamental O-H bending feature (Fig. 2.4b). The spectra are from the CSIRO/MXY spectral library, and have been displaced vertically for clarity.

REFLECTANCE SPECTRA - AL-OH MINERALS

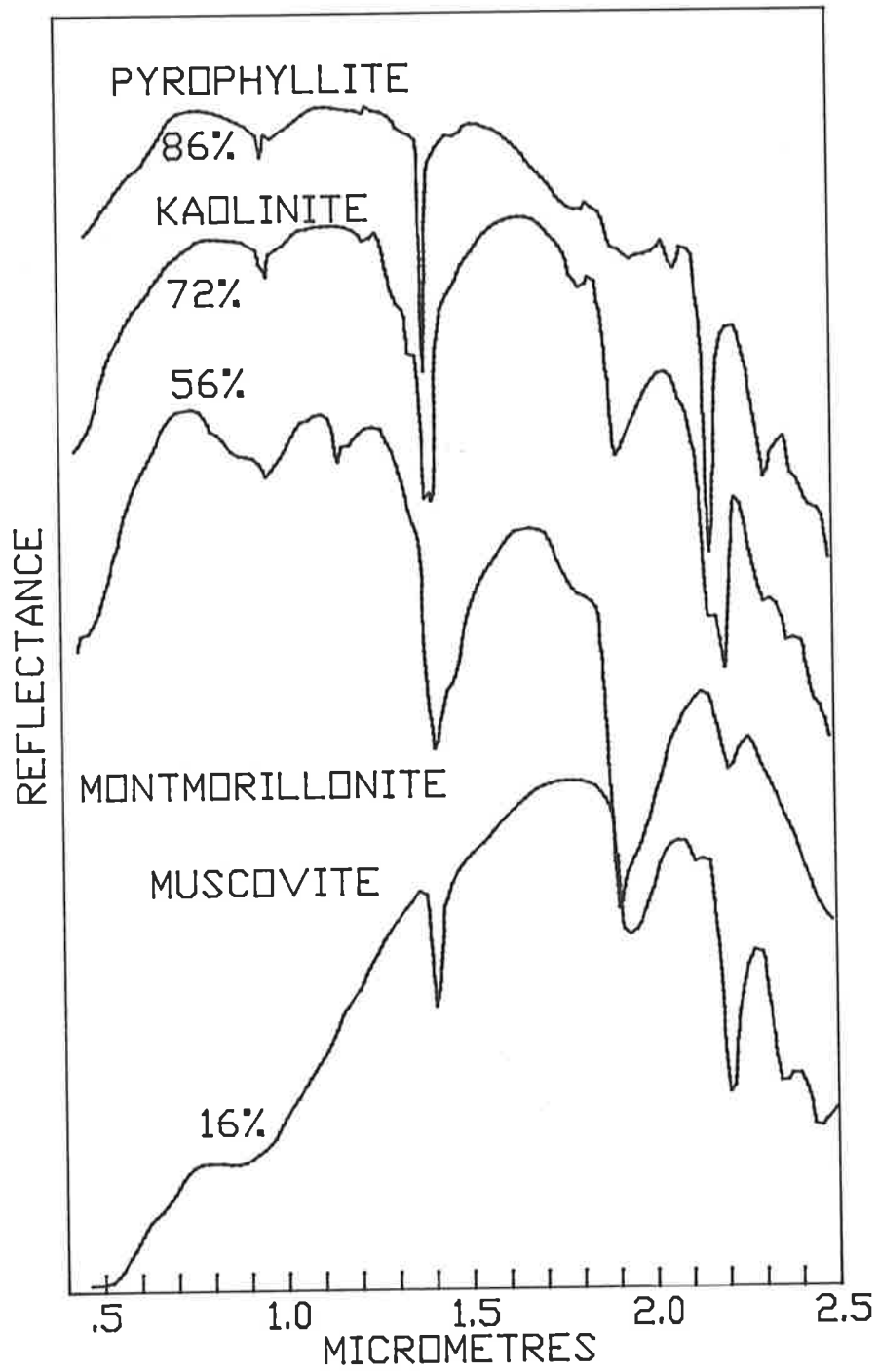
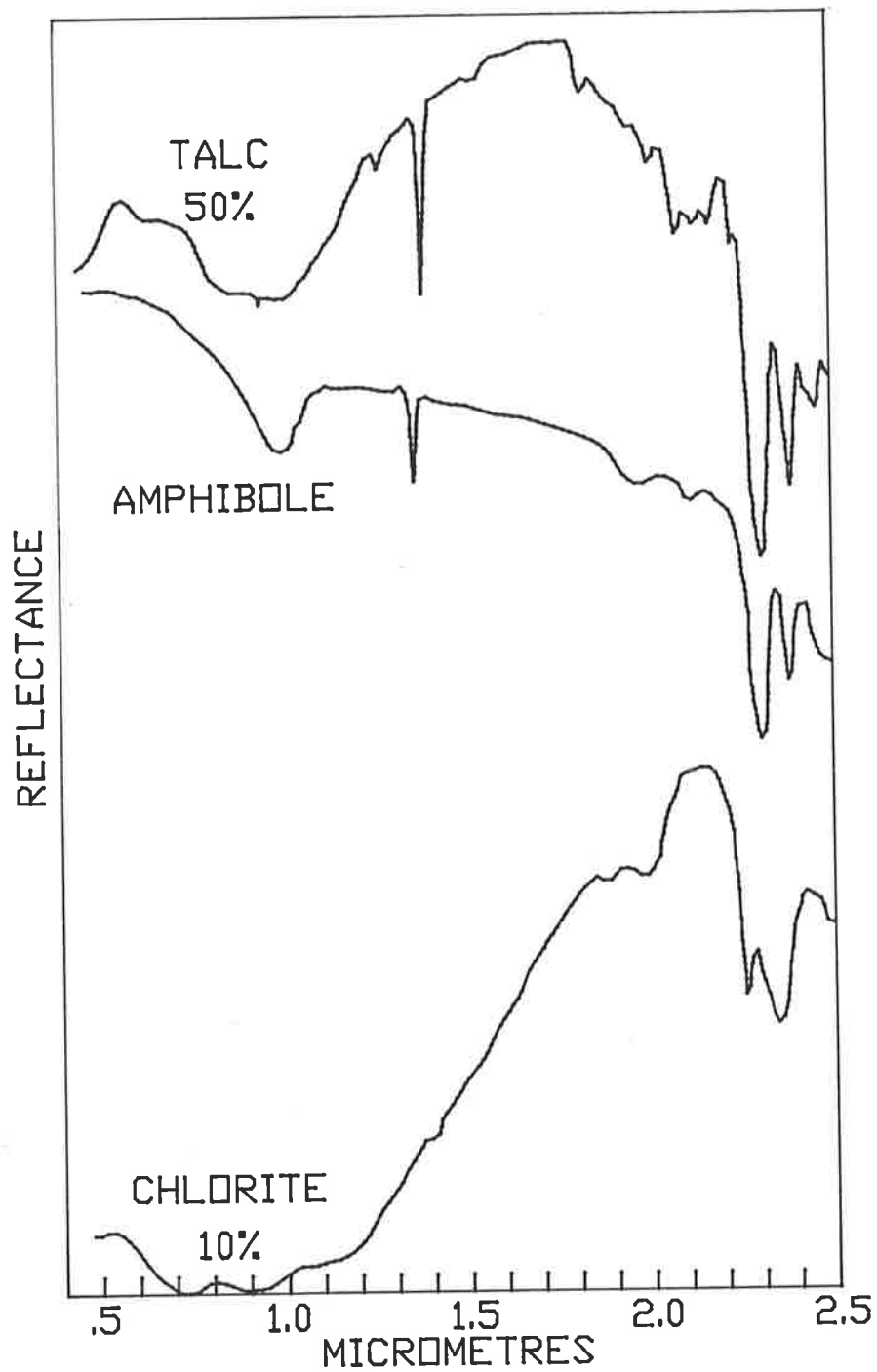


Figure 2.6 SWIR reflectance spectra - Mg-OH minerals.

The SWIR reflectance spectra of the trioctahedral (Mg-OH) phyllosilicates, talc and chlorite, and the Ca,Mg-bearing amphibole, tremolite, are depicted. These spectra are also from the CSIRO/MXY spectral library, and have been displaced vertically.

REFLECTANCE SPECTRA - MG-OH MINERALS



Figures 2.7a&b VNIR reflectance spectra of kaolinite.

Figure 2.7a shows the VNIR reflectance spectrum of a kaolinite sample from Alum Mountain, NSW. The $3\nu_{OH}$ absorption appears as a doublet centred on $.952\mu\text{m}$ and $.964\mu\text{m}$.

Figure 2.7b depicts the magnified inset of Fig. 2.7a. The absorption feature at $.75\mu\text{m}$ is barely detectable, but does represent significantly greater variation than the noise level. Although the theoretical position of the third overtone is $.71\mu\text{m}$, it is likely that this minimum represents the $4\nu_{OH}$ absorption feature.

VNIR REFLECTANCE SPECTRA OF KAOLINITE

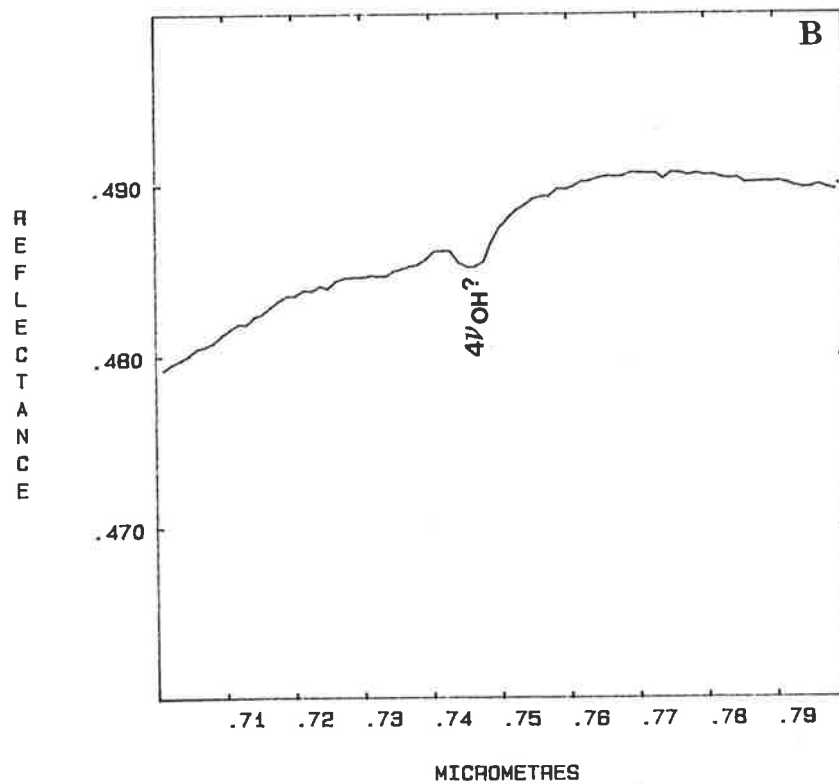
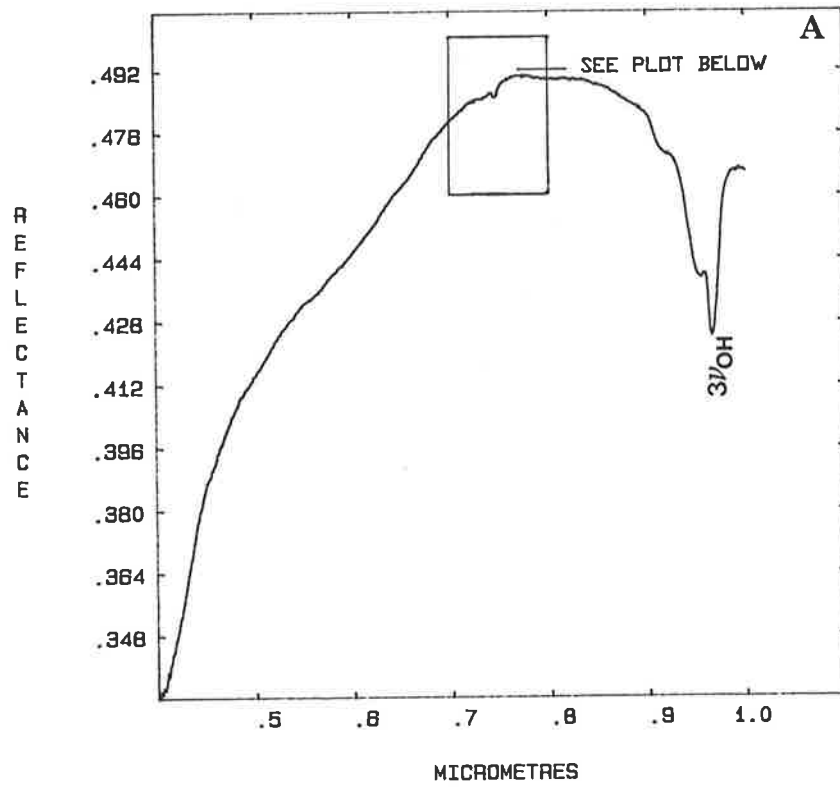


Figure 2.8 SWIR reflectance spectra - carbonates.

The SWIR reflectance spectra of magnesite (MgCO_3), dolomite ($(\text{Ca,Mg})\text{CO}_3$) and calcite (CaCO_3) are shown. Hunt and Salisbury (1971) ascribe the major absorption feature near $2.3\mu\text{m}$ to the third overtone of the antisymmetric mode of the carbonate ion ($3\nu_3$). The vertical line on the plot is centred on $2.32\mu\text{m}$, and is included as a reference point to highlight the progressive shift of the $3\nu_3$ minimum to longer wavelengths with increasing Ca content. The higher atomic weight of Ca relative to Mg leads to the slight decrease in frequency (or increase in wavelength) at which the carbonate ion vibrates. The spectra are from the CSIRO/MXY spectral library, and have been displaced vertically.

REFLECTANCE SPECTRA - CARBONATES

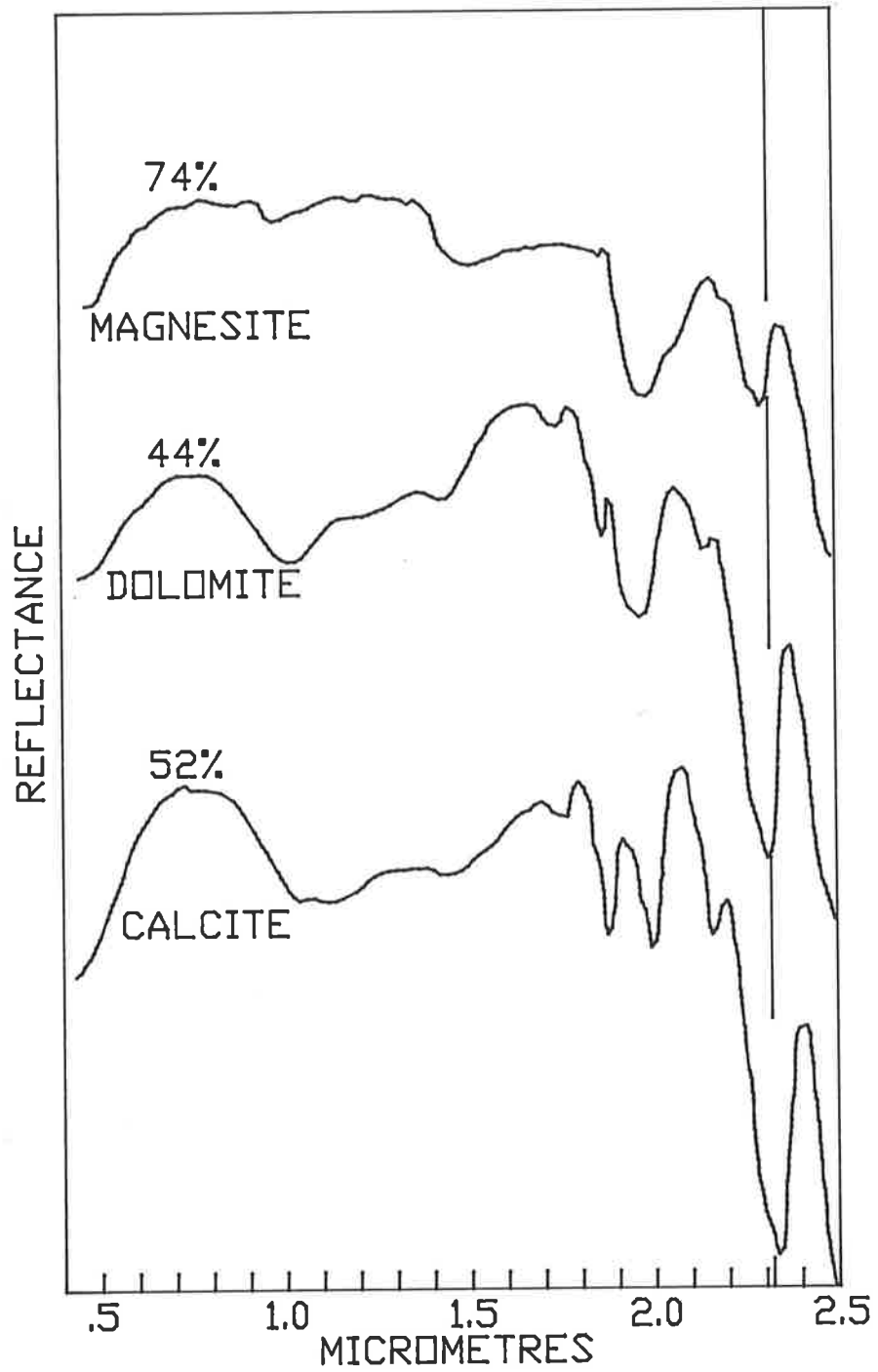


Figure 2.9 SWIR reflectance spectra - sulphates.

The SWIR reflectance spectra of jarosite ($\text{KFe}(\text{OH})_6(\text{SO}_4)_2$), alunite ($\text{Al}_3(\text{SO}_4)_2(\text{OH})_6$) and gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$) are depicted. Hunt et al. (1971) attribute all of the absorption features in alunite to various overtones and combination tones of Al-O-H, and those of gypsum to overtones and combination tones of water. The jarosite spectrum measured by Hunt et al. showed almost no absorption features, but those of the spectrum shown here are consistent with an origin due to Fe-O-H. (The absorption features of jarosite correlate well with those of alunite, but are shifted to longer wavelengths). Because the sulphate fundamentals occur at low frequencies (the highest being only 1100cm^{-1}), at least the fourth overtone would be required to produce features in the SWIR, making their detection unlikely. The spectra are from the CSIRO/MXY spectral library, and have been displaced vertically.

REFLECTANCE SPECTRA - SULPHATES

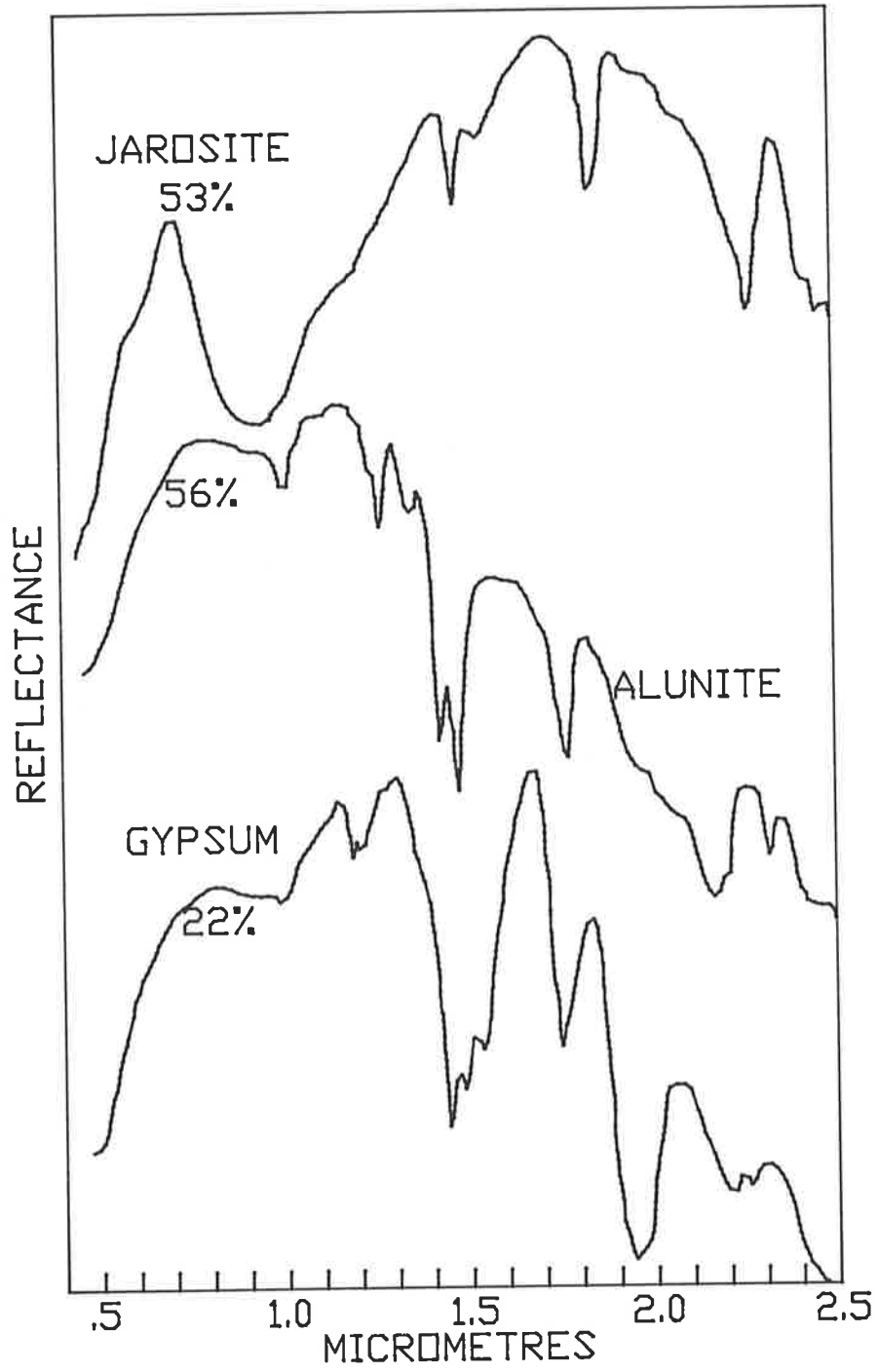


Figure 2.10 MIR spectra of felsic rocks (after Eberhardt et al., 1984).

The solid lines are emissivity curves for felsic rocks, which have an increasing feldspar content from the top curve downwards. They have been displaced for clarity, and superimposed on reflectance plots of a granite (crosses) and a quartzite (triangles).

Figure 2.11 MIR spectra of intermediate rocks (after Eberhardt et al., 1984).

The solid line is the emissivity spectrum of an andesite (plotted with a scale factor of 3). The reflectance spectrum of a diorite (diamonds) is shown for comparison.

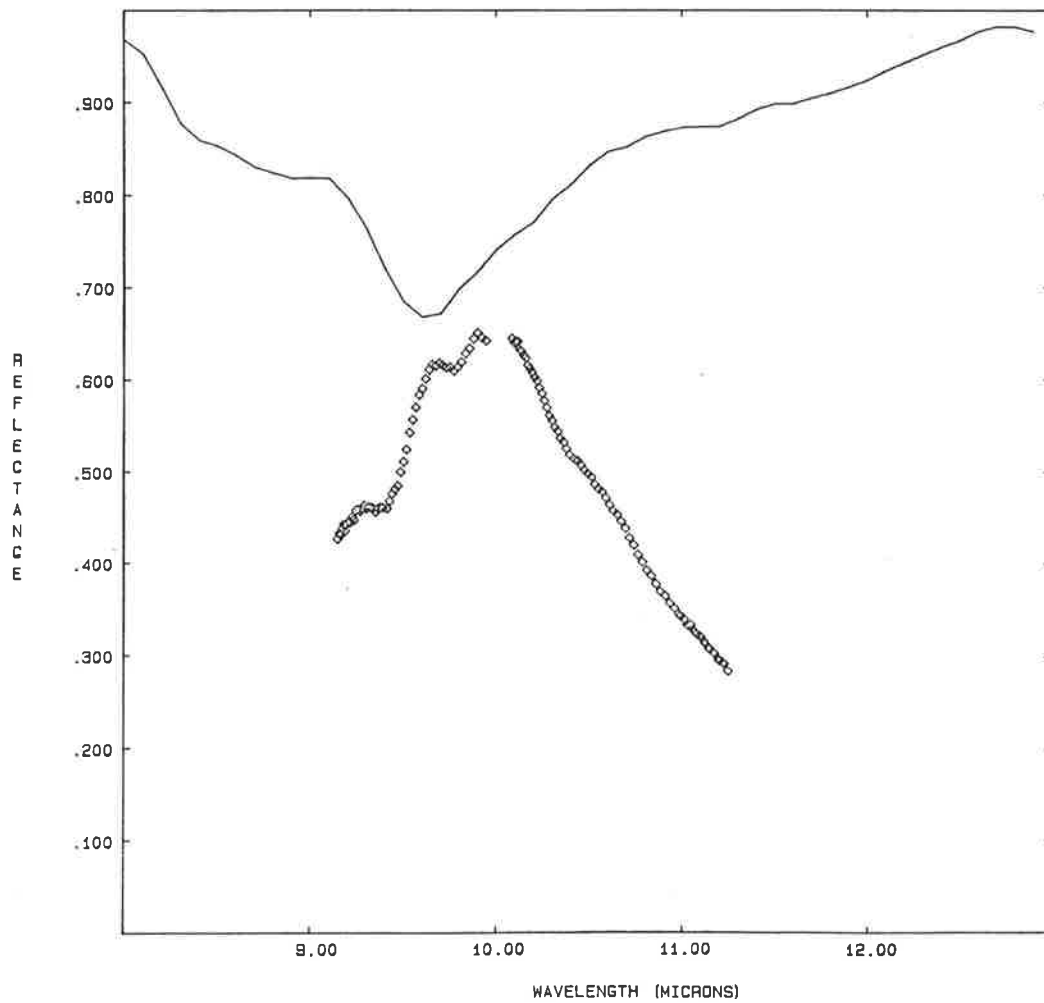
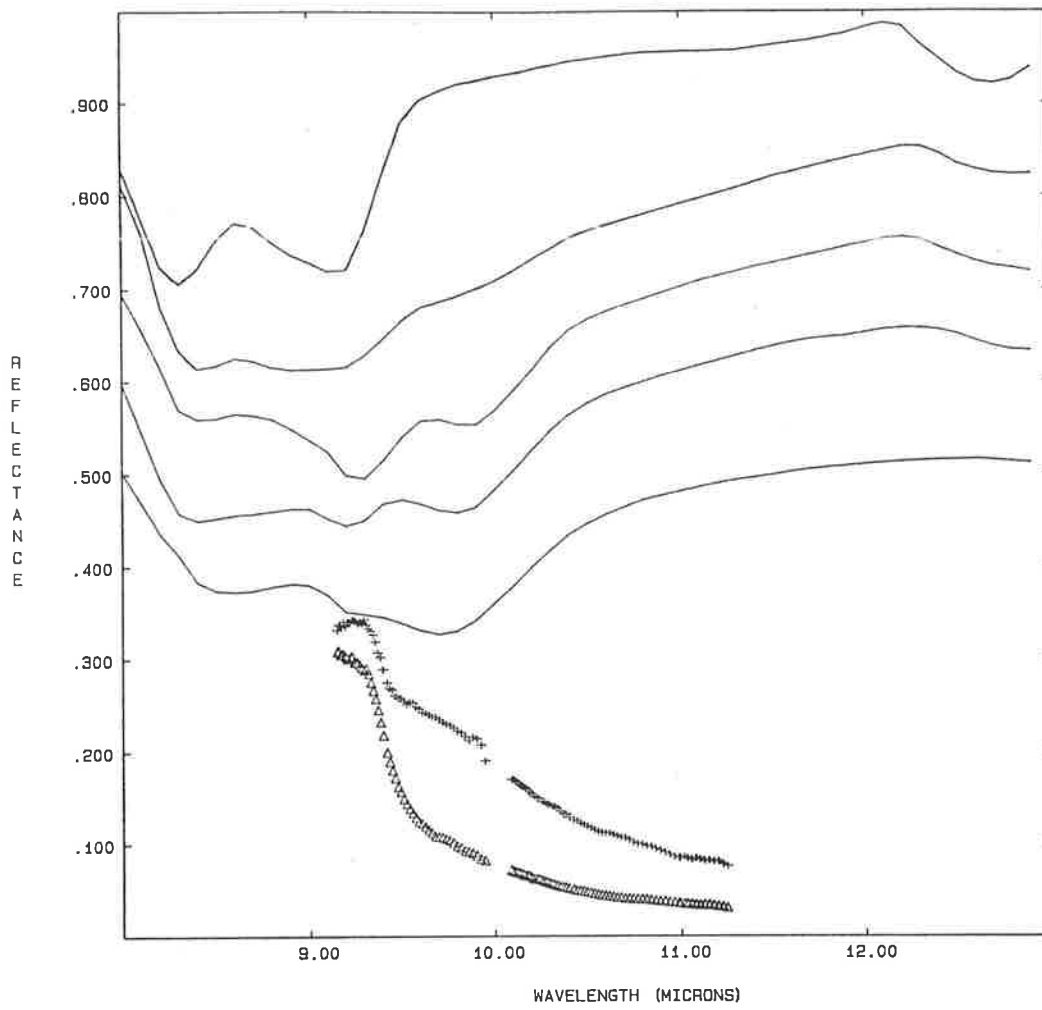


Figure 2.12 MIR spectra of serpentinites (after Eberhardt et al., 1984).

The emissivity spectrum (scale factor 3) shows very similar structure to the reflectance spectra of both the weathered (diamonds) and the fresh (crosses) surfaces of the serpentinite sample (both plotted with a scale factor of 2).

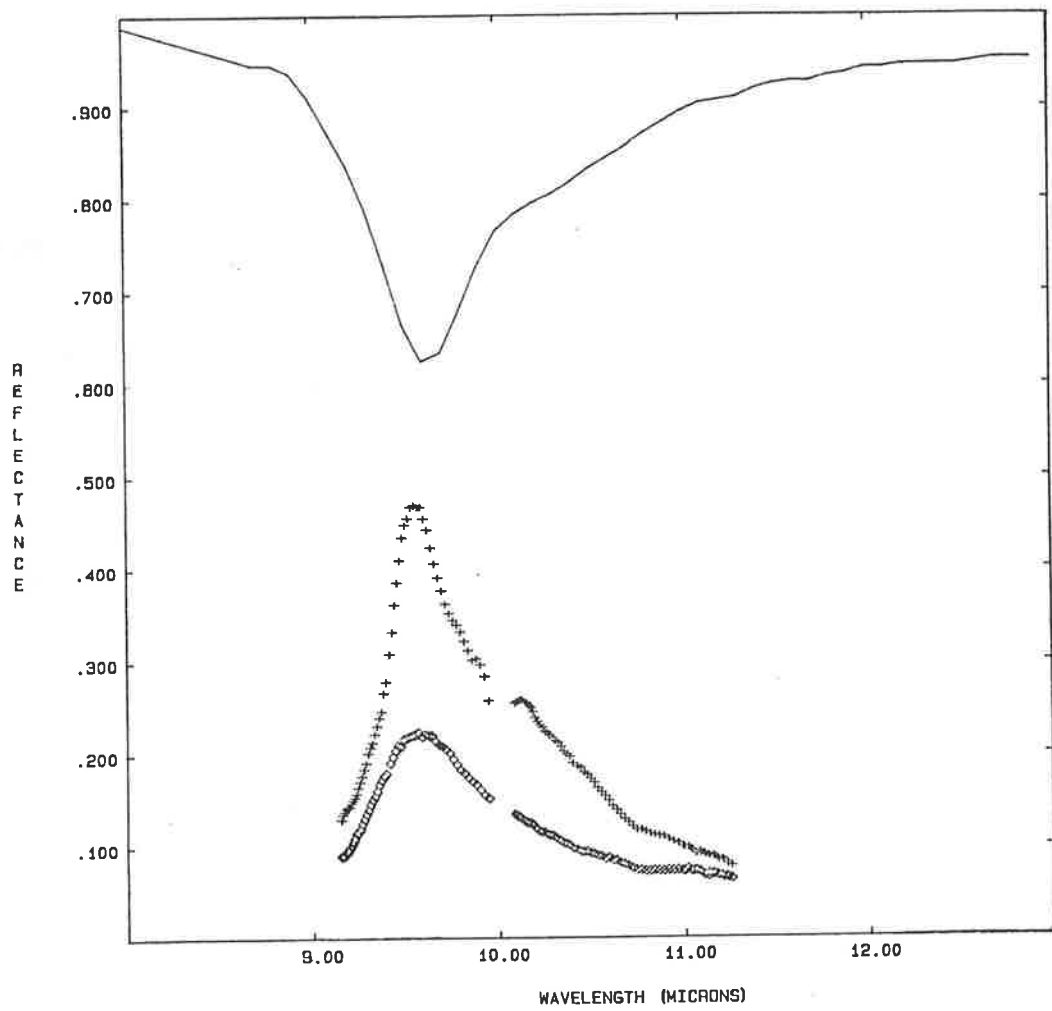


Figure 2.13 MIR spectra of ultramafic rocks (after Eberhardt
et al., 1984).

The solid line is an emissivity curve for a pyroxenite (scale factor 3). The reflectance spectra of a talc-carbonate rock (diamonds, scale factor 3.2) and a picrite, comprising amphibole and chlorite (crosses, scale factor 3) are shown for comparison.

Figure 2.14 MIR spectra of garnet-rich rocks (after Eberhardt
et al., 1984).

The emissivity spectrum (scale factor 5) and reflectance spectra (both scale factor 2.5) of weathered (diamonds) and fresh (crosses) surfaces of a garnetite are shown.

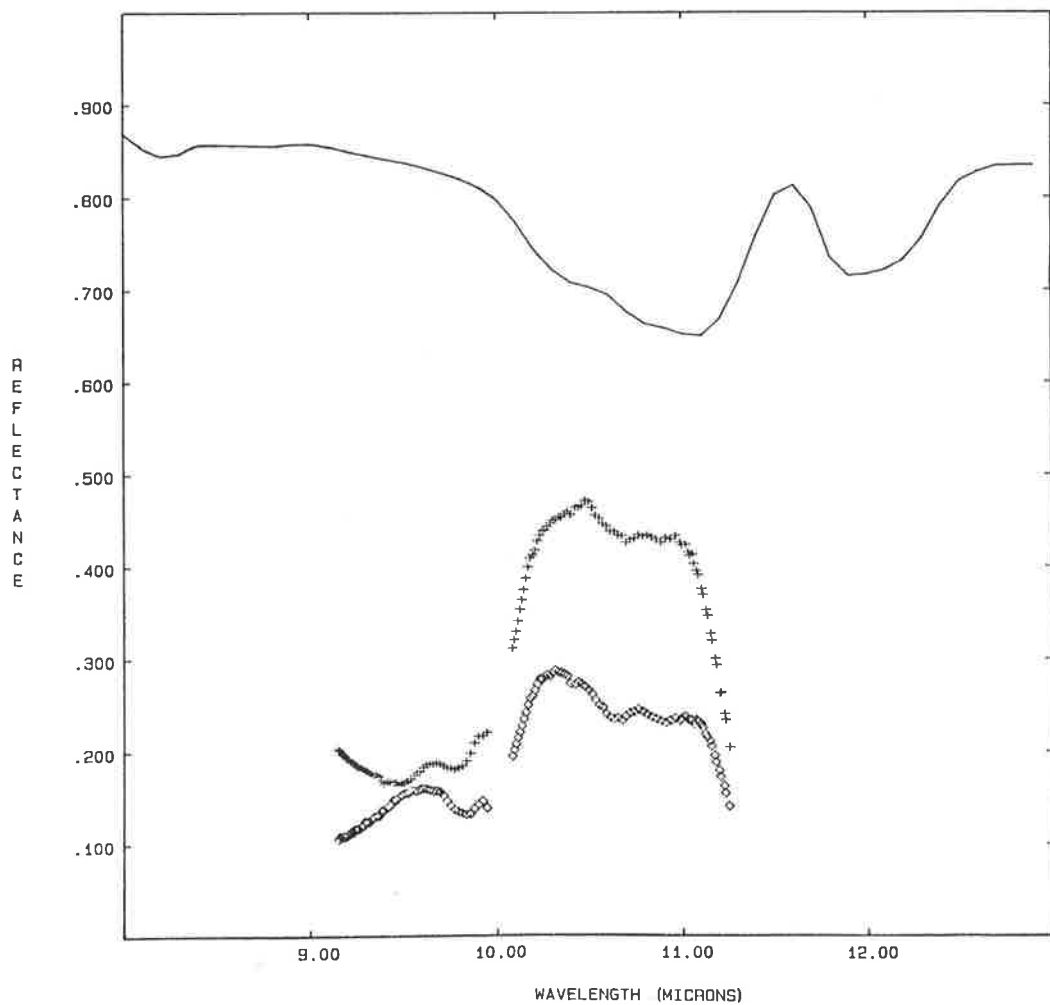
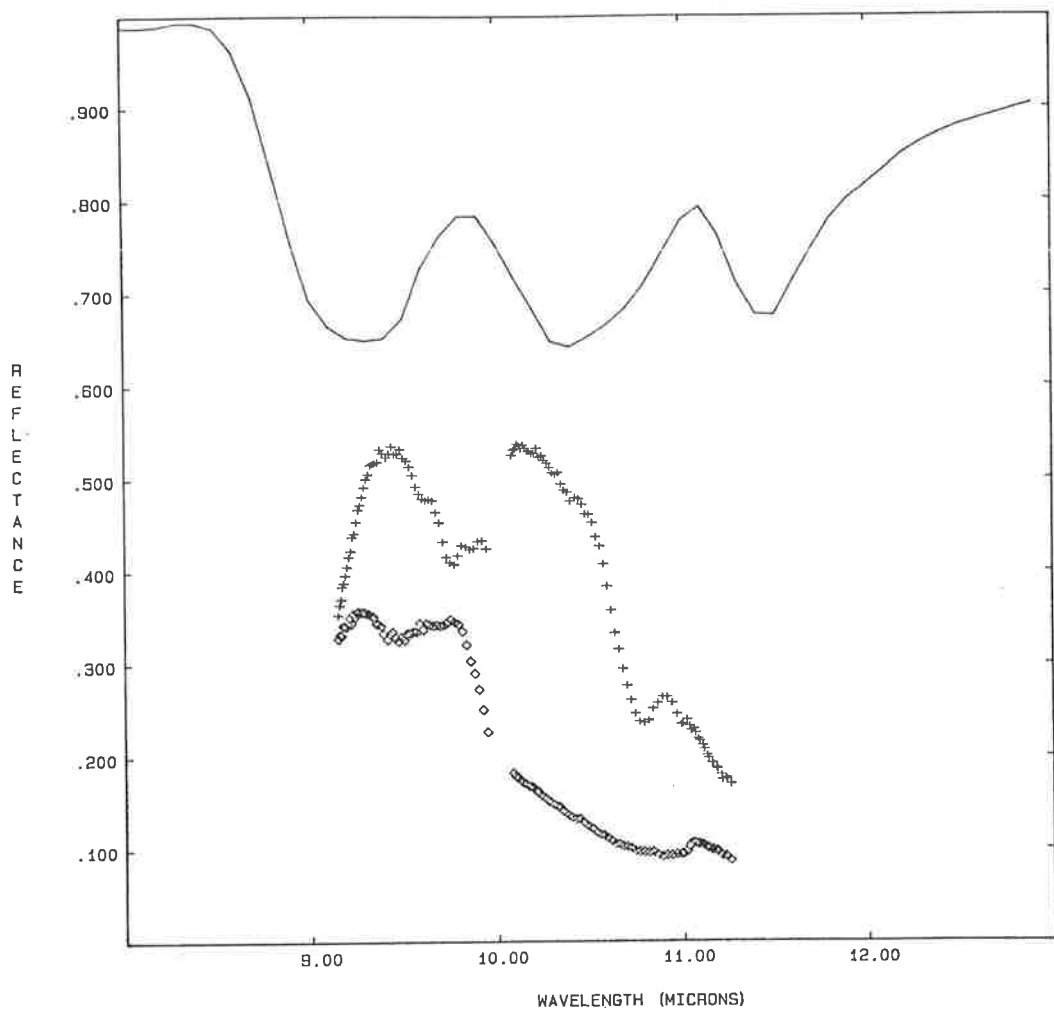
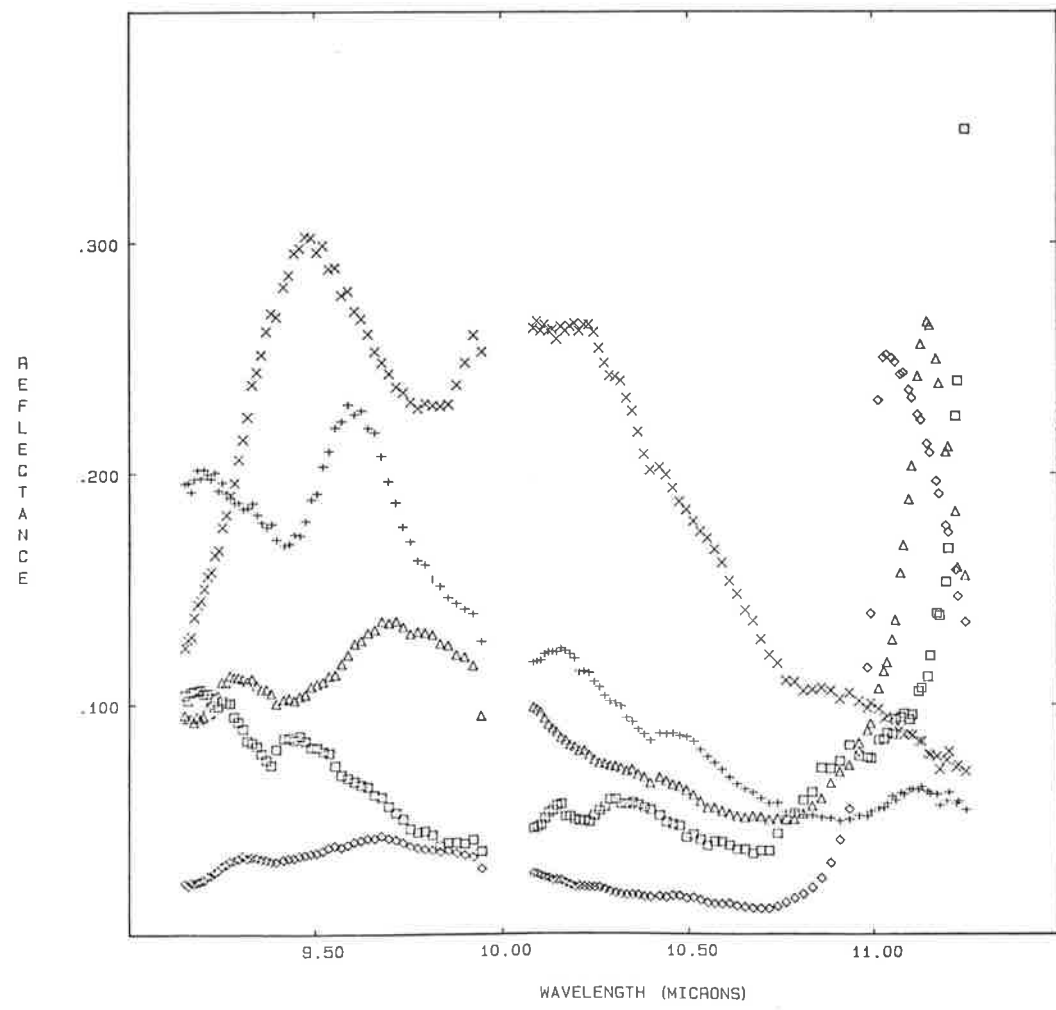
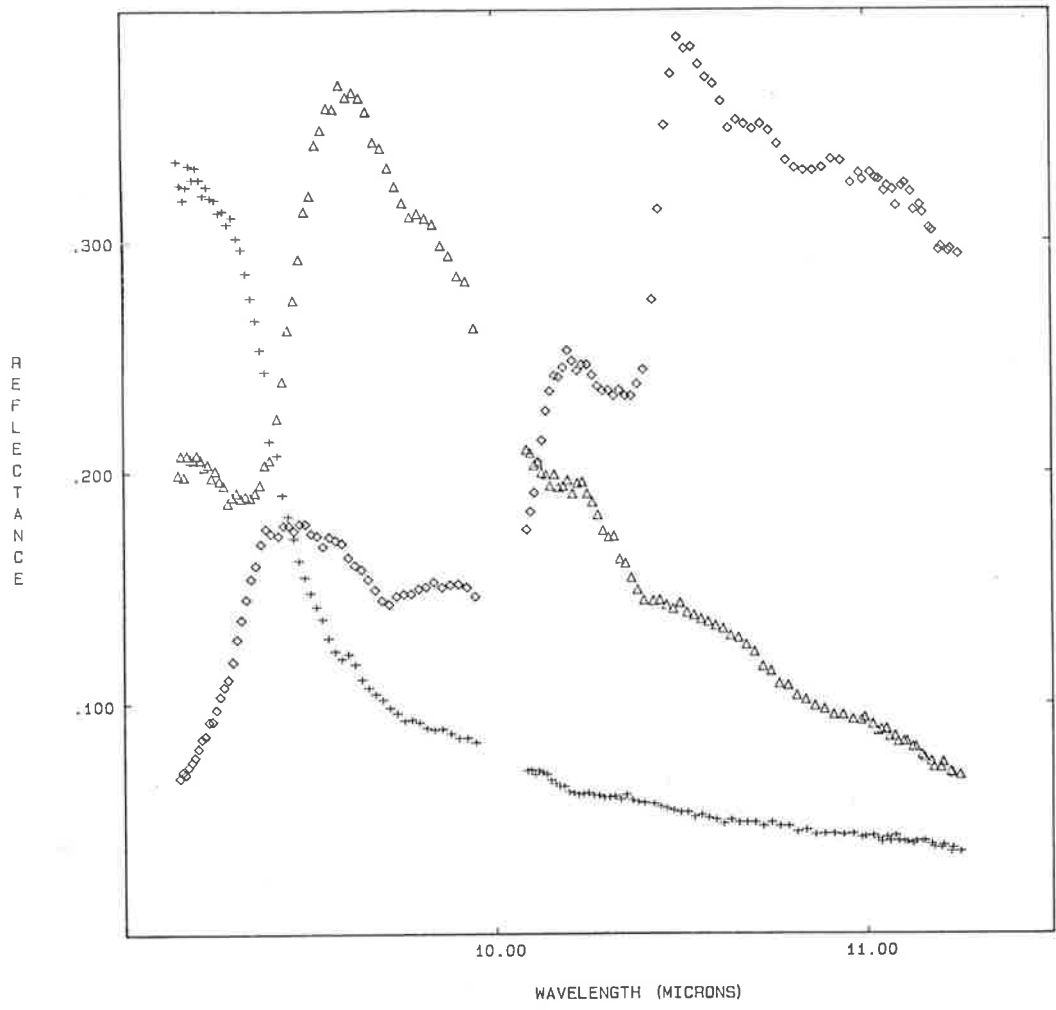


Figure 2.15 MIR reflectance spectra of minerals (after Eberhardt et al., 1984).

The reflectance spectra of quartz (crosses), plagioclase (triangles) and olivine (diamonds). The spectra have been rescaled (scale factors 4, 1 and 1.3 respectively) and offset (-5%, -10% and 0% respectively) for ease of display. Note that once again the wavelengths of the reflectance maxima increase with decreasing SiO_2 .

Figure 2.16 MIR reflectance spectra of minerals (after Eberhardt et al., 1984).

The reflectance spectra of muscovite (crosses), hornblende (x), calcite (squares), magnesite (diamonds) and dolomite (triangles). Spectra are rescaled (scale factors 2, 3, 2.5, 1.5 and 1.7 respectively) and offset (-10%, -10%, -15%, -7% and -25% respectively). Note the sharp reflectance maximum displayed by the carbonates near $11.1\mu\text{m}$, moving to progressively longer wavelengths with increasing Ca content (a similar effect to that observed in the SWIR ; see Fig. 2.8).

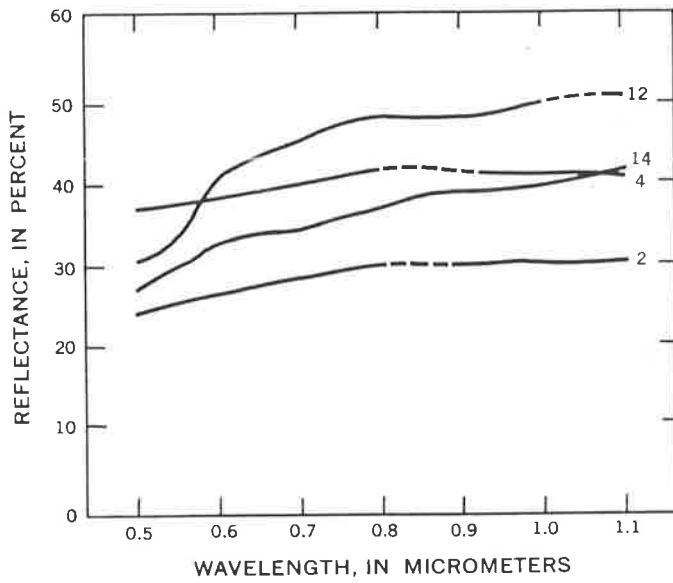


Figures 2.17a-c VNIR reflectance spectra of altered and unaltered rocks (after Rowan et al., 1974).

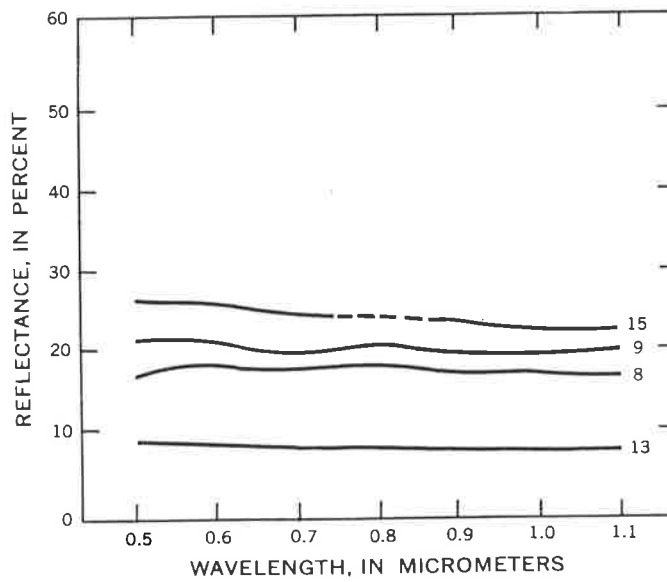
Figure 2.17a shows spectra of felsic rocks, rhyolite (12), granite (14), rhyolite (4) and granodiorite (2), all showing increasing reflectance with increasing wavelength.

Figure 2.17b depicts selected mafic rocks, serpentinite (15), gabbro (9), peridotite (8) and basalt (13) exhibiting overall decreasing reflectance with increasing wavelength.

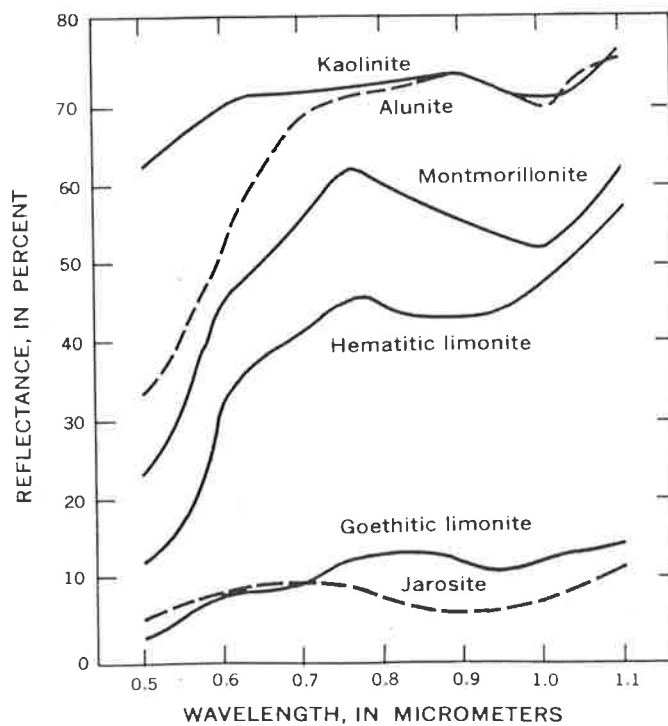
Figure 2.17c shows reflectance of some alteration minerals. The hematitic limonite spectrum was recorded in situ at Goldfield, Nevada ; other spectra are from various sources. Note the absorption features in the .8 to 1.0 μ m region.



A



B



C

Figure 2.18 VNIR and SWIR spectral reflectance curves of altered and unaltered rocks from Cuprite, Nevada (after Abrams et al., 1977).

These spectra show the presence of VNIR iron absorption features in argillically altered samples only. SWIR clay absorption features are present in opalite containing alunite, and in argillized rocks. Note that silicified rocks (no clays) and unaltered rocks (top curve) have no distinctive absorption features.

Figures 2.19a&b Aircraft spectra illustrating removal of atmospheric effects (after Marsh and McKeon, 1983).

Figure 2.19a shows a single airborne spectrum (solid line) and the average spectrum (dotted line) of an entire survey.

Figure 2.19b shows the difference (residual) of the two plots in Fig. 2.19a. The intensity scale is in $\text{mW/cm}^2/\text{steradian}$.

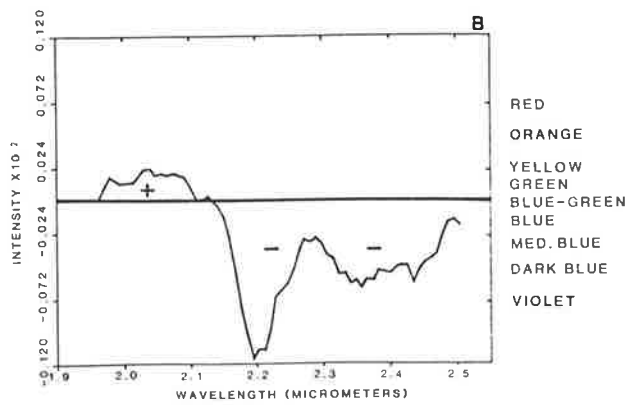
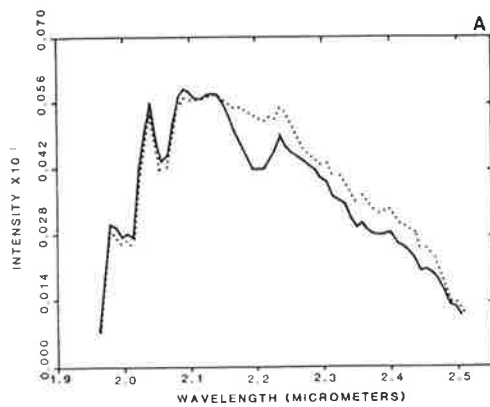
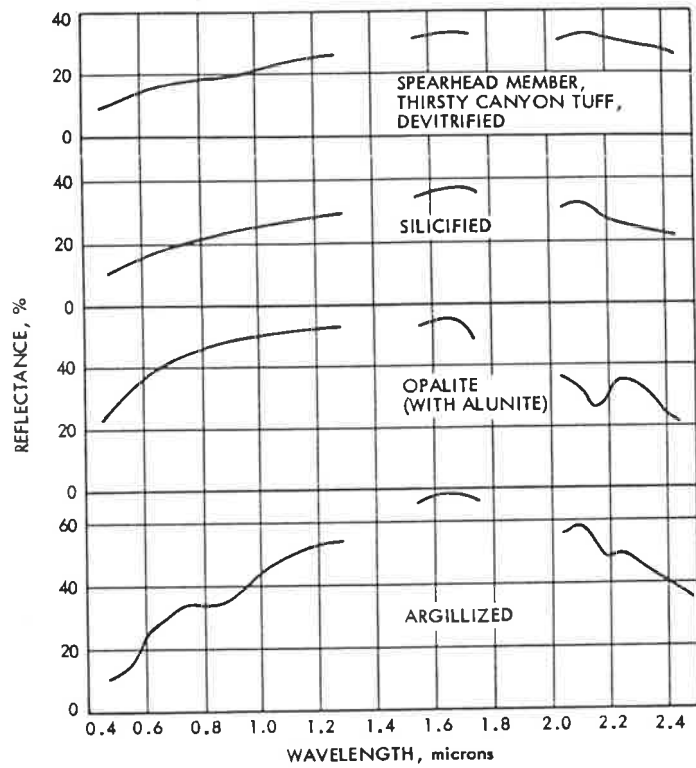


Figure 2.20 Classification of alteration at Oatman, Arizona (after
Marsh and McKeon, 1983).

Ratio of 2.16 μ m to 2.20 μ m bands of flight line OA2A. Categories A,
B and C were delineated using a waveform analysis technique.

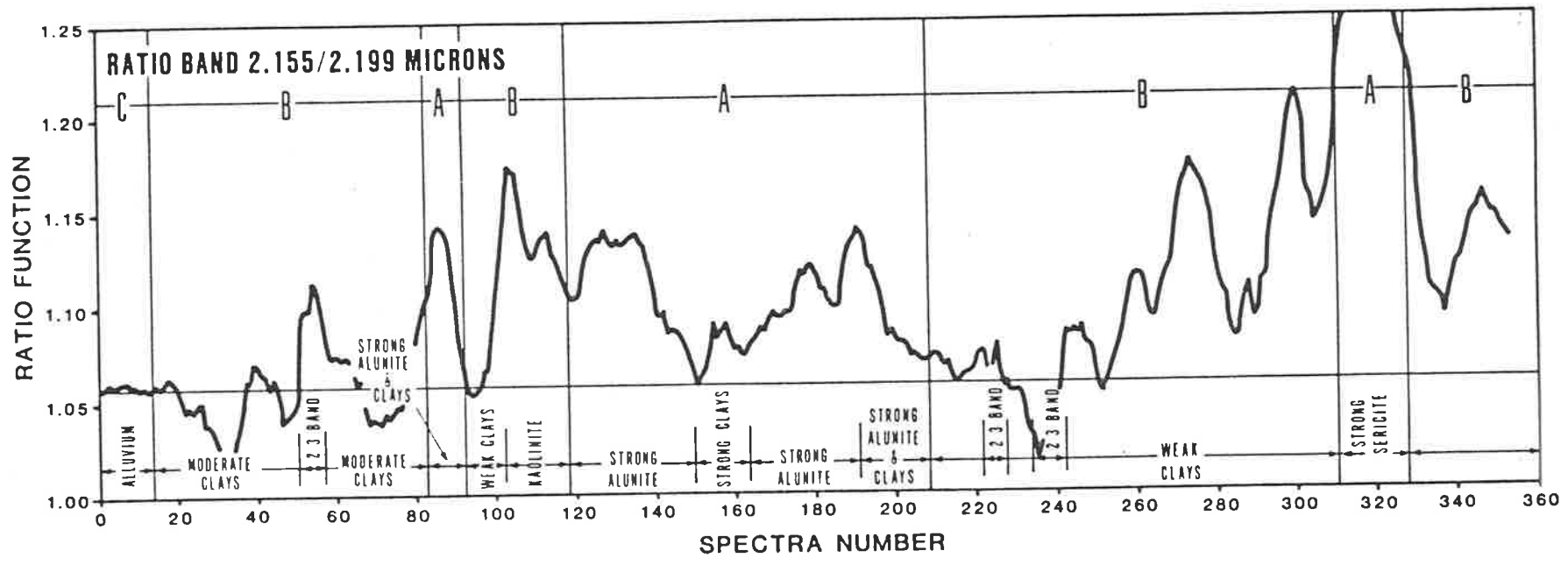


Figure 2.21 AIS data from Cuprite, Nevada (after Goetz, 1984).

Spectra derived from flight data are compared to laboratory spectra (dashed lines) of alunite and kaolinite samples collected from Cuprite, Nevada.

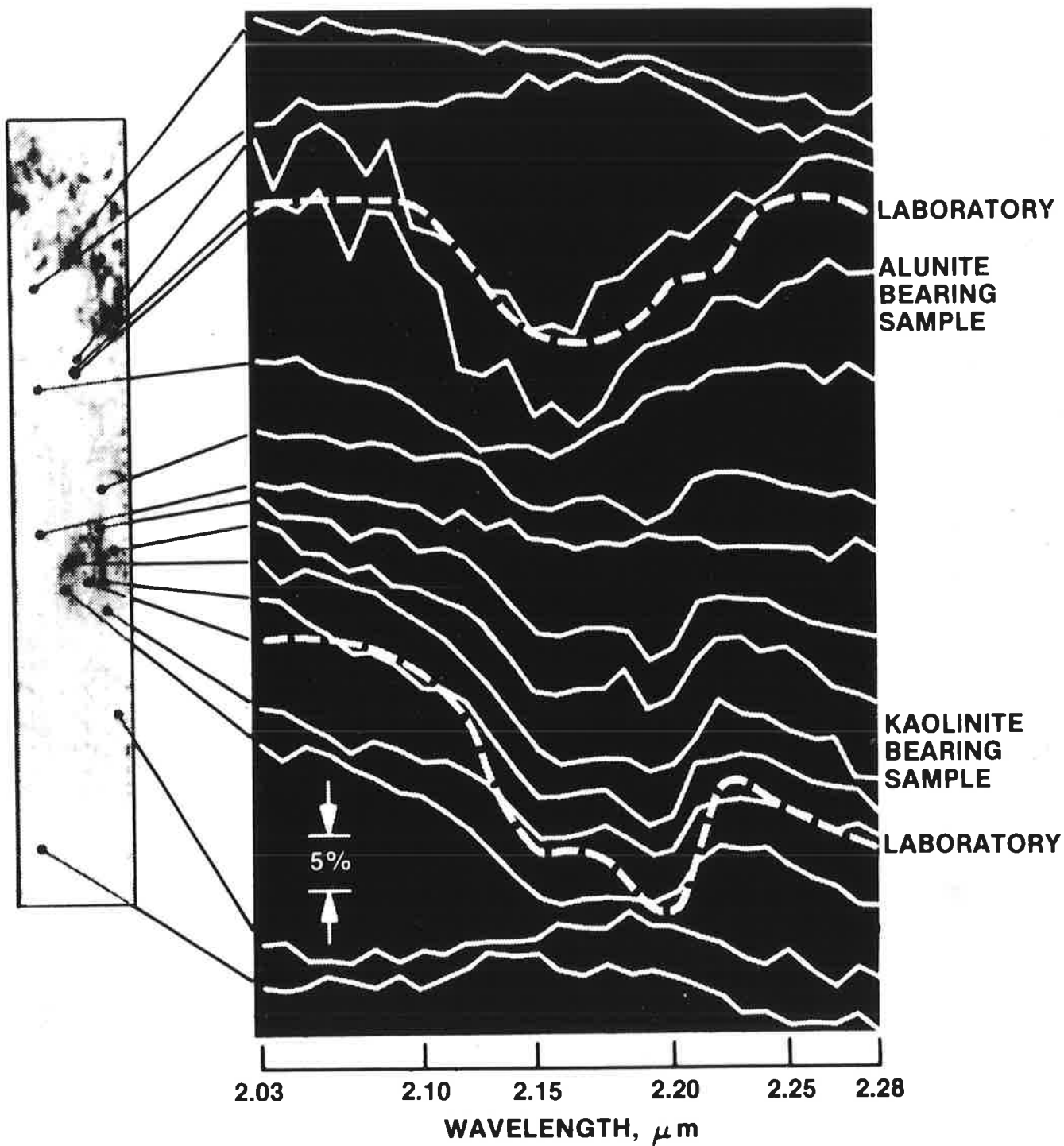


Figure 2.22 AIS data from the Wind River Basin, Wyoming (after
NASA, 1985).

The spectra shown were derived from AIS data over ground corresponding to the geologic units indicated in the stratigraphic column.

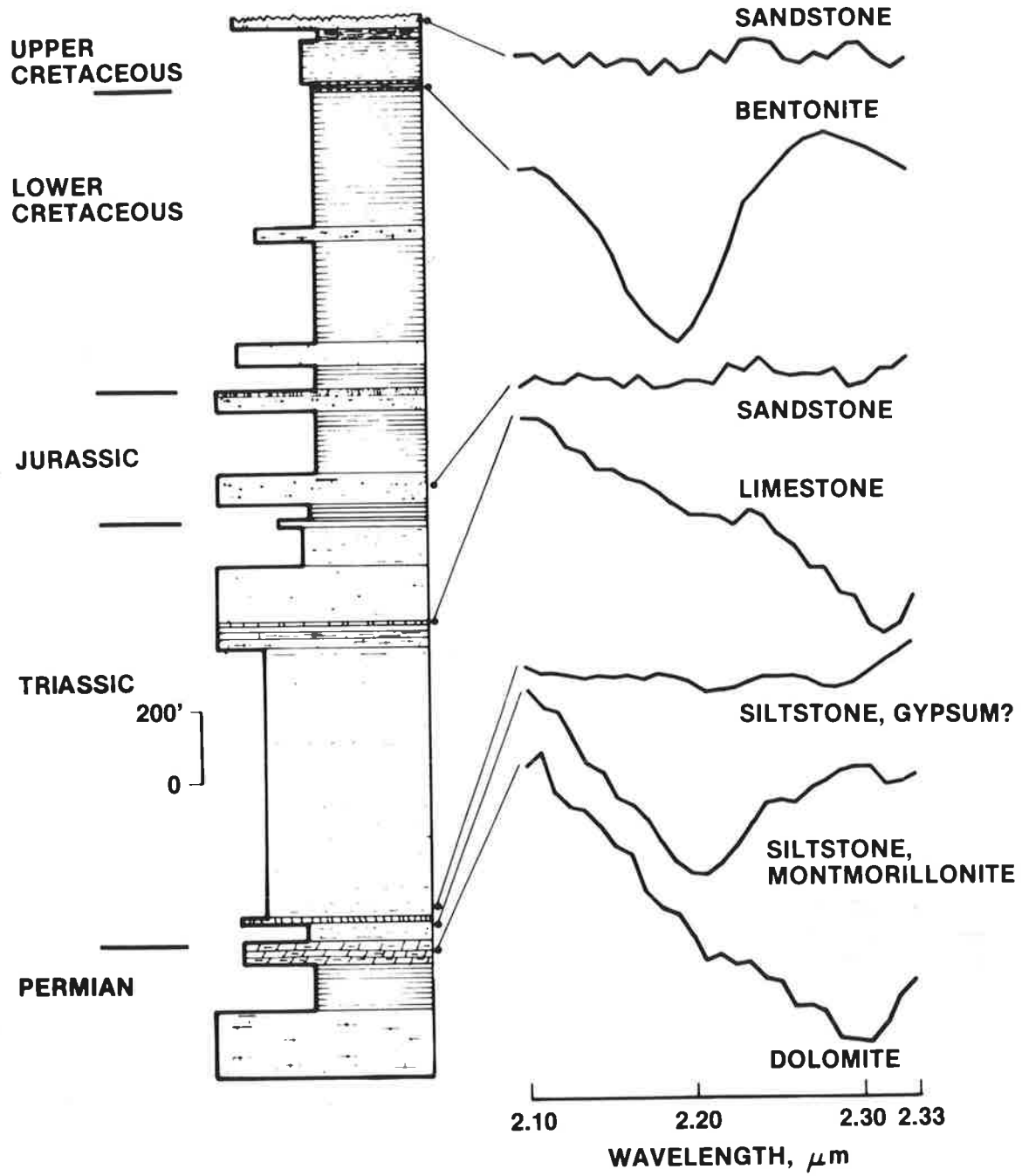


Figure 2.23 Preservation of the deep weathering profile in
Australia (after Butt and Smith, 1980).

The full profile is most commonly preserved in areas of low to moderate relief, as plateau remnants or beneath transported overburden. The weathered profile is vestigial or absent mainly in humid regions of moderate to high relief.

Figure 2.24 Development of the weathering profile (after
Butt,1983).

Figure 2.24 depicts profile formation, during lateritic weathering, modelled on the SW coast of Western Australia. The profiles are applicable to acid and basic rocks.

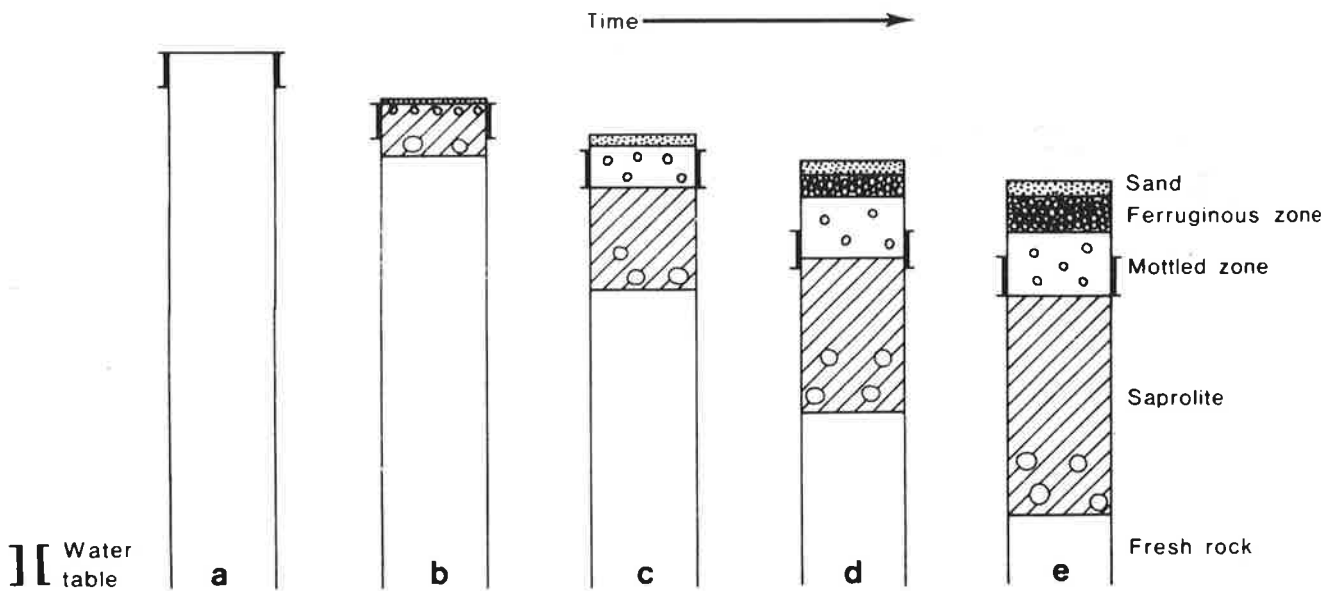
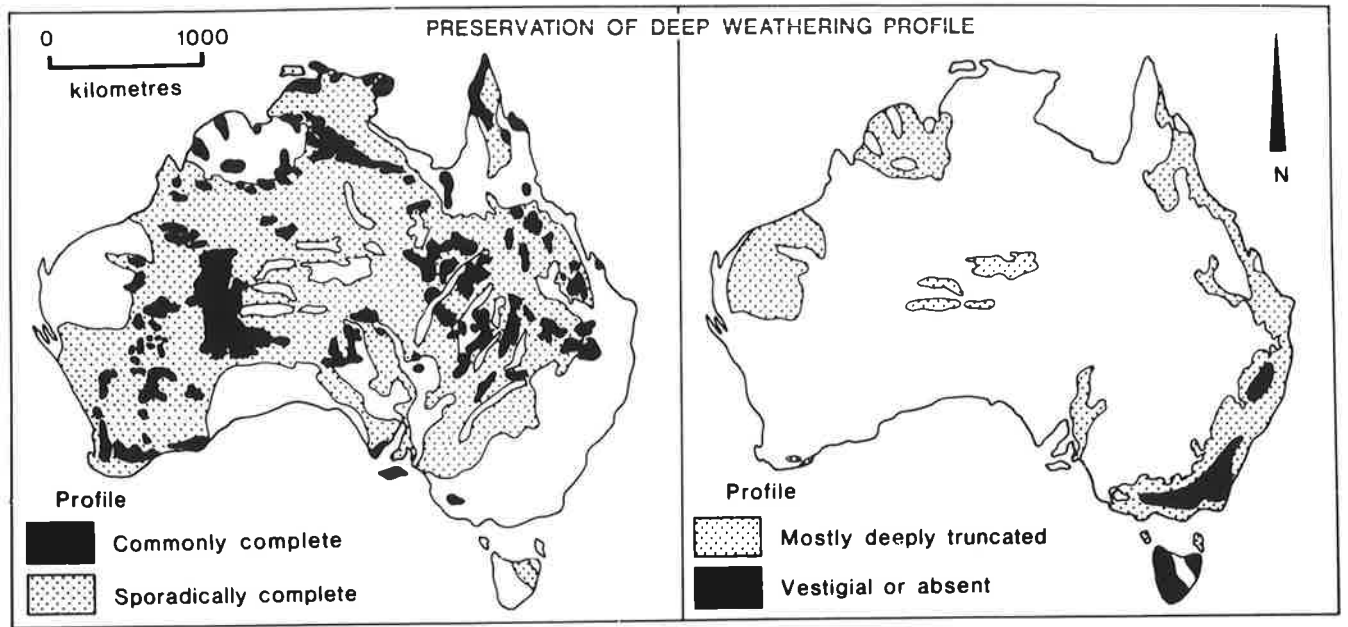


Figure 2.25 Composition of lateritic duricrusts (after Schellmann, 1979).

The triangular diagrams represent the various compositions of duricrusts derived from five different precursors. The compositions are expressed in terms of Fe_2O_3 , Al_2O_3 and SiO_2 end-members. Information about each of these end-members is available from the VNIR, SWIR and MIR portions of the spectrum, respectively, using remote sensing techniques.

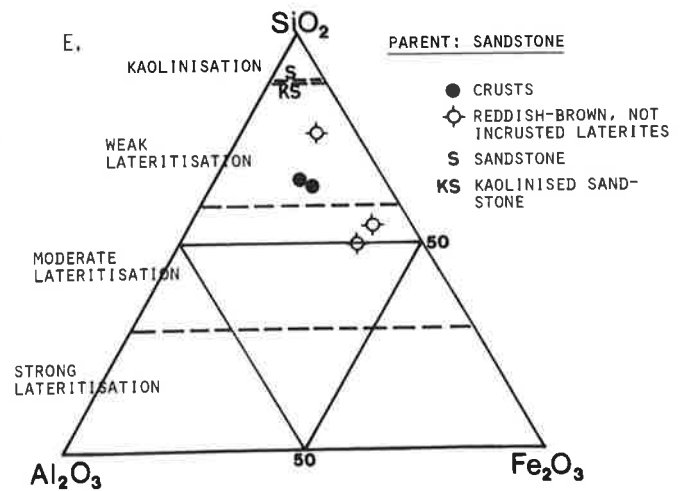
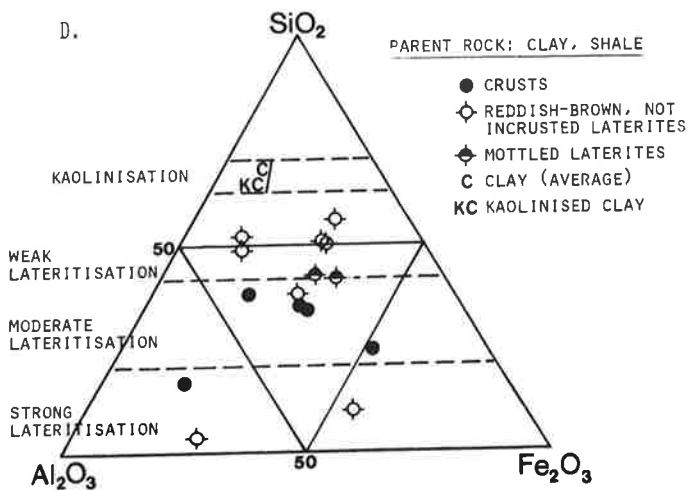
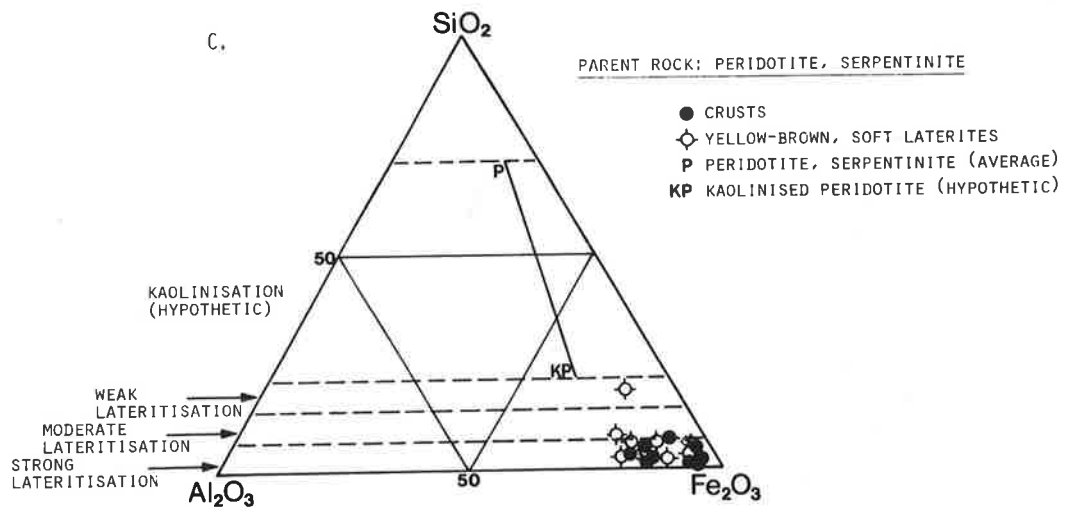
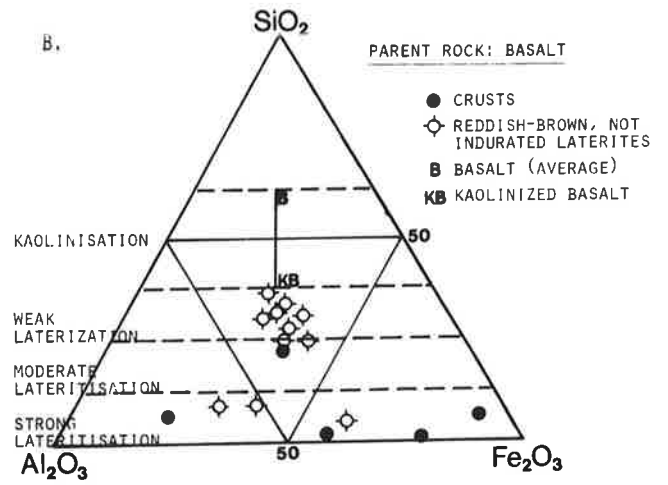
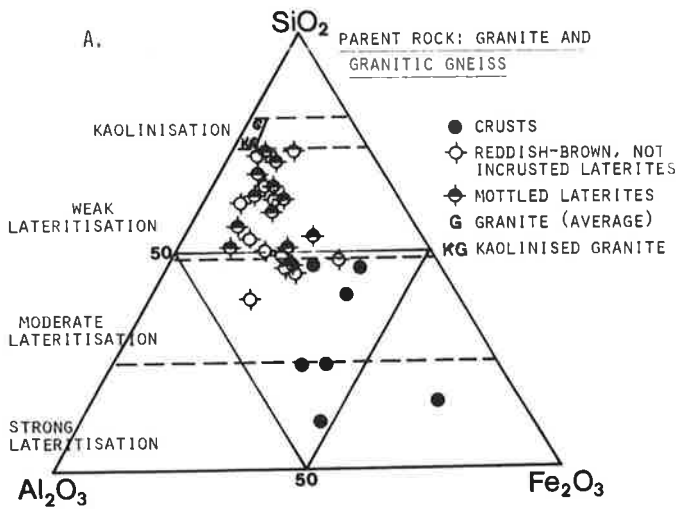


Figure 2.26a-c Weathering profiles developed on felsic, mafic and ultramafic rocks (after Butt, 1983).

The effects of convergent evolution are most pronounced near the top of each profile, mineralogical differences becoming greater with depth. A partially stripped profile will therefore offer a better chance of identifying the precursor by remote sensing techniques.

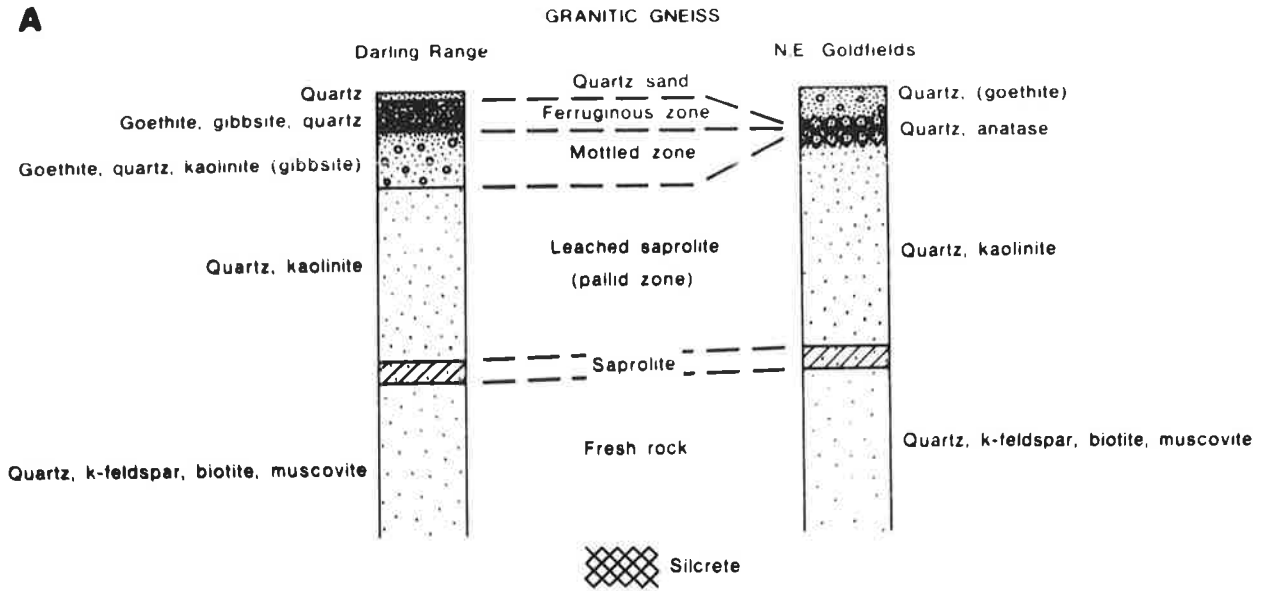
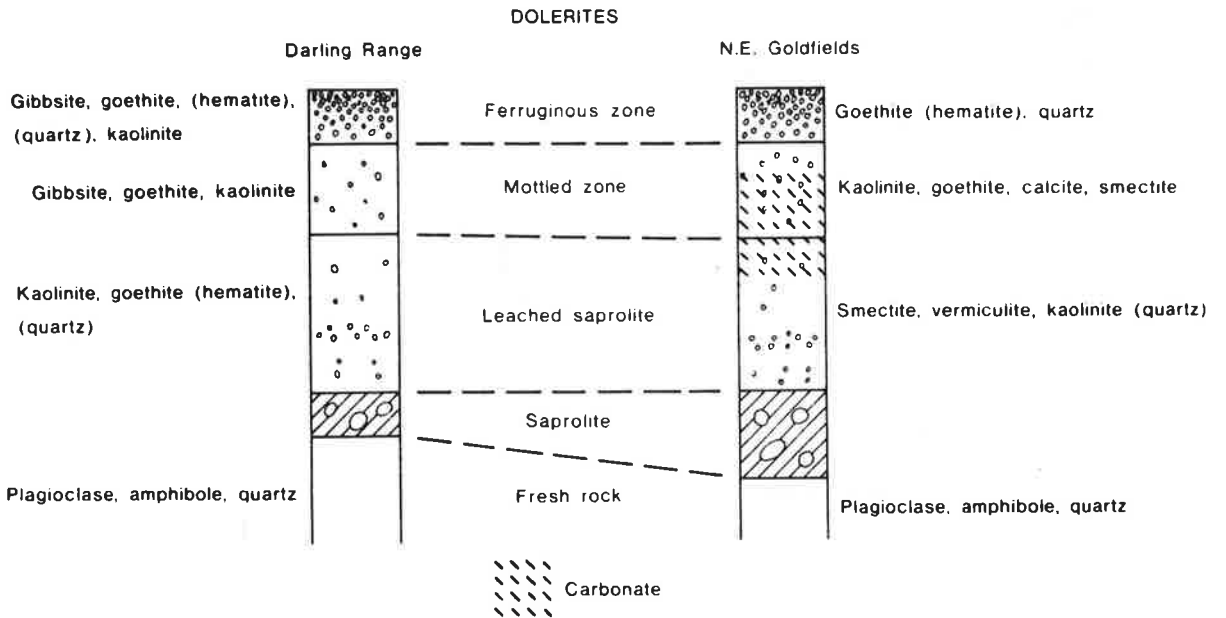
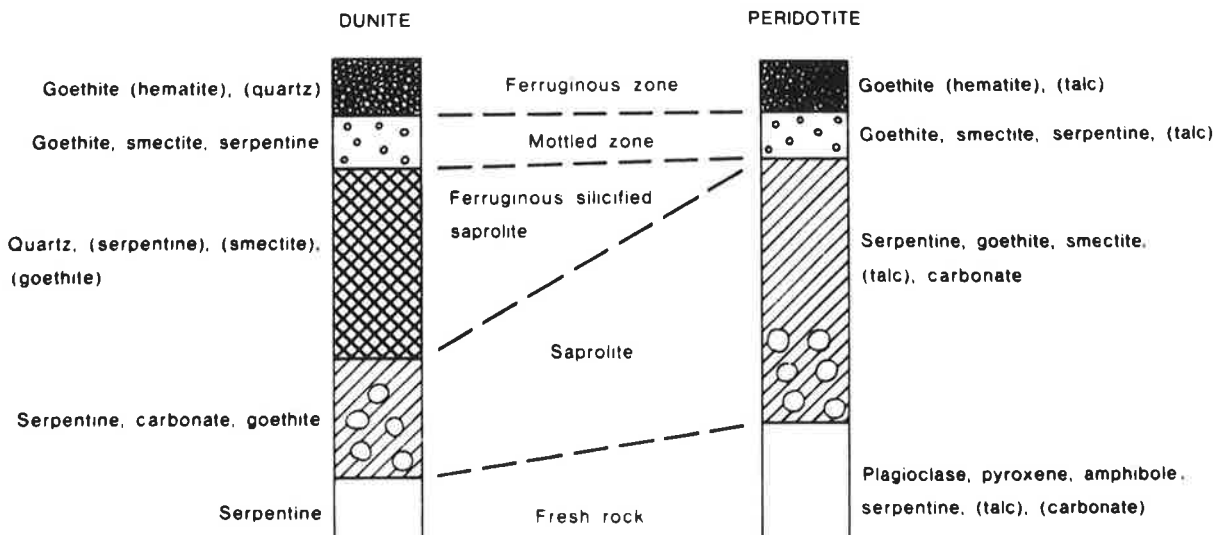
A**B****C**

Figure 3.1 Experimental arrangement for measurement of VNIR and SWIR spectra in the laboratory.

EXPERIMENTAL ARRANGEMENT FOR MEASUREMENT OF SPECTRA

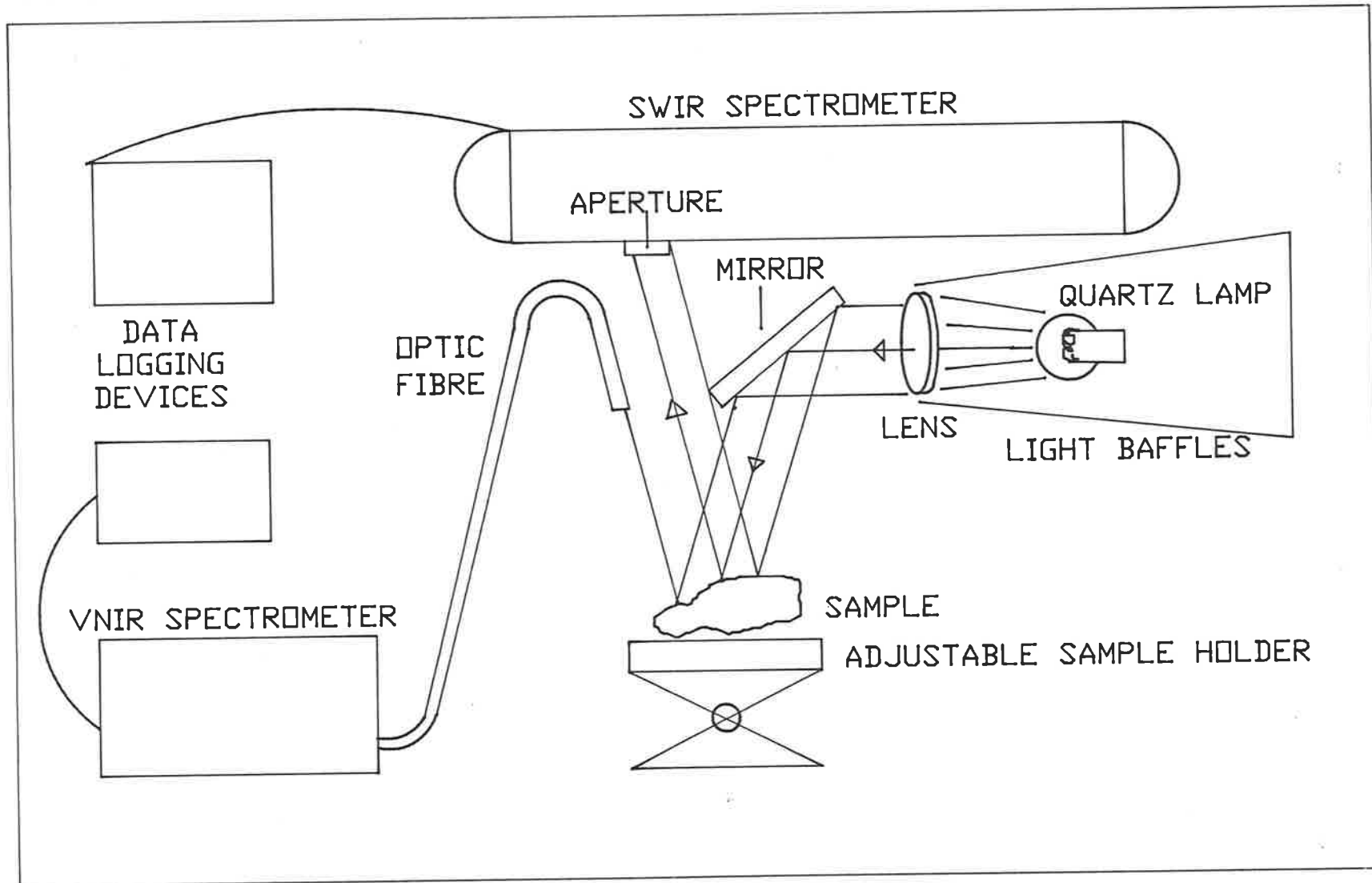


Figure 3.2 The Tracor Northern VNIR spectrometer.

The blue box (middle right) is a multichannel analyser, linked to the spectrometer head (grey box, centre of picture) by the thick, grey cable. An optic fibre bundle runs from the input slit of the spectrometer head to a position above the adjustable sample holder.

Figure 3.3 The SWIR spectrometer.

The CVF spectrometer is mounted on the inside (right-hand side) of the pod which can be attached to the exterior of an aircraft. The CVF has a resolution of approximately 1% of wavelength. A SWIR radiometer occupies the left-hand side of the pod, but was not used in the laboratory. A movable mirror allows the instrument to be used in either an upward-looking, or a downward-looking mode. The pod is approximately 1.8m long.

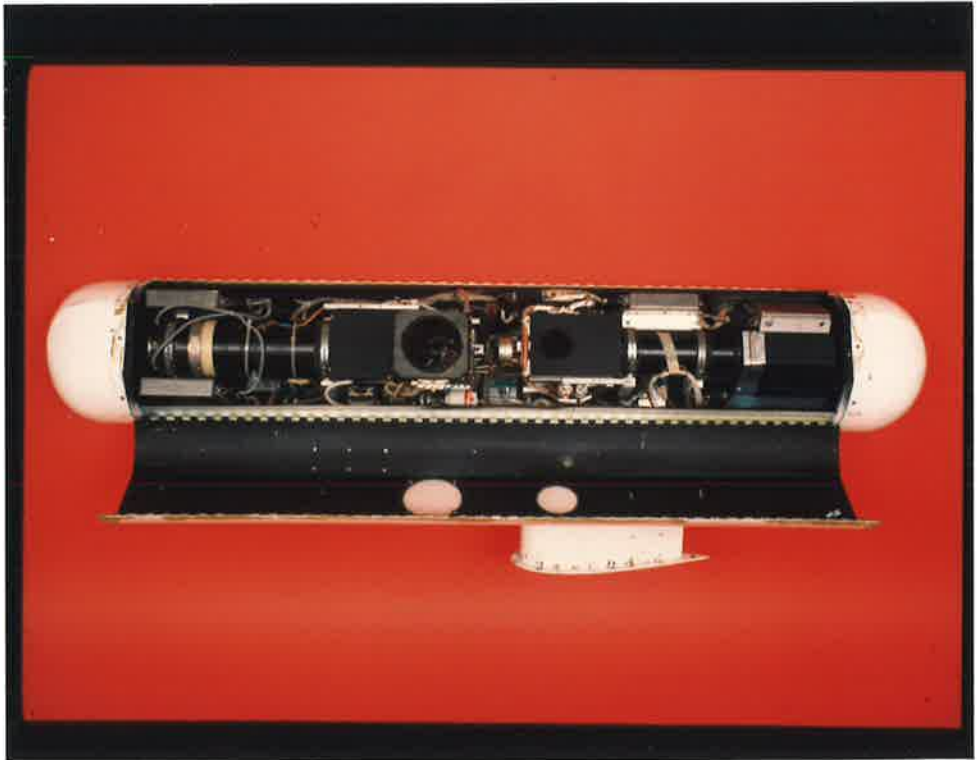
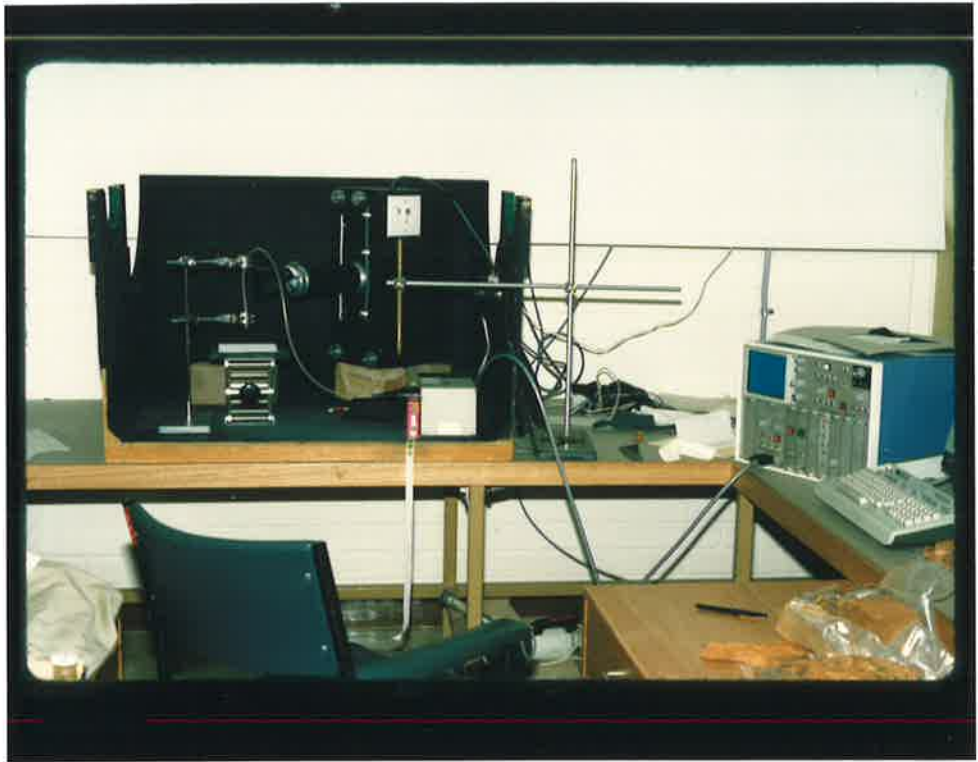


Figure 3.4 The CO₂ laser spectrometer (after Eberhardt et al., 1984).

The various components of the system are listed below :

1. Line-tuned continuous CO₂ laser.
2. Calorimeter
3. Visible He-Ne laser
4. ZnS splitter
5. Ge lens
6. HgCdTe photoconductive detector
7. Photoelectric trigger
8. Samples
9. Hexagonal mirror
10. Oscillation of hexagonal mirror axis

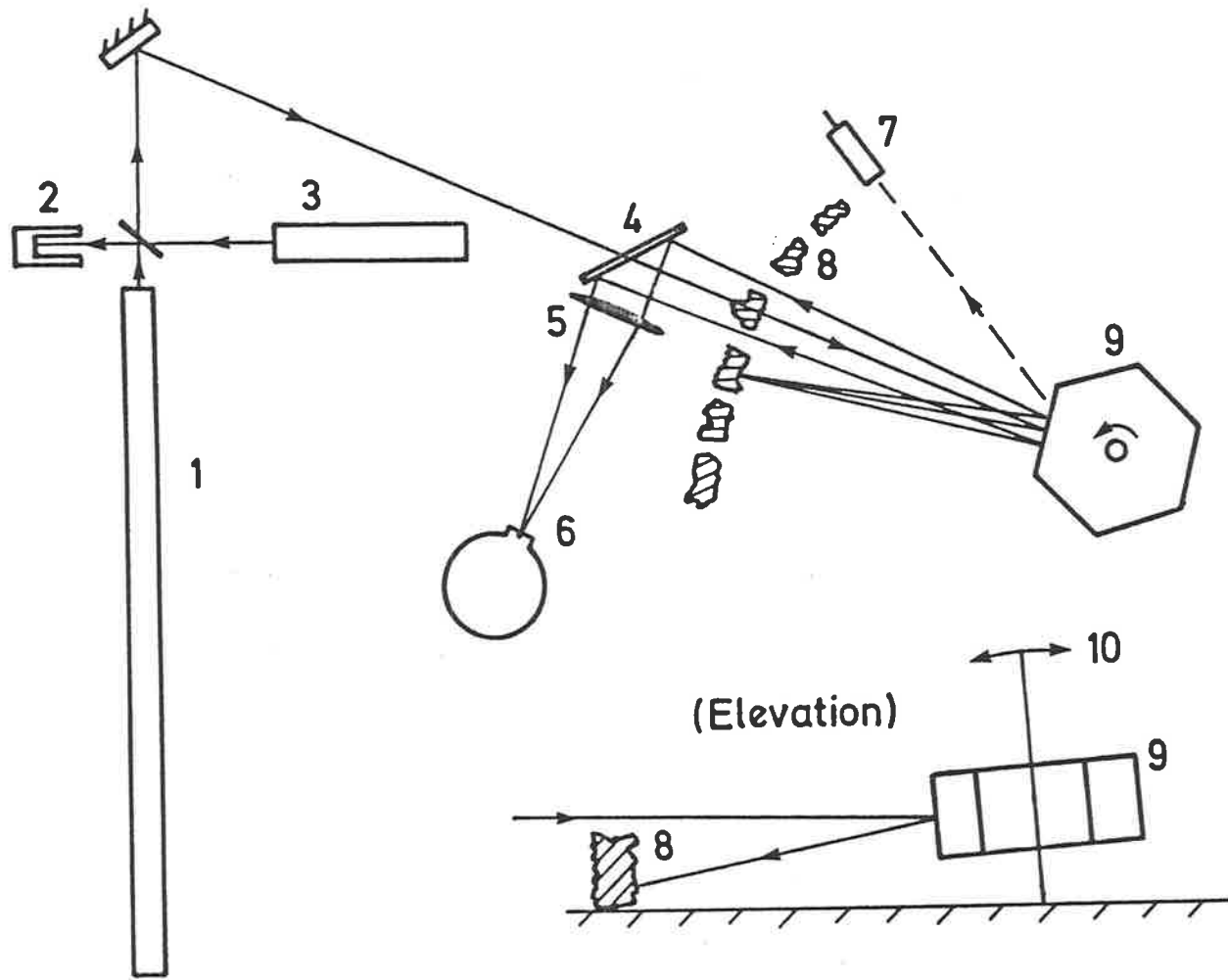


Figure 3.5 The image processing system.

The components of the system are the colour television monitor (left), a computer terminal (centre) and a remote operator panel (right). The remote operator panel controls cursor movement and graphics display. Refresh memory is housed with the host computer, located behind a wall to the left of the picture.

Figure 3.6 Typical structure of an XPUT file sample.

Each sample comprises a header, containing a label, and any number of groups (four are shown here). Each group contains a number of channels, and can consist of spectral data, derived data, or non-spectral information (such as coded geology). The trailer for each group can contain a flag used by several XBOSS programs when analysing the data.

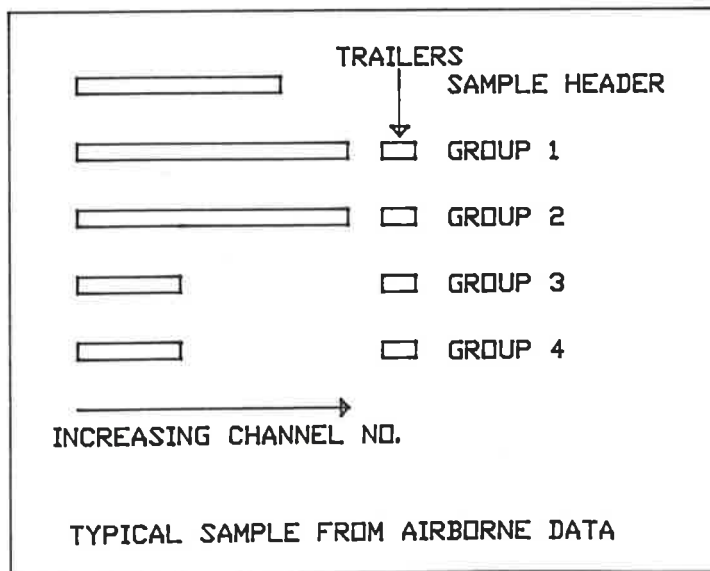


Figure 3.7 XBOSS menus (after Craig et al., 1984).

The three menus show the different types of operations which can be performed on XPUT files. Operations vary from simple display of data, to complex analysis and classification.

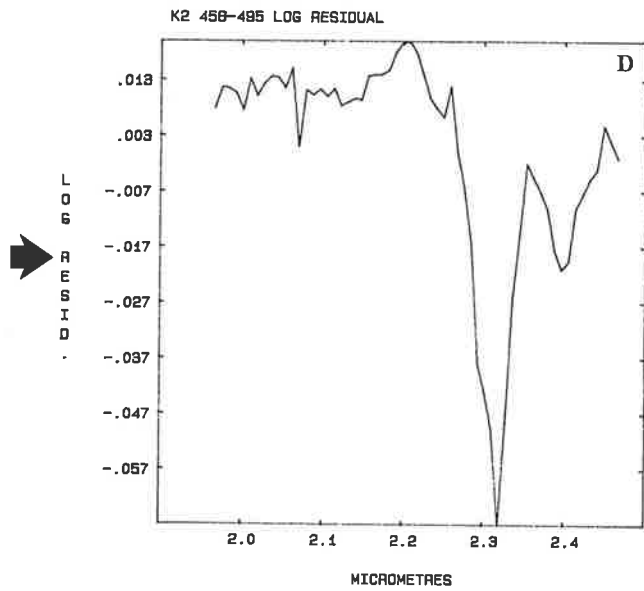
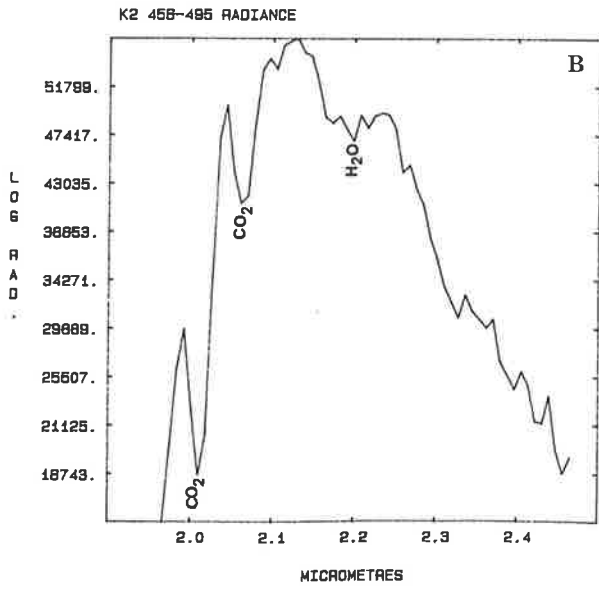
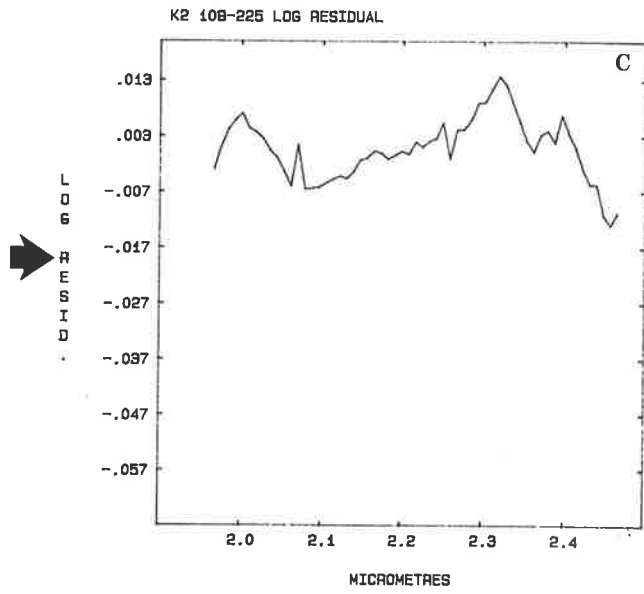
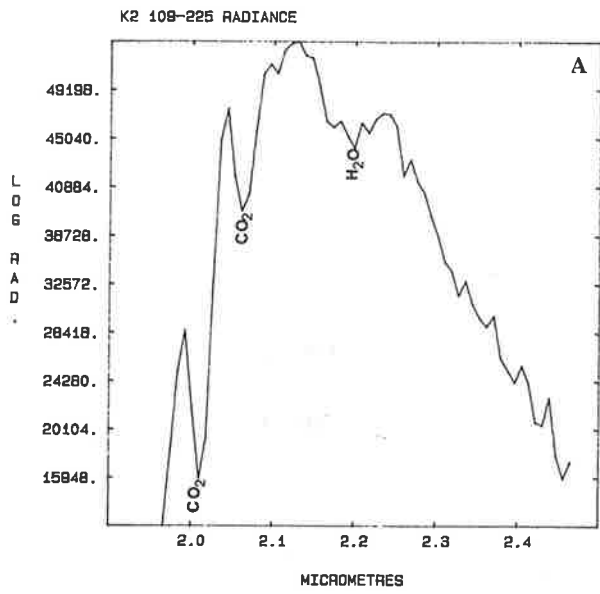
DL OR DL.FNAME.....ABRIDGED FILE PARAMETERS
PL.FNAME.....COMPLETE FILE PARAMETERS
LI.FNAME.....LIST FILE DATA
MO.FNAME.....MODIFY FILE DATA
SL.FNAME OR SL.FNAME,M,N..PRINT SAMPLE HEADERS M TO N
HL.FNAME OR HL.FNAME,N...PRINT HEADER INFORMATION
HS.FNAME.....MODIFY SAMPLE HEADER INFO
HM.FNAME,N.....MODIFY GROUP HEADER INFO
CR.....CREATE XPUT FILE
PU.FNAME.....PURGE XPUT FILE
LL,LU#.....CHANGE LIST DEVICE LU
ZC.FNAME.....ZAP BAD CHANNEL
BK.FNAME.....LIST PARBLOCK
EX (OR /).....EXIT PROGRAM

SA.....SAVE FILE ON MAG TAPE
RE.....RESTORE FILE FROM MAG TAPE
VI.FNAME.....COMPUTE VOLTAGE INTEGRAL
PB.FNAME.....SCHEDULE PROGRAM COLNS
FN.....SCHEDULE PROGRAM FINX
TU.FNAME.....SCHEDULE PROGRAM TUCOL
OK.FNAME.....SCHEDULE PROGRAM XFIX
LN.FNAME.....TAKE LOG NATURAL
XP.FNAME.....TAKE EXPONENTIAL
NT.FNAME.....EQUALIZE FLIGHT LINES (XTEND)
RC.FNAME.....RECONSTRUCT SPECTRUM (XPLIN)
RX.FNAME.....FLIGHT-LINE STAVES (XBRIX)
PN.FNAME.....GENERATE TCHEBYCHEV POLYNS
MP.....SCHEDULE PROGRAM XMAP
NX.FNAME.....ZEROS OF SPECTRA (XNIX)

CO.FNAME.....COPY GROUP TO ANOTHER FILE
SM.FNAME.....SMOOTH (SAMPLES OR CHANNELS)
AV.FNAME.....FILL NEW GROUP WITH LOC AV
MX.FNAME.....FILL NEW GROUP WITH LOC MAX
GF.FNAME.....GRAPHING FACILITY (XPOSE)
SC.FNAME.....SCATTER DIAGRAMS (XCAT)
RS.....SCHEDULE PROGRAM RSXXP
LR.....LOGARITHMIC RESIDUALS (XLOG)
XA.FNAME.....SPECTRAL STATISTICS (XTAT)
FU,FILEOUT.....ARITHMETIC FACILITY (XFUN)
D1.FNAME.....FIRST DERIVATIVE
D2.FNAME.....SECOND DERIVATIVE
TC.FNAME.....TCHEBYCHEV ANALYSIS
YZ.FNAME.....SPECTRAL ANALYZER
PR.FNAME.....SPECTRAL CLASSIFIER
ST.FNAME.....MULTIVARIATE STATS (XMVAR)
PC.FNAME.....PRINCIPAL COMPONENTS
CC.FNAME.....INTERPOLATE (CHANGE # CHANNELS)
TV.FNAME.....SCHEDULE PROGRAM XTV
TP.FNAME.....SCHEDULE PROGRAM XTVP
KT.FNAME.....SCHEDULE PROGRAM XDECT
KR,FILEOUT.....SCHEDULE PROGRAM XDECR

Figures 3.8a-d The log residual technique.

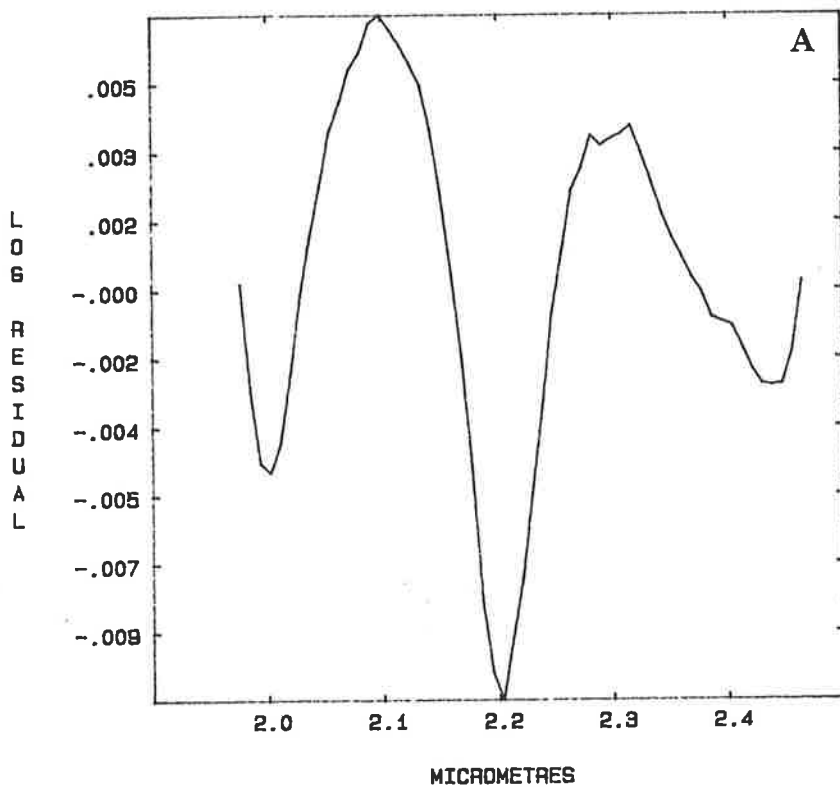
Figures 3.8a&b show airborne radiance spectra from the Kambalda region. The overall shape of the curve is influenced by the energy available to the detectors, and is controlled largely by atmospheric water absorption at both ends of the spectrum shown. In addition, sharp absorption features due to CO₂ and H₂O occur between 1.9 and 2.5 μ m. The log residual technique removes these effects, and variations due to terrain brightness, resulting in the enhancement of subtle mineralogical features, as shown in Figs.3.8c&d.



Figures 3.9a&b Positive and negative log residuals.

These plots show the effect of removing the flight-line mean during the log residual process. Muscovite is widespread throughout the flight line, and its effects (2.2 μ m absorption) are contained in the line mean. Figure 3.9a is derived from an area that contains more muscovite than the flight-line average, so it exhibits a log residual minimum (greater absorption) at 2.2 μ m. Figure 3.9b is derived from an area that contains less muscovite than the flight-line average, so it has a log residual maximum (less absorption) at 2.2 μ m.

STANDARD 14



STANDARD 8

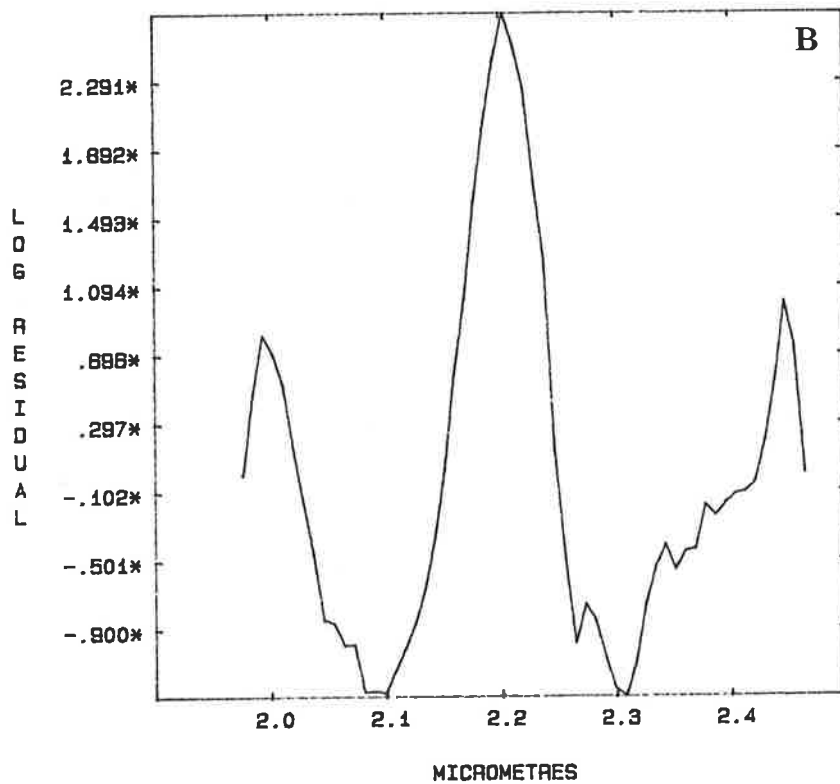


Figure 3.10 Log residual data in image format.

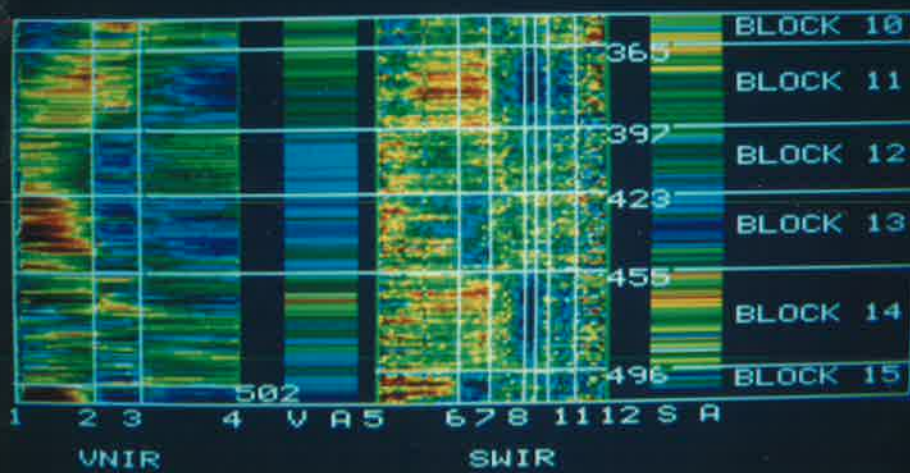
There are four columns of data in Fig. 3.10. From left to right they are VNIR log residuals, VNIR albedo (labelled V A), SWIR log residuals and SWIR albedo (labelled S A).

The data have been density sliced. Dark blue colours represent the lowest log residual values (deepest absorption features), the mid-green colours (see vertical strips at extremities of the SWIR log residuals) represent zero values, and deep red colours represent the highest log residual values. The block boundaries (horizontal lines) outline changes in the nature of the data along the flight line.

The data fall naturally into a number of 'bands' defined by the wavelength limits of the absorption features. The vertical lines outline these bands. Wavelengths of the band boundaries are ;

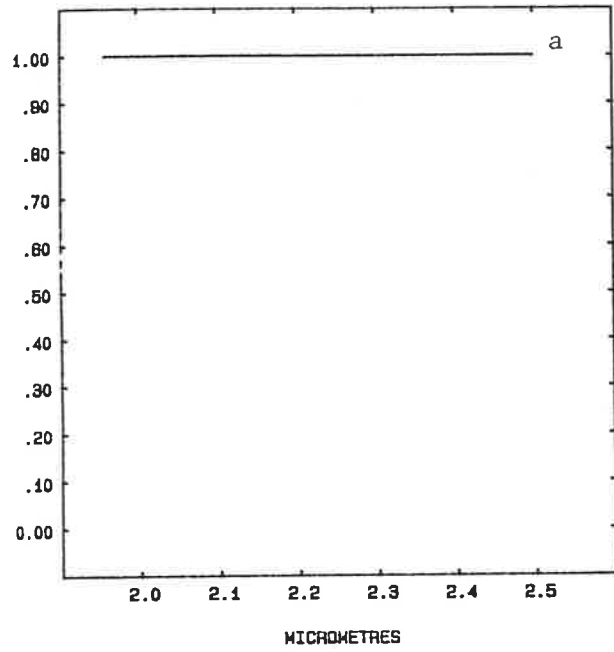
VNIR	1	0.327 μ m (start)
	2	0.635 μ m
	3	0.784 μ m
	4	1.012 μ m (end)
SWIR	5	1.954 μ m (start)
	6	2.147 μ m
	7	2.216 μ m
	8	2.302 μ m
	9	2.328 μ m
	10	2.353 μ m
	11	2.422 μ m
	12	2.500 μ m (end)

LOG RESIDUAL IMAGE DATA

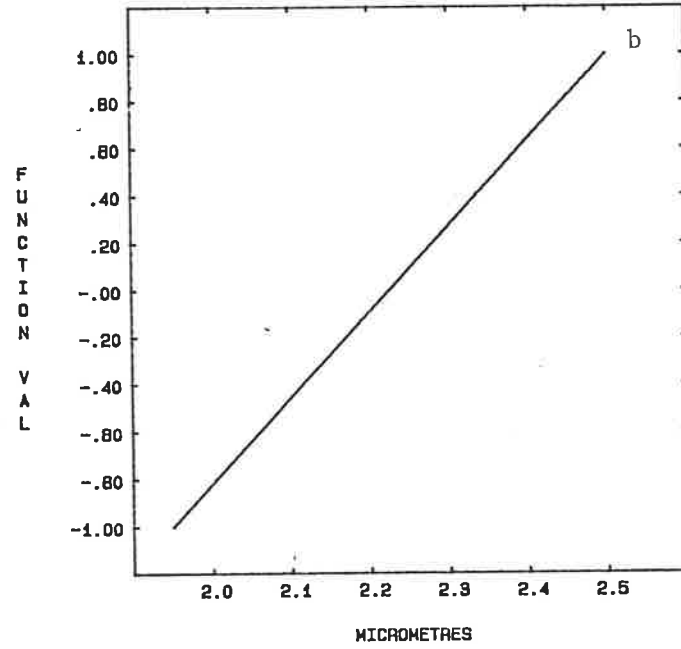


Figures 3.11a-f Tchebychev polynomial functions - degree 0 to 5.

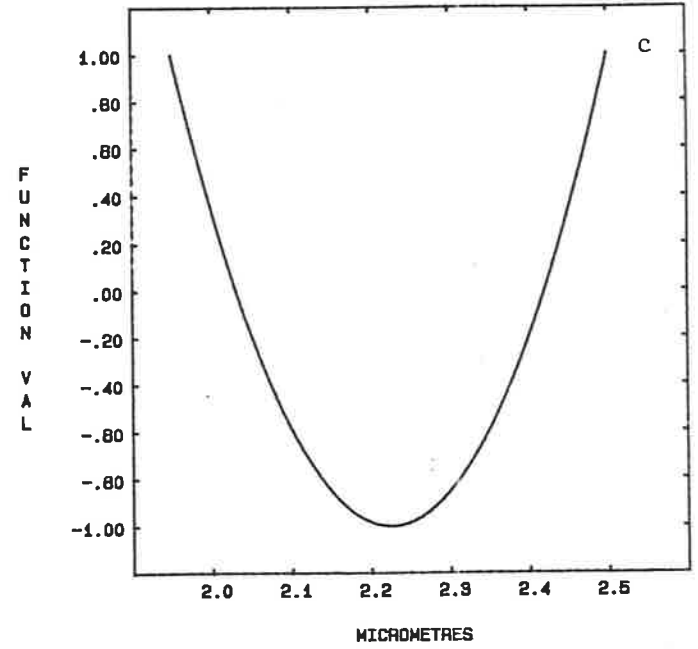
1 A DEGREE 0



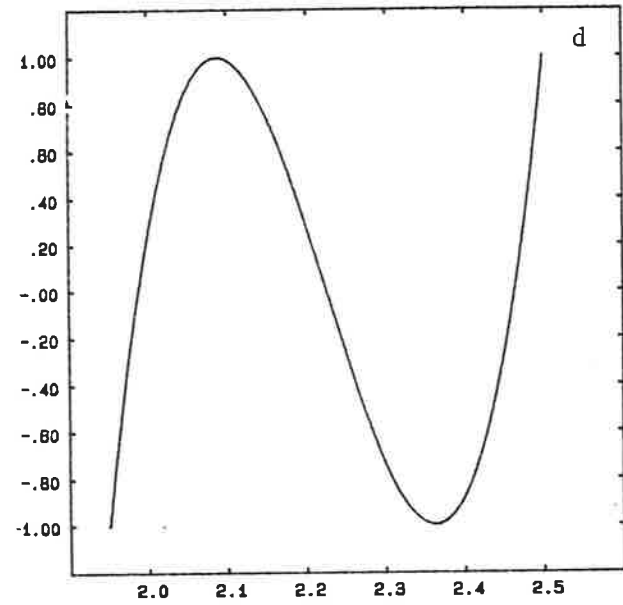
2 B DEGREE 1



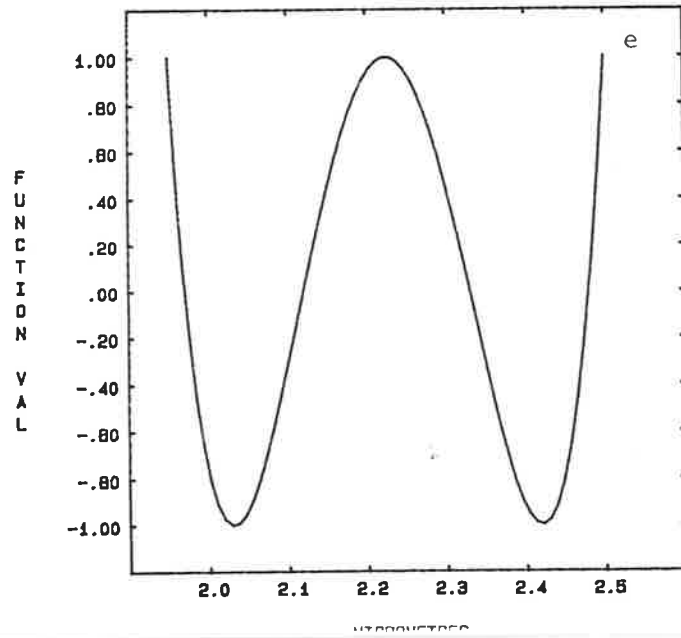
3 C DEGREE 2



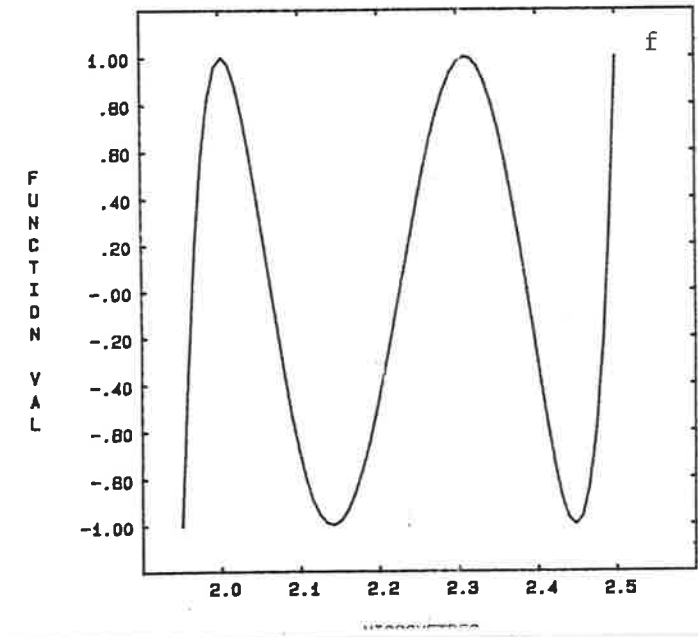
4 D DEGREE 3



5 E DEGREE 4

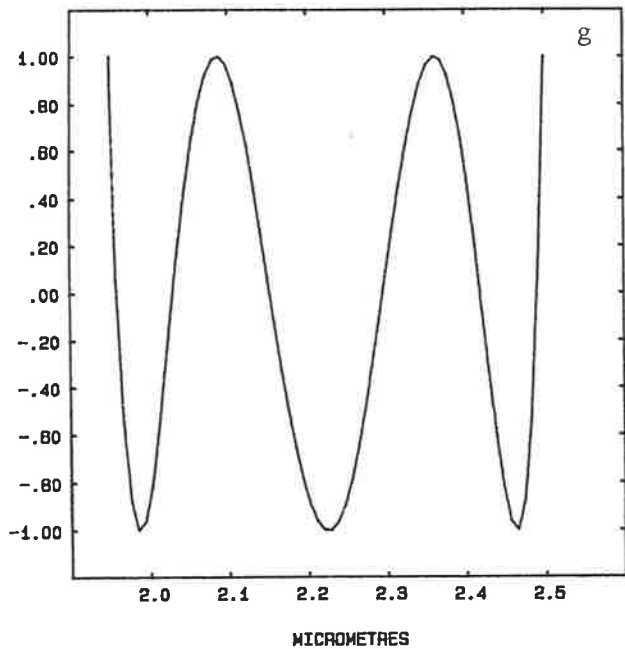


6 F DEGREE 5

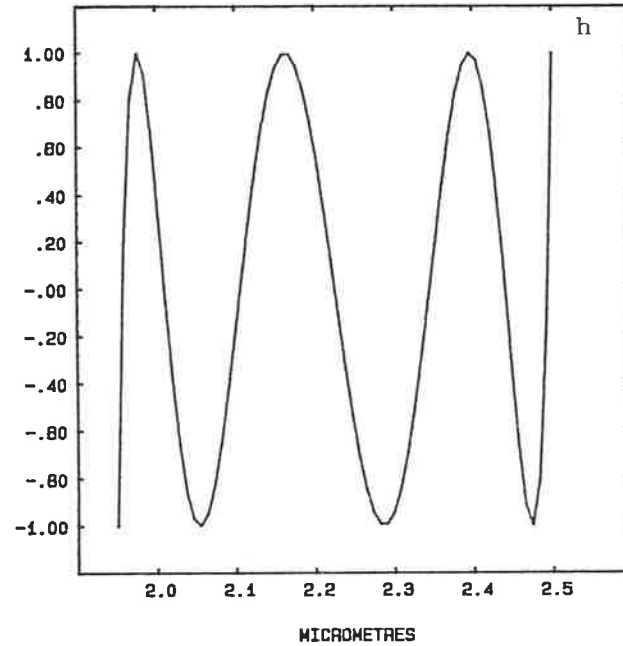


Figures 3.11g-1 Tchebychev polynomial functions - degree 6 to 11.

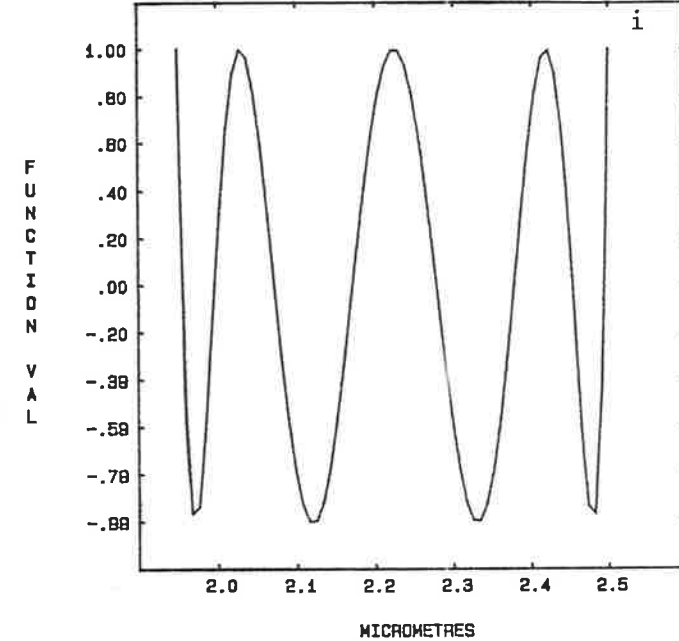
7 G DEGREE 8



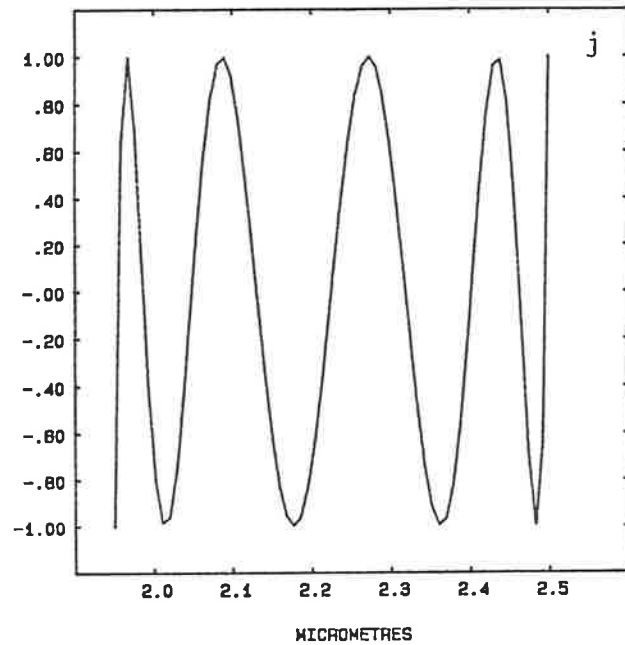
8 H DEGREE 7



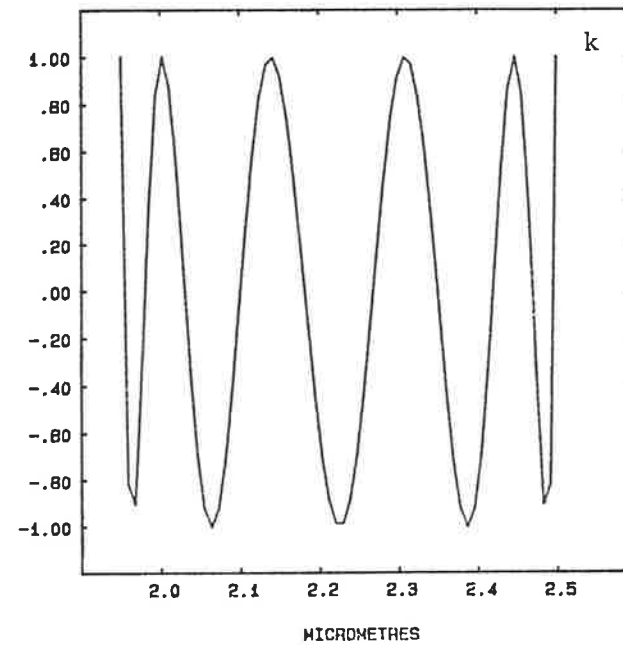
9 I DEGREE 6



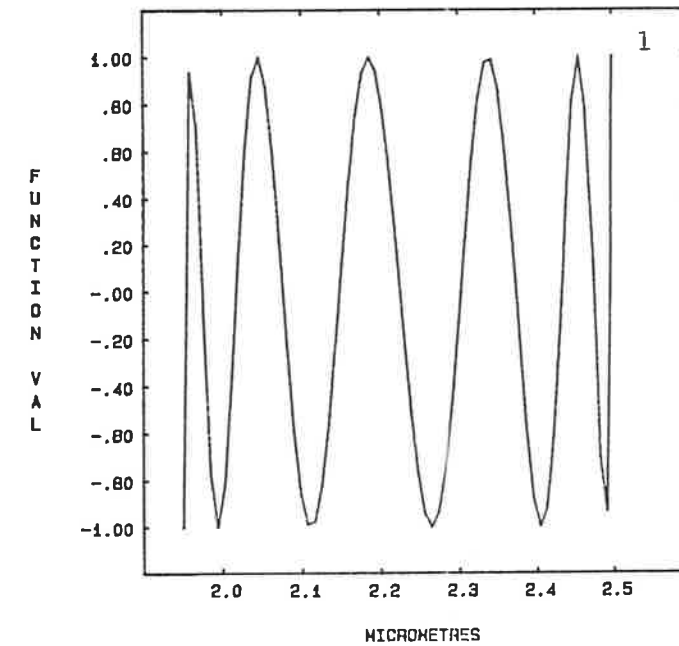
10 J DEGREE 5



11 K DEGREE 4



12 L DEGREE 3



Figures 3.12a-d Interpretation of spectra by smoothing using Tchebychev polynomials and calculation of first and second derivatives.

Figure 3.12a is a VNIR laboratory reflectance spectrum consisting of 512 data channels. It has been smoothed by Tchebychev 'reconstruction' (using 12 polynomial terms) to produce Fig. 3.12b. Figure 3.12c&d are the first and second derivatives, respectively, of Fig. 3.12b. Zero values for the first derivative represent inflection points in the original spectrum, while zero values for the second derivative represent maxima or minima in the original spectrum (thus defining the positions of absorption features).

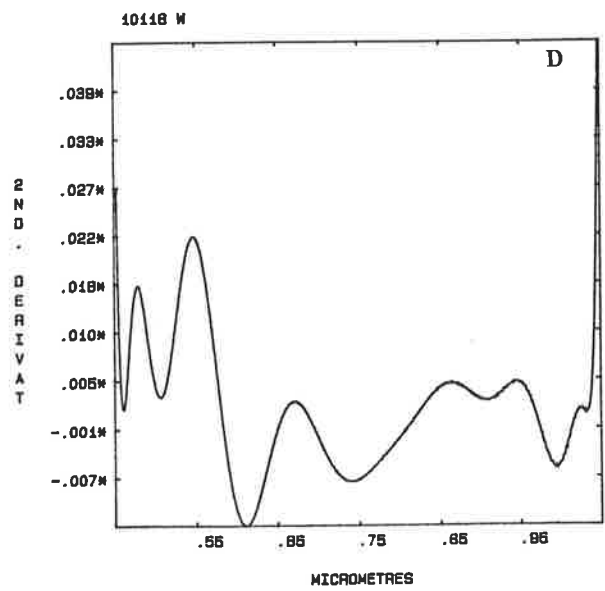
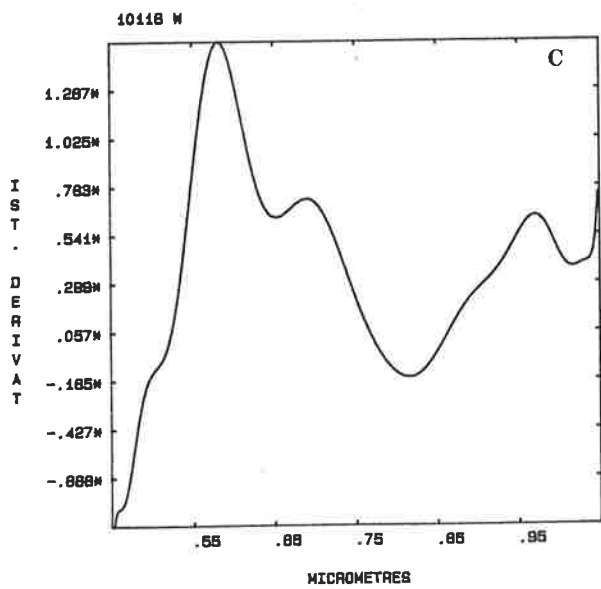
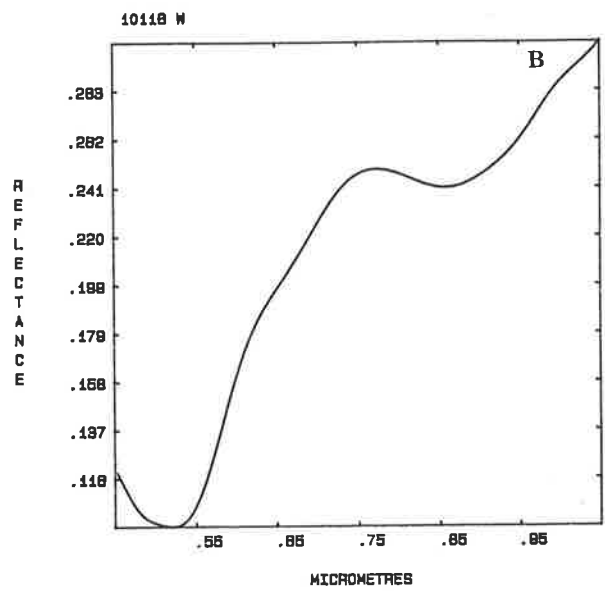
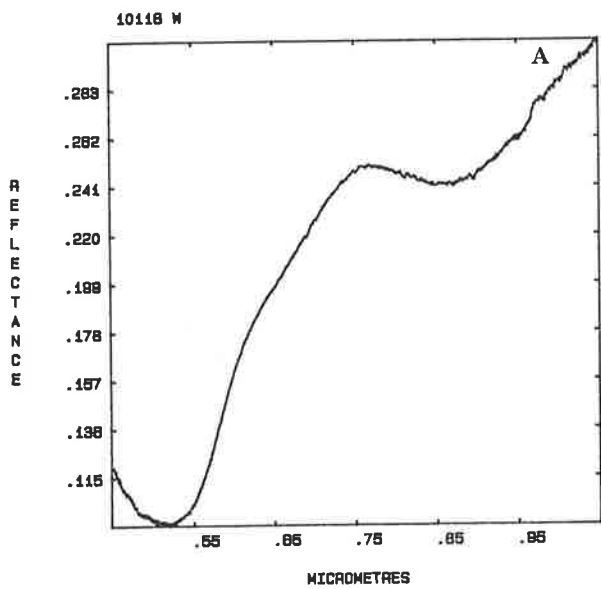
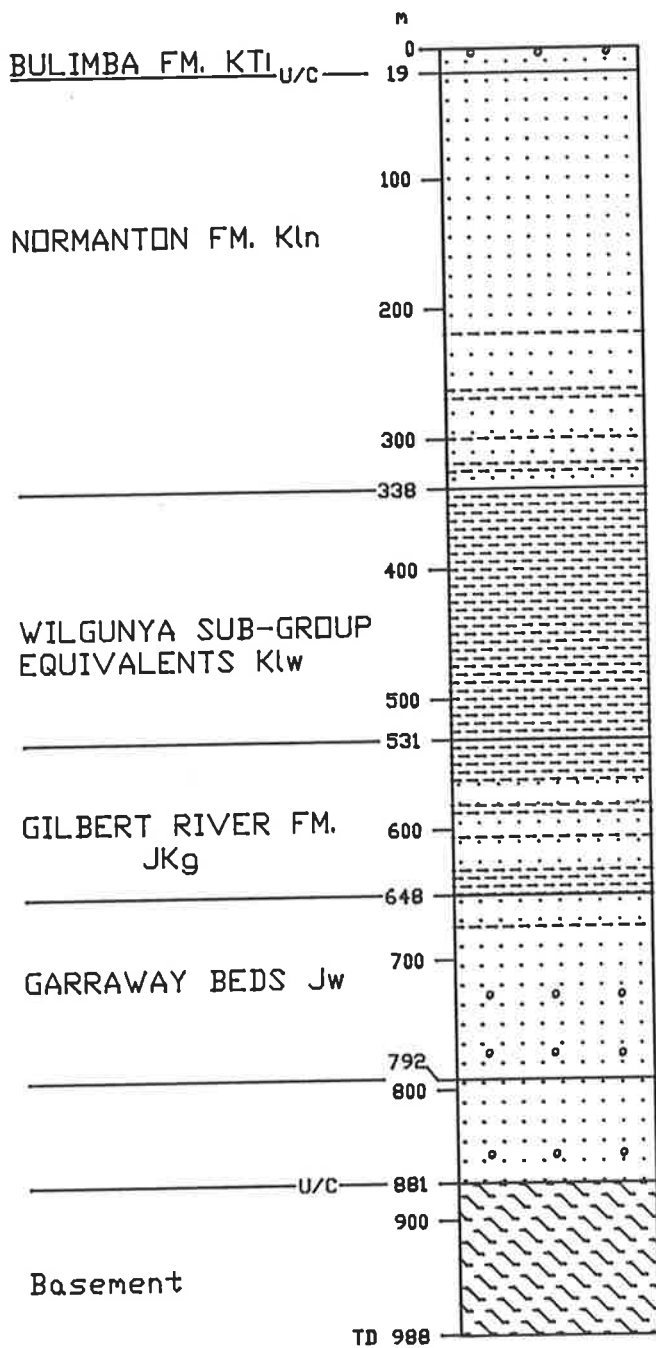


Figure 4.1

Summary of stratigraphy on Cape York Peninsula (after
Doutch et al., 1973).



0-3m Bauxite and ferricrete
 3-19m Poorly sorted, angular, very clayey quartz sandstone.
 19-26m Yellow-brown stained, labile sandstone.

26-338m Fine to medium-grained grey-green glauconitic labile sandstone with minor inter-beds of grey silty shale

338-531m Grey micaceous silty shale

531-656m Grey marine silty shale with interbeds of medium-grained, grey-green glauconitic quartzose sandstone.

656-709m Grey-green, well-bedded glauconitic quartzose sandstone with minor pyrite, mica. Some interbeds of silty shale.

709-792m Coarse-grained poorly sorted, subangular clayey feldspathic quartzose sandstone, a few conglomerate and clay bands. Minor coal below 783m.

792-881m Sandstone as above, some conglomerate and clay. Becoming increasingly coaly and micaceous towards basement.

881-988m Deeply weathered metamorphic sediments composed of quartz, chlorite, muscovite, magnetite and pyrite, with

Figure 4.2 Distribution of laterite on Cape York Peninsula (after
Evans, 1972).

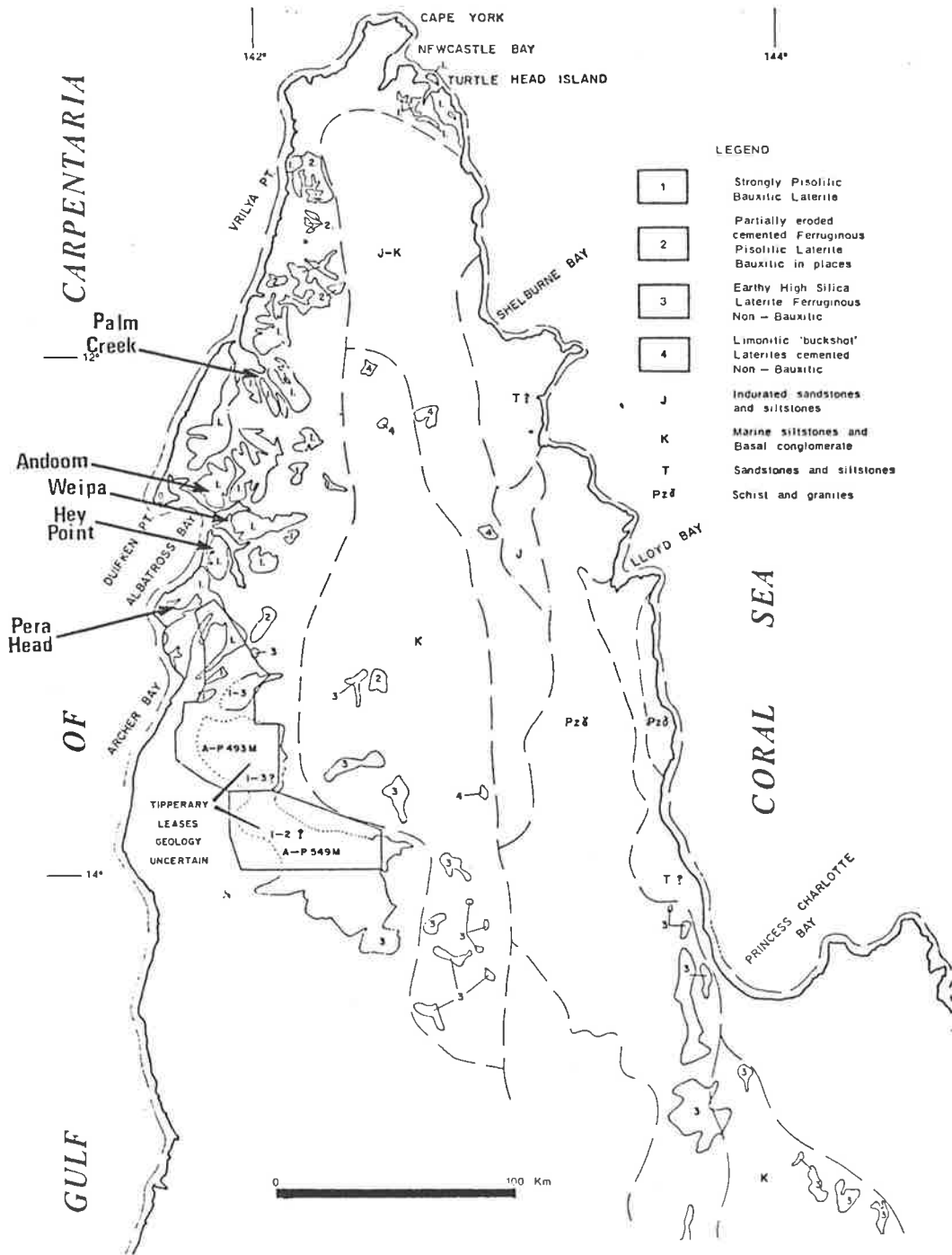


Figure 4.3 Typical stratigraphic log from the Weipa Peninsula
(after Coffey and Hollingsworth, 1971).

Figure 4.4 Cross section through the Weipa Peninsula from
hydrologic bore data (after Coffey and Hollingsworth,
1971).

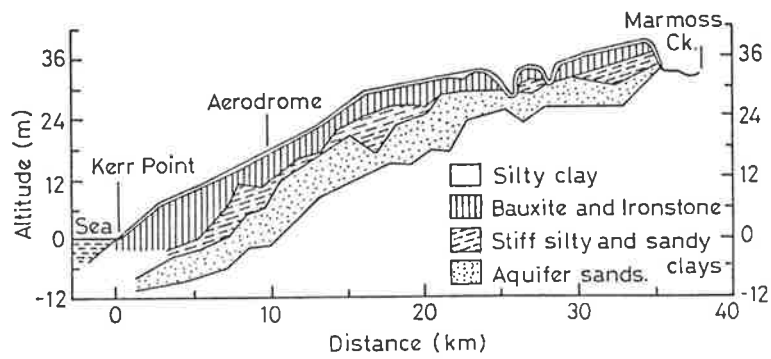
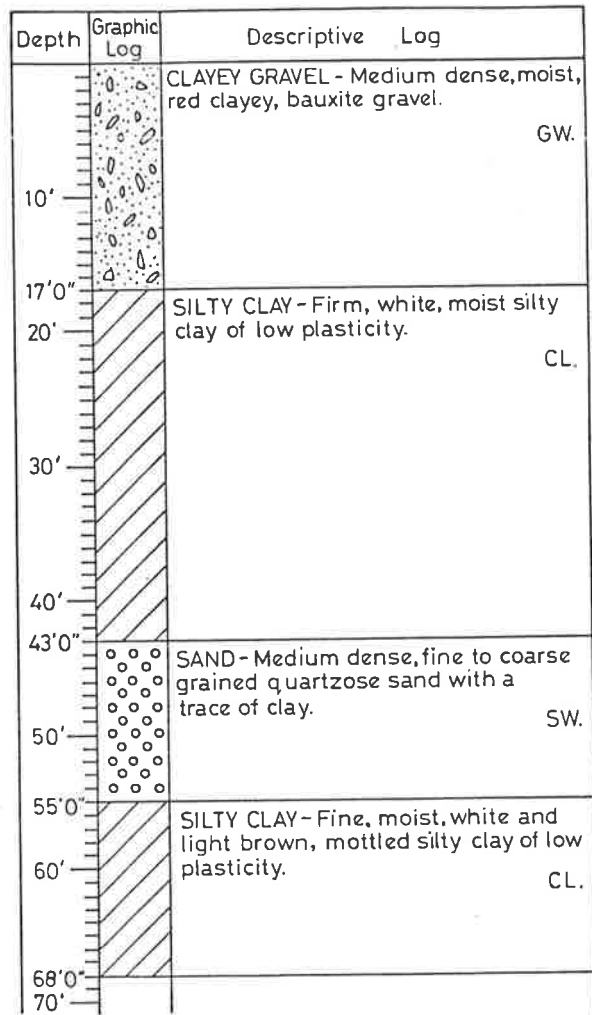


Figure 4.5 Weipa area vegetation map (after Specht et al., 1977).

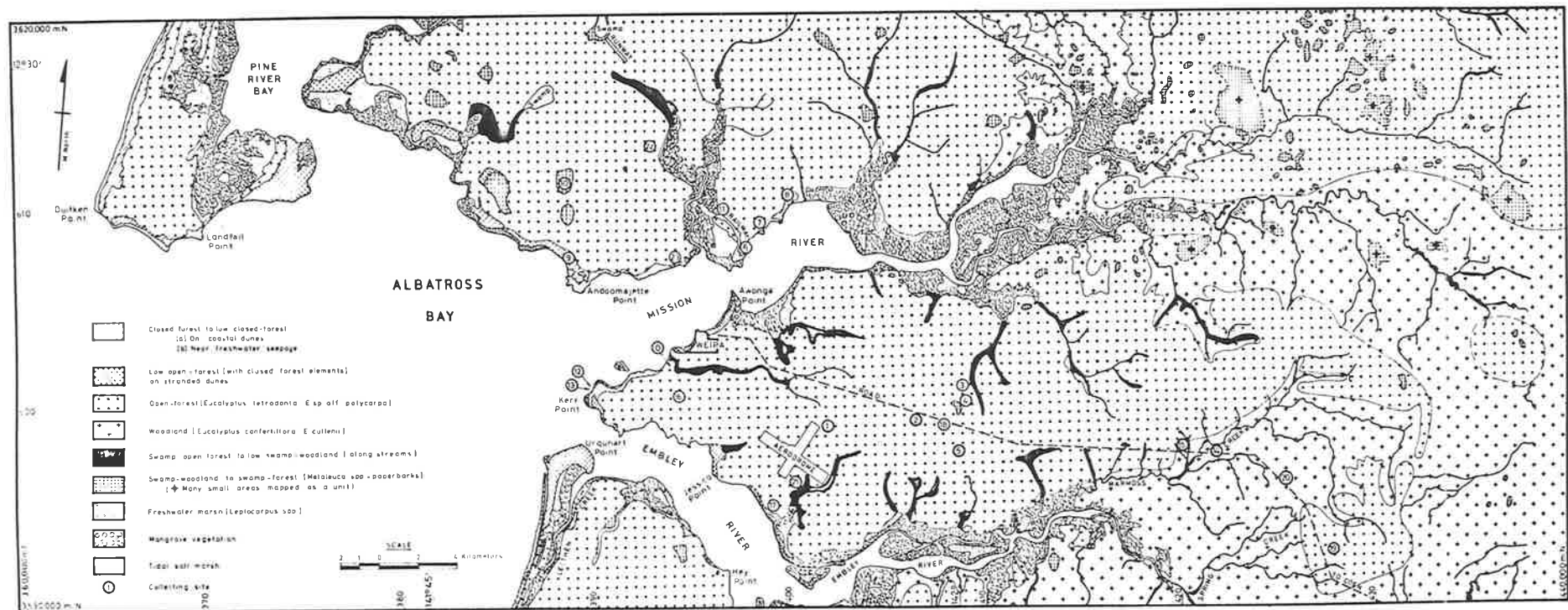


Figure 4.6 Graph of Moisture Index and Net Photosynthetic Index
from Weipa Peninsula (after Specht et al., 1977).

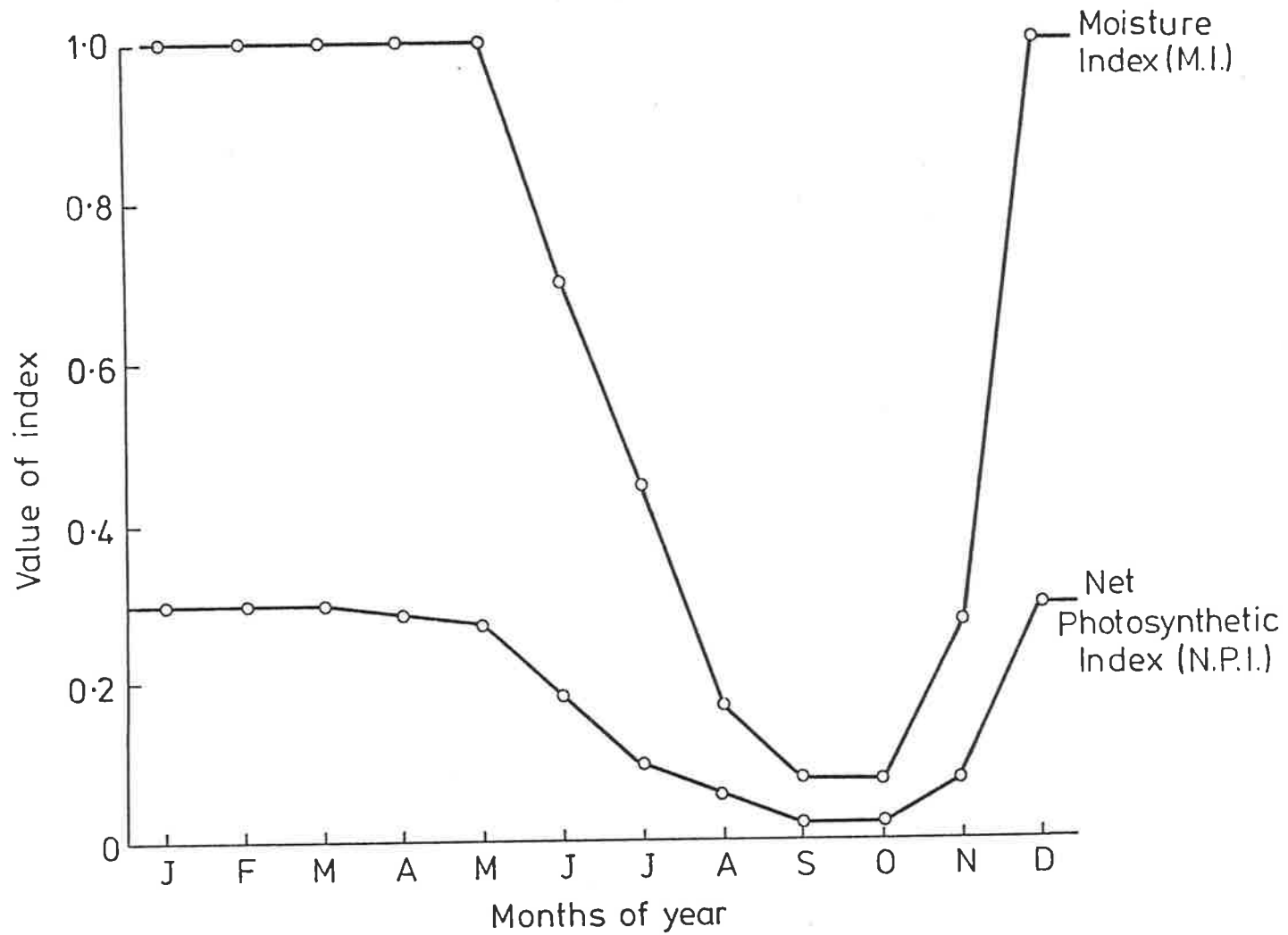


Figure 4.7 Landsat false colour composite - sub-scene of Cape Weymouth - linear stretch of raw bands 4, 5 and 7, dry season image (after Huntington et al., 1982).

Image ID No. 01036-00004, 26 July, 1973. Scale is approximately 1:1,000,000, or 10km between scale ticks along top and bottom of the image. The large central white area is Weipa mine site, while black to green-brown patches to the east and southeast are bushfire burn scars. The bauxite plateaux boundaries are just detectable on this image.

Stretch limits for this scene were derived from the land subscene shown on the overlay, corresponding to: FL 600, LL 2340, FP 900, LP 2100. DS stretch limits used were: 4 17-34, 5 9-29, 7 0-20.

CSIRO - MINERAL PHYSICS 01006-00004-00000 DE 4
WEIPA BAND 3 DS STRETCH = 00-20

00-10-80 10.32

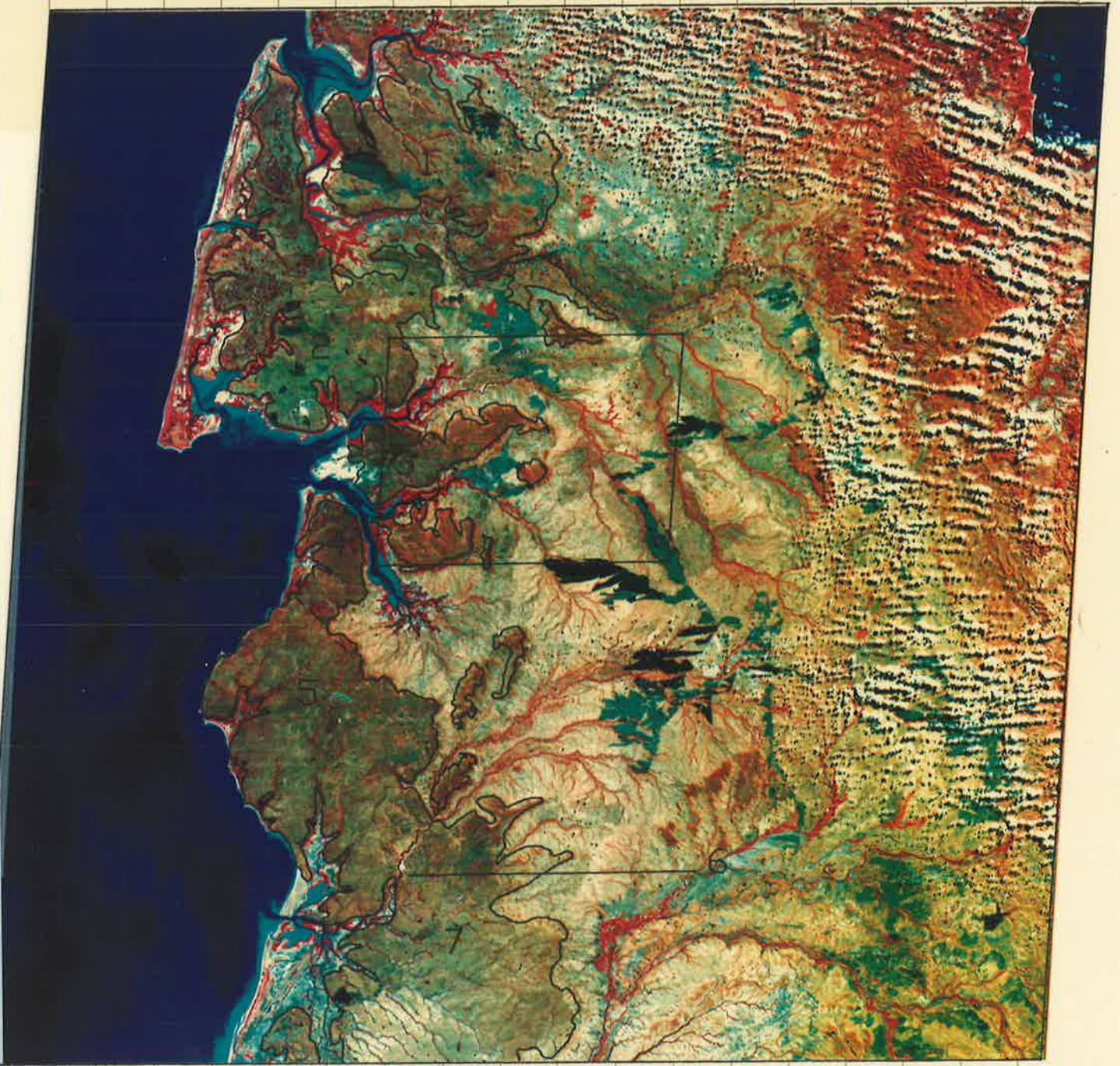


Figure 4.8 Landsat false colour composite - Cape Weymouth scene -
 histogram normalised, bands 4, 5 and 7, dry season
 image (after Huntington et al., 1982)

This image is the optimum enhancement of the Cape Weymouth scene, produced by histogram normalisation. The input file used for this process includes all data from the rectangular area shown on the overlay. Image ID and scale as for Fig. 4.7. The improvement over Fig. 4.7 is dramatic. The bauxite surfaces are clearly delineated (boundaries are shown on the overlay), and individual plateaux show variations in red and green colour tones. Intense red along creeks are mostly mangrove swamps.

KEY

- 1 Palm Creek Block
- 2 Andoom Block
- 3 Weipa Block
- 4 Hey Point Block
- 5 Pera Head Block
- 6 Watson River
- 7 Watson River Block



WEYMOUTH, QLD. HG.
MINERAL PHYSICS 01036-00004-00000 LS ■ 26JUL73 18- 1-80 B.02

Figure 4.9 Density sliced Transformed Vegetation Index image of
the Weipa Peninsula (after Huntington et al., 1982).

The image has been smoothed with a low-pass filter. Note the east-west trend in the index on the plateau top. Blue = low TVI7 values, dark red = high TVI7 values. High values depict areas of most intense (most luxuriant, most dense) vegetation response. M = mangrove areas. Image ID as for Figs. 4.7 and 4.8. Scale as shown.

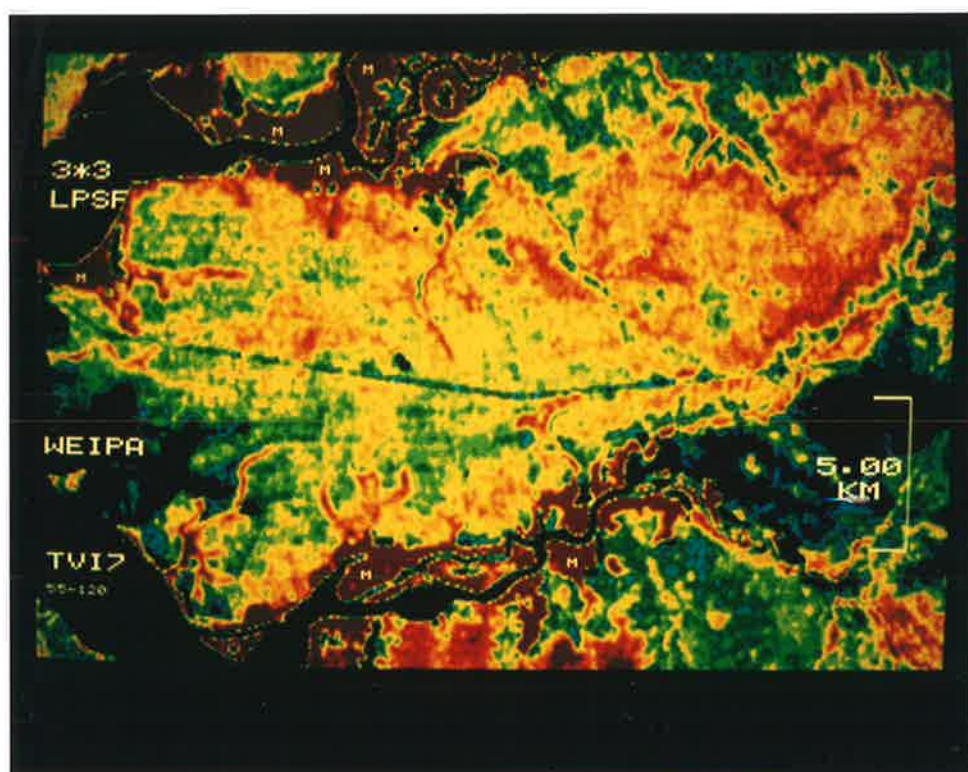


Figure 4.10 Landsat false colour composite - sub-scene of Cape Weymouth - linear stretch of bands 4, 5 and 7, wet season image.

Image ID No. 22289-23511, 29 April, 1981. Technique used for production of this image is essentially the same as for Fig. 4.8, but the image shown is sub-sampled and photographed from a TV screen. Note overall increased redness of the image compared to Fig. 4.8, and larger black 'swamp' areas on Weipa Peninsula.

Figure 4.11 Landsat false colour composite - Weipa Peninsula - linear stretch, bands 4, 5 and 6.

Image ID as for Figs. 4.7 and 4.8. The image has been photographed from a full resolution TV display. The same trends are present as in Fig. 4.8, but they are not as well defined.

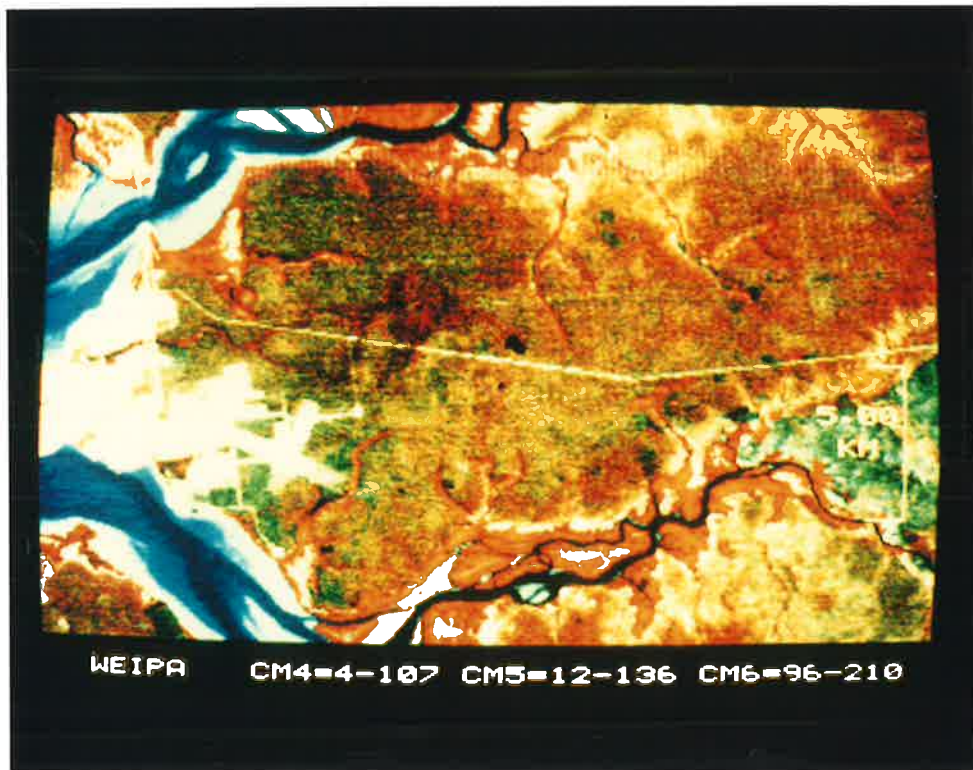
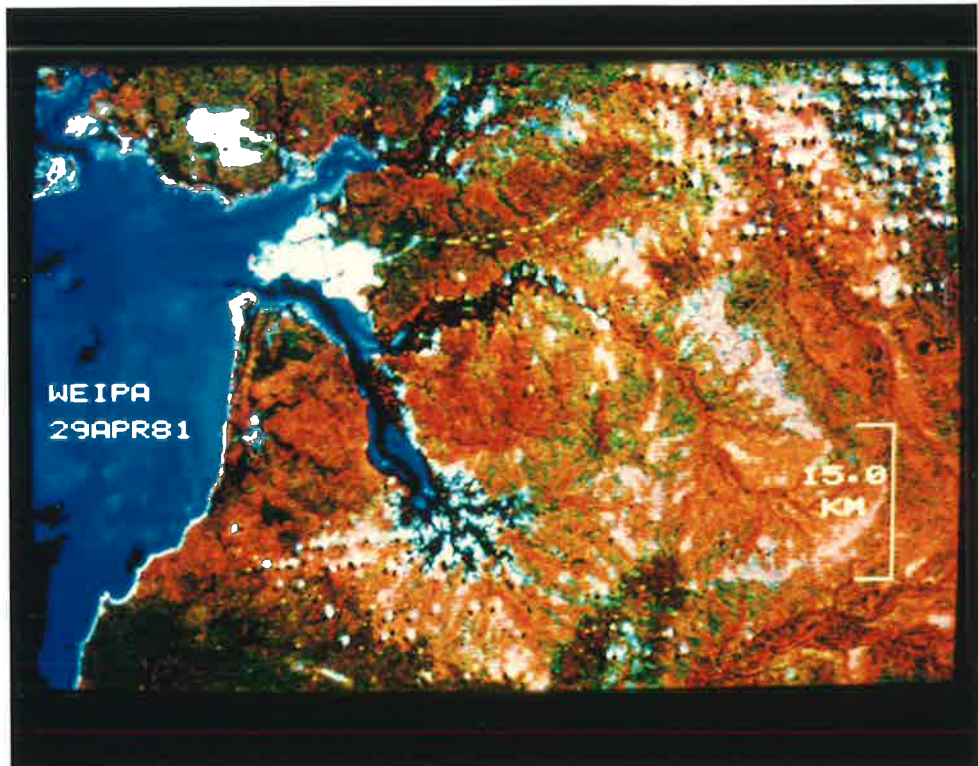


Figure 4.12a&b Bauxite isopachs from a portion of Weipa Peninsula
(after Evans, 1972 and O'Sullivan, pers. comm.,1982).

Figure 4.12a shows the interpretation of Evans (1972), indicating a structural control (fold axes) on ore thickness.

Figure 4.12b depicts Evans' 'synclinal axes' in highlighted colour. The 'synclinal axes' are more likely to be channels of an older drainage system, while 'anticlinal axes' would be interflaves (K. O'Sullivan, pers. comm., 1982).

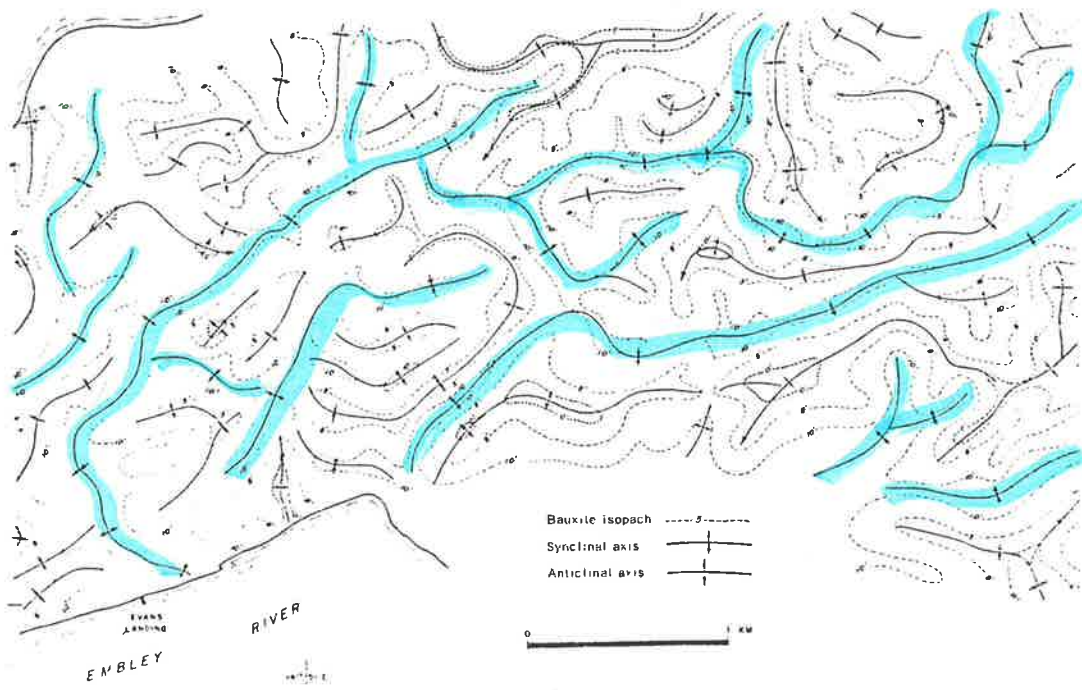
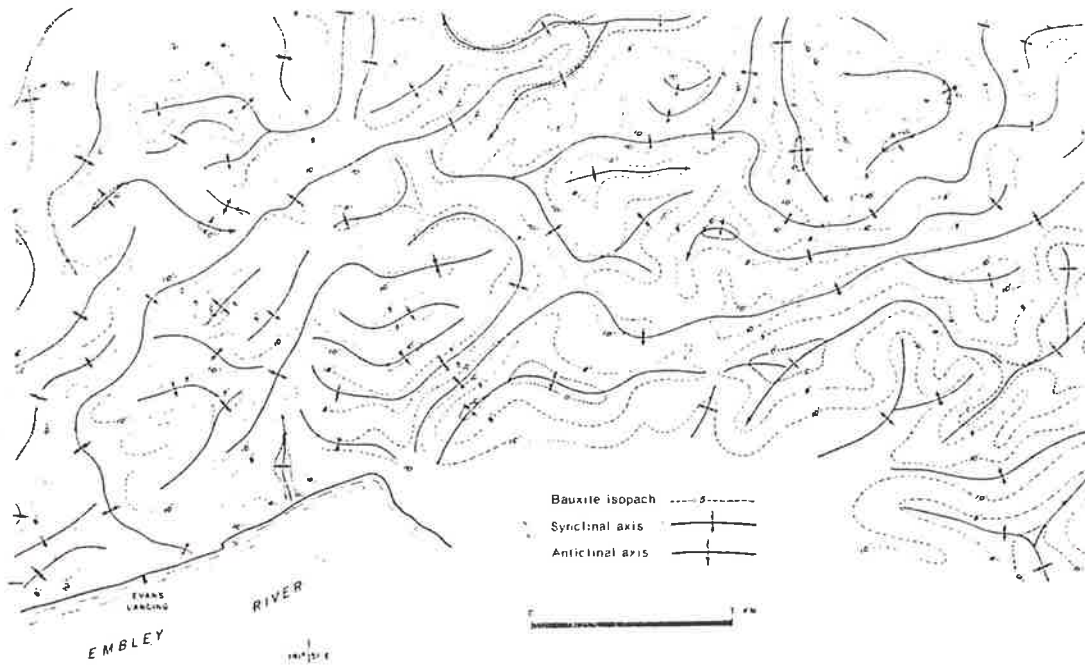


Figure 4.13 Landsat false colour composite - Weipa Peninsula -
linear stretch, bands 4, 5 and 7, dry season image.

Image ID as for Fig. 4.7. Contrast modification limits as shown. Arrows point to drilling-grid lines. East-west arrows are aligned along coordinate 20,000N. Note again the gradational colour change from B (green) to A (red) along the peninsula. BF = bush fire scar. The overlay indicates NNW-SSE linear trend.

Figure 4.14 Landsat false colour composite - Weipa Peninsula -
linear stretch, bands 4, 5 and 7, wet season image.

Band 4 has been smoothed with a 4*4 low pass box filter to reduce residual striping. Image ID as for Fig. 4.10. Note the red fringe to the bauxite plateau.

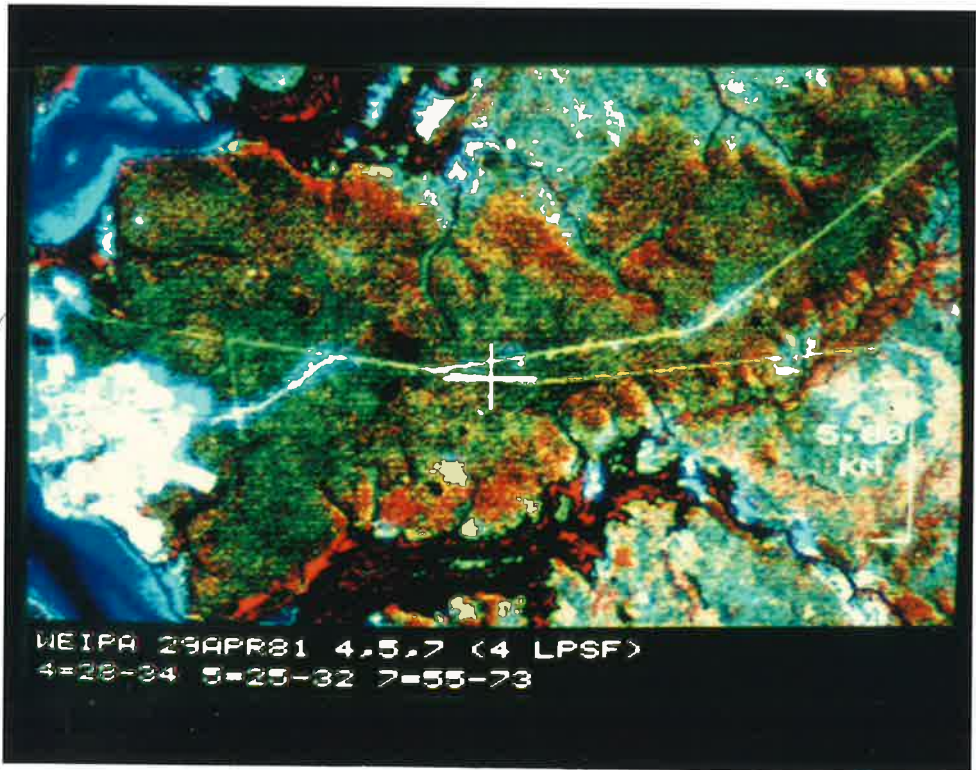
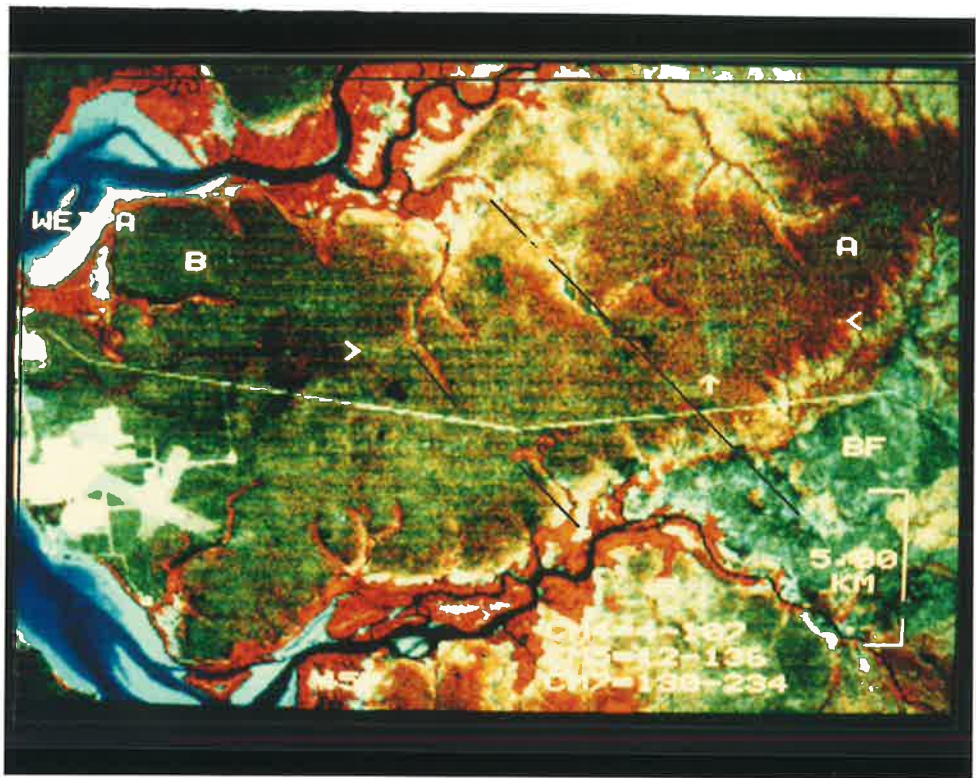


Figure 4.15 Landsat colour ratio composite - Weipa Peninsula -
 histogram normalised, wet season image.

The ratios used were 4/5 (blue), 5/6 (green) and 6/7 (red). Note that the colour variation is mainly from the edges towards the centre of the plateau, rather than along the peninsula (as is the case with dry season imagery).

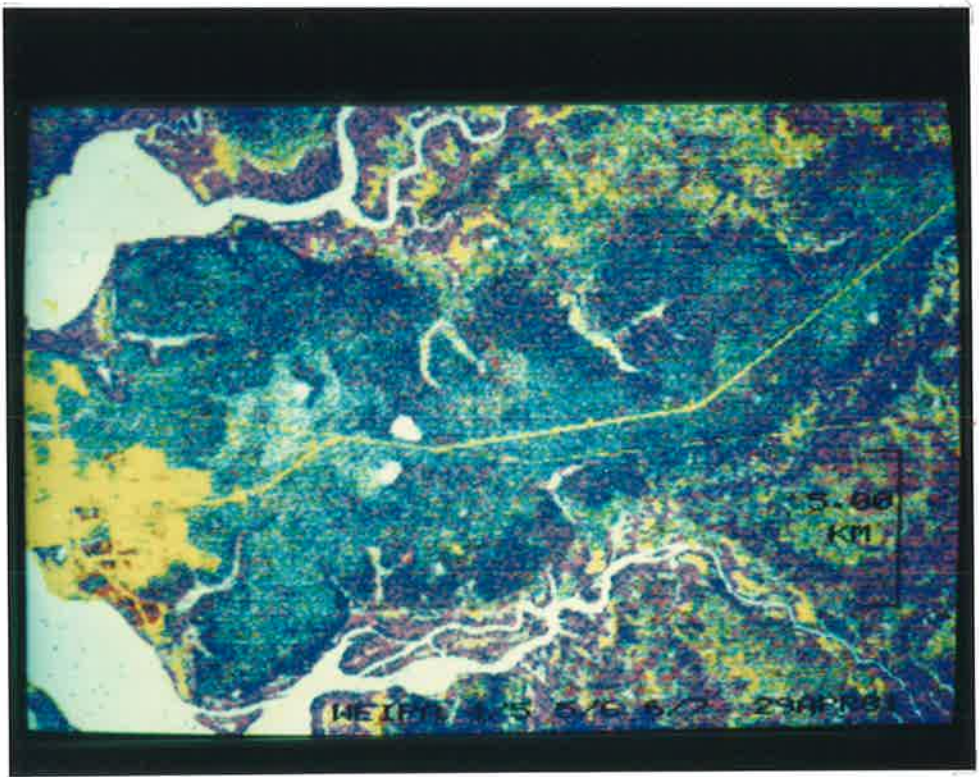
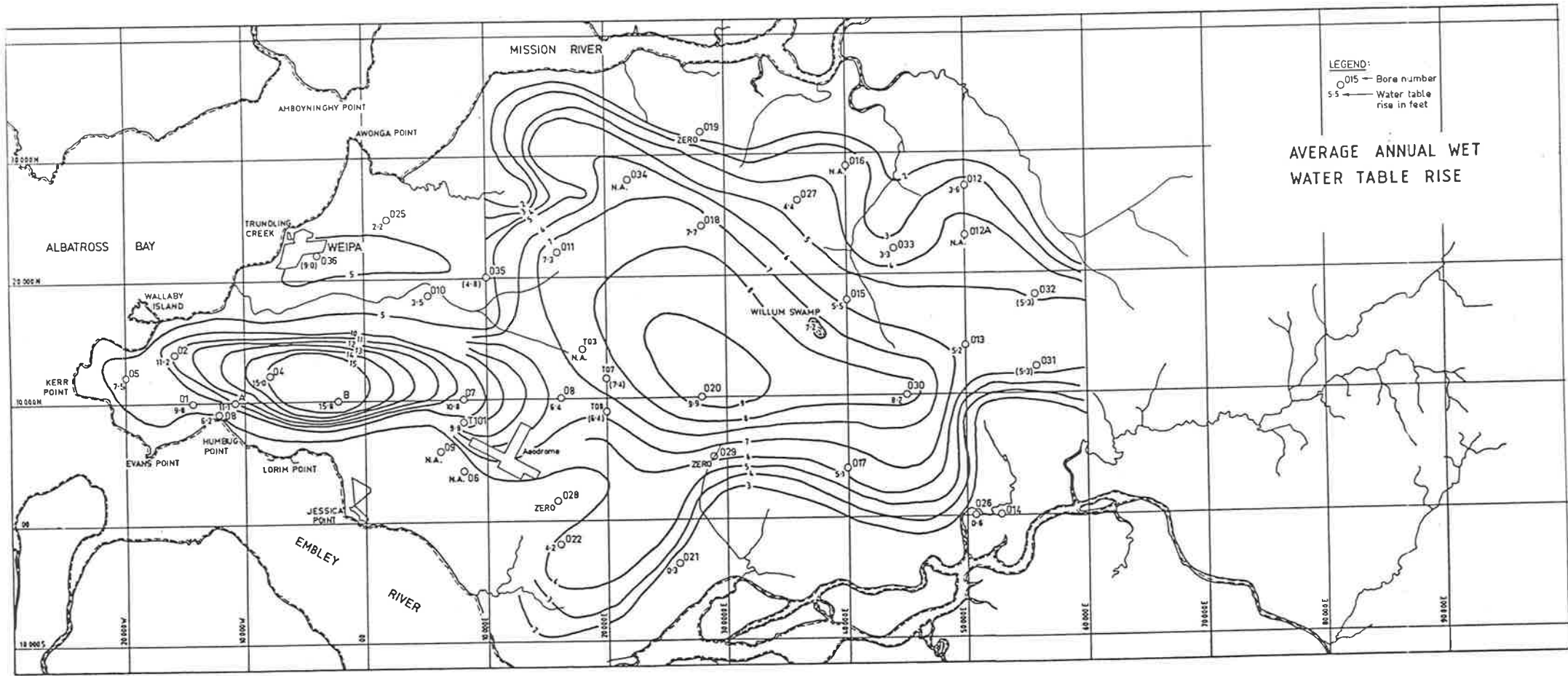


Figure 4.16 Average annual wet season water table rise over Weipa Peninsula (after Coffey and Hollingsworth, 1971).

Note the similarity of the shape of the contours and the distribution of colour variations in Fig. 4.15.



LEGEND:
 ○ 015 — Bore number
 5.5 — Water table rise in feet

AVERAGE ANNUAL WET WATER TABLE RISE

Figure 4.17 Variation of SiO_2 content of soil samples along Weipa Peninsula.

For Figs. 4.17 to 4.20 data come predominantly from the 20,000N line (east of 50,000E), and the 10,000N line (west of 50,000E). Precise locations are documented in Table 4.1. The regression lines have been calculated for the y-direction.

SiO_2 values show a strong trend to increasing values from west to east along the peninsula.

WEIPA SOIL SiO2 VERSUS E-W GRID POSITION

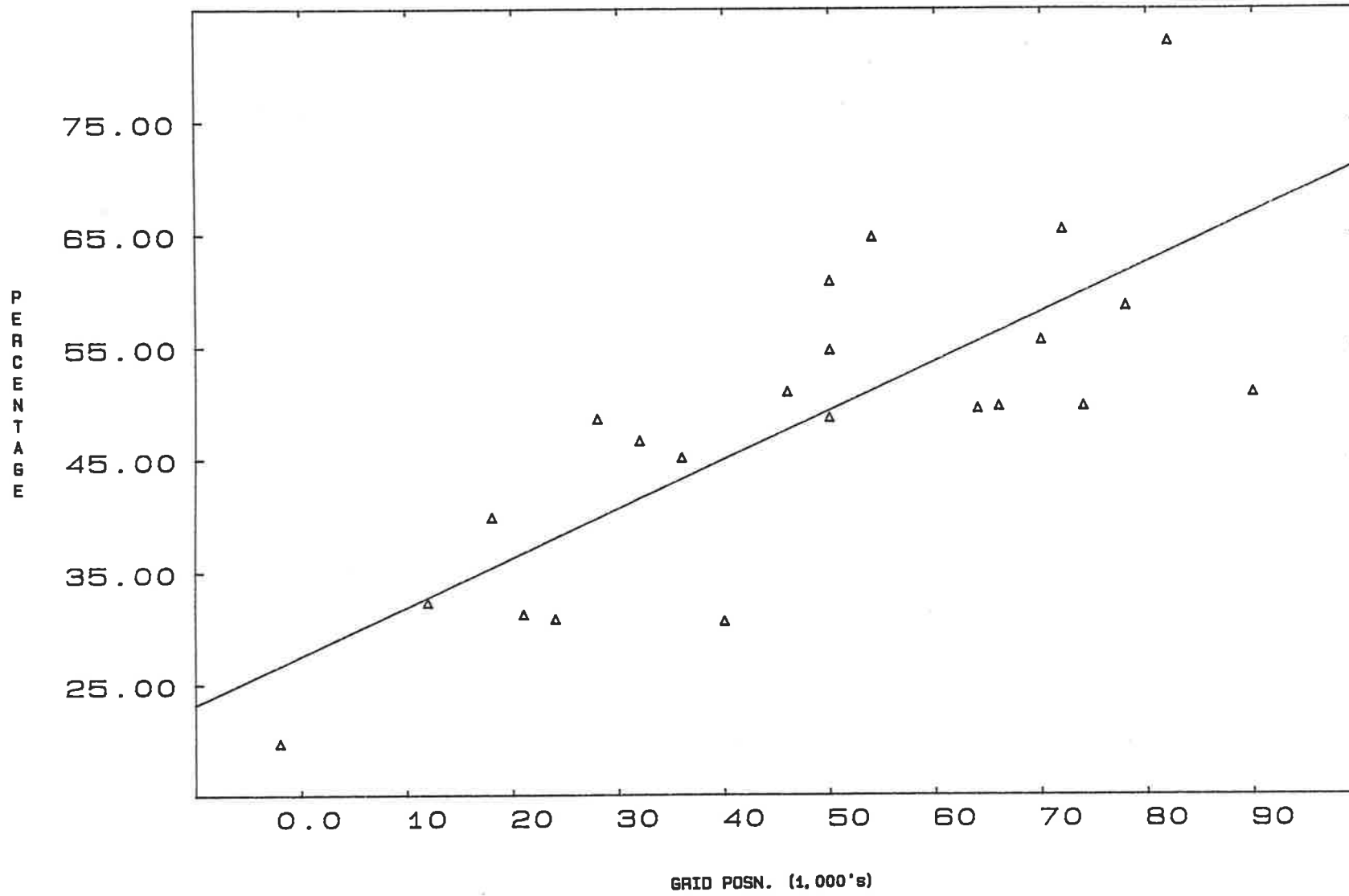


Figure 4.18 Variation of Al_2O_3 content of soil samples along Weipa Peninsula.

Note that the trend is the inverse of that exhibited by SiO_2 in Fig. 4.17.

WEIPA SOIL AL2O3 VERSUS E-W GRID POSITION

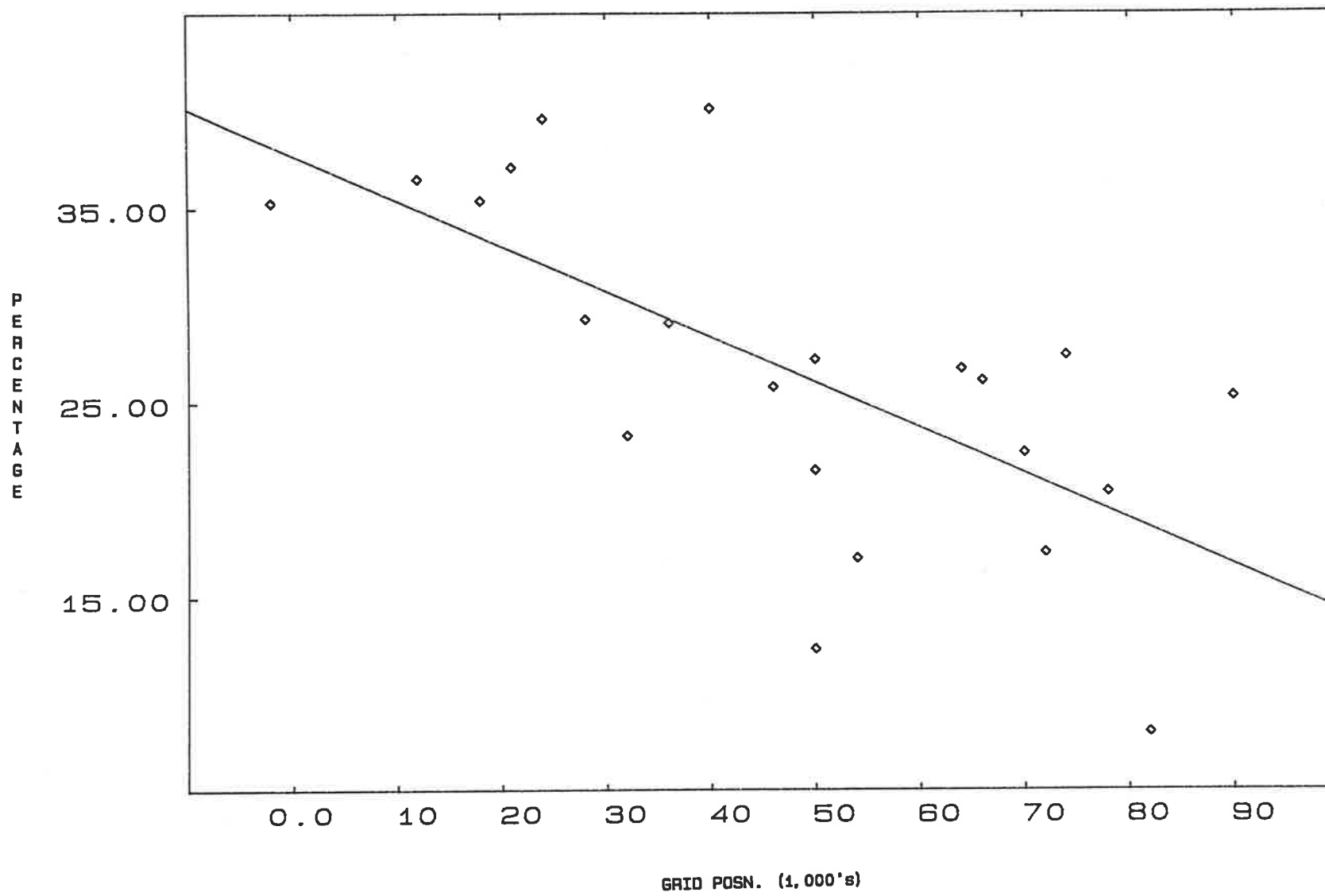


Figure 4.19 Variation of TiO_2 content of soil samples along Weipa Peninsula.

WEIPA SOIL TI02 VERSUS E-W GRID POSITION

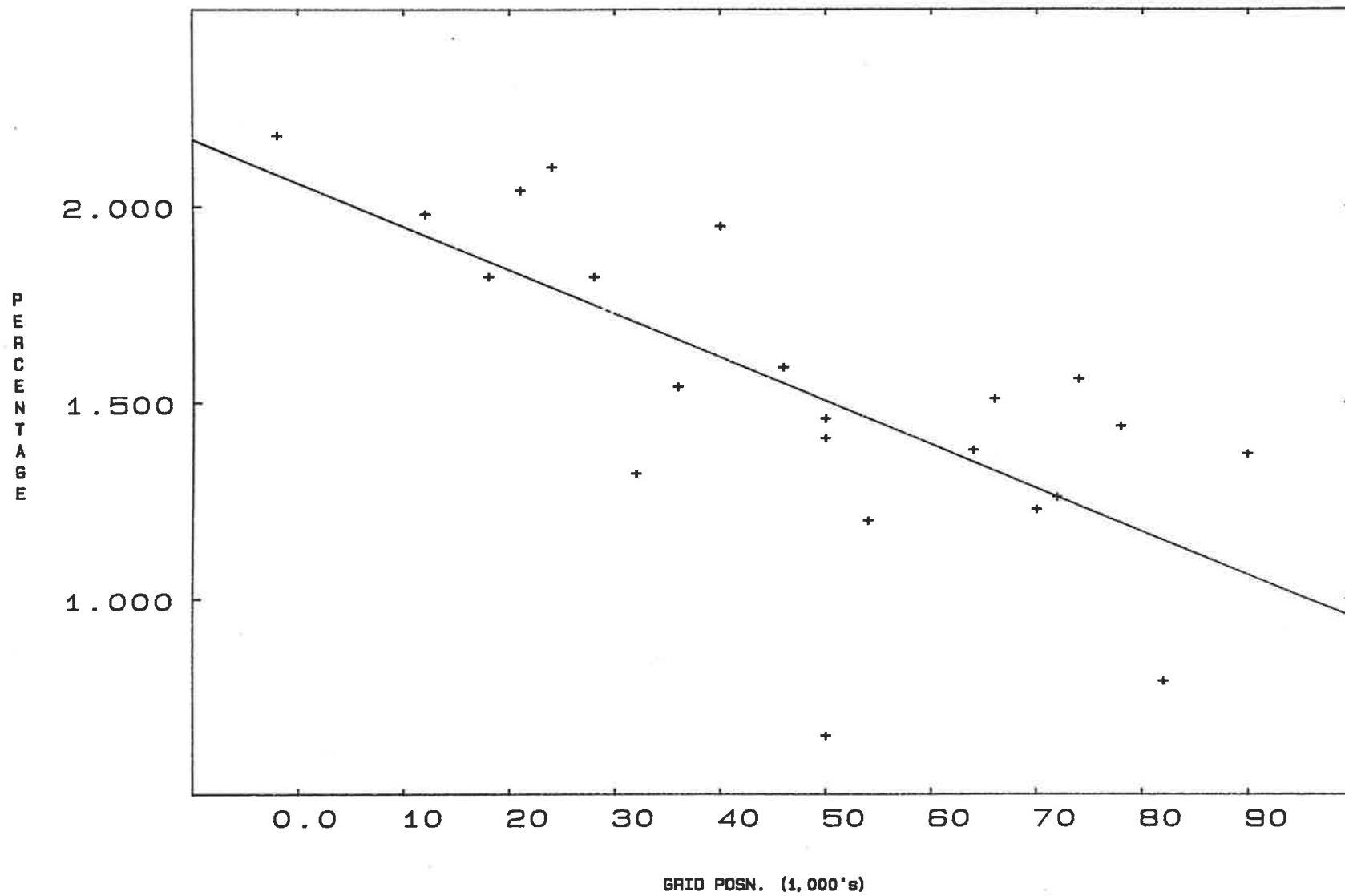


Figure 4.20 Variation of Fe_2O_3 content of soil samples along Weipa Peninsula.

Note that the two outliers at 2,000W and 50,000E have been included in the regression calculation. Exclusion of these values will decrease the slope of the line, but still leave a weak trend in the same direction. All trends in Figs. 4.17 to 4.20 are compatible with decreased leaching of residual soil from west to east along the peninsula.

WEIPA SOIL FE2O3 VERSUS E-W GRID POSITION

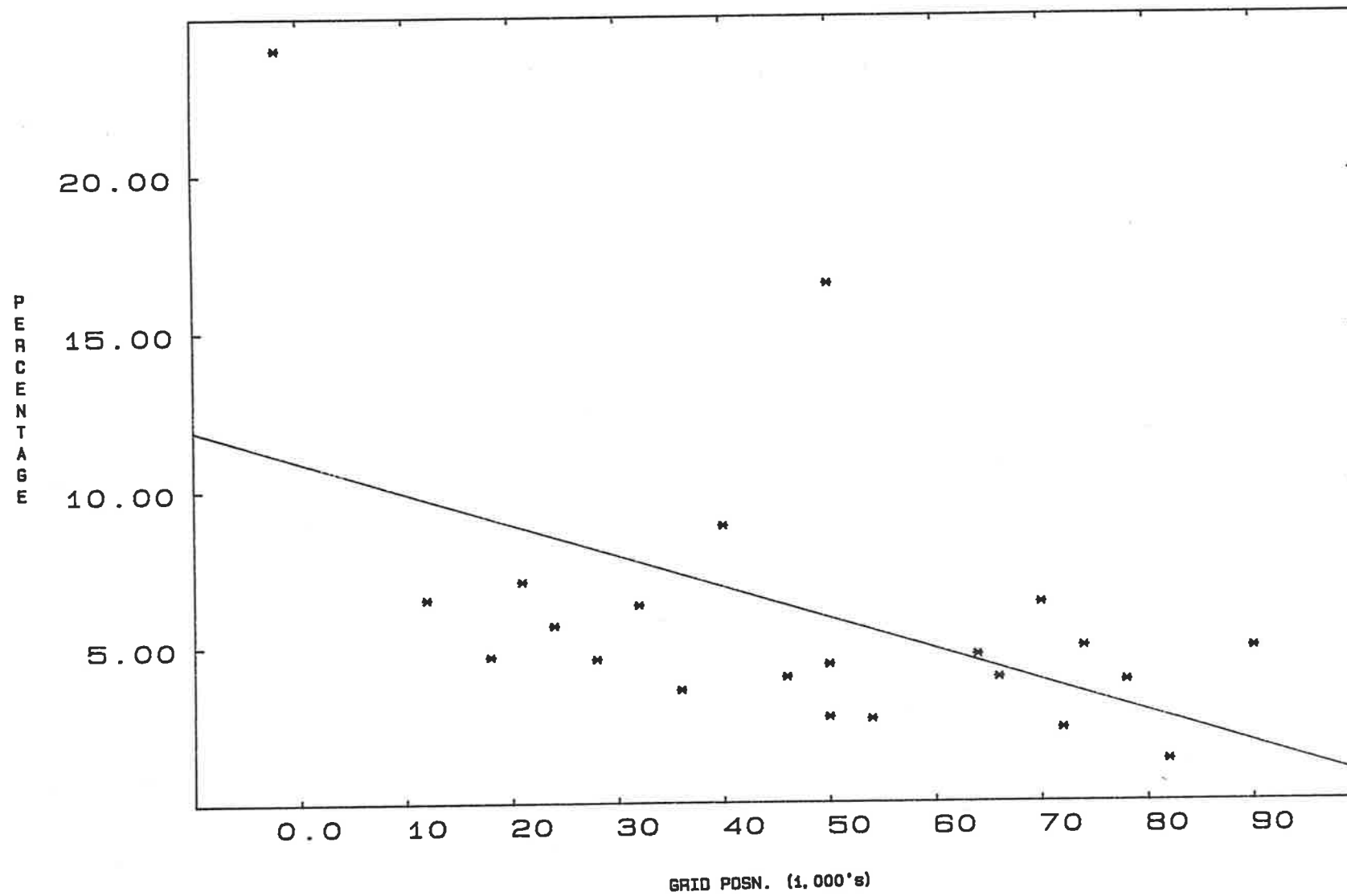
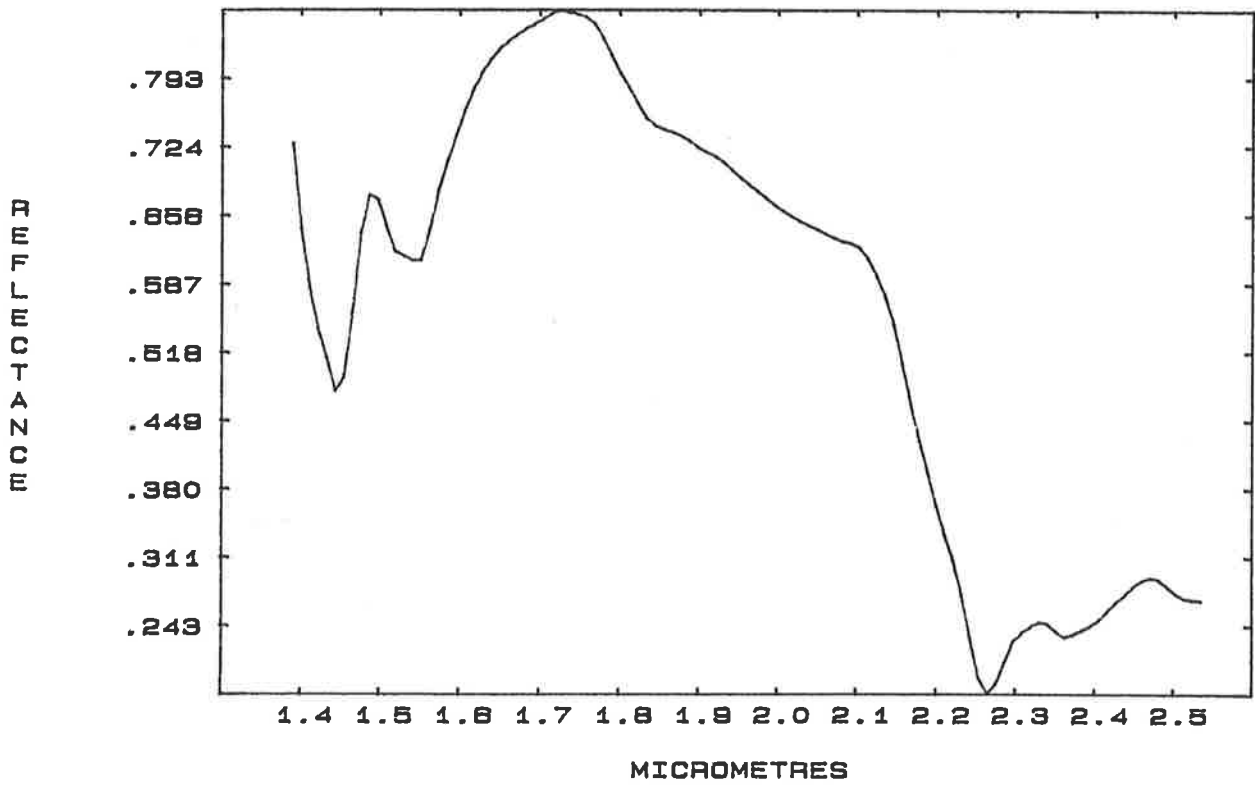


Figure 4.21 SWIR spectrum of synthetic gibbsite powder.

The spectrum shown was measured from a synthetic sample of gibbsite (aluminium trihydrate ($\text{Al}(\text{OH})_3$)). The spectral shape is almost identical to that of the gibbsite sample measured by Hunt et al., 1971. Note the unusual wavelength of the major absorption feature ($2.27\mu\text{m}$).

SYNTHETIC GIBBSITE



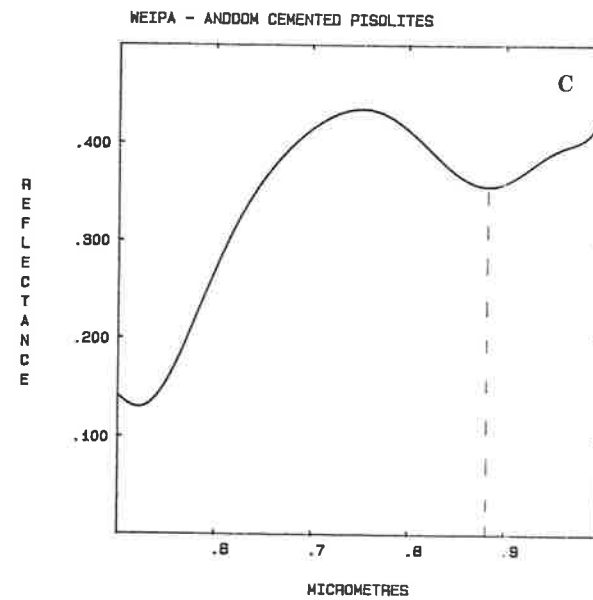
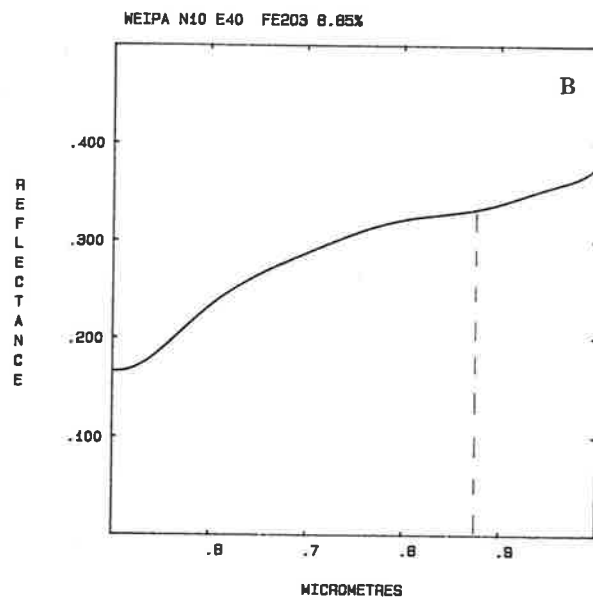
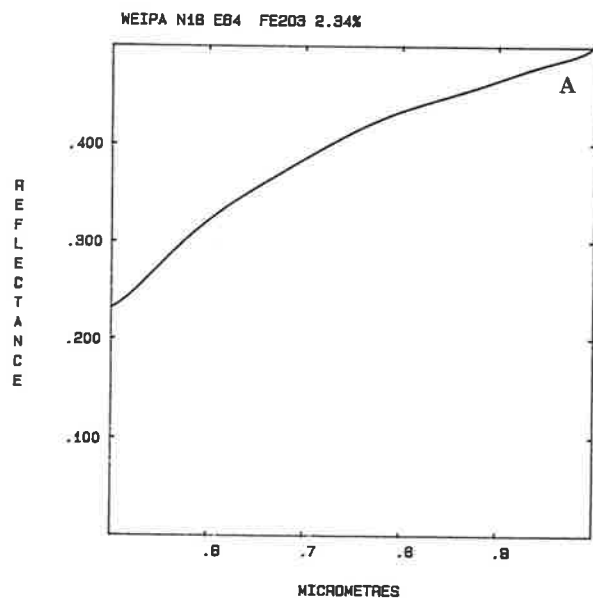
Figures 4.22a-c Representative VNIR laboratory spectra of Weipa soils.

All spectra have been smoothed by reconstruction using Tchebychev polynomials.

Figure 4.21a is representative of the majority of VNIR soil sample spectra, with increasing reflectance from .5 to 1.0 μm , and no iron absorption features. All samples containing less than about 6% Fe_2O_3 fall into this category.

Figure 4.21b is an example of one of the few spectra which exhibit any iron oxide spectral features. Even so, the absorption is very subtle, and takes the form of a minor dip centred near .85 μm .

Figure 4.21c is the spectrum of a haematite-rich sample of cemented pisolite ore. This sample contains 24% Fe_2O_3 .



Figures 4.23a-d Representative SWIR laboratory spectra of Weipa soils.

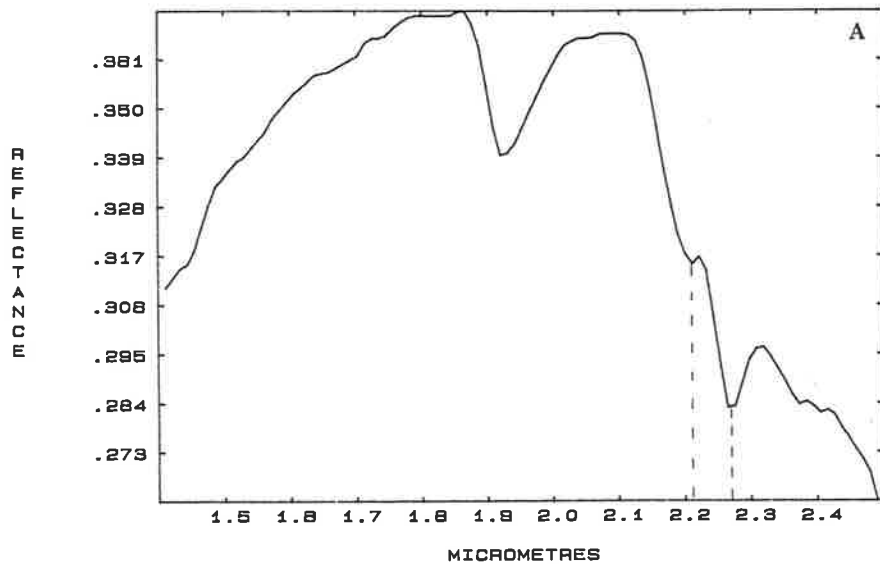
Figure 4.23a depicts a spectrum which has a particularly strong (for the Weipa samples) absorption feature in the $2.27\mu\text{m}$ region, indicative of the presence of bauxite minerals.

Figure 4.23b shows an unusually strong $2.2\mu\text{m}$ absorption feature, with a weak shoulder developed at $2.17\mu\text{m}$. These features are attributed to kaolin.

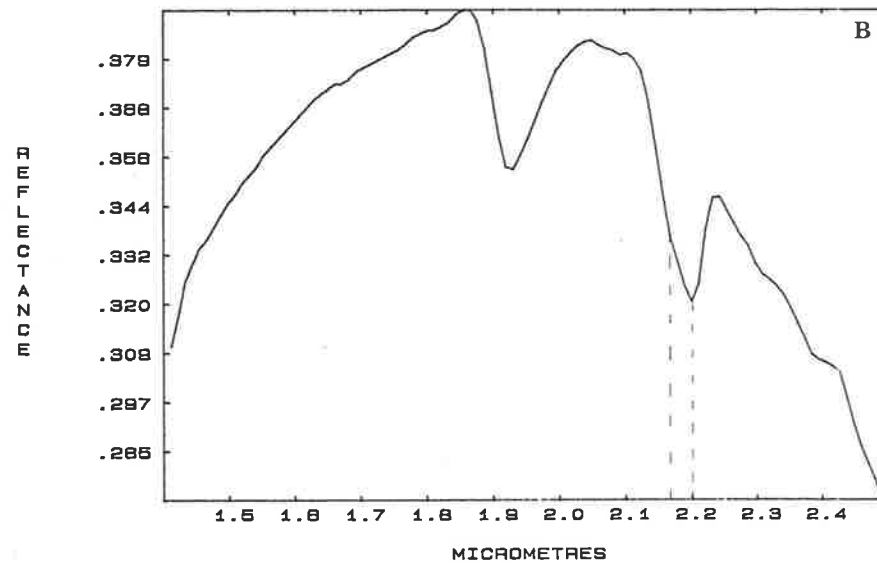
Figure 4.23c is representative of the majority of the Weipa soil samples. It falls between the extremes depicted in Figs. 4.23a&b, and has only weak to moderate absorption in both the 2.2 and $2.27\mu\text{m}$ regions.

Figure 4.23d is the spectrum of a pisolitic ore sample from Andoom, which is composed largely of gibbsite, boehmite and iron oxides.

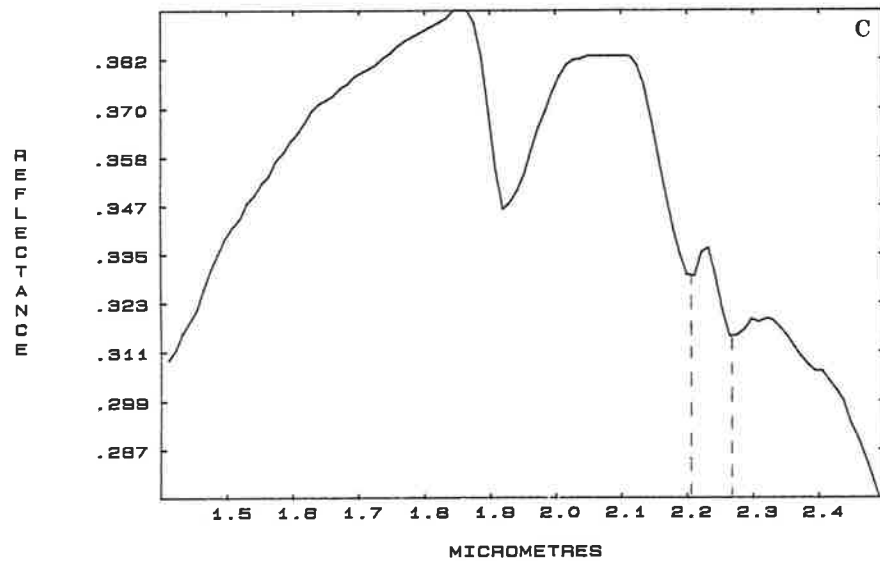
31/7/84 WEIPA N10 E12



31/7/84 WEIPA N30 E50



31/7/84 WEIPA N20 E58



31/7/84 ANDOOM N21 W13 CEMENTED PISOLITE

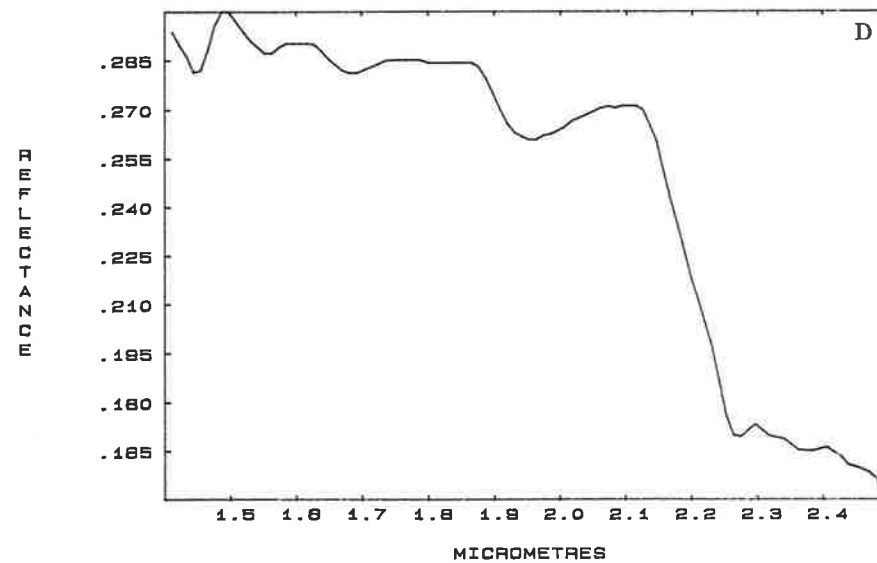
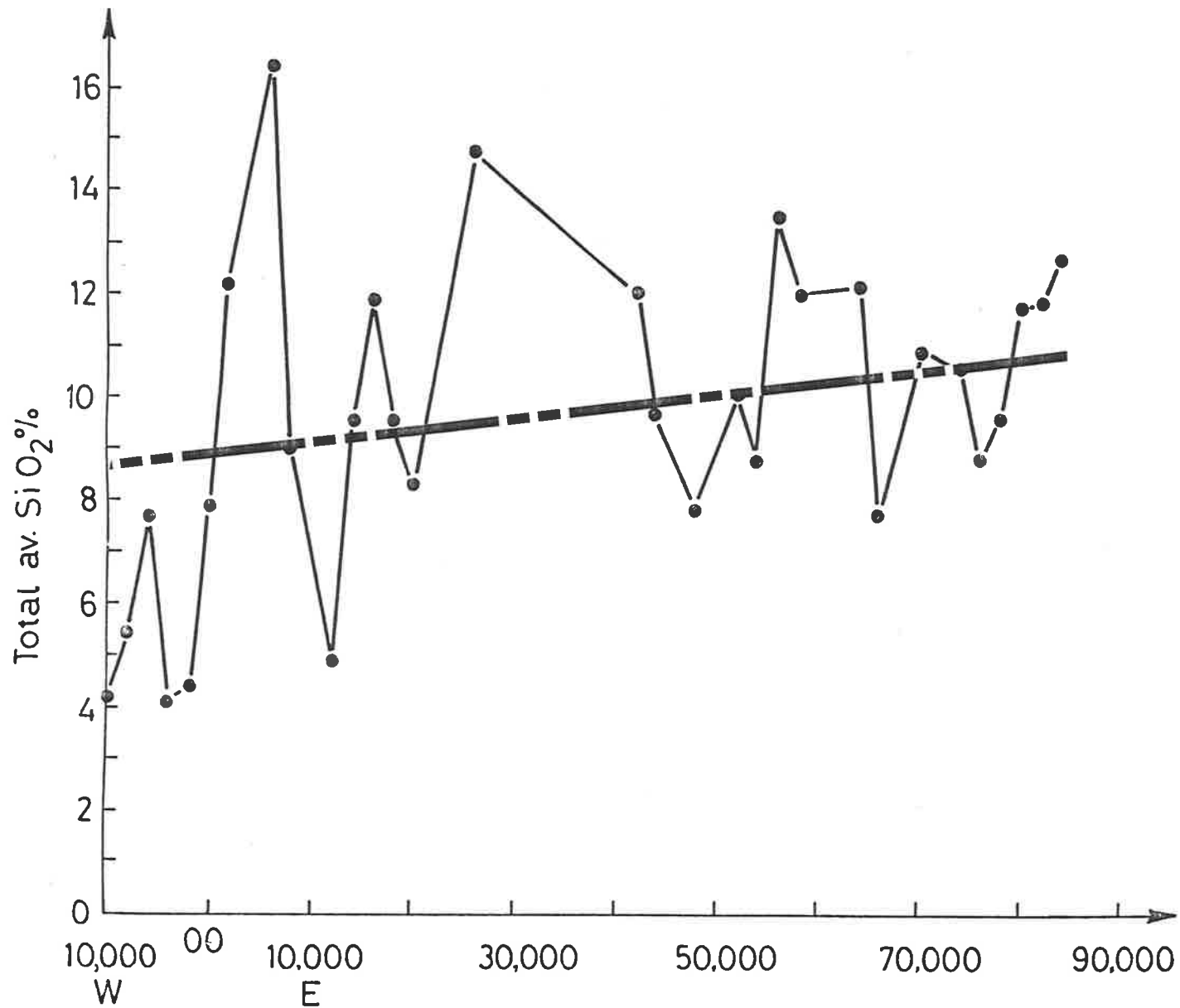


Figure 4.24 Variation of SiO_2 content of ore on the 2000N line
along Weipa Peninsula (Comalco drilling data).

Data points plotted in Figs. 4.24 and 4.25 are averages of multiple analyses of samples taken from each drill hole. Each point should therefore be representative of the entire ore zone at that location. In both plots the thick line represents a line of best fit.

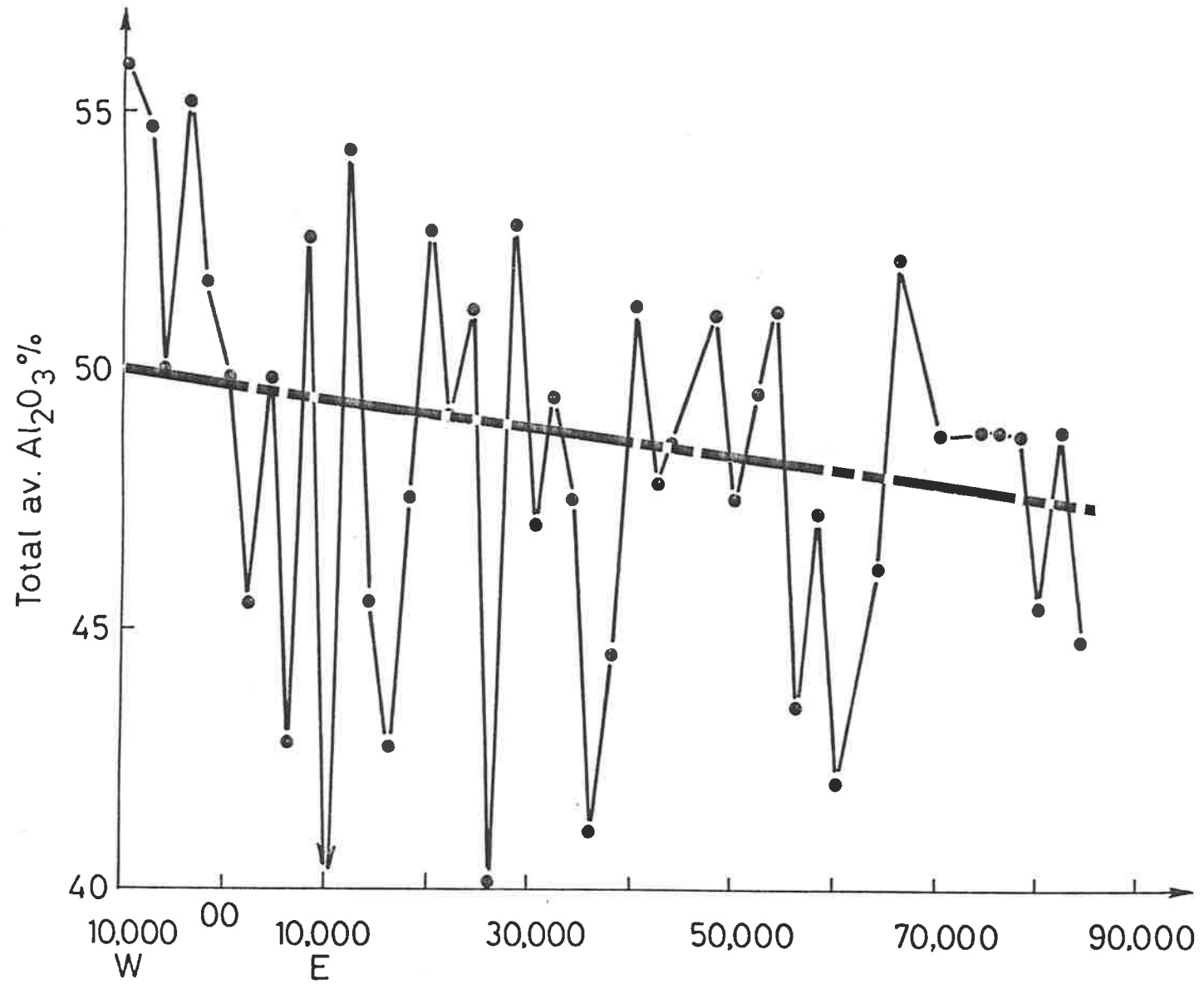
Although there is considerable scatter in the SiO_2 data, a trend of increasing silica content from west to east is evident.



Total av. Silica % vs E-W grid position

Figure 4.25 Variation of Al_2O_3 content of ore on the 2000N line
along Weipa Peninsula (Comalco drilling data).

Like Fig. 4.24 there is considerable scatter in the data, but an inverse trend is apparent, with decreasing alumina content from west to east.



Total av. Al₂O₃ % at 180°C (nearest avail. data) vs E-W grid positin.

Figure 4.26 Variation of SiO_2 content of ore (N-S lines averaged)
along Weipa Peninsula (Comalco drilling data).

Figures 4.26 and 4.27 are plots of ore geochemistry, but all data for each north-south grid line has been averaged to a single value. In both plots the thick line is a line of best fit.

For SiO_2 the trend of increasing values to the east is stronger, and there is less scatter in the data.

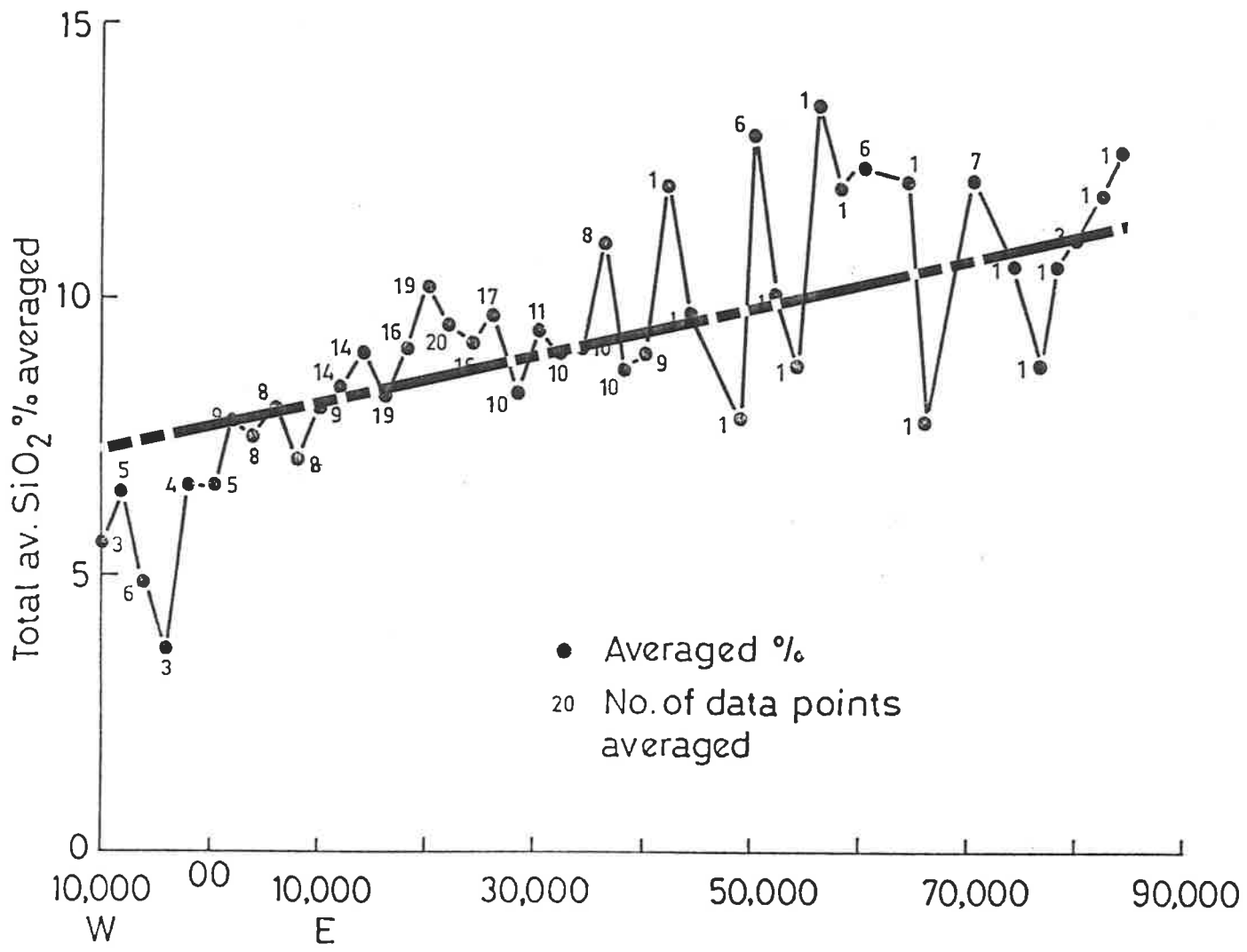


Figure 4.27 Variation of Al_2O_3 content of ore (N-S lines averaged) along Weipa Peninsula (Comalco drilling data).

Like SiO_2 the Al_2O_3 values show a stronger trend, with less scatter in the data, once the north-south lines have been averaged.

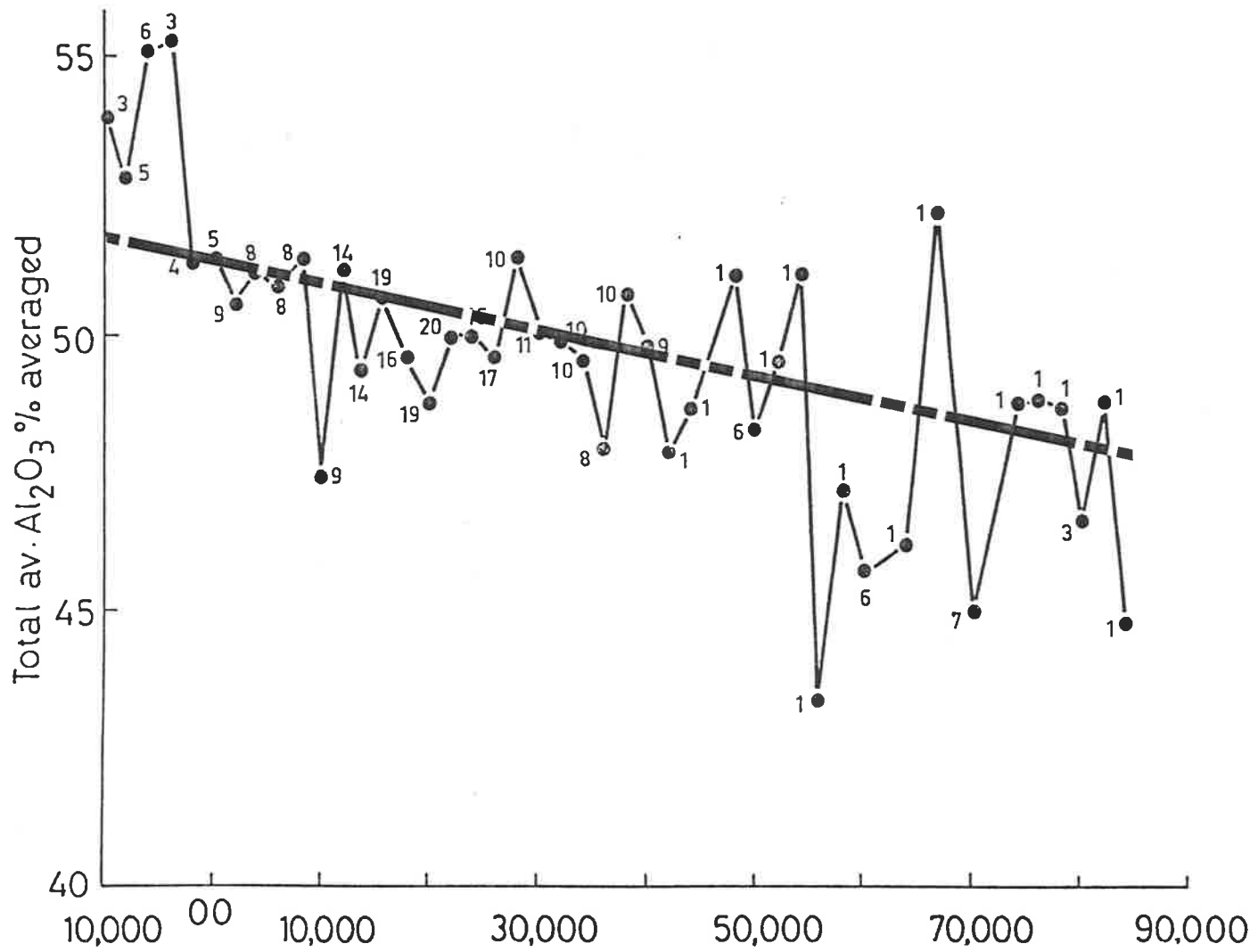


Figure 4.28 Image representation of Al_2O_3 content of ore over Weipa Peninsula (Comalco drilling data).

The image has been contrast stretched and density sliced. Each coloured pixel represents one 2,000 foot drilling block. Al_2O_3 values range from 36 to 55%. Red equals high values, yellow medium values and green and blue low values. Note the gradient from high to low values from west to east. The longest east-west data line is located on the 20,000N grid line. North is vertical.

Figure 4.29 Geometrically resampled image of Al_2O_3 content of ore over Weipa Peninsula (Comalco drilling data).

Same as Fig. 4.29, but the data has been geometrically resampled and registered to fit Landsat scene 1036-00004. North is now roughly 10° west of the vertical.

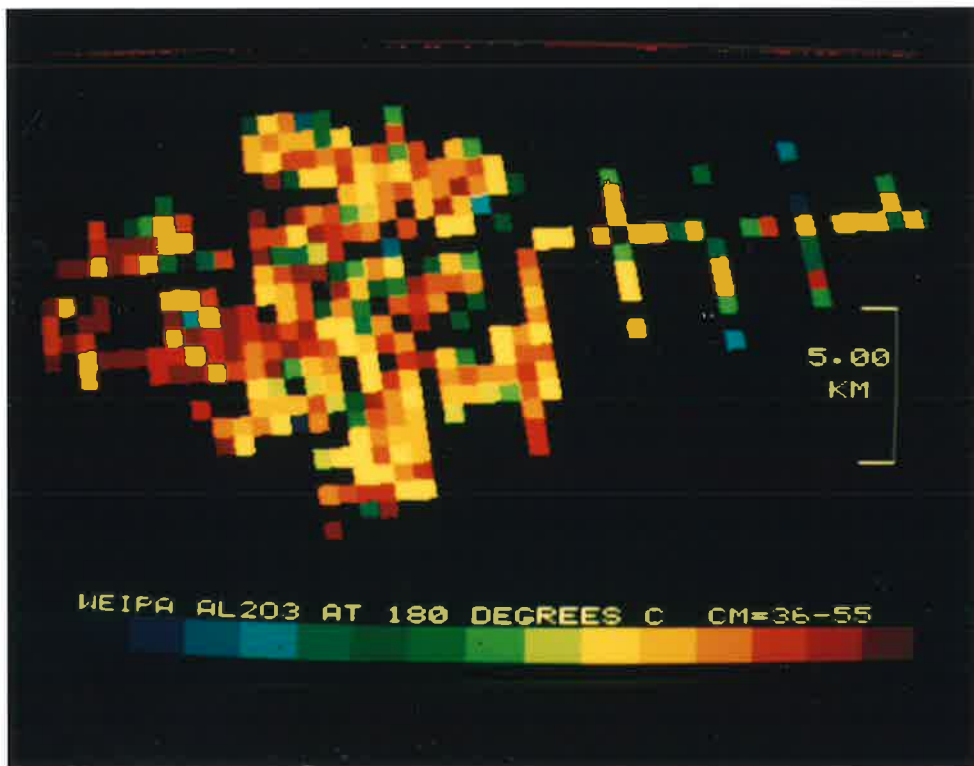
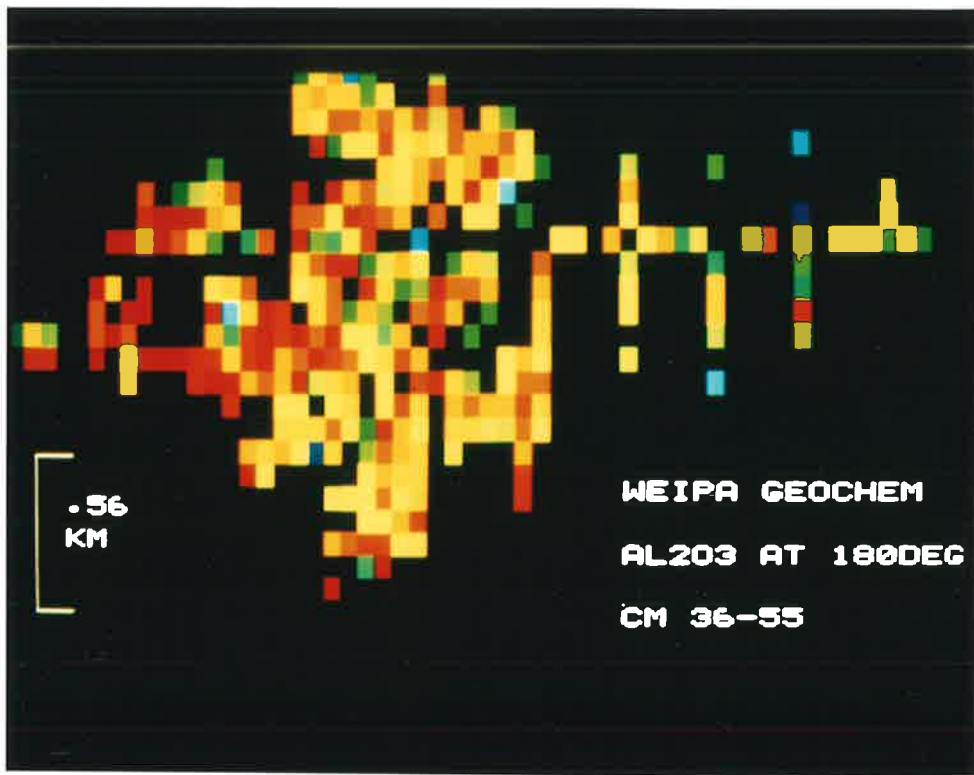


Figure 4.30 Geometrically resampled image of SiO₂ content of ore over Weipa Peninsula (Comalco drilling data).

The image has been density sliced, and values range from 15% (red) to 3% (blue). Note the inverse trend to Fig. 4.29.

Figure 4.31 Geometrically resampled image of ore thickness over Weipa Peninsula (Comalco drilling data).

The image has been density sliced, with values ranging from 14 feet (red) to 1 foot (blue). Where local variations in ore thickness are likely to be related to palaeotopography (see Fig. 4.12b), the regional trend depicted here could either be due to regional palaeotopographic variation or to a weak structural deformation post-dating ore formation. The author favours the latter alternative.

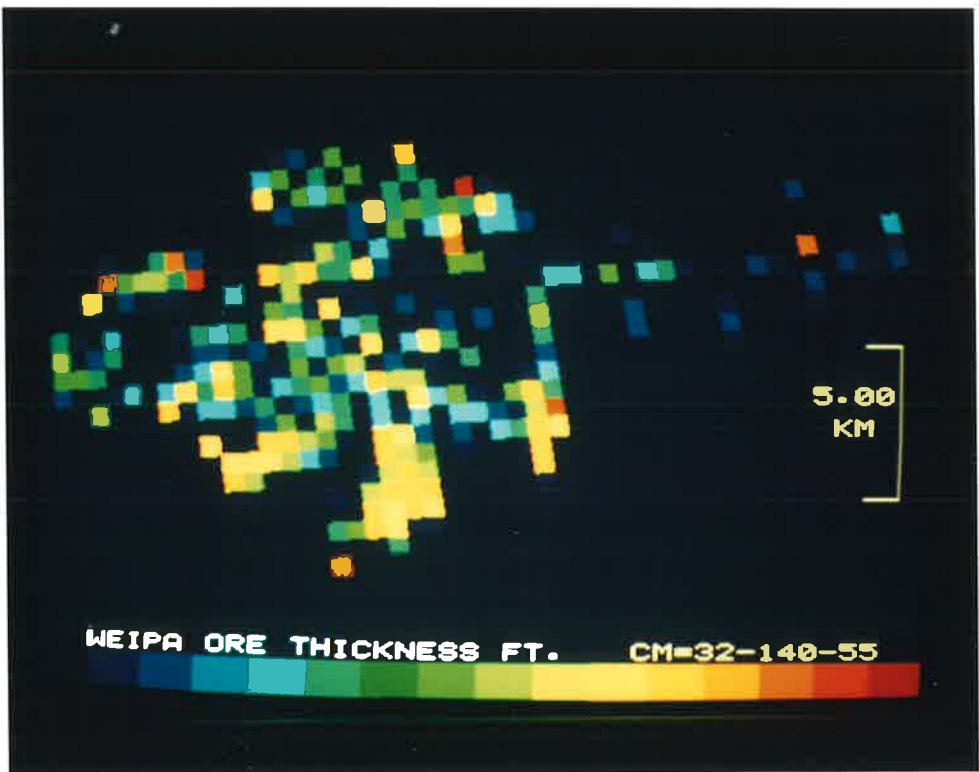
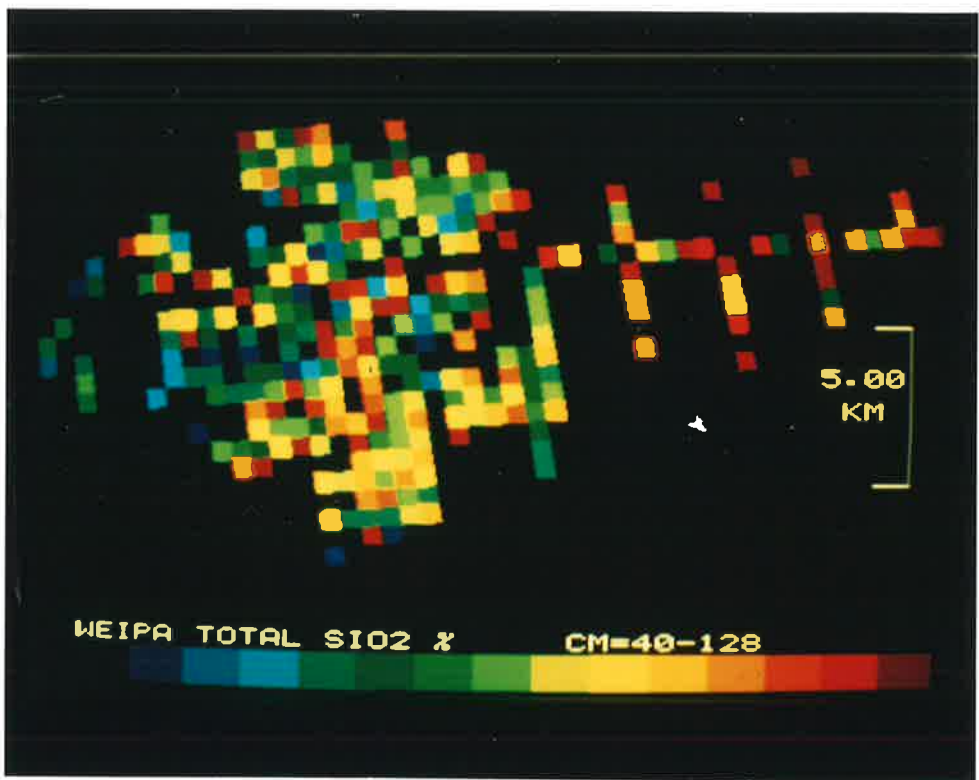


Figure 4.32 Landsat and geochemistry training areas and geochemical grid over Weipa Peninsula.

Training areas are shown as numbered, white boxes, while geochemistry grid is shown in blue. Landsat MSS bands 5 and 7 are displayed as green and red, respectively.

Figure 4.33 Landsat false colour composite - Weipa and Andoom mine sites - highlighting vegetation regeneration - linear stretch, bands 4, 5 and 7, wet season image.

The image ID No. is 22289-23511 (29th April, 1981). The different stages of regeneration are evident as areas of varying shades of red on the white background of the mines. Grey-blue areas have recently had their tree cover felled, but not removed.

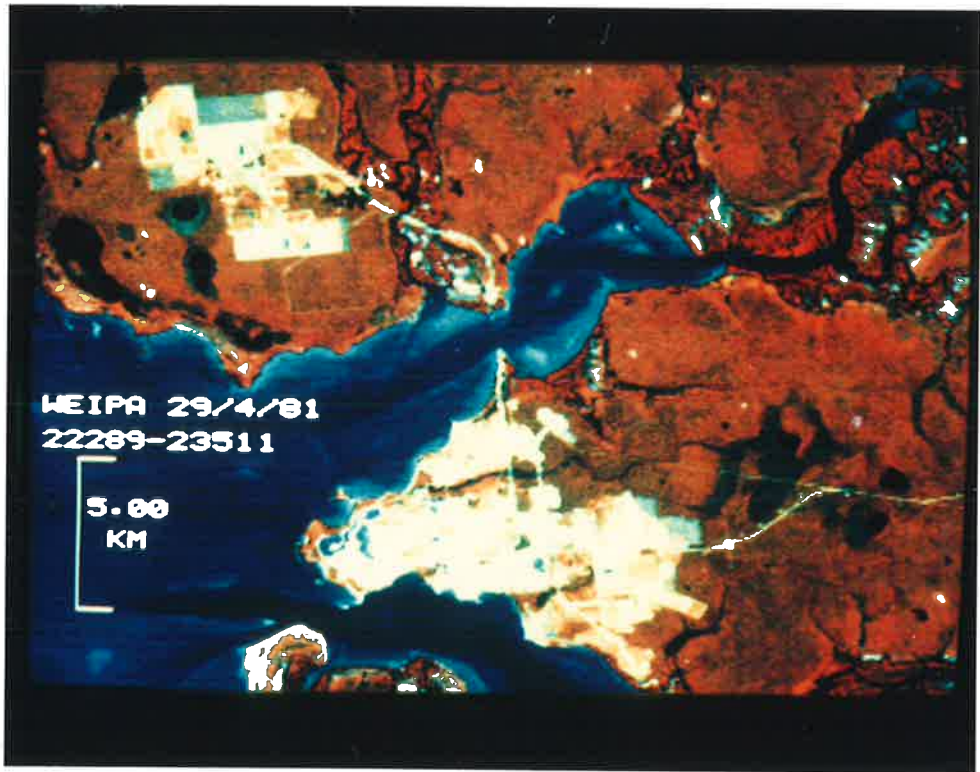
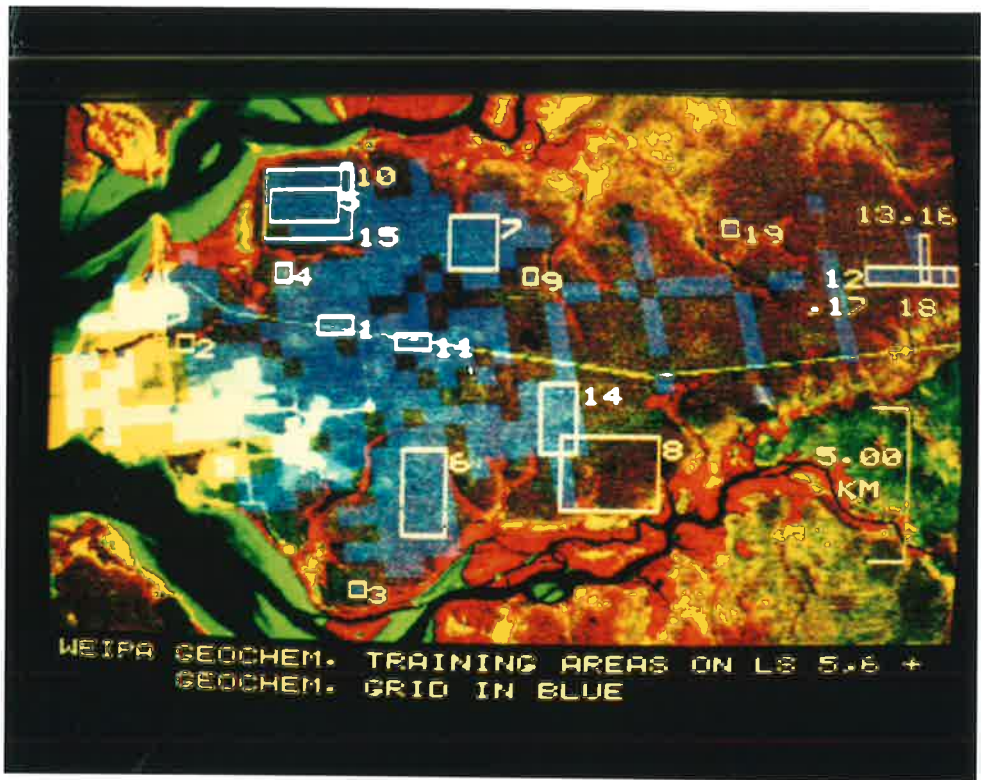
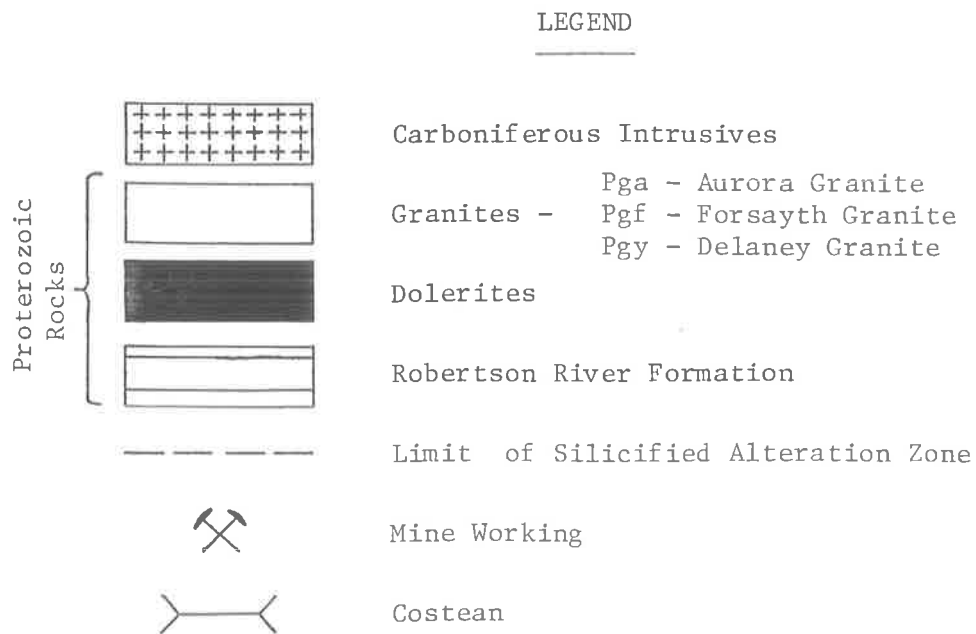


Figure 5.1 Geology map of Mt Turner.

The map has been compiled and simplified from BMR 1:25,000 compilation sheets. White mica (muscovite from the weathering of Proterozoic granites and Robertson River Formation metasediments) and sericite (from altered felsic intrusives) are widespread throughout the map area.



MOUNT TURNER GEOLOGY

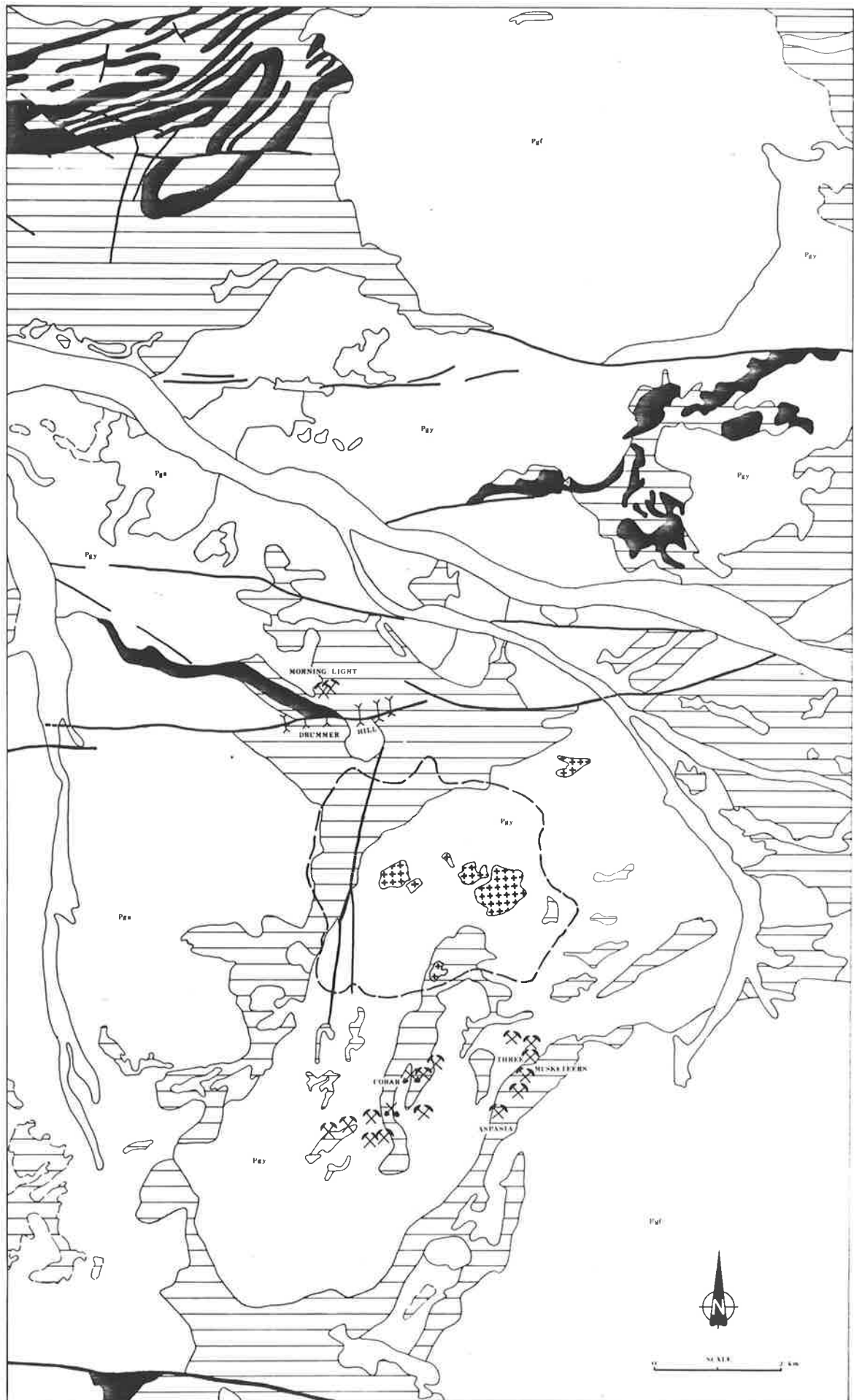
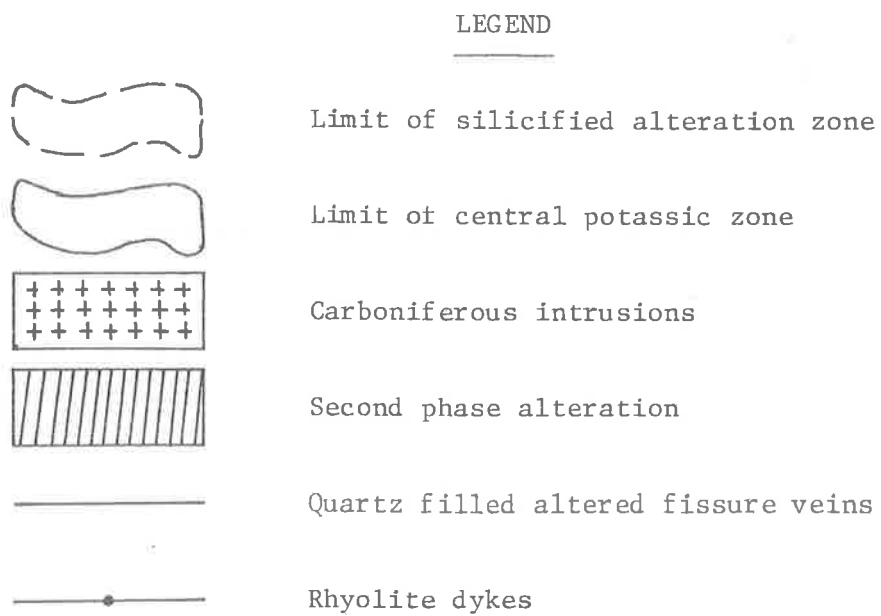


Figure 5.2 Photocopy of RC9 1:80,000 black and white aerial photograph of Mt Turner, with alteration (A) and flight line location (B) overlays.

Note that fissure veins and rhyolite dykes are best developed within the silicified zone, but have been omitted from the overlay for clarity.



2
RGETOWN



49-99
RUN 3



27-8-1;67

CAB
330

25,000



Ca 612
SAg 65
f-88.41



Figure 5.3 Landsat false colour composite - Mt Turner area -
linear stretch, bands 4, 5 and 7 (after Huntington,
pers. comm. 1983).

The vegetation anomaly over Mt Turner results in the high band 7
(red) response. The dark green areas are bushfire scars.

Figure 5.4 Landsat band ratio image - Mt Turner area - density
slice, ratio bands 5/6 (after Huntington, pers. comm.,
1983).

The vegetation anomaly over Mt Turner stands out as blue colours.
Also shown is the boundary of most intense "limonite" (derived from a
Landsat MSS 4/5, 4/6, 6/7 ratio image) resulting from alteration, mafic
intrusives and Robertson River Formation metasediments.

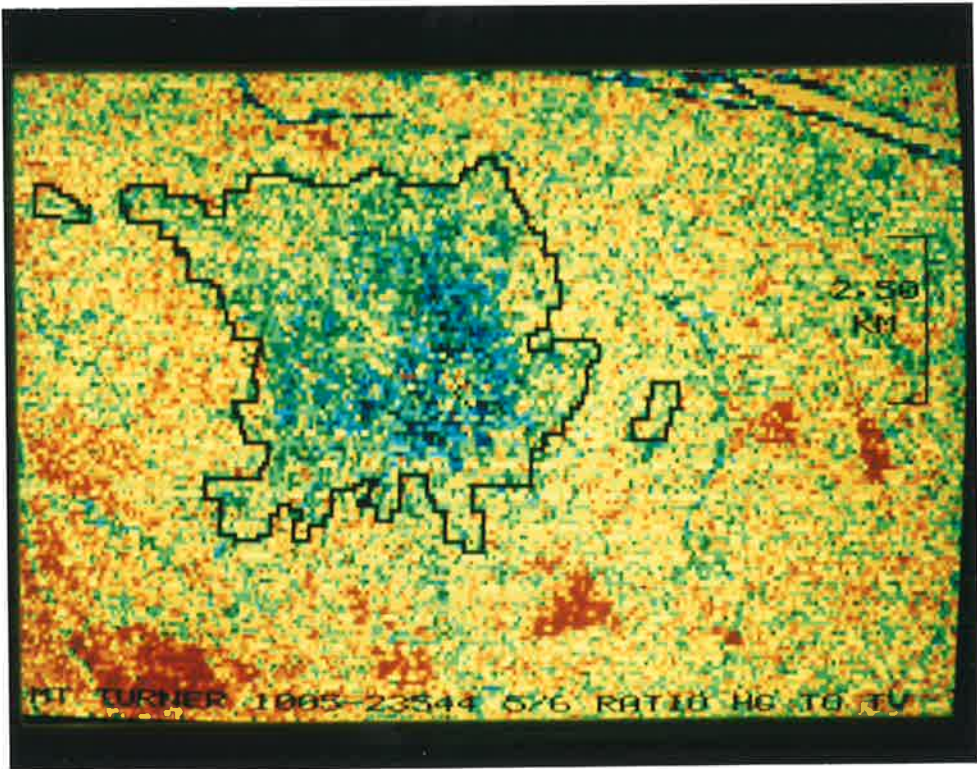
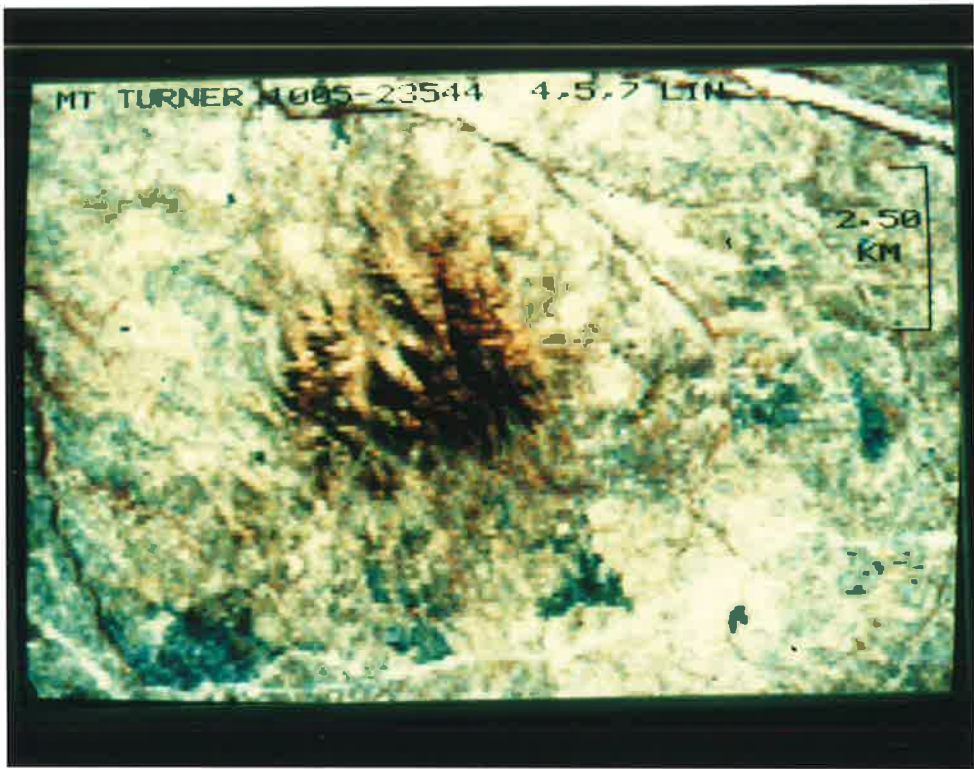
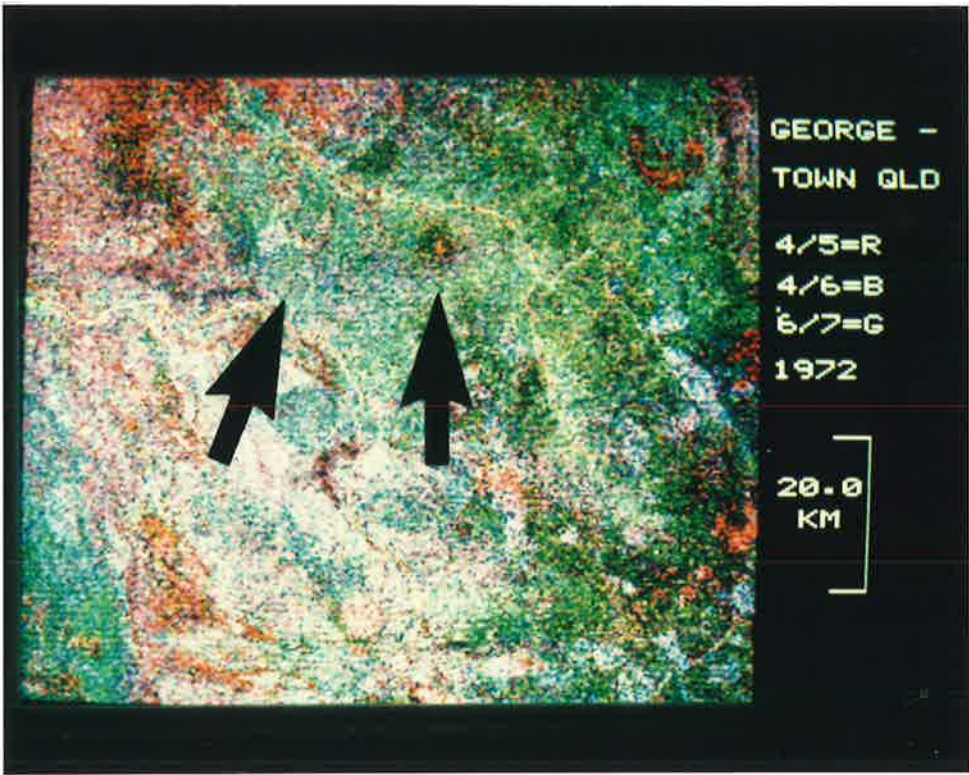


Figure 5.5 Landsat colour ratio composite - Mt Turner area -
histogram normalisation of bands 4/6 (blue), 6/7
(green) and 4/5 (red), (after Huntington, pers. comm.,
1983).

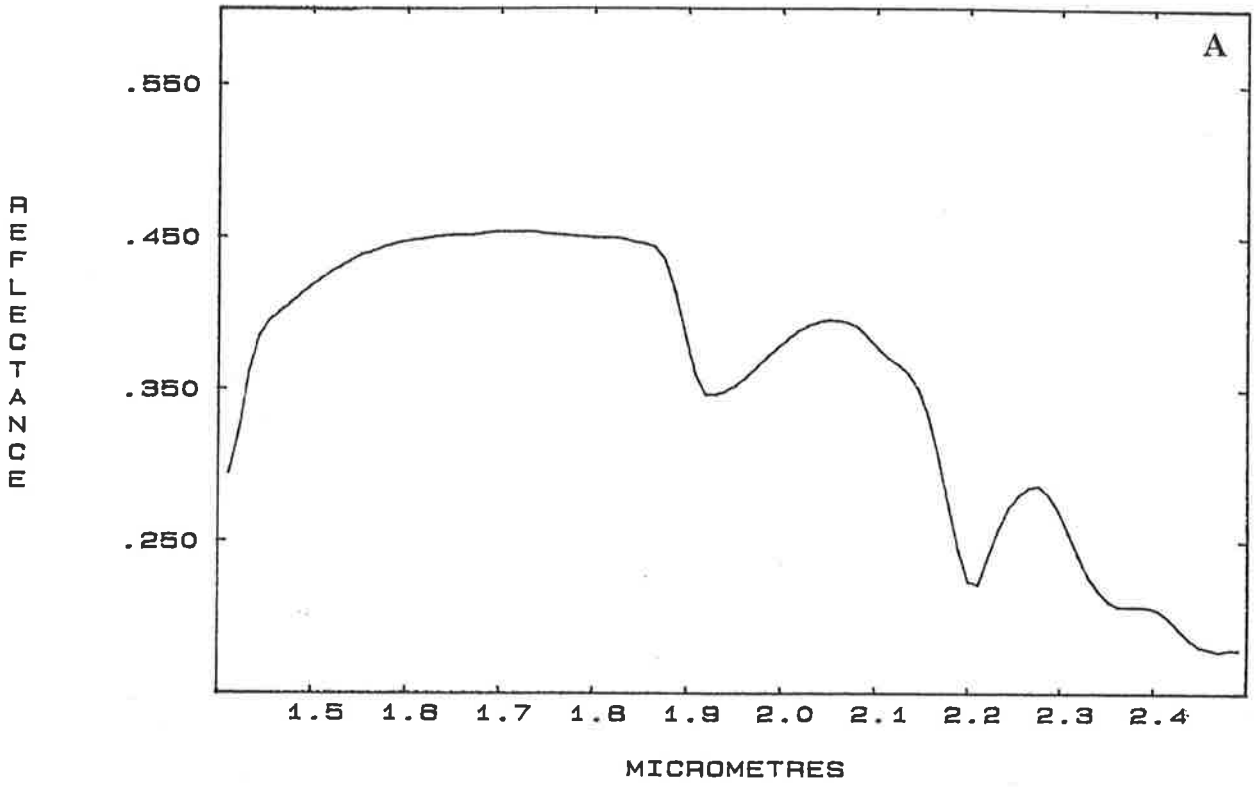
The image depicts limonite in shades of green, and vegetation in shades of red. Mt Turner (arrowed to the east), is differentiated from its immediate surrounds, but not from other iron-rich rocks (for instance dolerites in the Robertson River Formation to the NW). The Phyllis Mae area (arrowed to the west) shows no anomalous features.



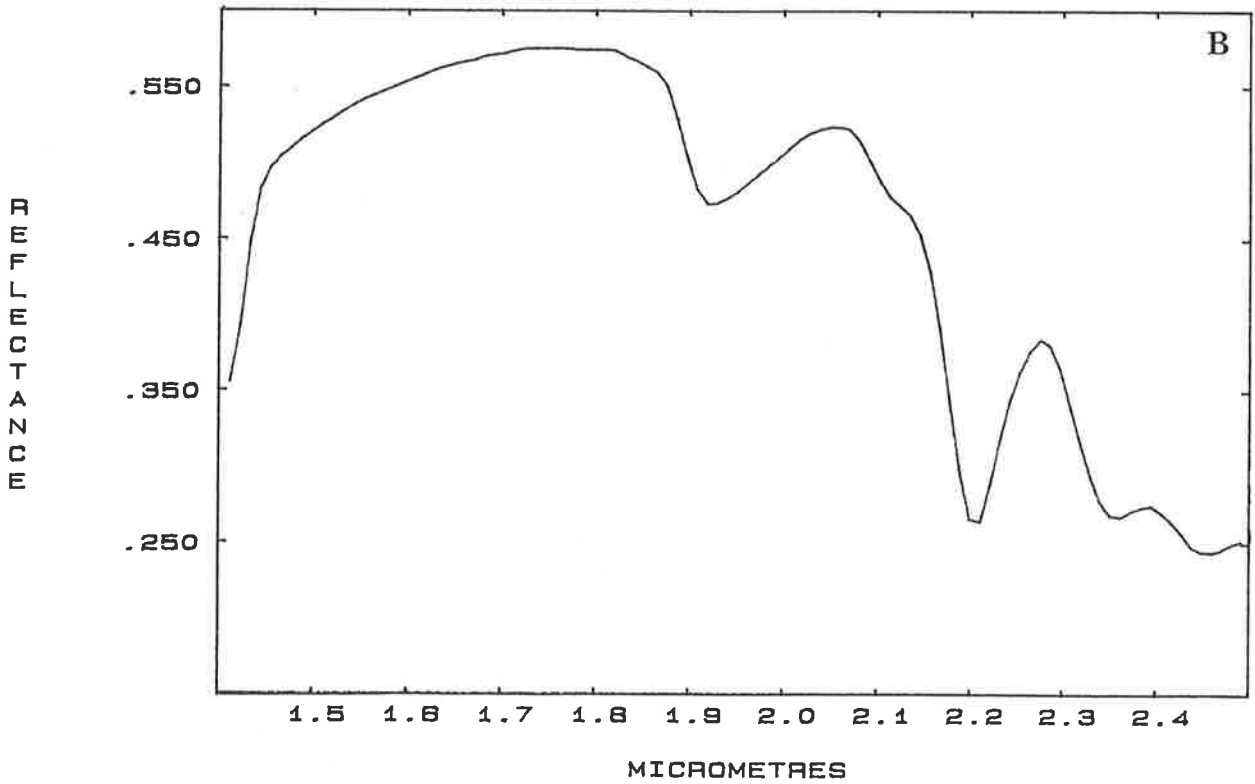
Figures 5.6a&b SWIR laboratory reflectance spectra of weathered and fresh surfaces of the same altered sample.

Figure 5.6a is the spectrum of the weathered surface, while Fig. 5.6b is the fresh surface spectrum. The sample contains quartz, sericite and minor kaolin. Note the similarity of the spectral features in both plots.

23-8-83 MT4/174/1 W



23-8-83 MT4/174/1 F

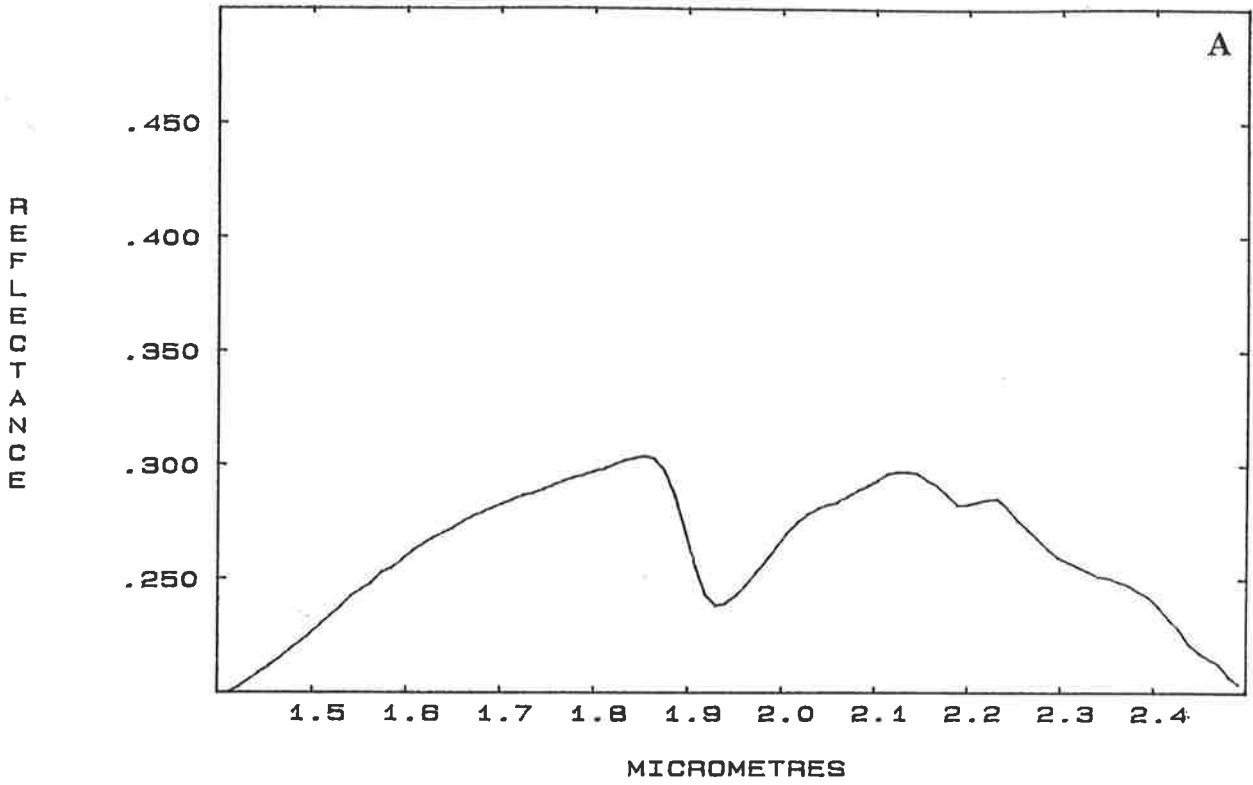


Figures 5.7a&b SWIR laboratory reflectance spectra of a weathered, lichen-covered surface and a fresh surface of the same altered sample.

Figure 5.7a depicts the spectrum of the weathered surface and shows the effects of an even cover of black lichen.

Figure 5.7b depicts the fresh surface spectrum, with strong $2.2\mu\text{m}$ absorption. This sample is an altered rhyolite composed of quartz and sericite.

15-8-83 MT2/258/1 W



15-8-83 MT2/258/1 F

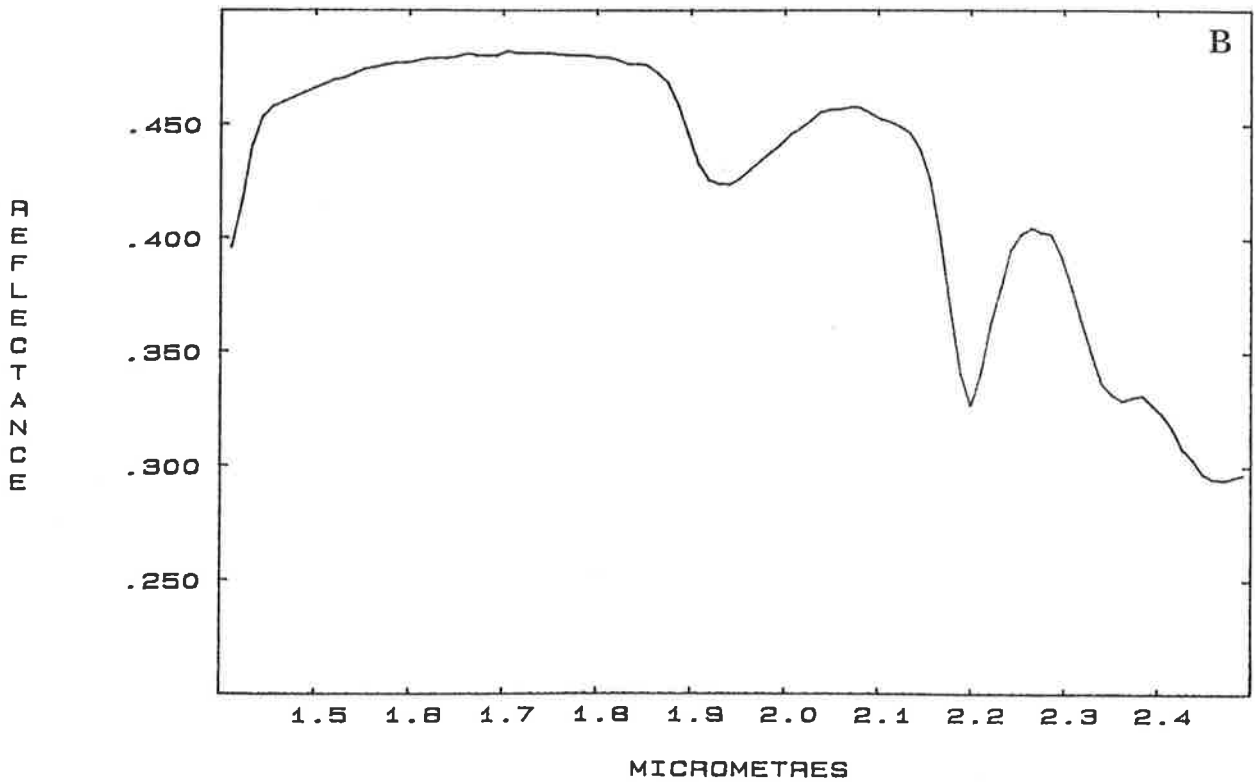
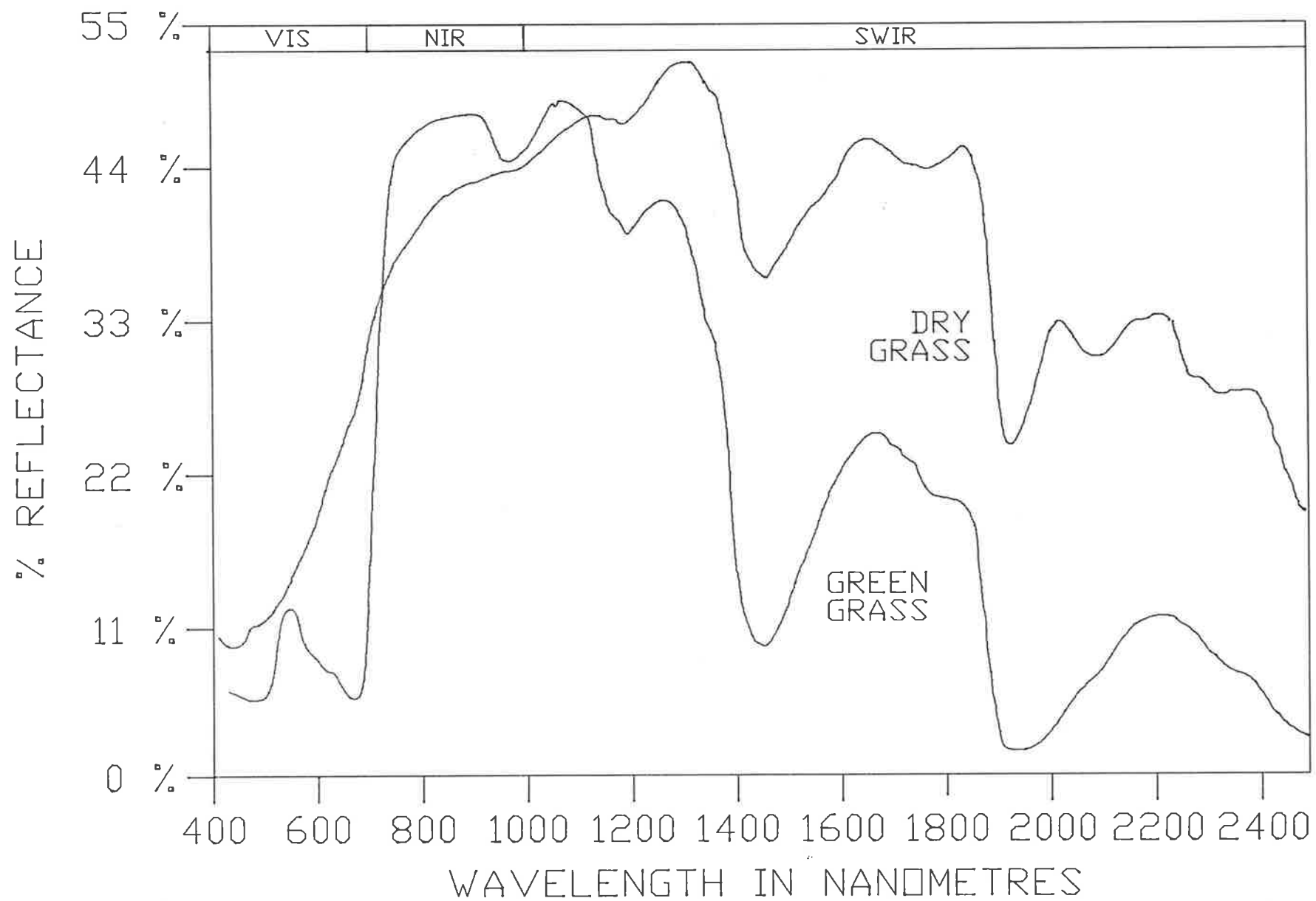


Figure 5.8 SWIR laboratory reflectance spectra of healthy and dry vegetation.

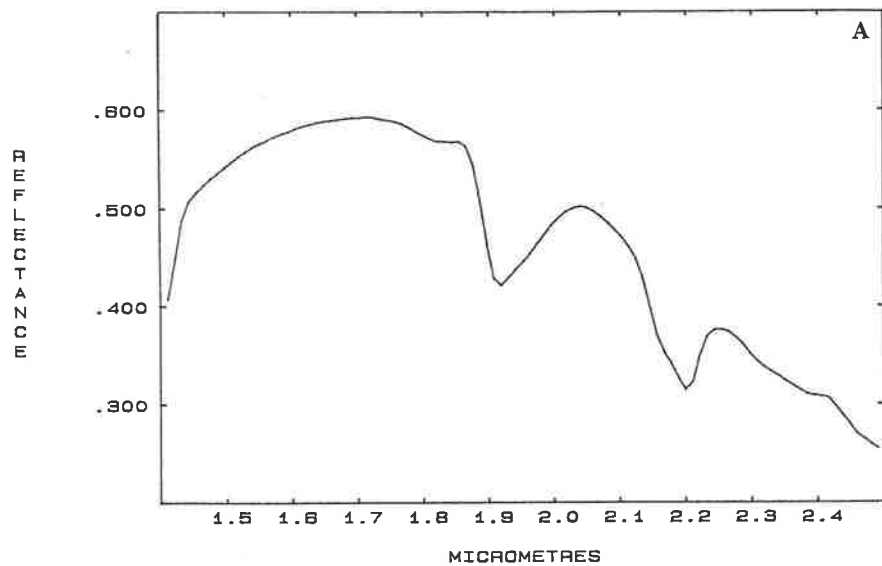
The spectrum of healthy vegetation is affected mainly by the moisture in the foliage. The absorption features at 2.1 and 2.3 μm in the spectrum of dry vegetation are caused by the cellulose present in the plant. These features become apparent only when the water content is low.



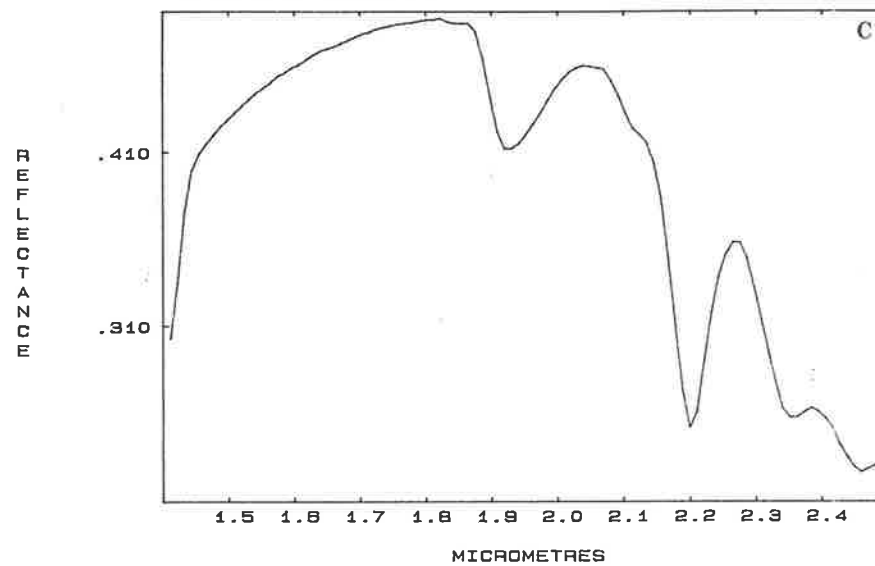
Figures 5.9a-d The four major spectral types from SWIR laboratory reflectance measurements of Mt Turner samples.

Kaolin-type (a) occurs only when there is significant kaolin in the sample. Sericite-type (b) is from altered rocks, and has secondary absorption features near 2.35 and 2.45 μm . Muscovite-type (c) is similar to Sericite-type, but does not have strong secondary absorption features. Mafic-type spectra (d) have their major absorption features in the 2.3 μm region, and can result from a variety of primary mafic or propylitic alteration minerals.

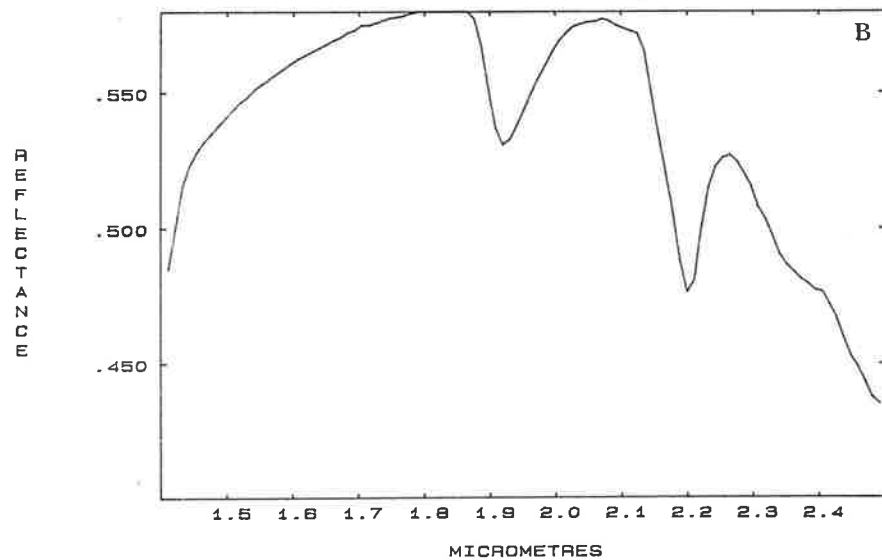
16/8/83 MTPM/281/1 W



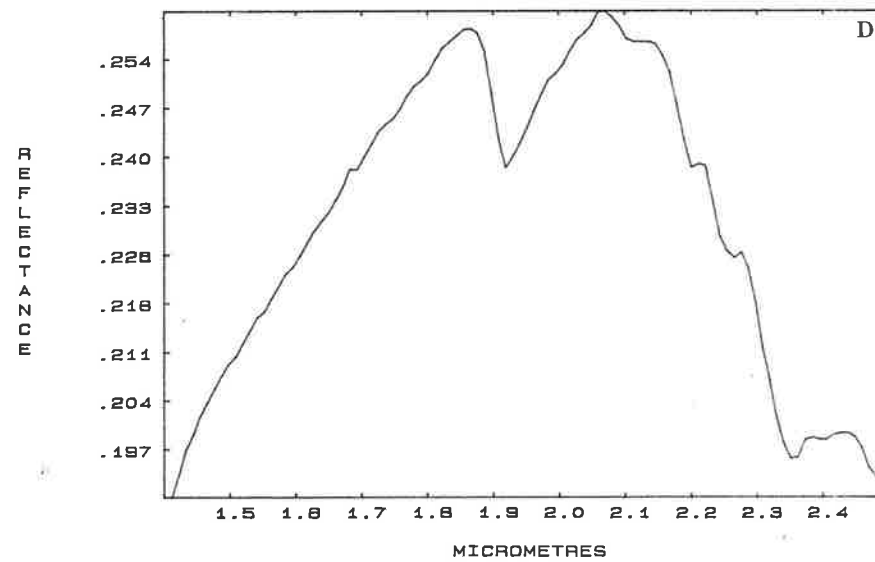
17/8/83 MT1/851/2 W



8-8-83 MT8/243/S



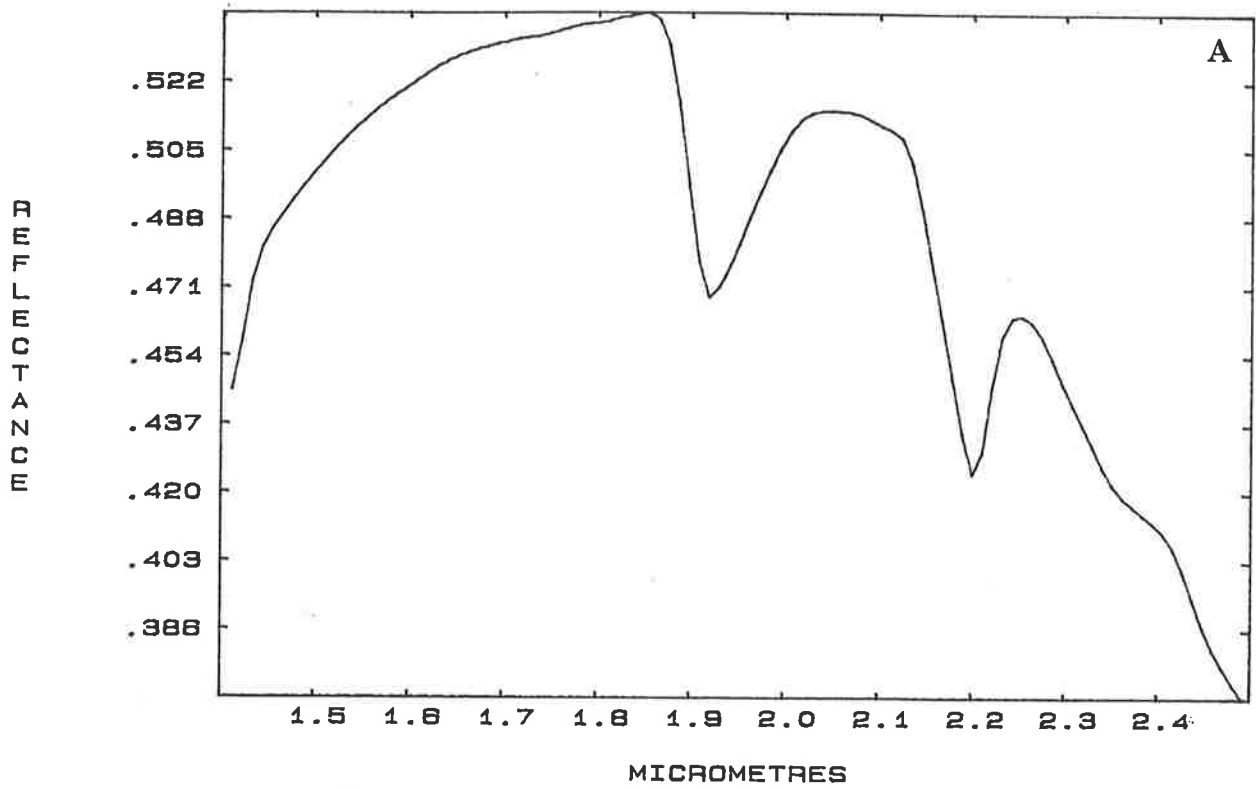
22-8-83 MT4/47/1 W



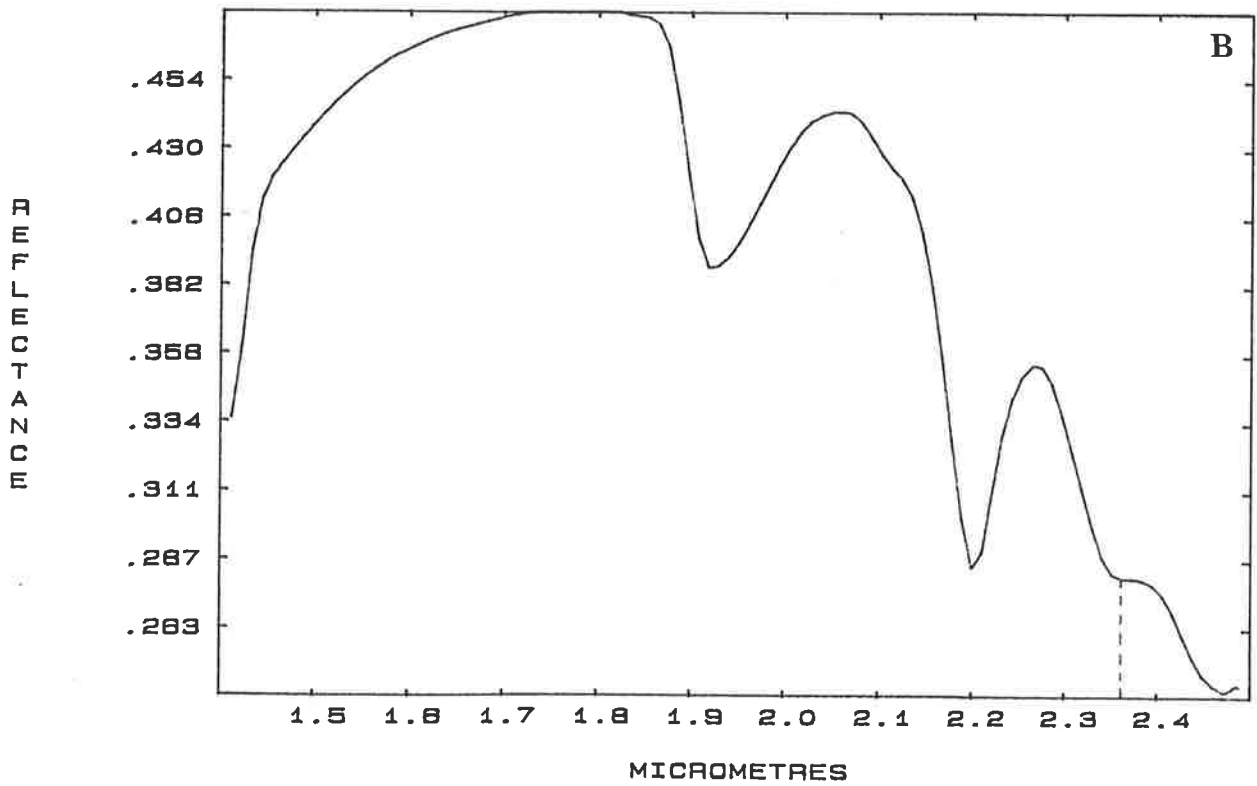
Figures 5.10a&b Averages of 'altered' and 'unaltered' SWIR laboratory reflectance spectra of Mt Turner samples.

The major absorption feature at $2.2\mu\text{m}$ is deeper in the spectrum of the averaged altered samples (10% for altered versus 4% for unaltered). Secondary features near 2.35 and $2.45\mu\text{m}$ are also better developed in the altered spectrum.

AVERAGE UNALTERED



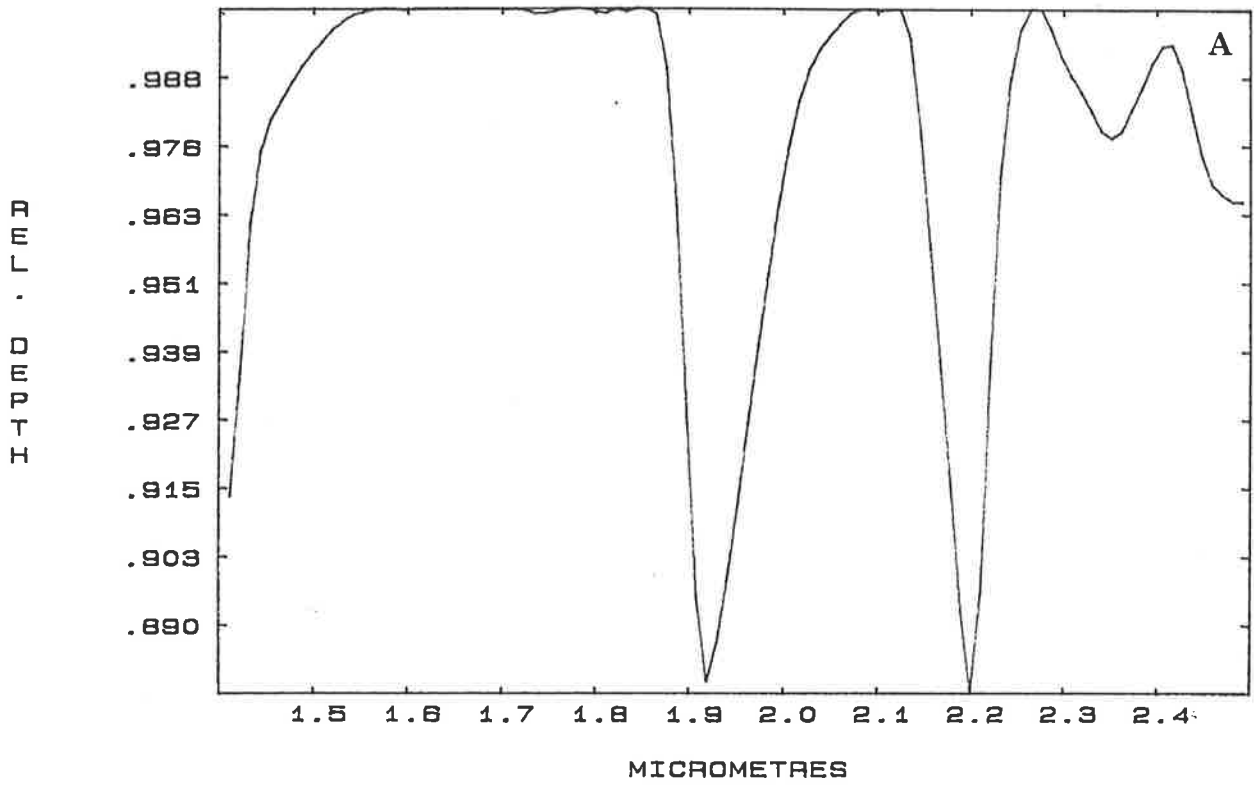
AVE. ALT.



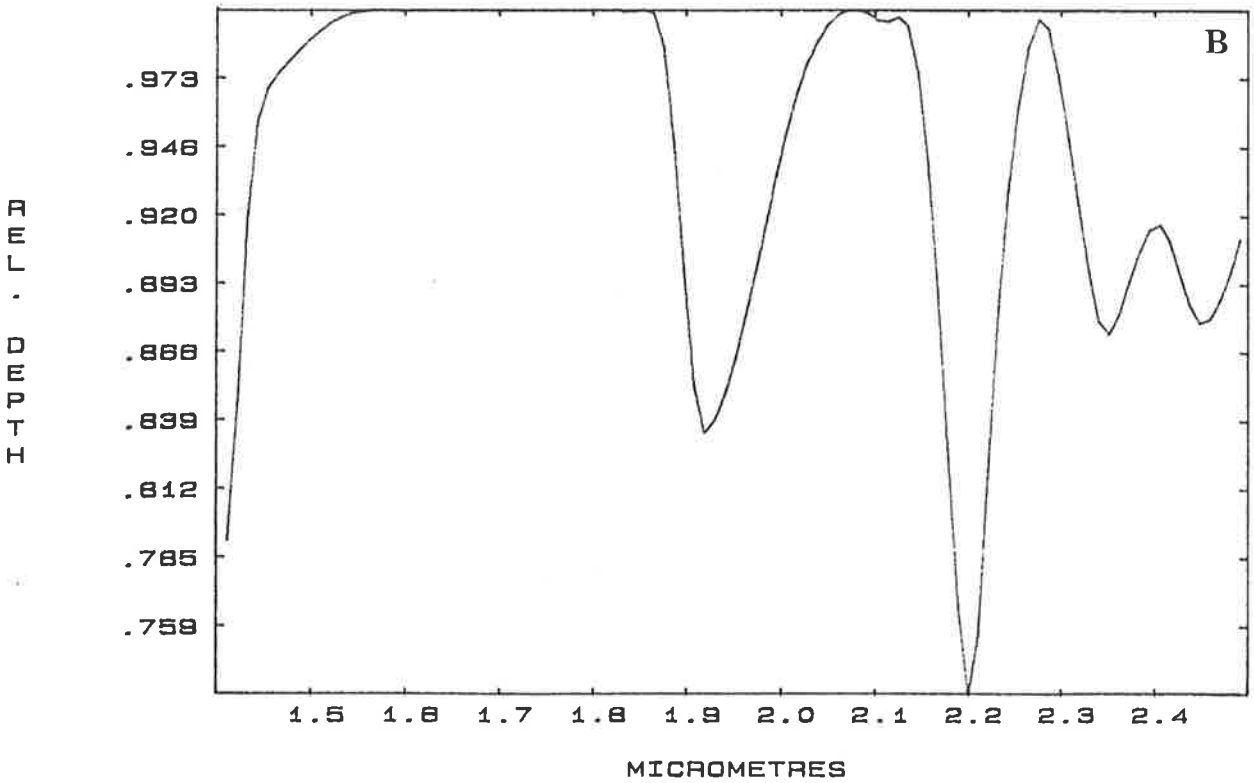
Figures 5.11a&b Hull residuals of average altered and unaltered SWIR laboratory reflectance spectra.

Once the background slope has been removed using the hull residual technique, the presence of features near 2.35 and 2.45 μm in the average unaltered spectrum become apparent (although they are not as well developed as the corresponding features in the average altered spectrum).

HULL RESID. AVE. UNALT.



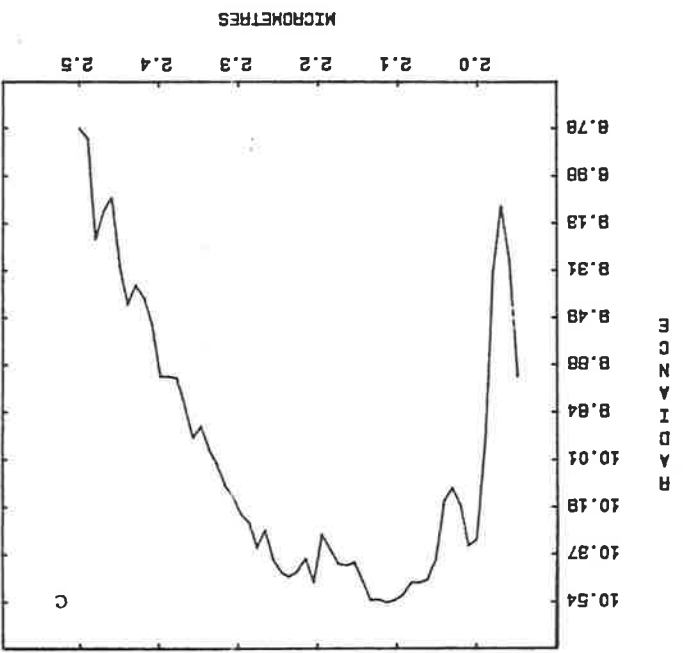
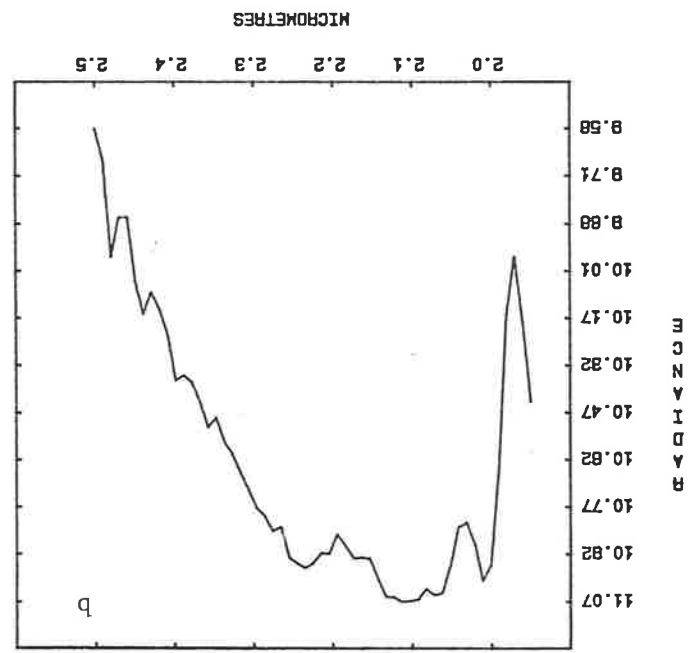
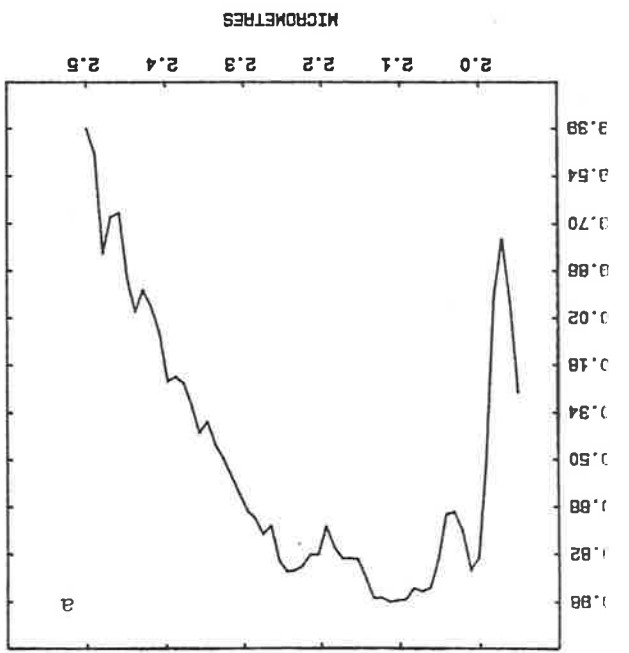
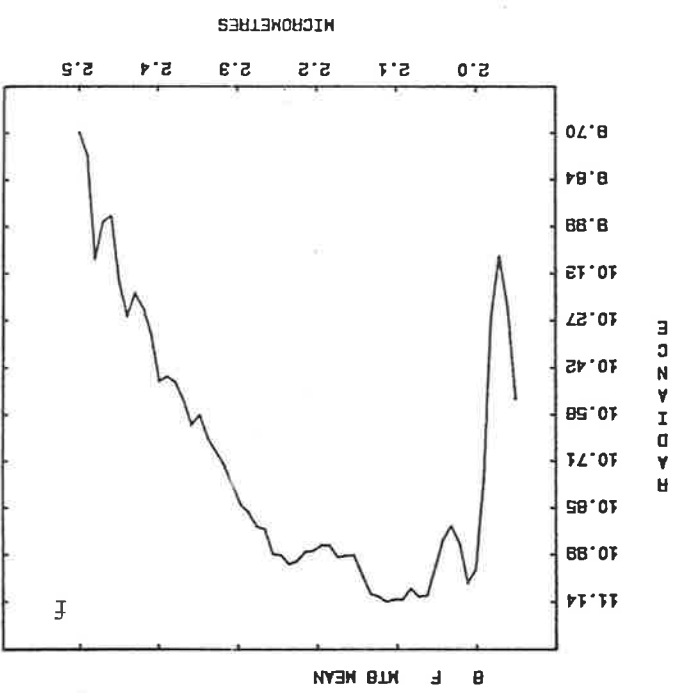
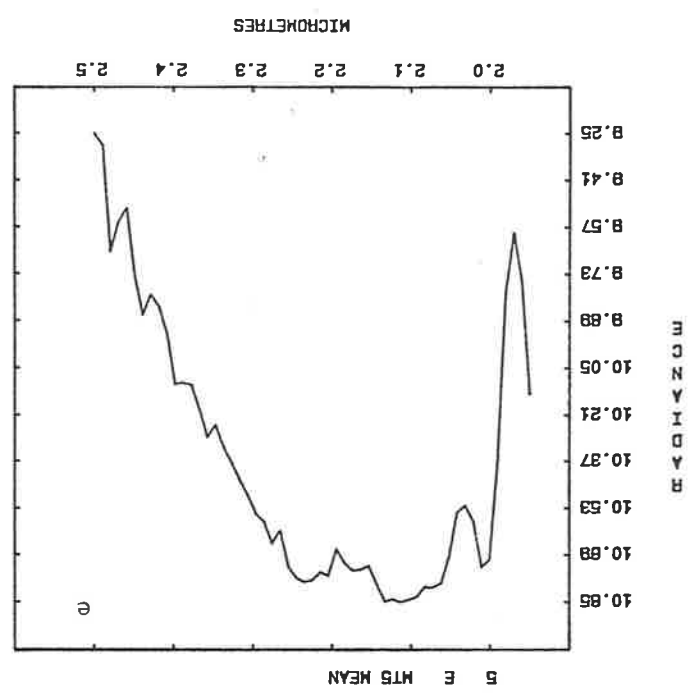
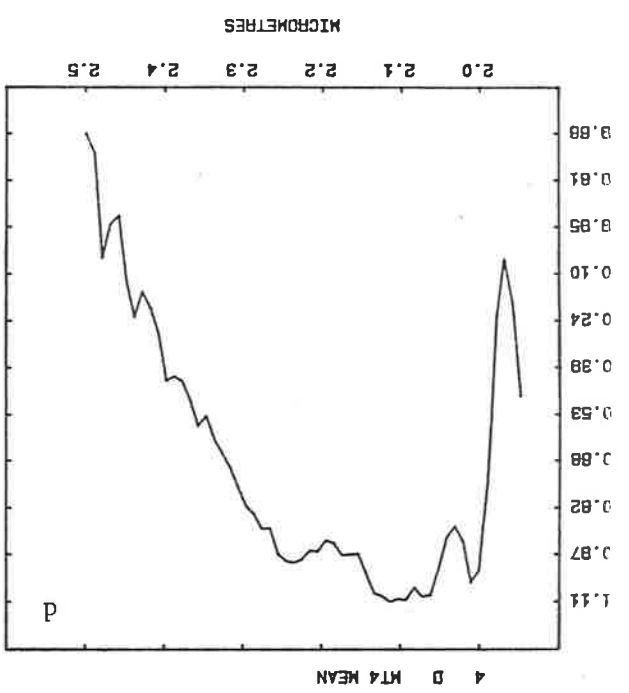
HULL RESID. AVE. ALT.



Figures 5.12a-k Mean radiance curves for each Mt Turner flight line.

Note the presence of the $2.2\mu\text{m}$ feature in every curve (due to atmospheric water absorption), and the varying strength of the shoulder near $2.17\mu\text{m}$. This is a result of the influence of mineralogical (lithospheric) absorption (muscovite/sericite and kaolin).

(continued over)



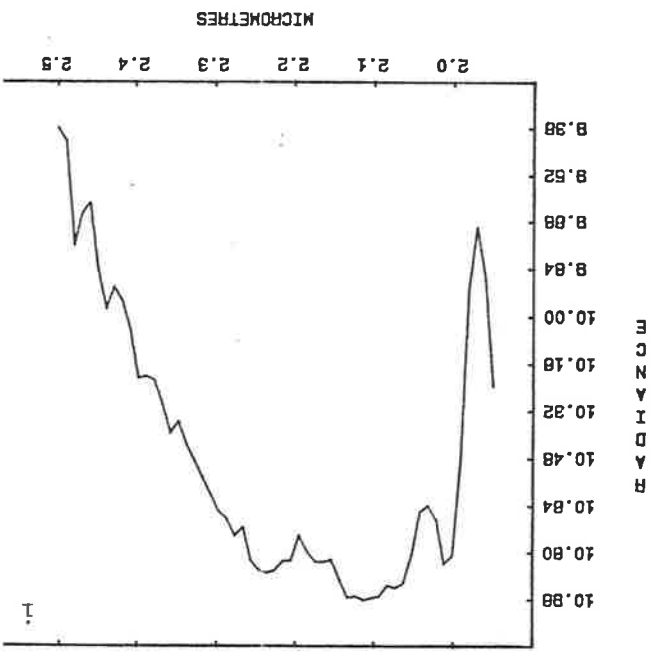
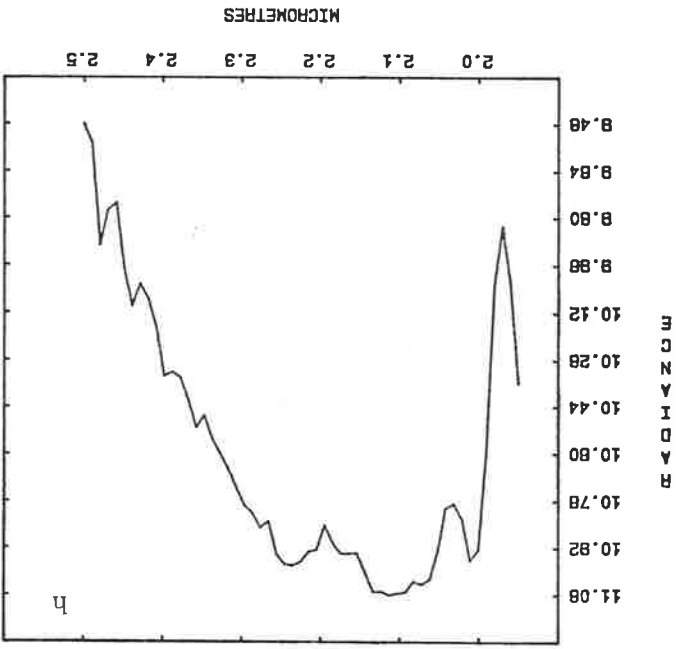
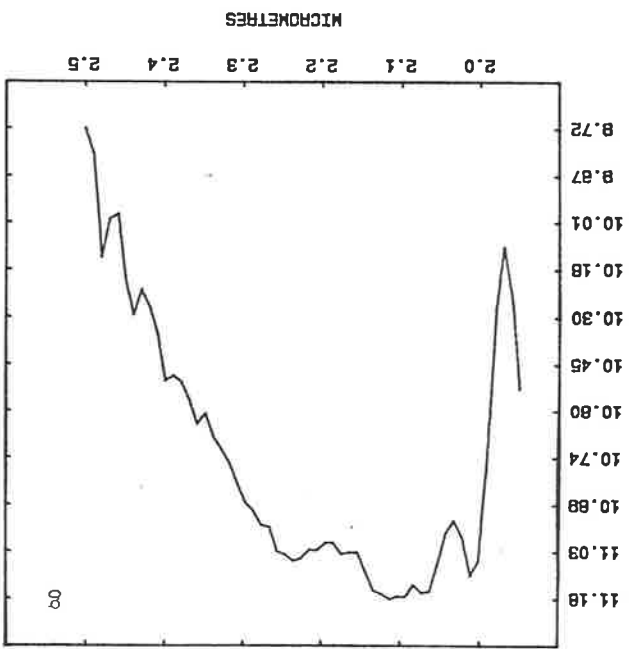
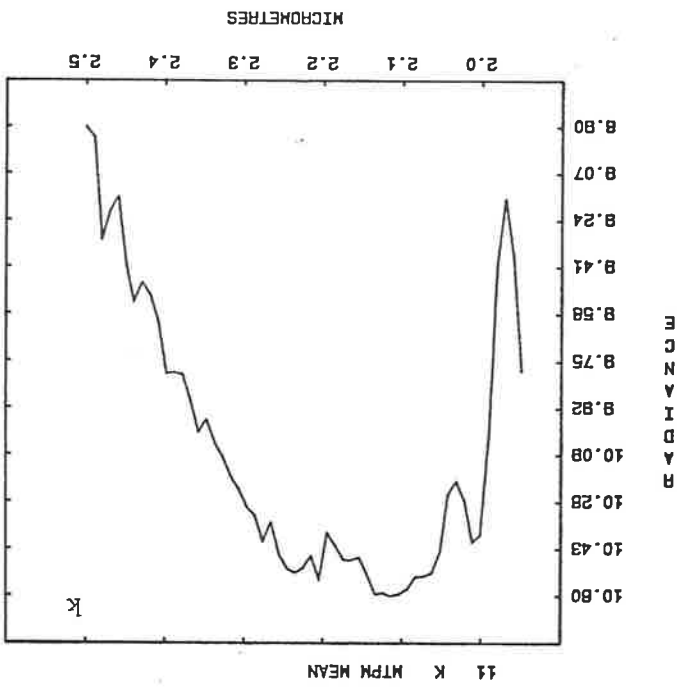
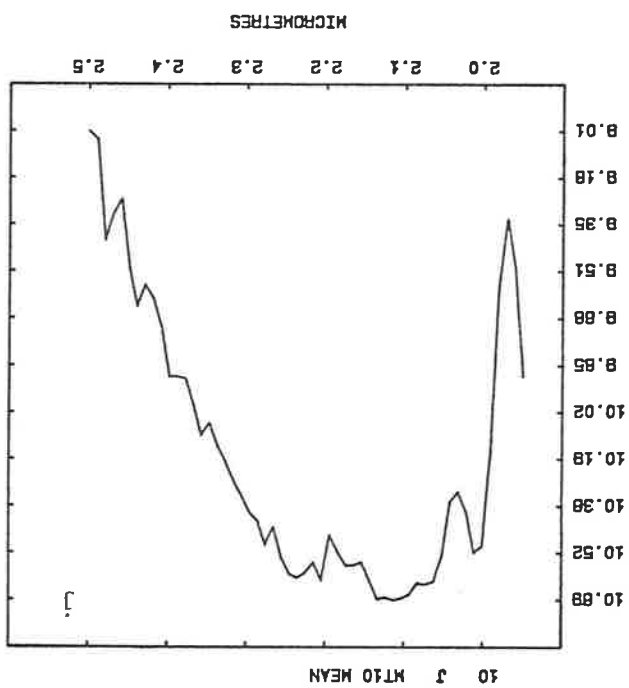
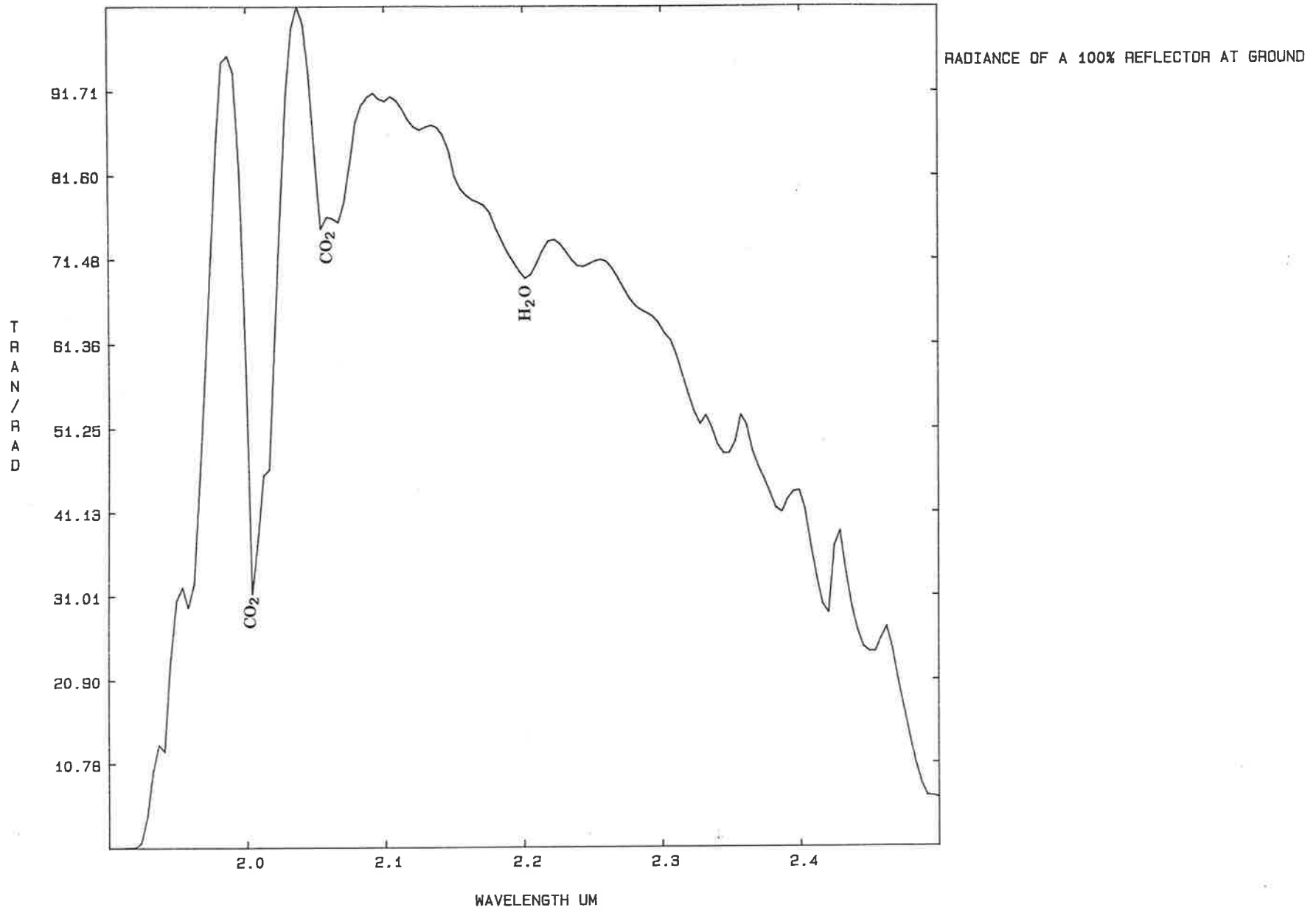


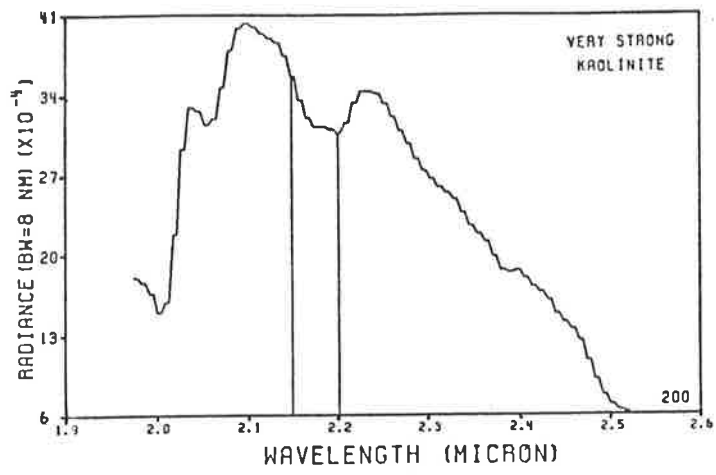
Figure 5.13 Radiance of a 100% reflector at sea level, calculated using the LOWTRAN model.

The major features present in this spectrum are due to the presence of CO₂ and H₂O in the atmosphere.

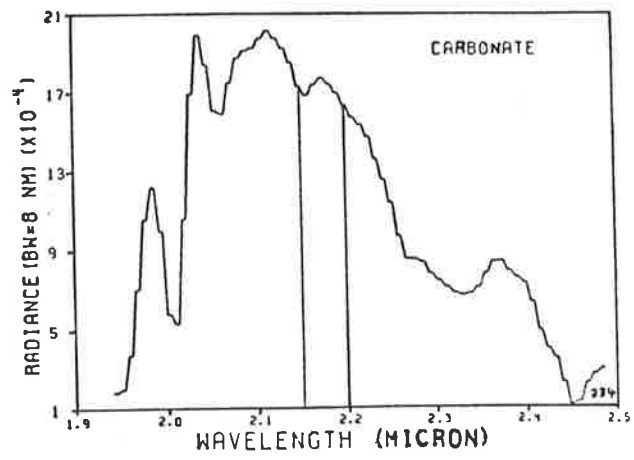


Figures 5.14a&b Radiance curves from U.S. surveys (after Collins
et al., 1981).

Note the depth of absorption features at both 2.2 μ m and 2.3 μ m.
These are much stronger than any mineralogical effects detected at these
wavelengths with the same instrument in Australia.



a



b

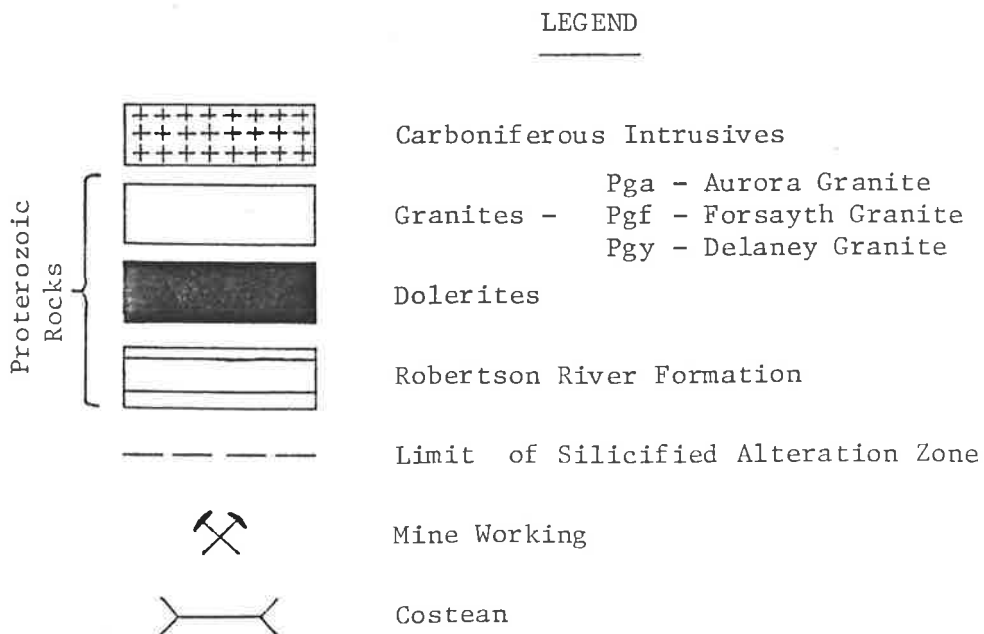
Figure 5.15 Log residuals of all Mt Turner flight lines in image format, with overlay showing major spectral types.

For each flight line there are six columns of data. From left to right, they are VNIR (0.4 μm to 1.0 μm) log residuals, VNIR Albedo, SWIR log residuals (1.95 μm to 2.5 μm), SWIR Albedo, a vegetation index derived from VNIR data, and coded geology. VNIR data are labelled Reticon Log Residuals and SWIR data are labelled Pbs Log Residuals.

The overlay shows typical examples of Types A, B and C spectra for line MT9 (blocks labelled B1 and B2 show varying strengths of the same Type spectrum). Note that MT9 was flown from south to north. The log residual image data are arranged in order of increasing pixel number, from top to bottom, so the top of the MT9 image data is south, bottom is north. For flight directions of other lines, see overlay B to Fig. 5.2.

Figure 5.16 Classification of spectrometer data based on visual examination of log residuals in image format, overlain on the geology of Mt Turner.

The nature of the surface (e.g. bare soil, rock, or vegetation), rather than variations in the mineralogy, appears to be the major factor determining the category into which each spectral 'block' falls.



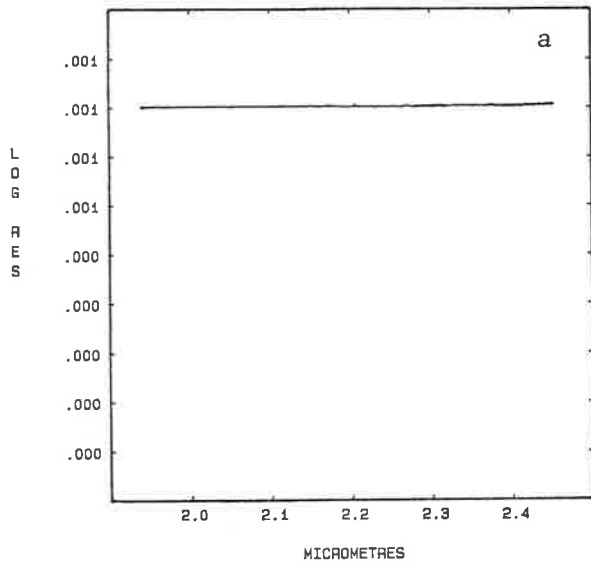
MOUNT TURNER GEOLOGY



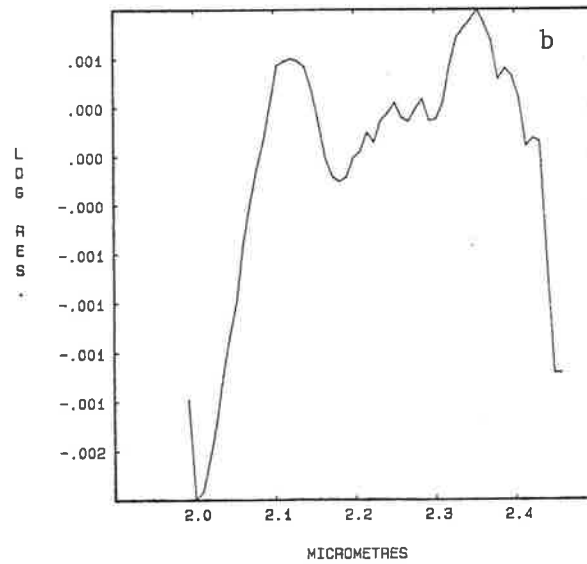
Figures 5.17a-f Spectral 'standards' derived from plots of log residual spectra.

Standard 1 is included as an offset used in the classification procedure. All other spectra have been produced by averaging 'blocks' from flight line MT4, and discarding very close matches.

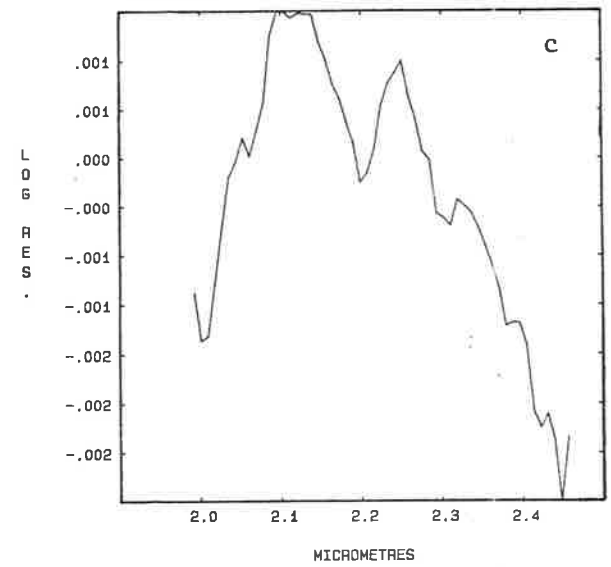
STANDARD 1



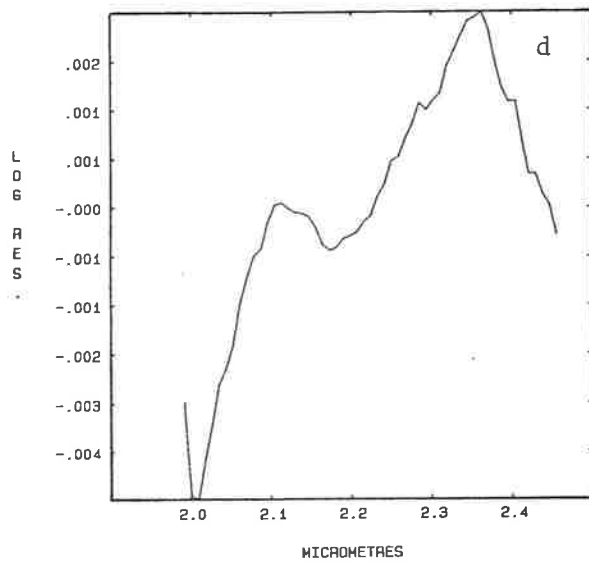
STANDARD 2



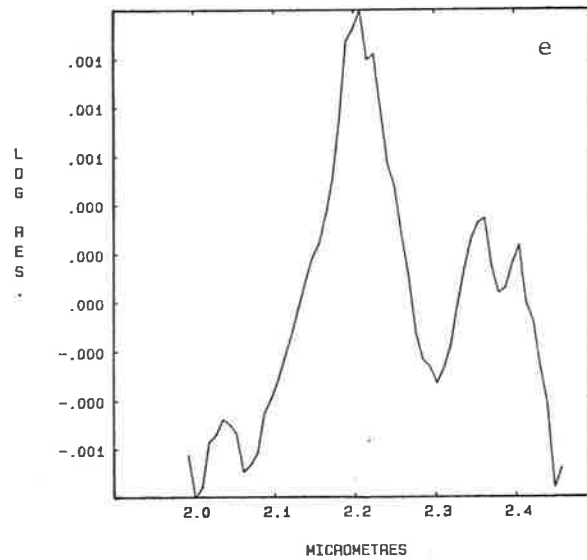
STANDARD 3



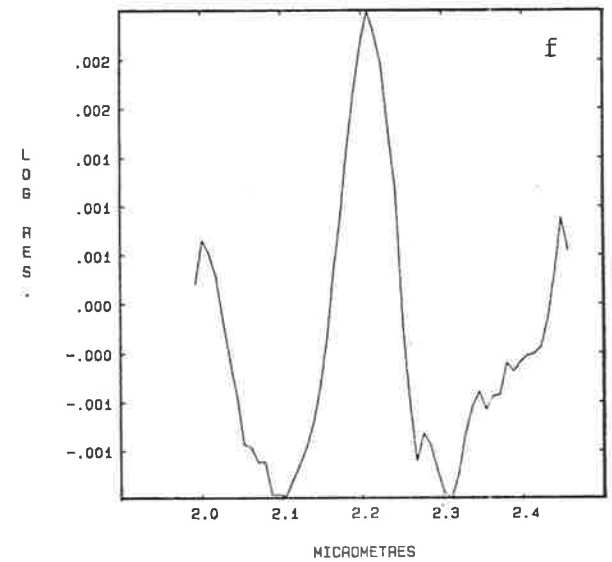
STANDARD 4



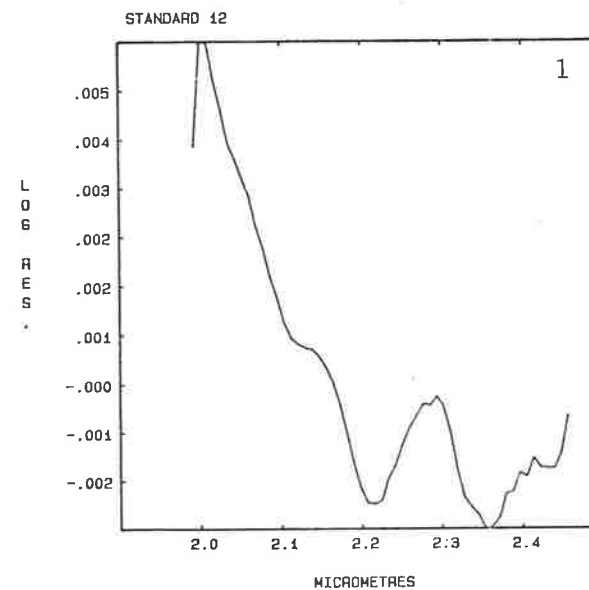
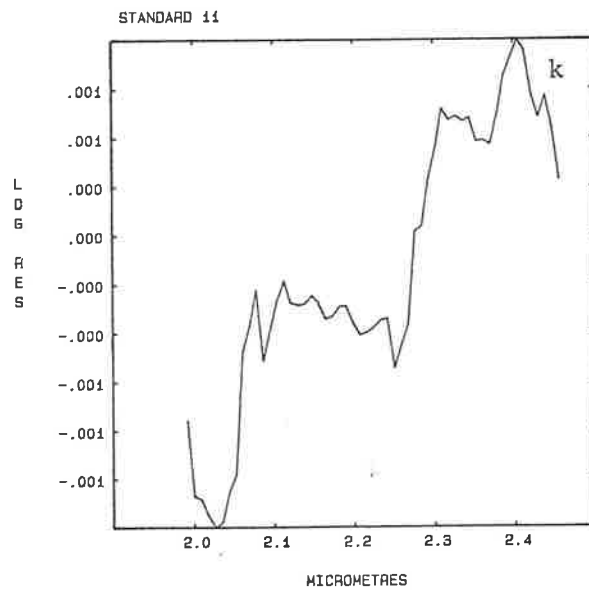
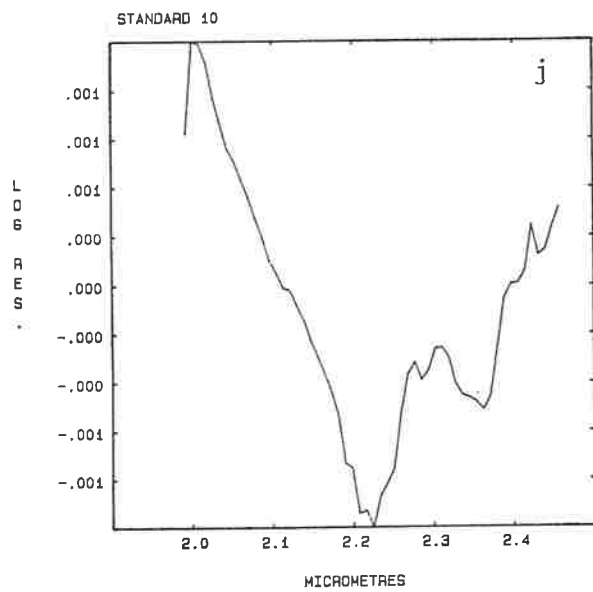
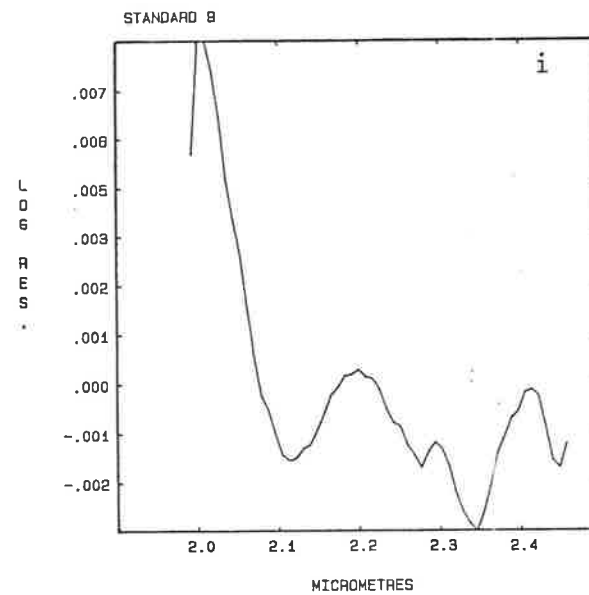
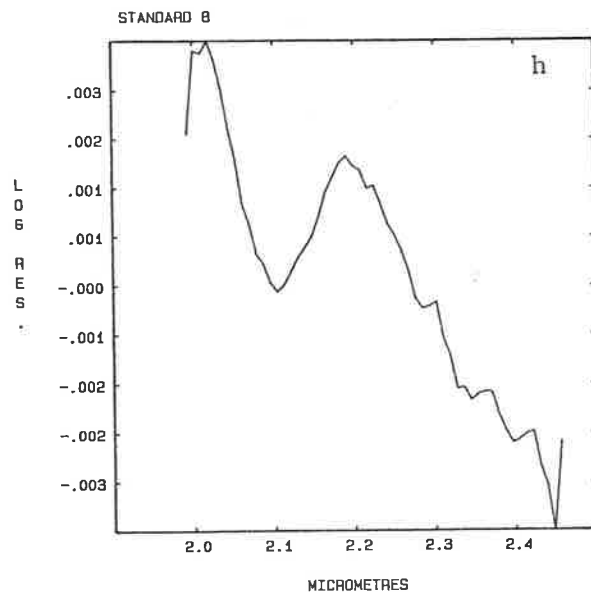
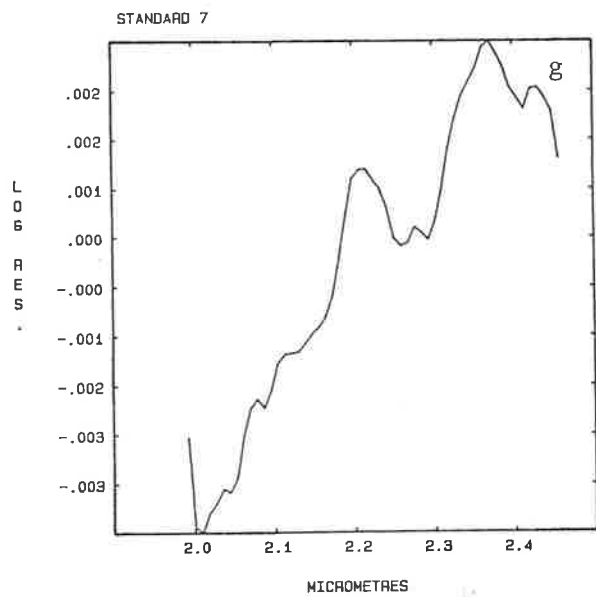
STANDARD 5



STANDARD 6



Figures 5.17g-1 Spectral 'standards' derived from plots of log residual spectra.



Figures 5.17m&n Spectral 'standards' derived from plots of log residual spectra.

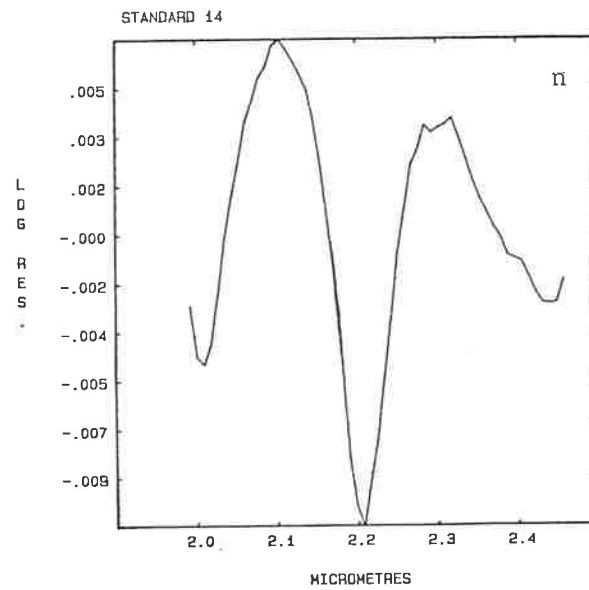
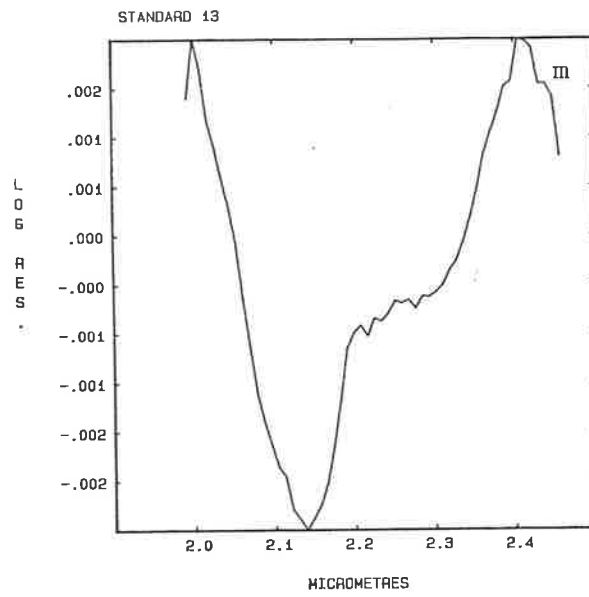
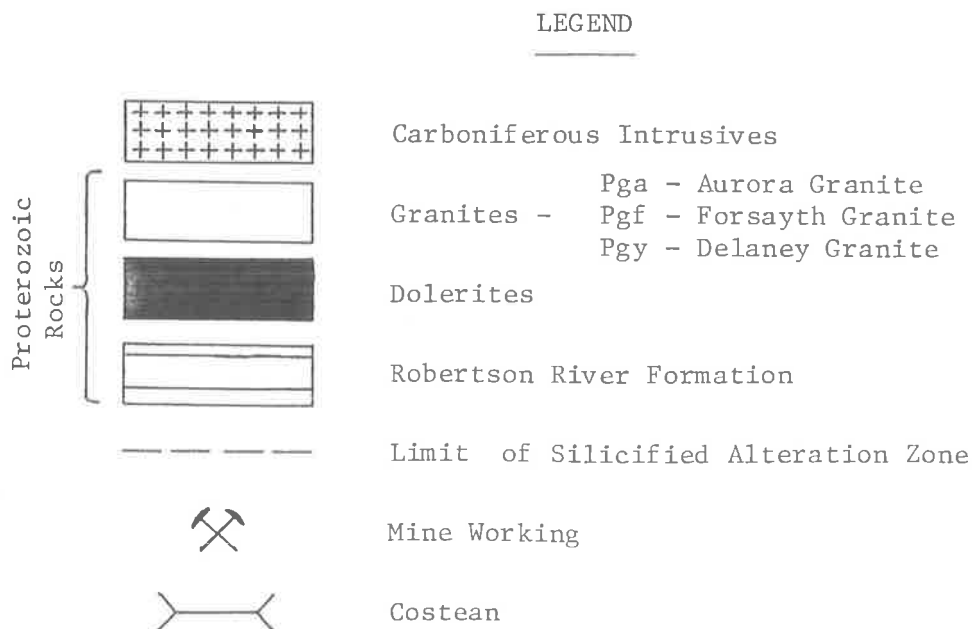


Figure 5.18 Output of portion of the least-squares classification
of flight line MT8.

Solid vertical lines coincide with boundaries easily visible on
Fig. 5.15. Dashed vertical lines represent boundaries which have been
defined using the classification output itself.

Figure 5.19 Least-squares classification of log residual spectra, overlain on the geology map of Mt Turner.

Note that there are differences between this classification and Fig. 5.16. Mineralogical variations appear to have more of an influence on the result in this case. Although the technique is not perfect, and has resulted in a number of misclassifications, areas of sericitic alteration have generally been discriminated successfully.



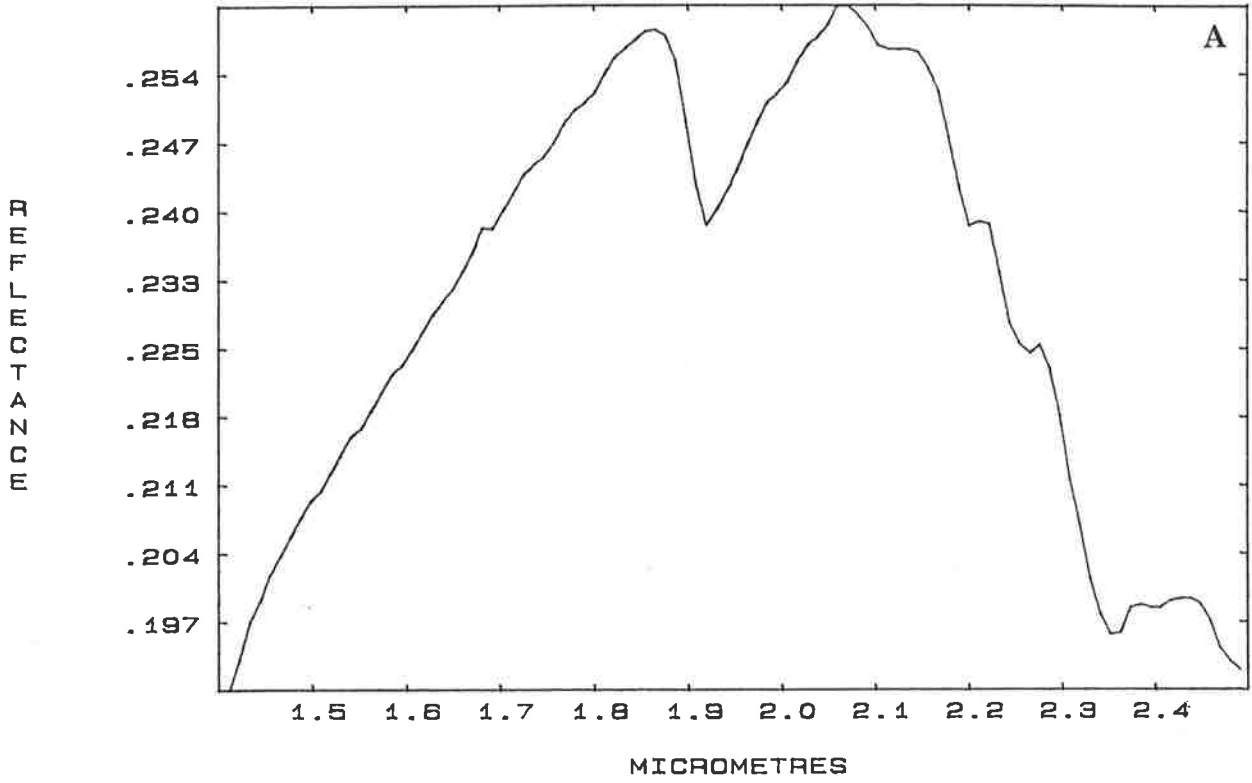
MOUNT TURNER GEOLOGY



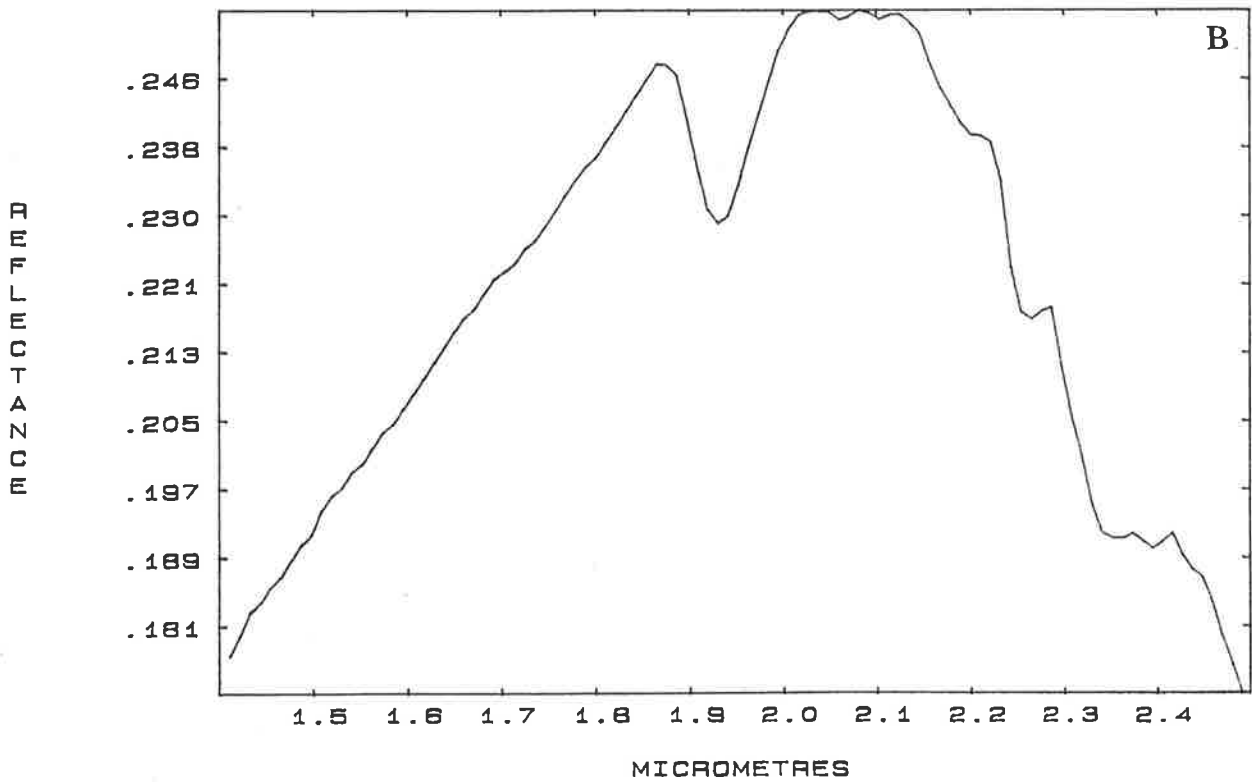
Figures 5.20a&b SWIR laboratory reflectance spectra of propylitically altered samples collected south of Mt Turner.

Both spectra shown have weak absorption features in the 2.3 μ m region. However, these features are not developed strongly enough to allow identification of the mineralogy. This is probably a contributing factor in the failure of the classification to discriminate propylitic alteration from dry vegetation.

22-8-83 MT4/47/1 W



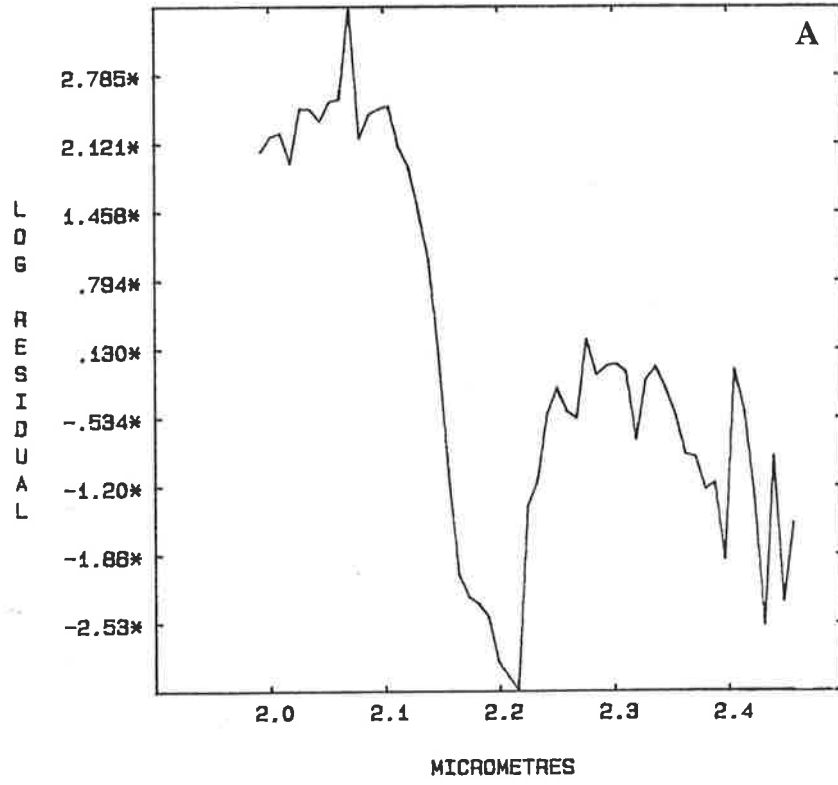
23-8-83 MT4/58/1 W



Figures 5.21a&b Airborne SWIR log residual and laboratory reflectance spectra of kaolin.

The kaolin spectra depicted here show much better definition of absorption features than those from other areas covered at Mt Turner.

M26, 28



24-8-83 K-N

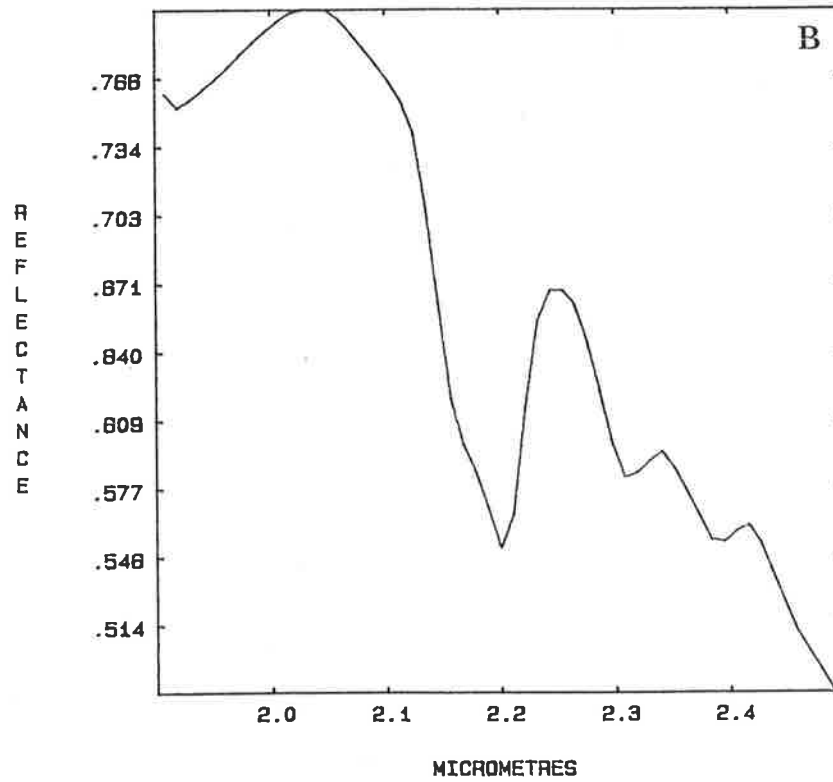
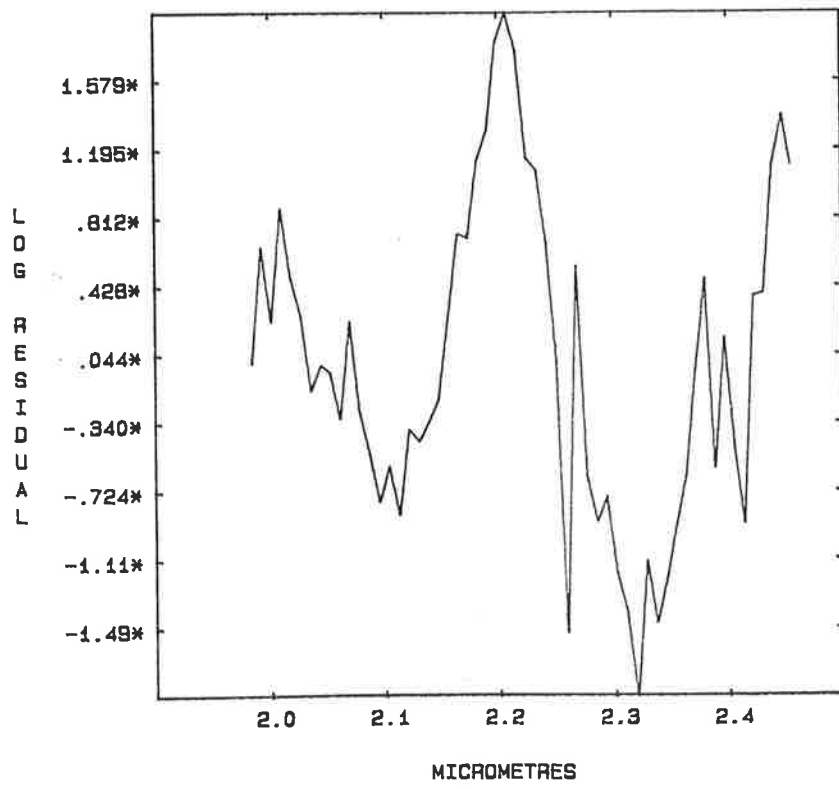


Figure 5.22 Airborne SWIR log residual spectrum of
chlorite/amphibole? mixture from line MT5.

The airborne spectrum was collected over an outcrop of amphibolite on flight line MT5. The sharp feature at $2.26\mu\text{m}$ is indicative of chlorite, while the noisy minimum near $2.4\mu\text{m}$ may indicate the presence of amphibole. The absorption feature near $2.3\mu\text{m}$ is likely to be the result of the combined influence of both minerals.

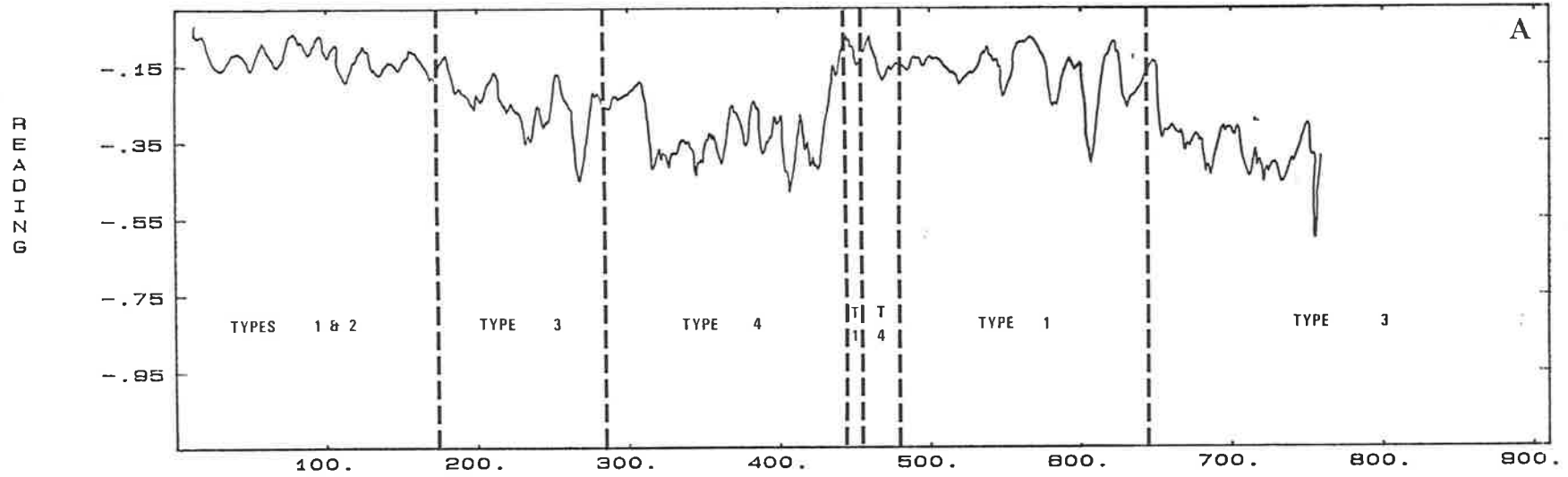
M479, 481



Figures 5.23a&b Tchebychev coefficient ratio and simulated narrow-band ratio plots for line MT9.

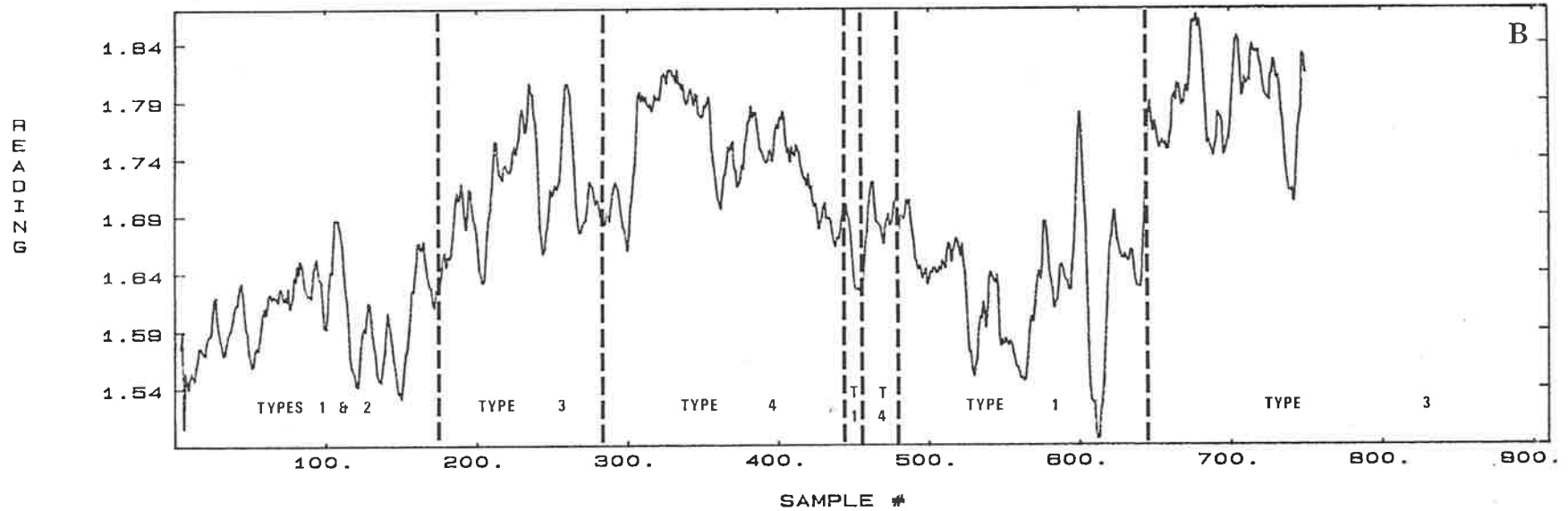
The dashed vertical lines are boundaries that have been determined from the classification scheme (Fig. 5.19). Boundaries of the major sericitic alteration zone are shown by the arrows.

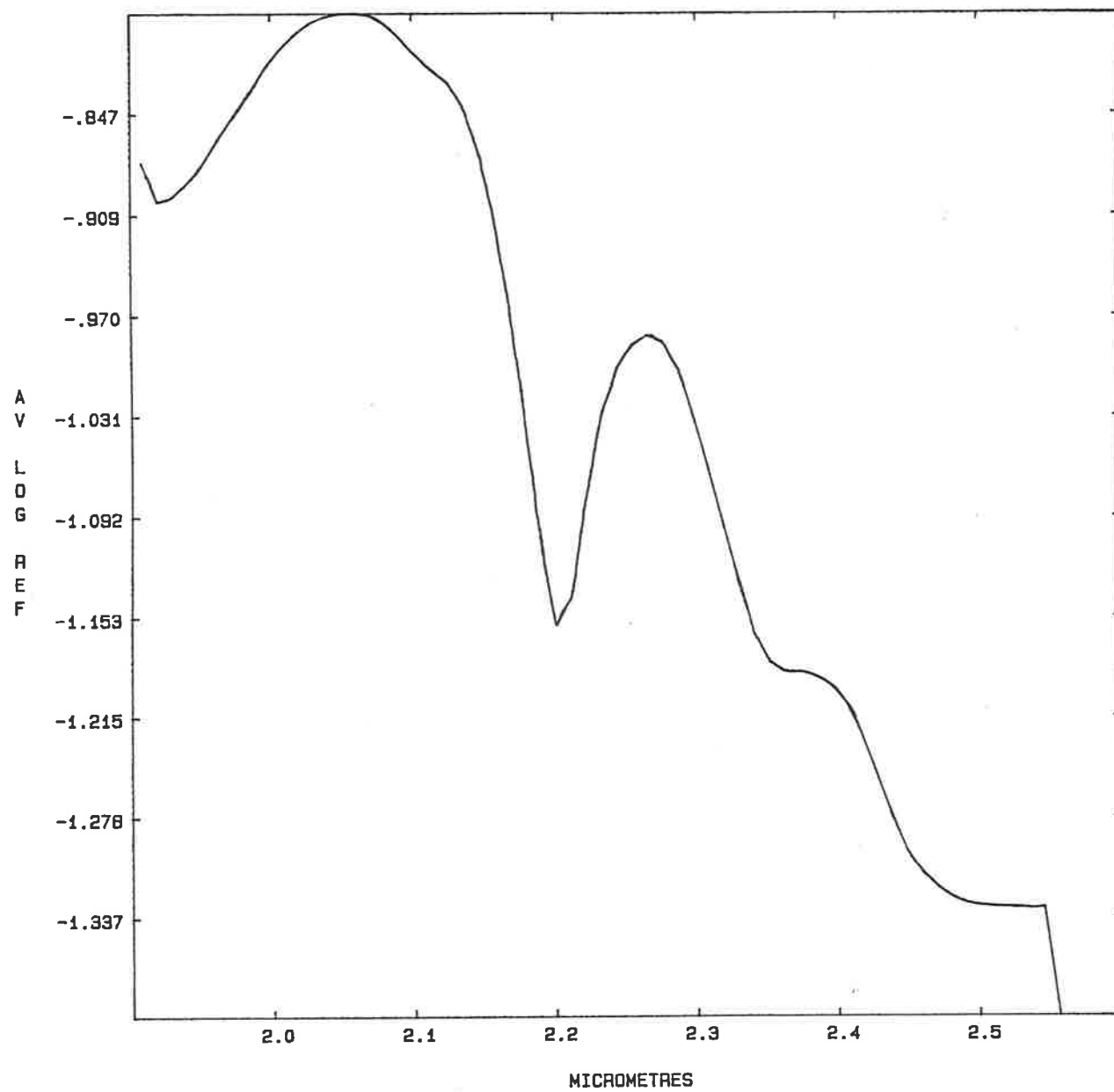
MT9 TCHEBYCHEV COEFFICIENT RATIO C4/C5 SM. BY 8



SAMPLE
ALTERATION
ZONE

BAND RATIO 2.158-2.237UM/2.314-2.374UM SM. BY 8

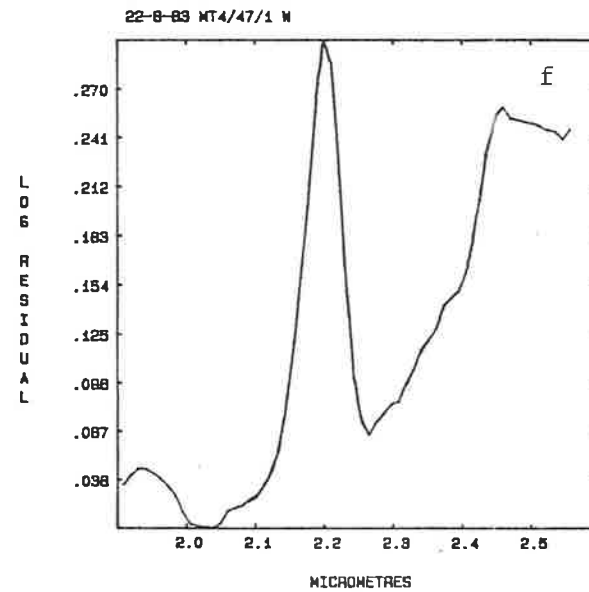
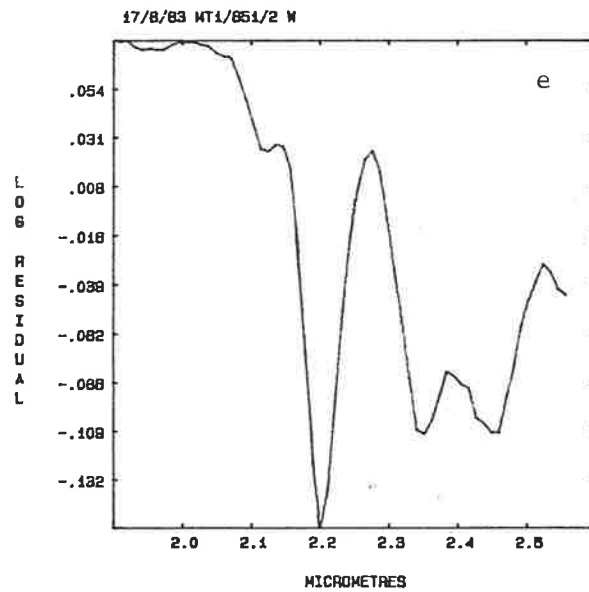
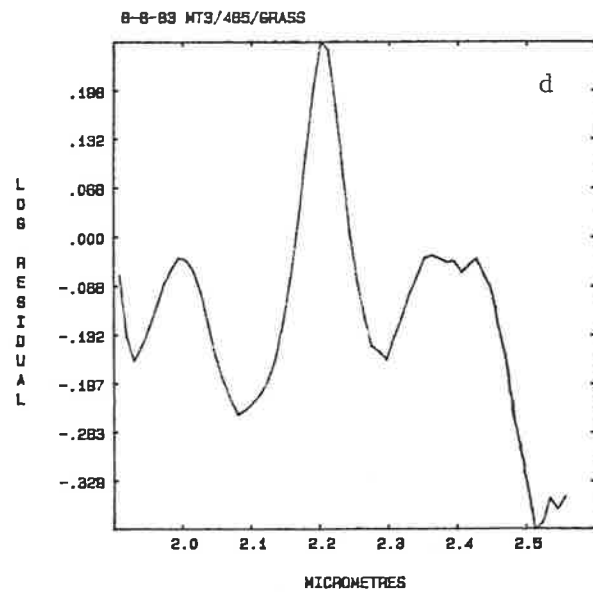
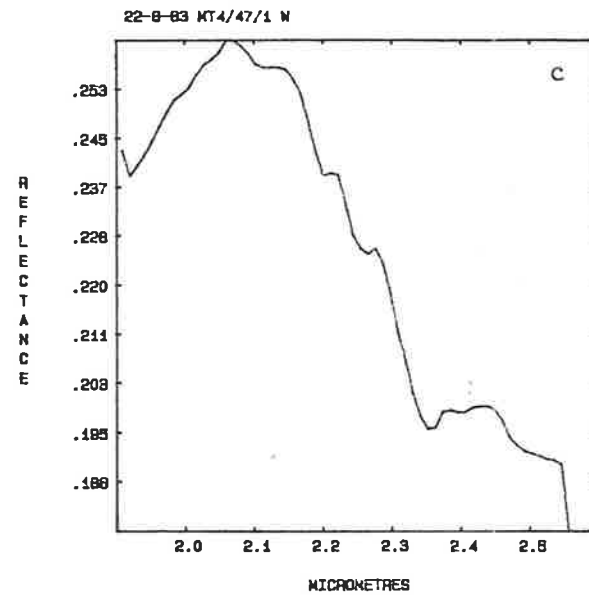
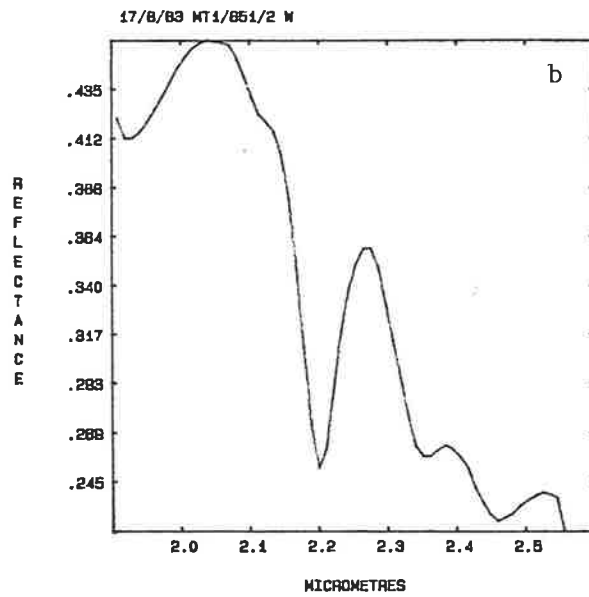
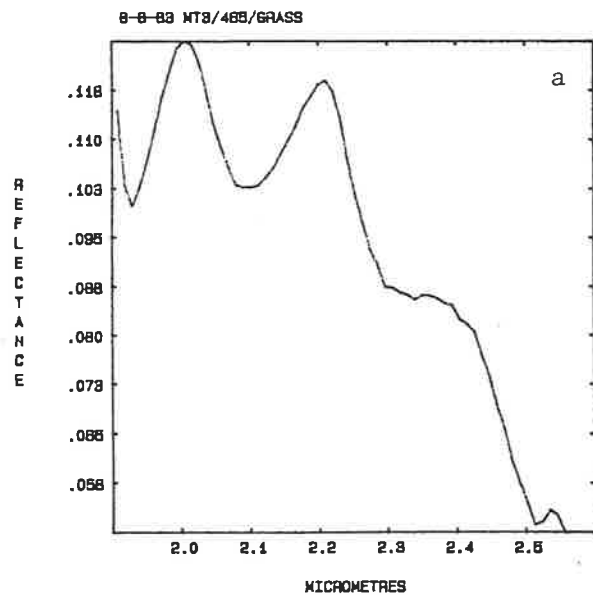




MT. TURNER LAB. SPECTRA MEAN

Figures 5.25a-f Selected SWIR laboratory reflectance spectra and their 'log residuals'.

These plots illustrate the difficulty of separating the effects of dry vegetation from propylitic alteration. Similarly-shaped 'log residuals' (Figs. 5.25d&f) have resulted from the spectra of these two very different materials.



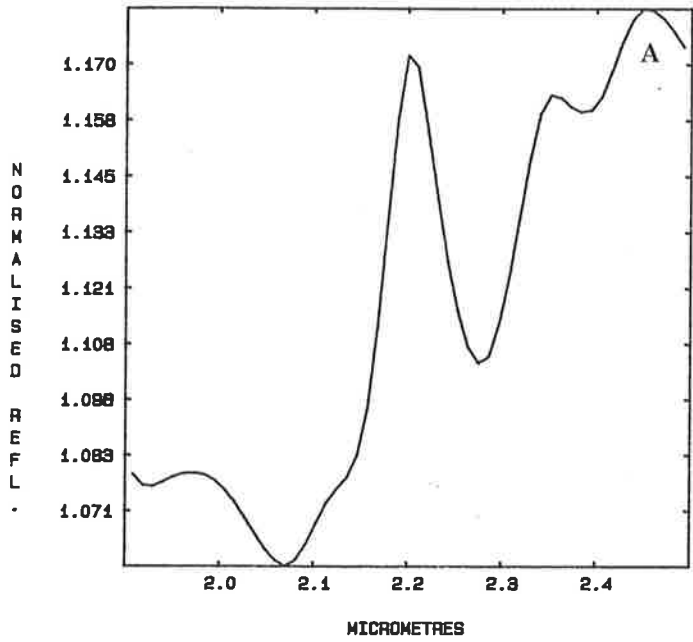
Figures 5.26a-d 'Log residual' spectra calculated from various mixtures of SWIR laboratory reflectance spectra.

Figures 5.26a&b are synthetic 'log residual' spectra of 1:1 mixtures of vegetation with the average muscovitic, and the average sericitic, spectra respectively.

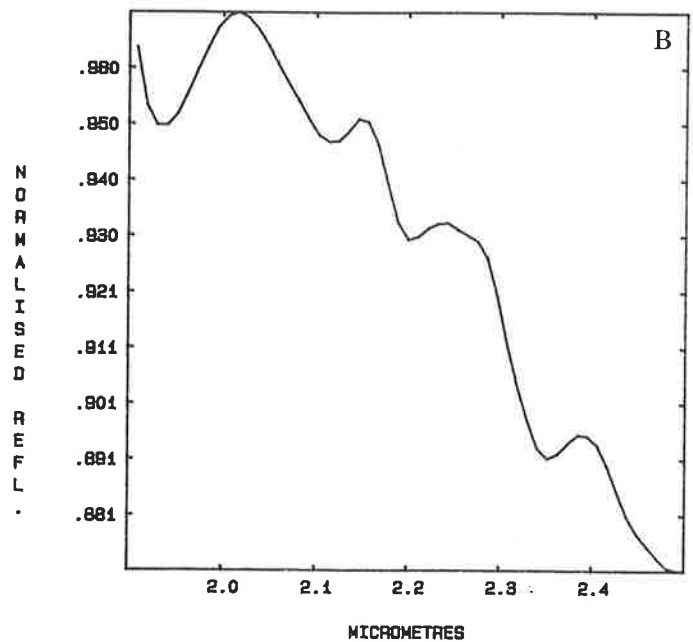
Figures 5.26c&d are the equivalent synthetic 'log residuals' for 1:1.5 mixtures.

Note that both the sericite + vegetation log residuals have a 2.35 μ m feature, and minima at 2.2 μ m. Corresponding muscovite + vegetation log residuals still have positive 2.2 μ m features (maxima) because they do not absorb strongly enough to overcome the effect of mixing with vegetation.

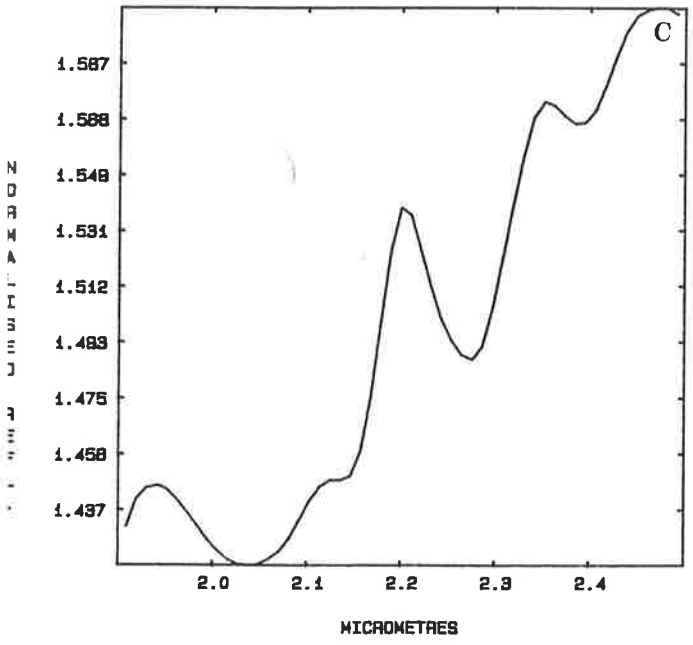
VEG.+MUSCOVITE 1:1



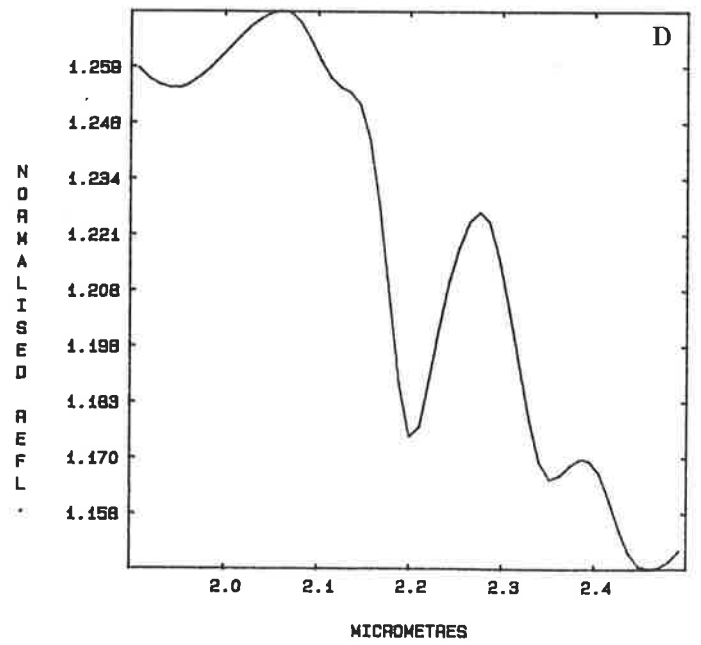
VEG.+SERICITE 1:1



VEG.+MUSCOVITE 1:1.5



VEG.+SERICITE 1:1.5

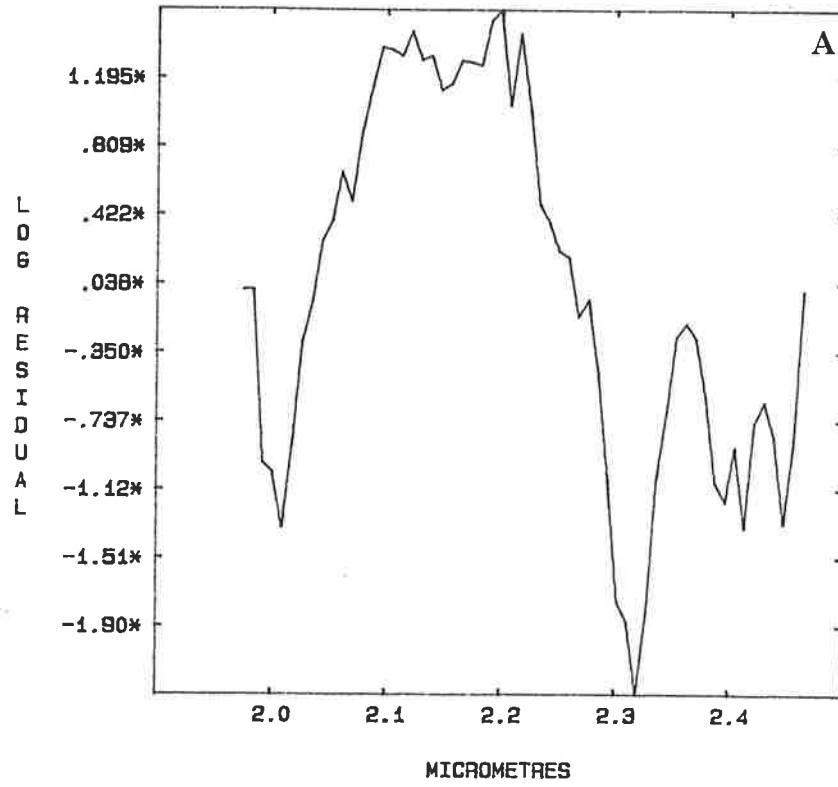


Figures 5.27a&b Airborne SWIR log residual and laboratory reflectance spectra of talc-rich soil from line MT8.

Figure 5.27a is the airborne spectrum of a talc-rich soil (flight line MT8, samples 447-450).

Figure 5.27b is the laboratory spectrum of a representative field sample from pixel 448 on line MT8.

SAMPLES 447, 450



MTB/448/S

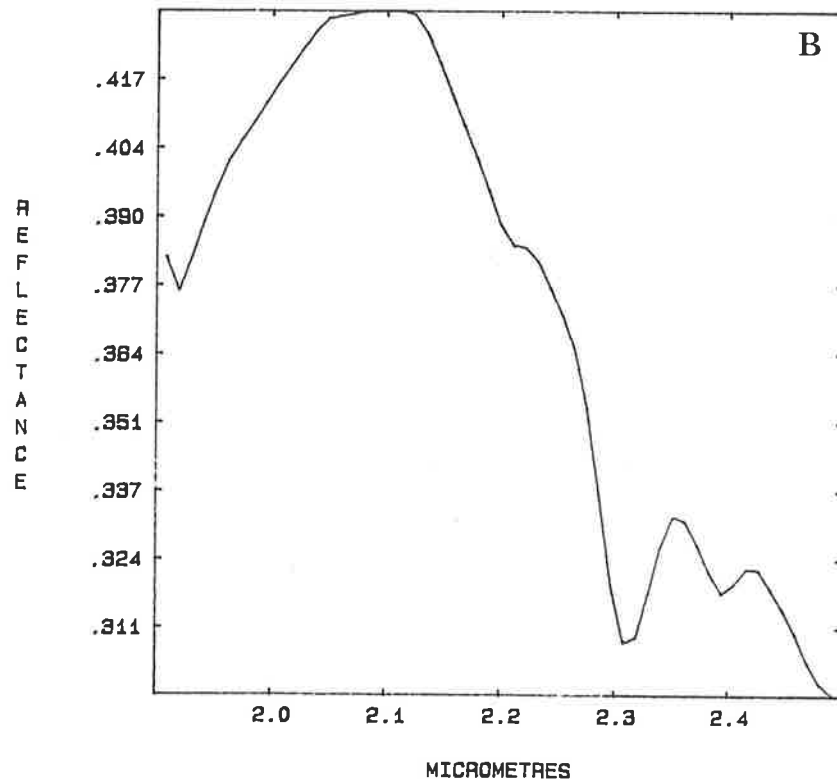
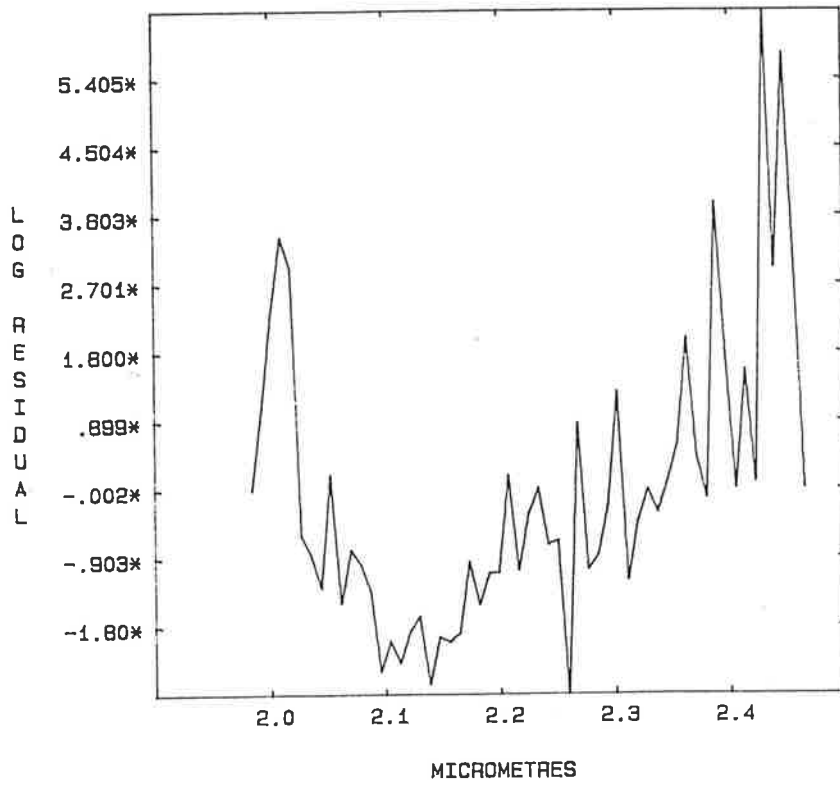


Figure 5.28 Airborne SWIR log residual spectrum of line MT10,
pixels 347-349.

This spectrum is extremely noisy, and there is no sign of any absorption feature that could be attributed to the presence of topaz.

LINE MT10 PIXELS 347-349

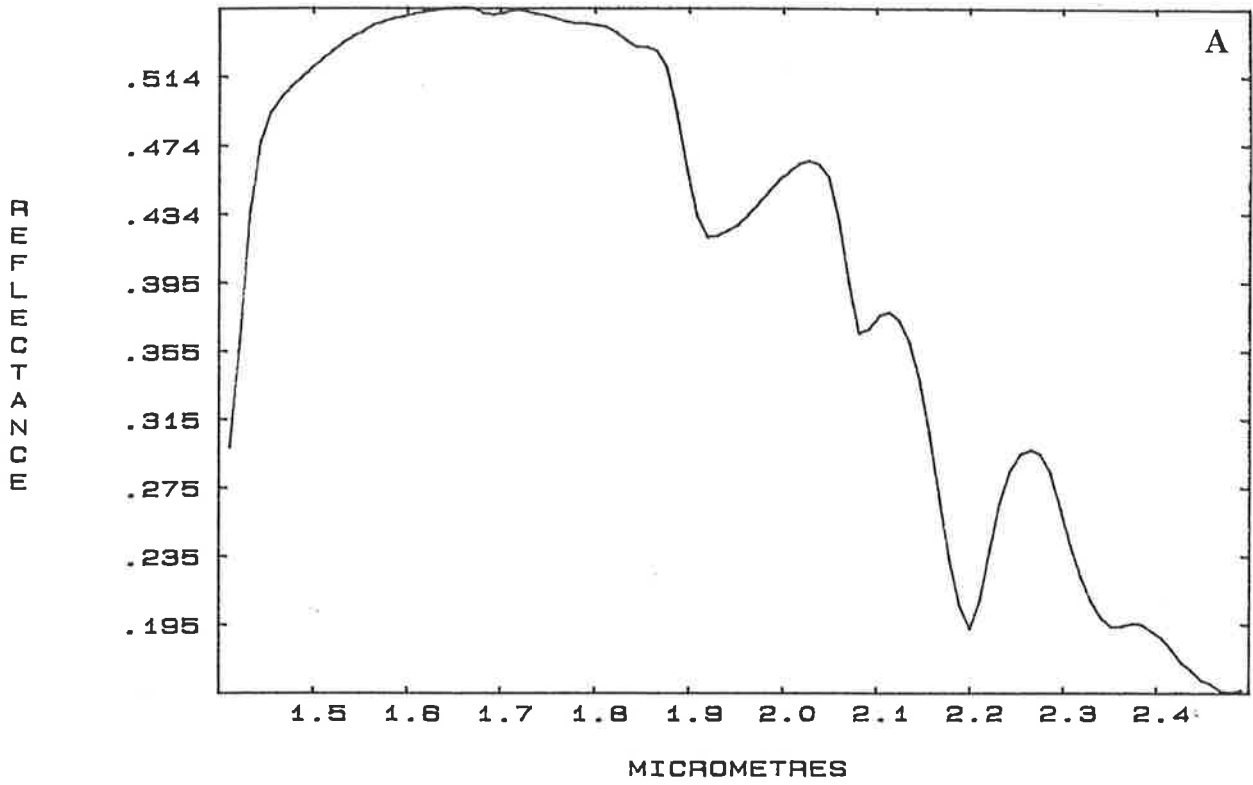


Figures 5.29a&b SWIR laboratory reflectance spectra of weathered and fresh surfaces of topaz-rich sample.

Figure 5.29a is the laboratory spectrum of a weathered surface of a rock containing between 5% and 20% topaz.

Figure 5.29b is the laboratory spectrum of a fresh surface of the same sample. Note that sericite-type features are present in the spectrum as well as the sharp 2.1 μ m absorption due to topaz.

11-8-83 MT10/348/3 W



11-8-83 MT10/348/3 F

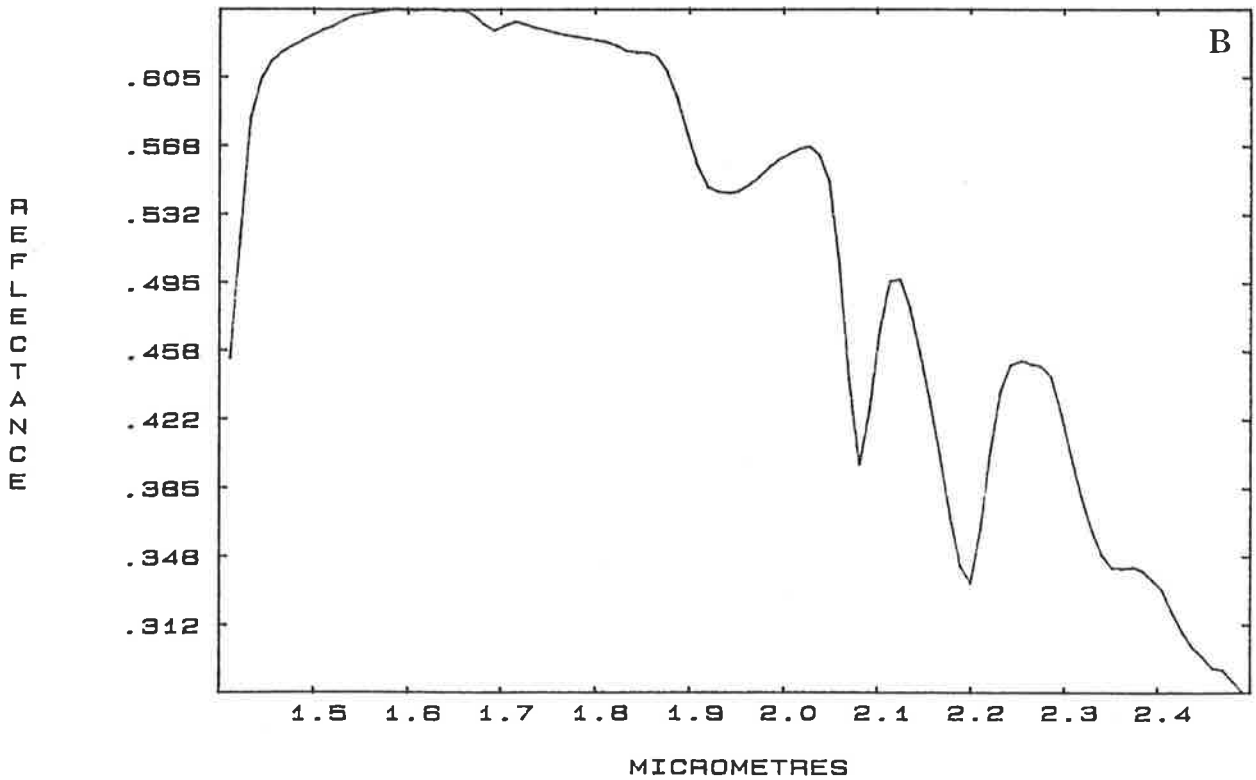


Figure 5.30 Field spectra of altered and unaltered rocks from
Silver Bell, Arizona (after Abrams and Brown, 1985).

Note that absorption features at all wavelengths are either much weaker, or entirely absent, in unaltered equivalents of the altered rocks. Spectra were acquired in situ using the Jet Propulsion Laboratory's Portable Field Reflectance Spectrometer (PFRS).

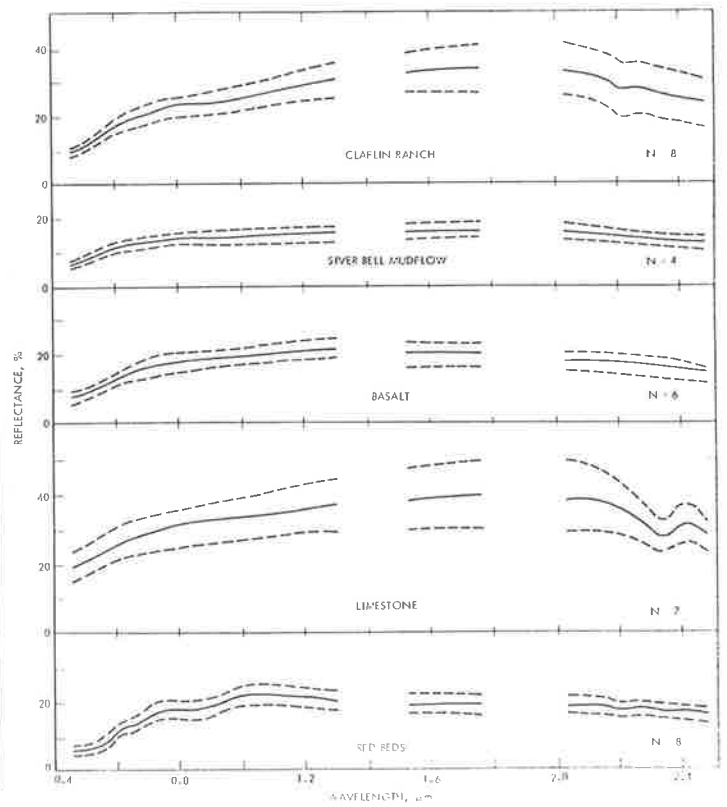
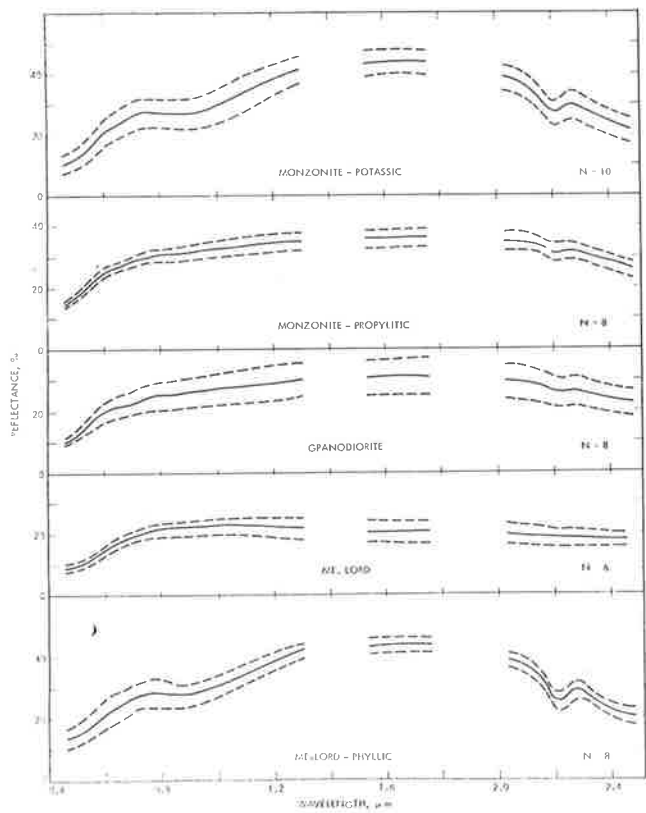
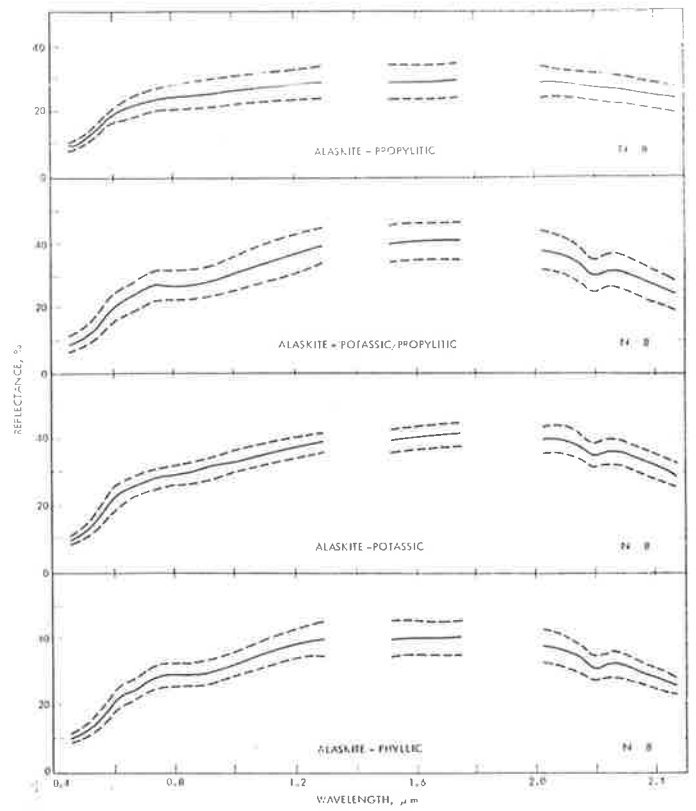
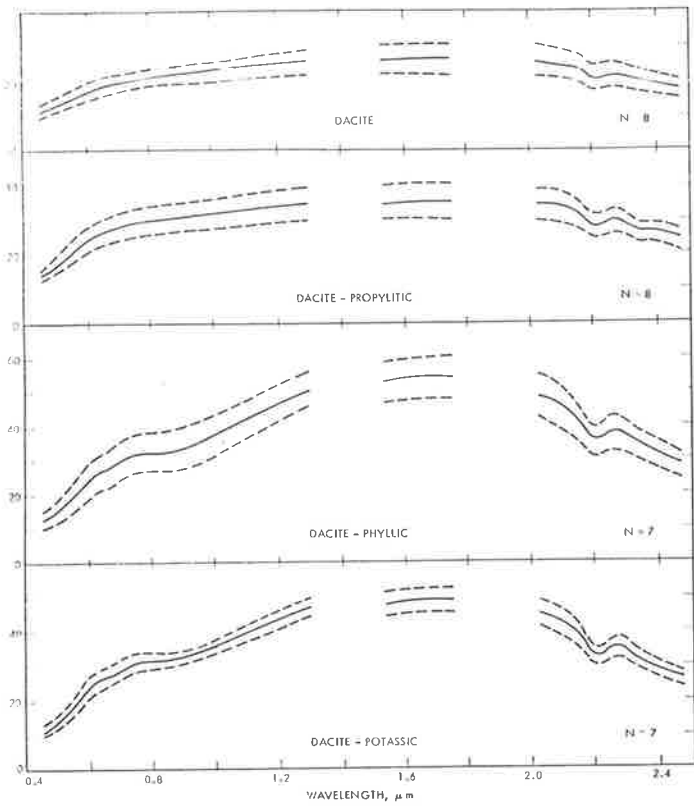
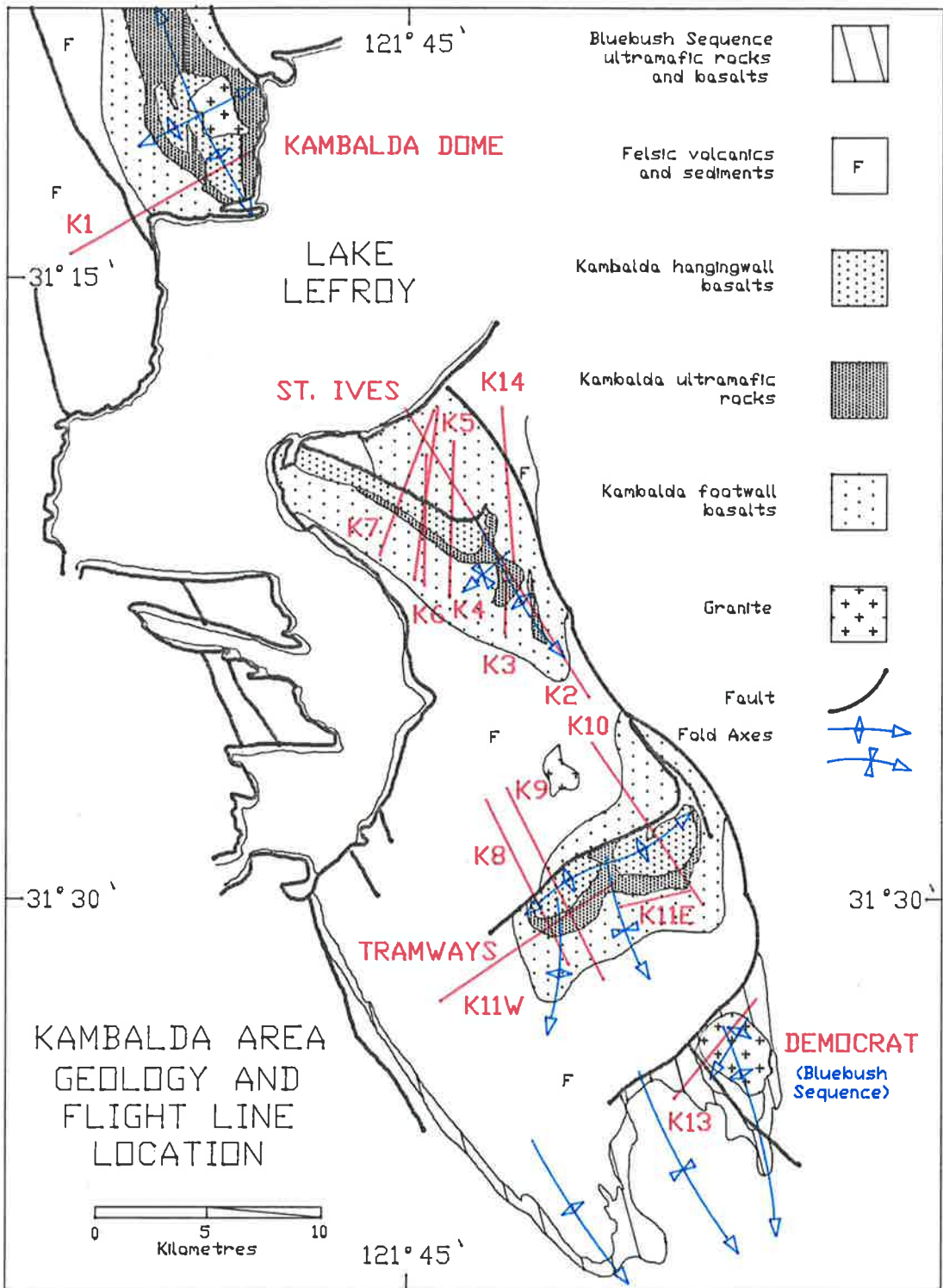


Figure 6.1 Kambalda area geology and flight line locations
(modified after Gresham and Loftus-Hills, 1981).

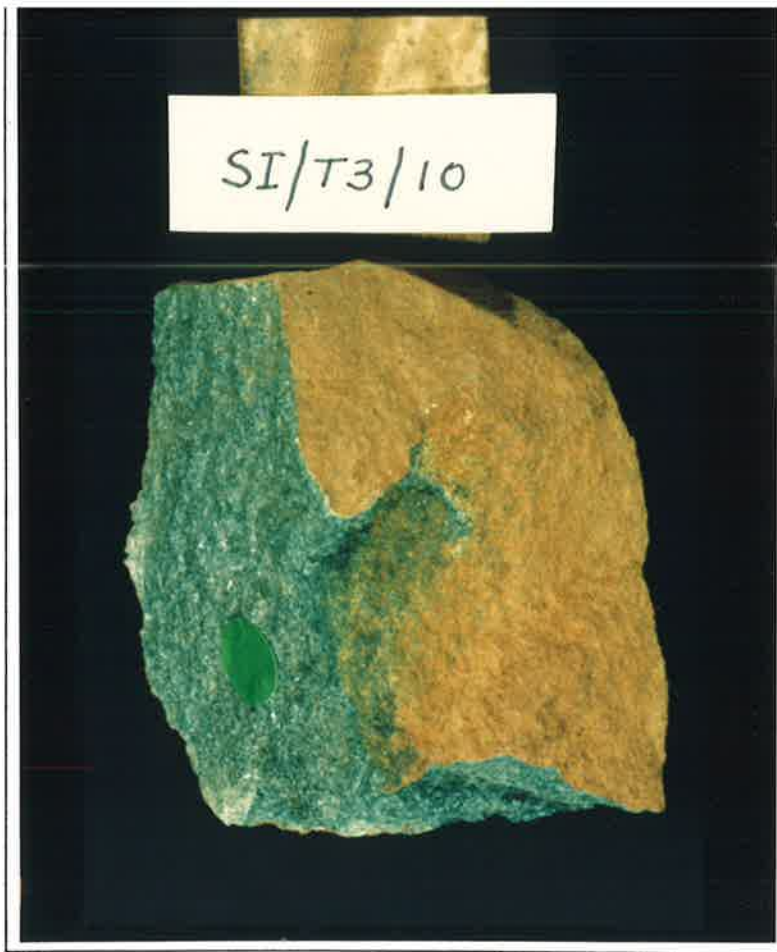


Figures 6.2a-c Photograph and SWIR laboratory reflectance spectra of weathered and fresh surfaces of a basalt sample from Kambalda.

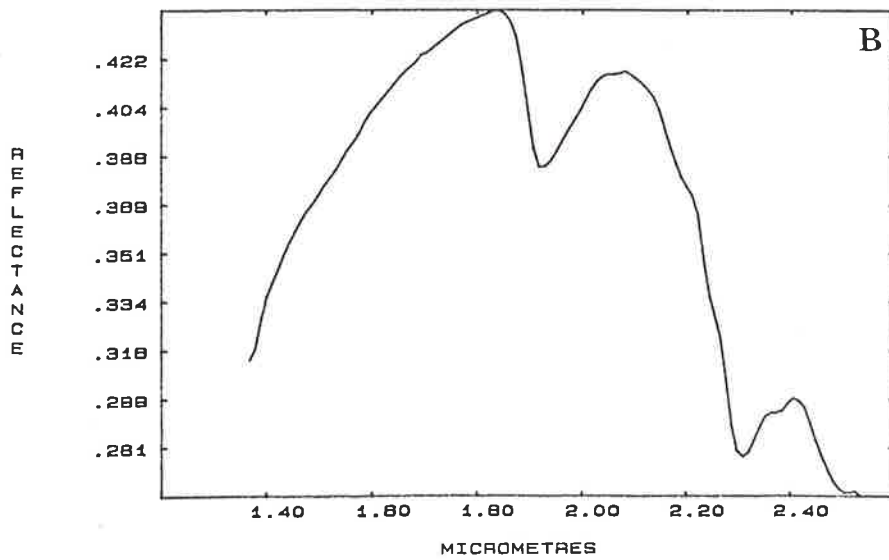
Figure 6.2a is a hand specimen of basalt with weathering rind.

Figure 6.2b is the spectrum of the weathered surface of the sample shown in Fig. 6.2a. Spectral features are those of amphibole and chlorite (responsible for the position of the major feature).

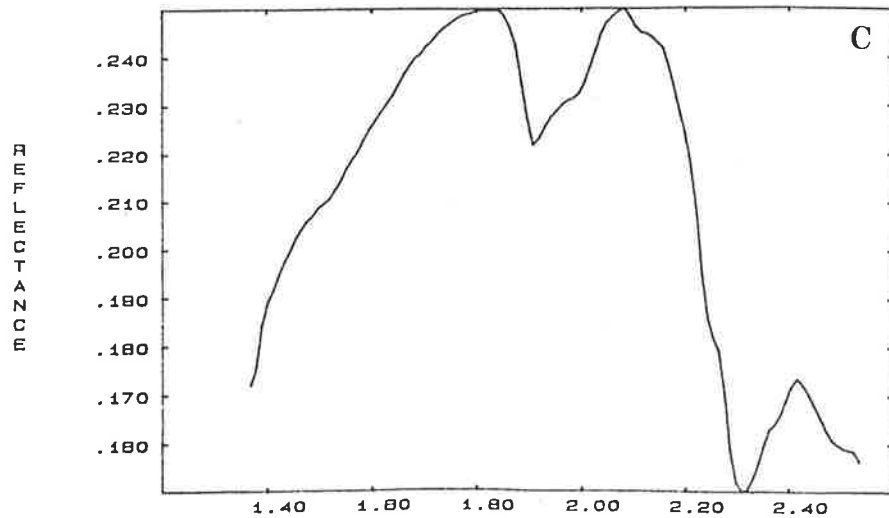
Figure 6.2c is the spectrum of the "fresh" surface of sample shown in Fig. 6.2a. Note the similarity to the 'weathered' spectrum. The 2.0 μ m feature in this spectrum is due to chlorite.



23/1/84 SI/T3/MV/10/W



23/1/84 SI/T3/MV/10/F



Figures 6.3a&b Photograph and SWIR laboratory reflectance spectrum of a lichen-covered soil from Kambalda.

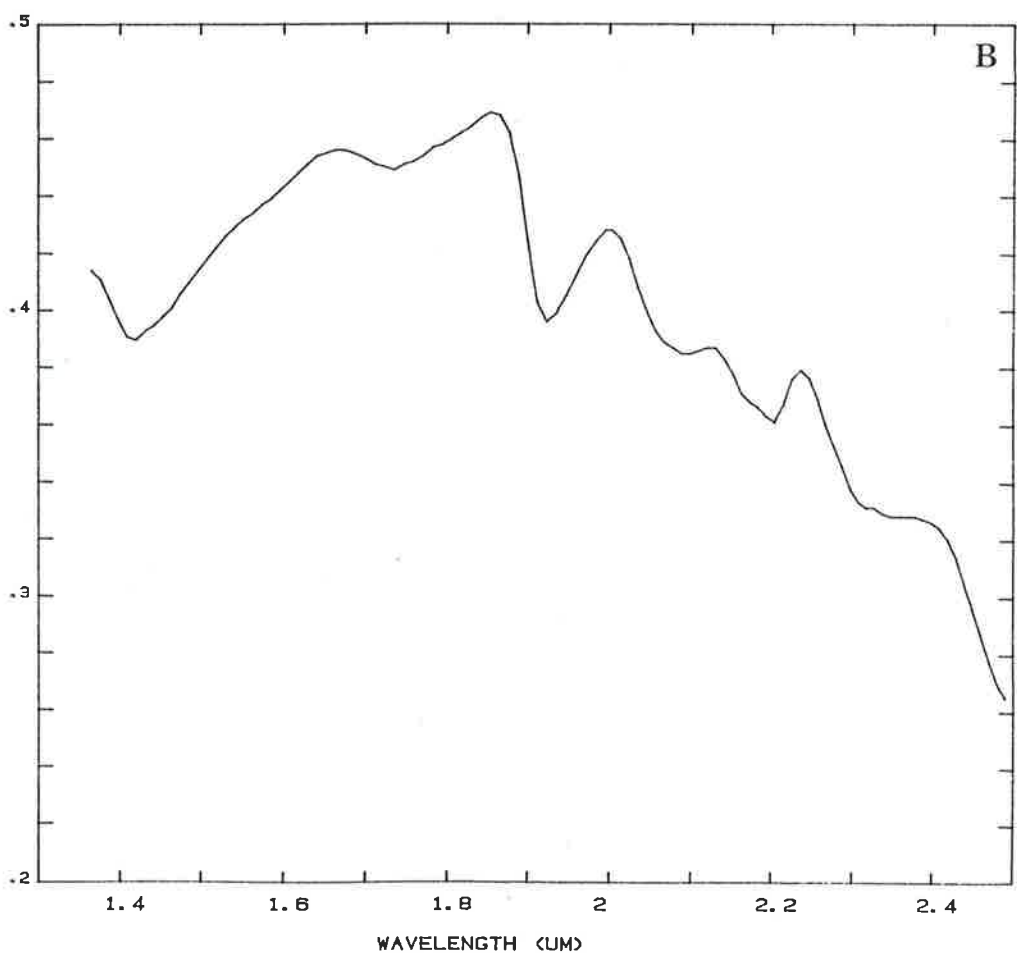
Figure 6.3a shows the lichen 'crust' and, to the left, the soil beneath the lichen.

Figure 6.3b is the spectrum of soil-lichen showing kaolin absorption features. Familiar 2.1 and 2.3 μ m absorption features typical of dry vegetation are caused by the lichen. Fine kaolin on the surface is also contributing to the spectrum with a 2.17 and 2.208 μ m doublet.

A



SWIR BIDIRECTIONAL REFLECTANCE

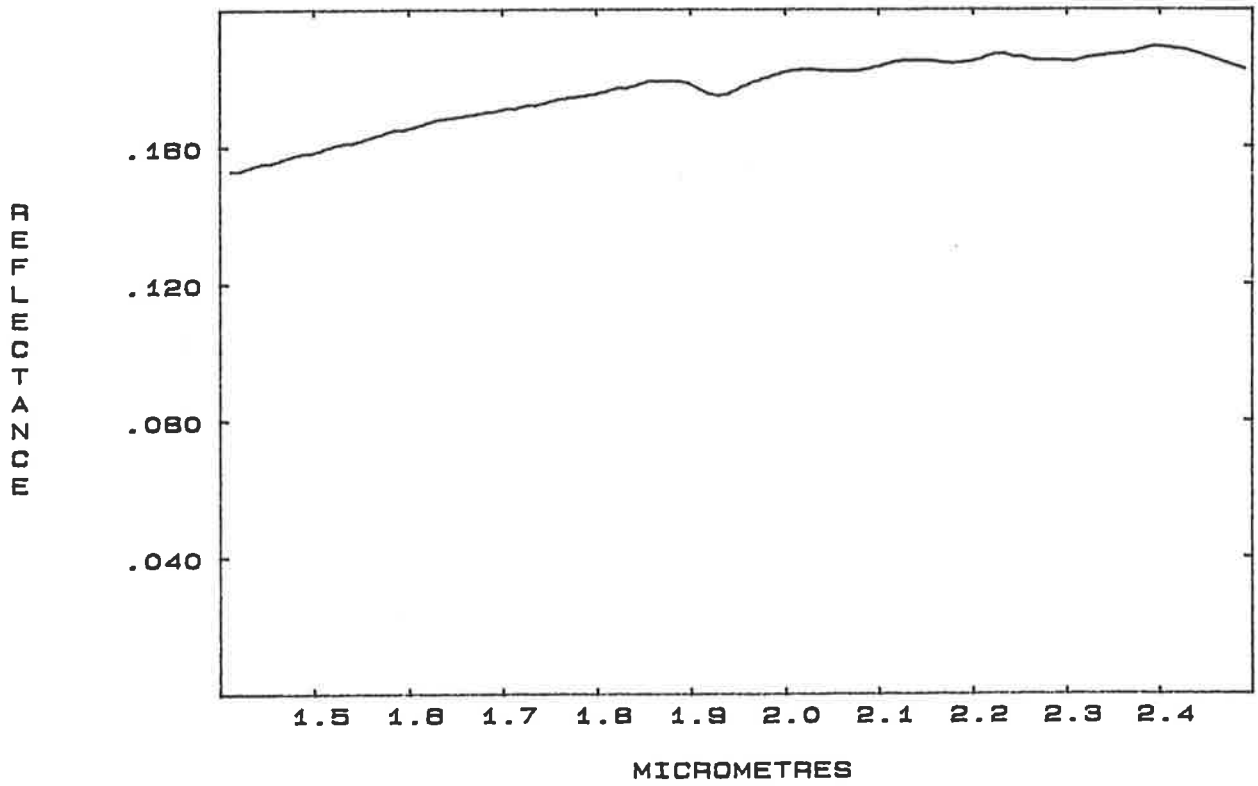


B

Figure 6.4 Flat SWIR laboratory reflectance spectrum from
Kambalda.

The dark colour of the sample, and a lack of strongly absorbing minerals results in a flat, noisy spectrum.

K2/440/F Q100 BLACK CHERT

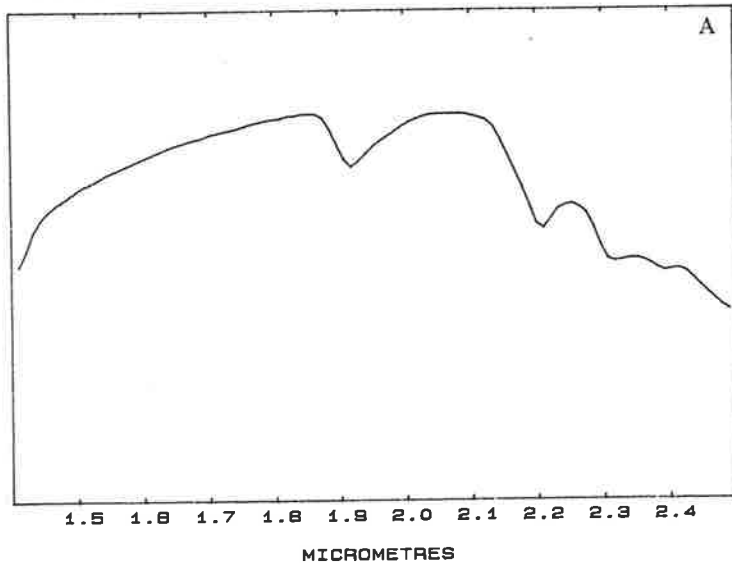


Figures 6.5a-c SWIR laboratory reflectance spectra of samples with varying quartz content from Kambalda.

Figure 6.5a has moderate $1.9\mu\text{m}$ (water) absorption, while Figs. 6.5b&c have strong and very strong $1.9\mu\text{m}$ (water) absorption features, respectively. The strength of these features is correlated with the percentage of quartz in the samples (see plot titles).

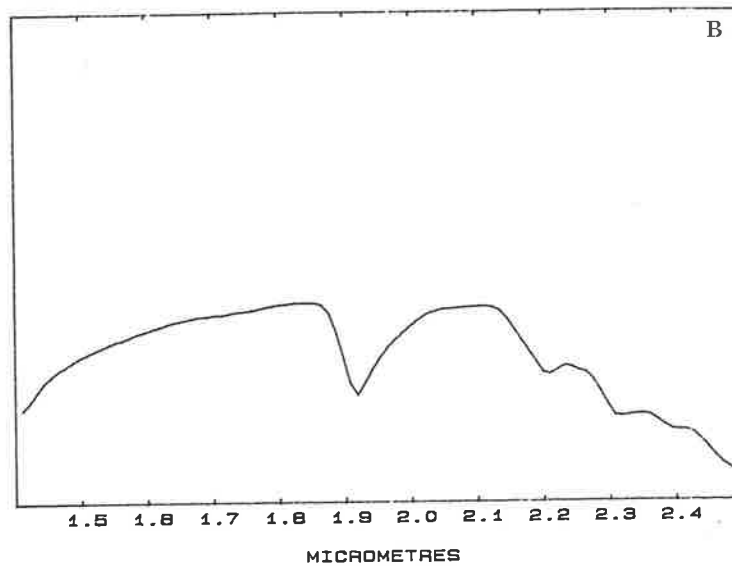
K1/30/S AB48 Q38 TC8 CHL4 AMP3 BI1/HEM<5

REFLECTANCE



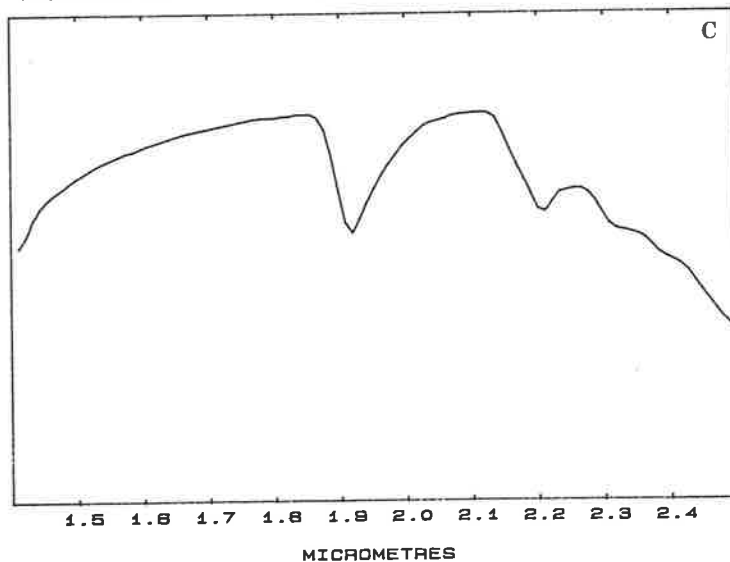
K1/131/S Q68 AB11 AMP8 TC7 CHL6 BI1/HE5

REFLECTANCE



K2/280/S Q91 CAL5 AB4/HEM<5

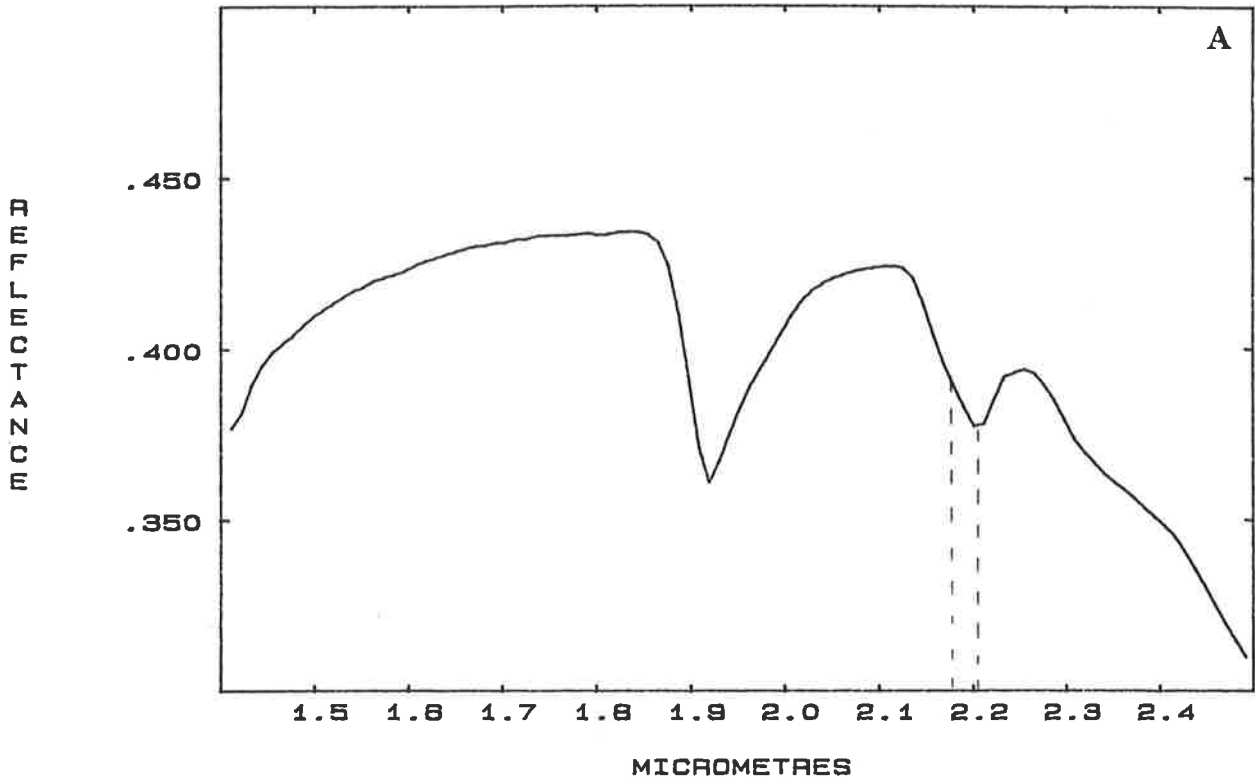
REFLECTANCE



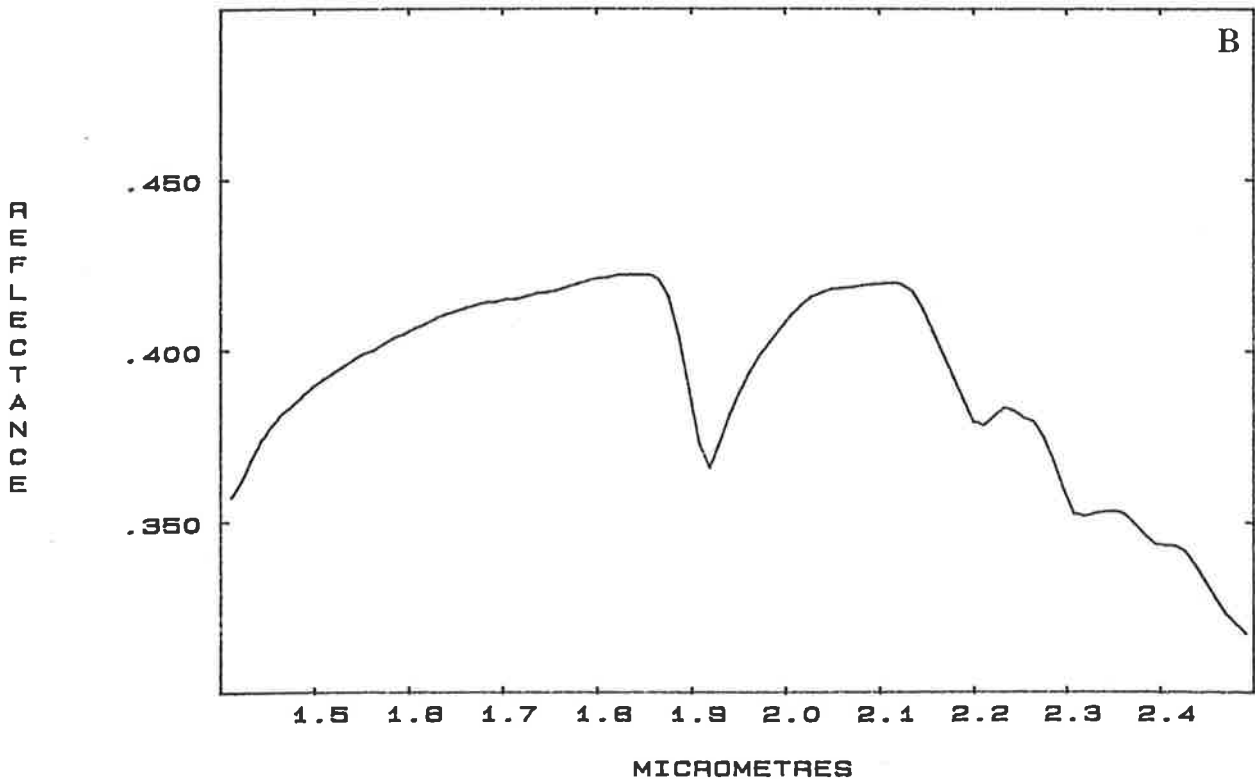
Figures 6.6a&b SWIR laboratory reflectance spectra of kaolin-bearing samples from Kambalda.

Kaolin's 2.17 μ m absorption feature is generally only well developed where kaolin crystallinity is high. At Kambalda, the 2.17 μ m feature usually only causes an asymmetry in the major absorption feature. There are many cases at Kambalda where weak 2.2 μ m absorption features appear in the spectrum, but no mineral which absorbs in that region is detected by XRD. Where 2.2 μ m features are strongest, the characteristic kaolin shape is seen. Figure 6.6a depicts one example of a stronger kaolin spectrum, while Fig. 6.6b depicts a spectrum where the mineralogy cannot be accurately determined, but is likely to be kaolin.

K7/310/S Q88 AB2/HEM<5



K1/131/S Q88 AB11 AMP8 TC7 CHL6 BI1/HE5

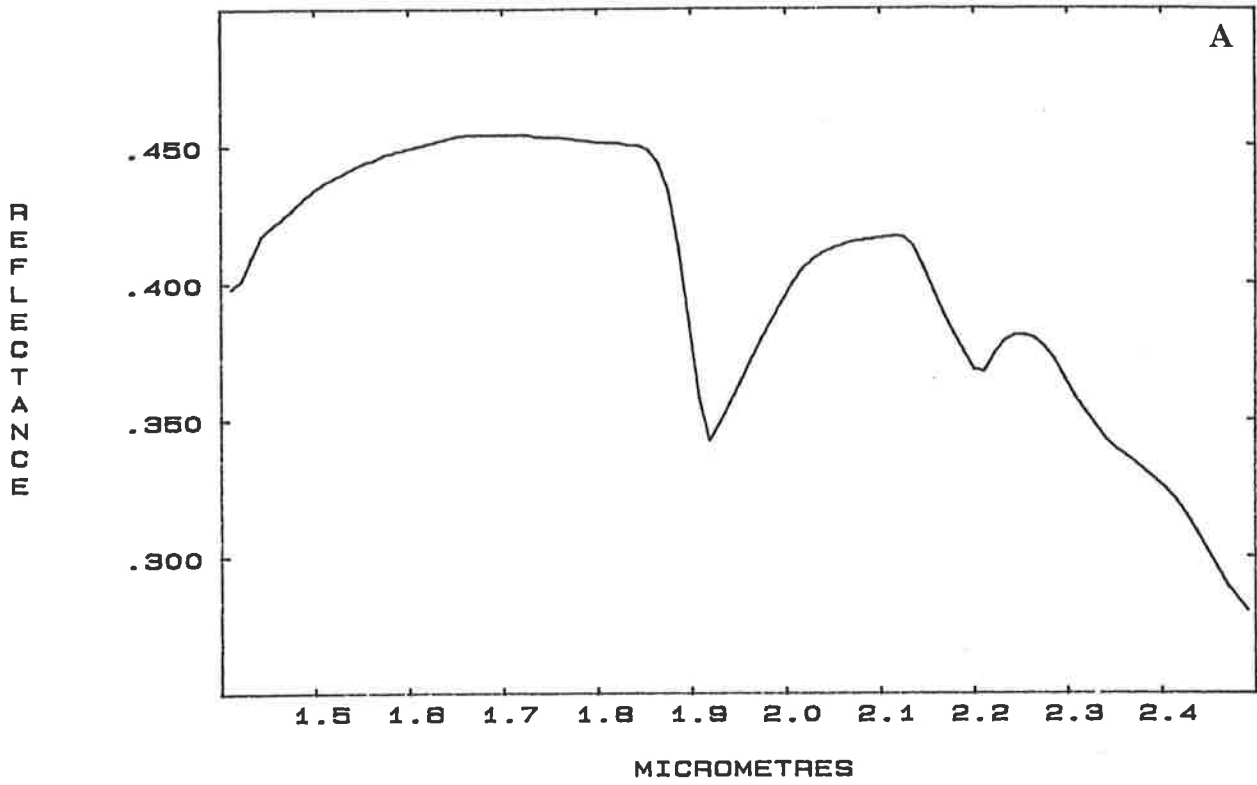


Figures 6.7a&b SWIR laboratory reflectance spectra of intact and powdered calcrete nodules from Kambalda.

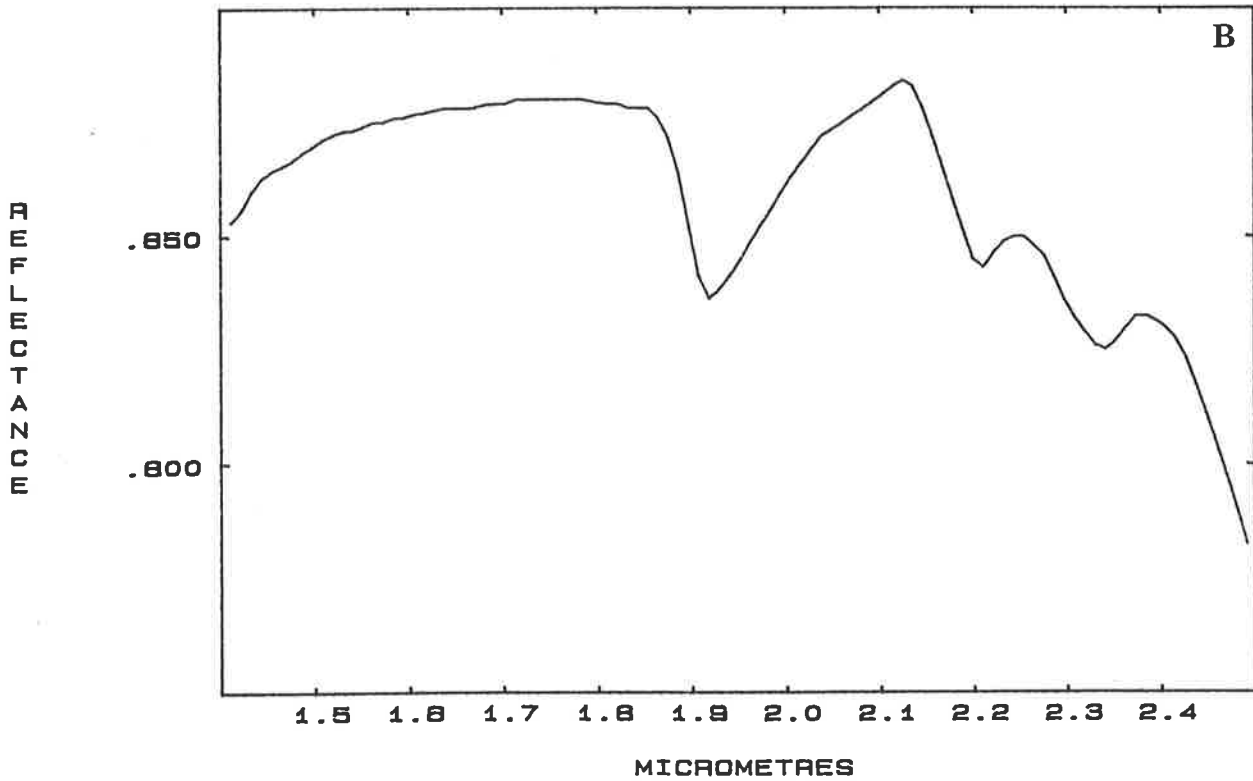
Figure 6.7a depicts the spectrum of calcrete nodules showing kaolin spectral features. The failure of XRD analysis to detect kaolin is attributed to its poorly crystalline form. Al-OH bonds are still present, however, and this is sufficient to produce the observed absorption features.

Figure 6.7b is the spectrum of the same calcrete in powdered form, showing kaolin and calcite absorption features. The presence of calcite features in the crushed sample suggests that their absence in Fig. 6.7a is a surface phenomenon.

K14/31/S CAL75 Q20 TC5



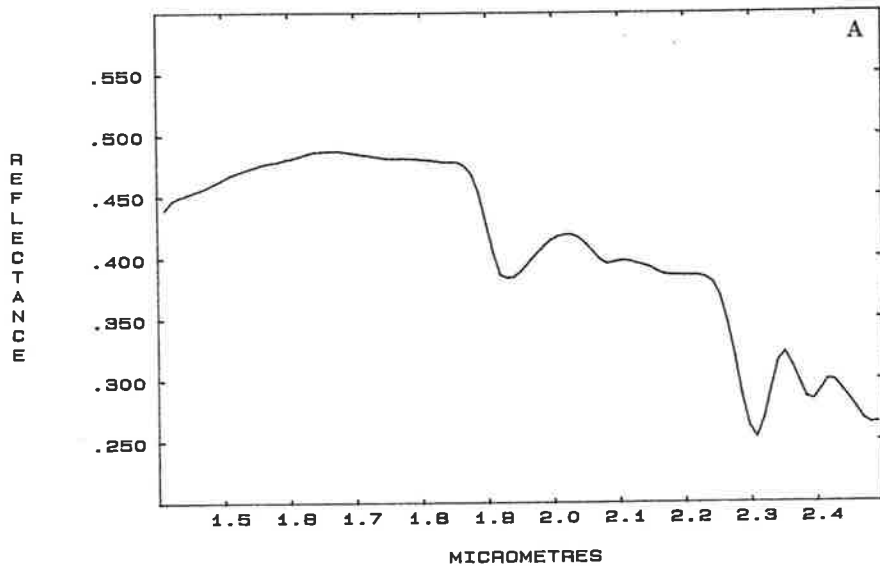
K14/31/S/P CAL75 Q20 TC5



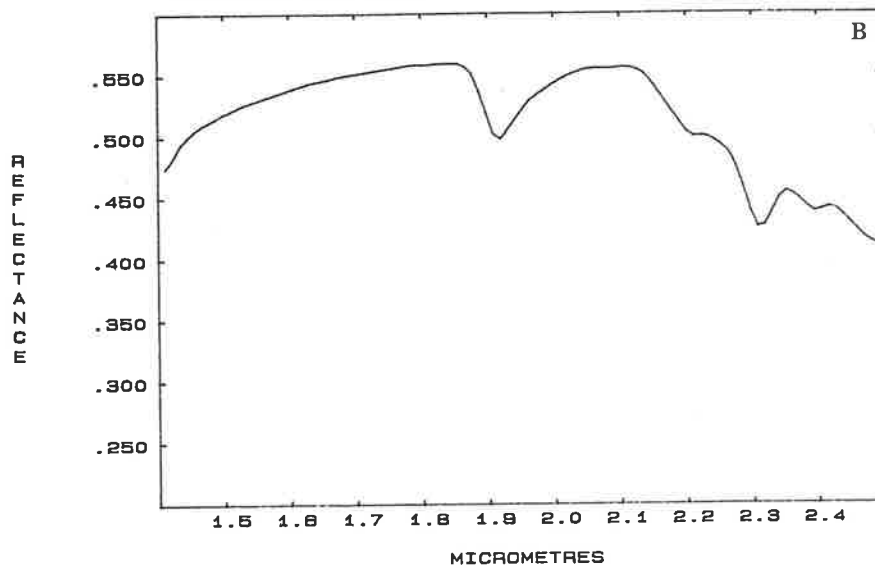
Figures 6.8a-c SWIR laboratory reflectance spectra of talc-bearing samples from Kambalda.

Talc absorption features can be detected even at low percentages (depending on the other minerals present and their percentages). Expansion of the y-axis scales would make the absorption features in Figs. 6.8b&c look stronger, but they are all presented at the same scale for comparison of their relative strengths.

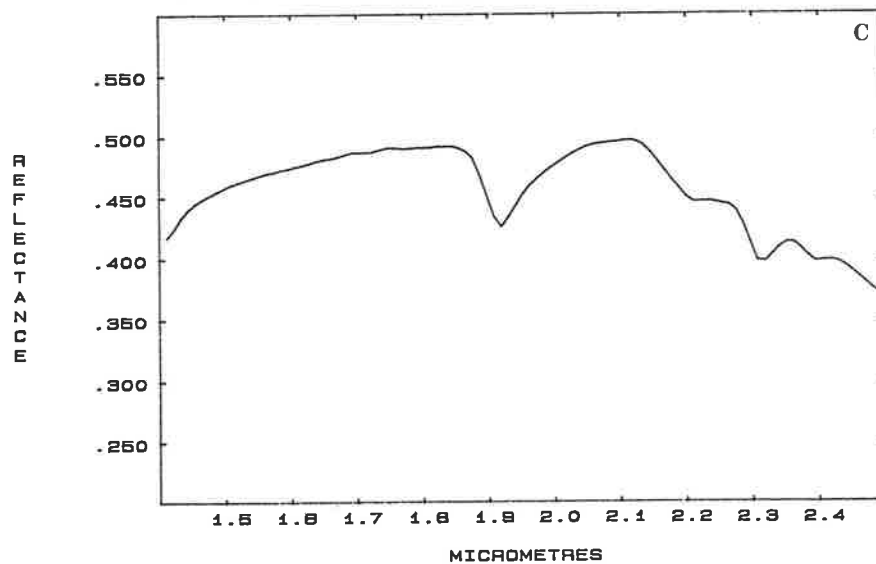
K10/304/F2/W TC98 CHL3 MAG1/HEM<5



K11W/87/S Q65 TC19 AB10 CHL8



K3/88/S Q78 TC8 AB8 AMP4 CHL3/KA0<5, HE<5



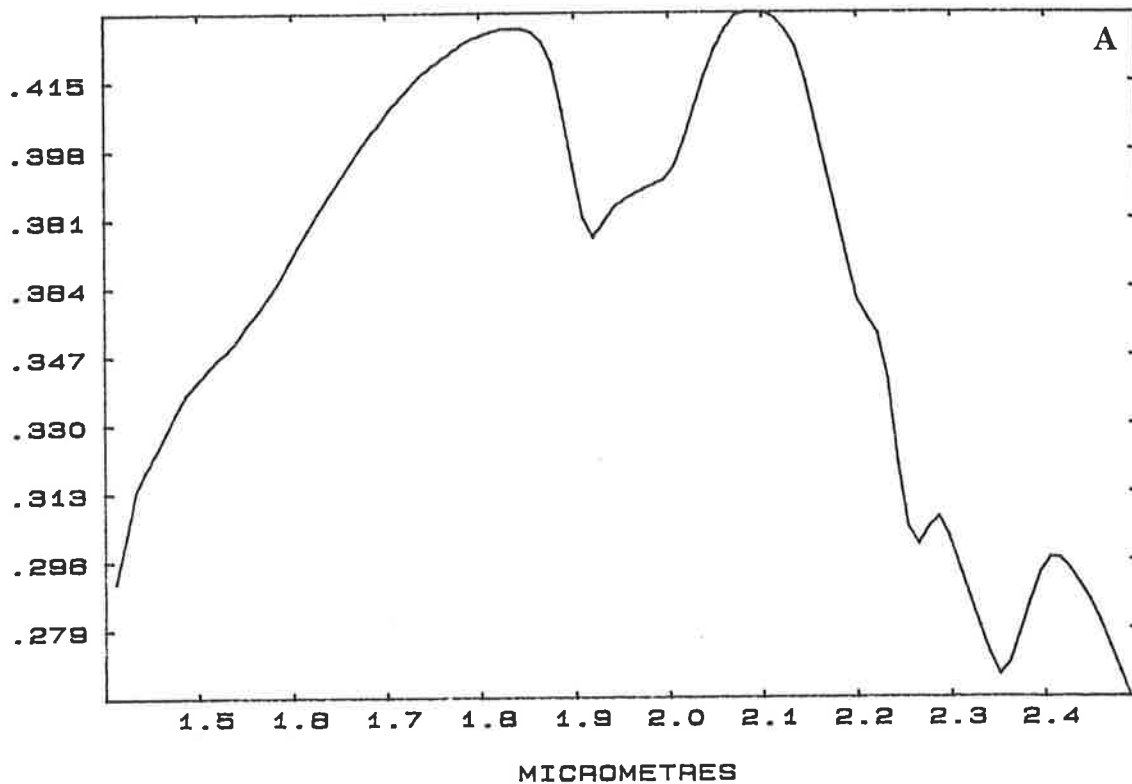
Figures 6.9a&b SWIR laboratory reflectance spectra of chlorite- and amphibole-bearing samples from Kambalda.

The chlorite spectrum (Fig. 6.9a) has a more pronounced background curve than talc. Diagnostic absorption features are at $2.0\mu\text{m}$ (causing the asymmetry of the $1.9\mu\text{m}$ water absorption feature), 2.26 and $2.34\mu\text{m}$.

Like chlorite, amphibole (Fig 6.9b) has a pronounced background curve. The asymmetry of the $1.9\mu\text{m}$ water absorption feature in this spectrum is caused by the presence of 7% chlorite. Amphibole's absorption features are at $2.32\mu\text{m}$ and a shoulder near $2.4\mu\text{m}$, but in this case the spectrum shows a minimum at $2.33\mu\text{m}$. This may be due to the influence of chlorite, which tends to absorb at wavelengths slightly longer than amphibole.

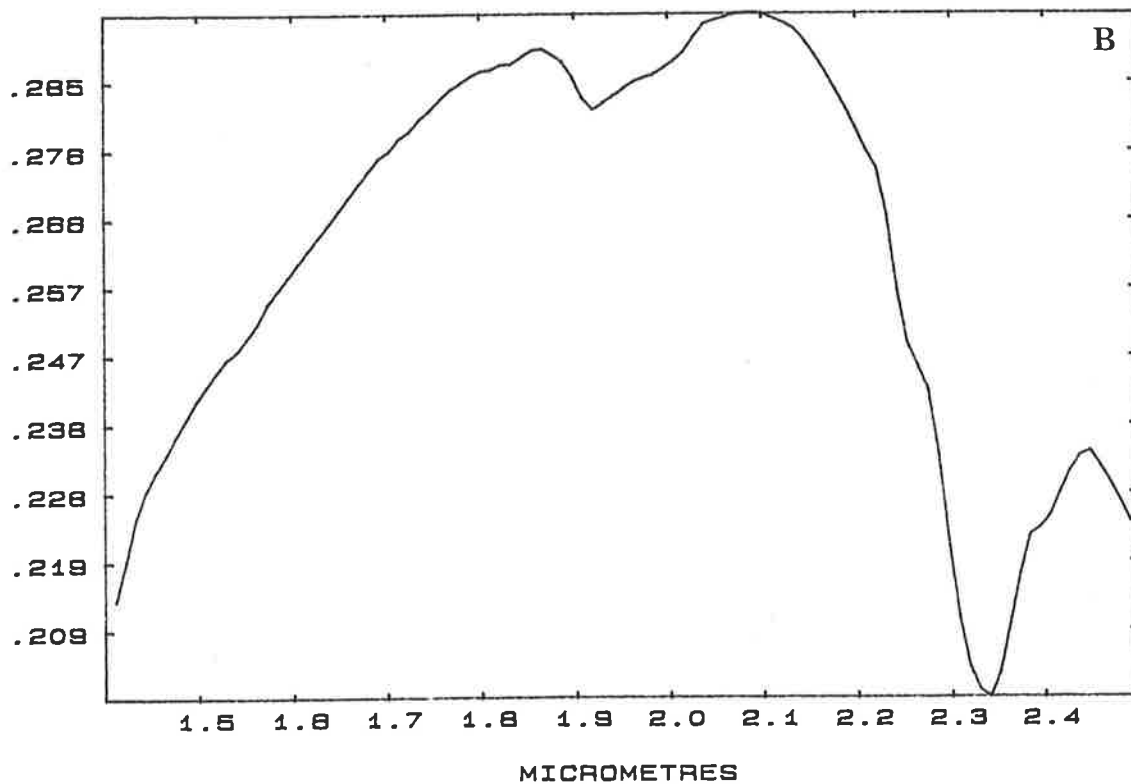
K4/123/F/W CHL42 Q34 PL23

REFLECTANCE



K1/74/F1/W AMP70 PL19 CHL7 Q5

REFLECTANCE



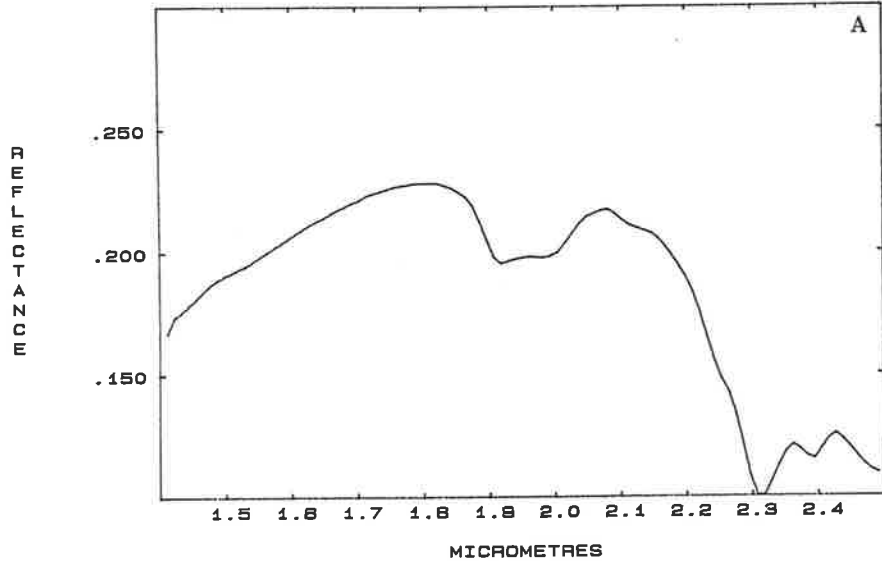
Figures 6.10a-d SWIR laboratory reflectance spectra of typical mixtures from Kambalda.

The spectra shown in Figs. 6.10a&b, of an amphibole-chlorite-talc rock and a talc-chlorite rock, respectively, show that talc dominates the spectrum, even though (in the case of Fig. 6.10a) it forms the smallest proportion of the rock. Any contribution by amphibole to both spectra is swamped by the talc, but chlorite still produces an absorption feature at $2.0\mu\text{m}$.

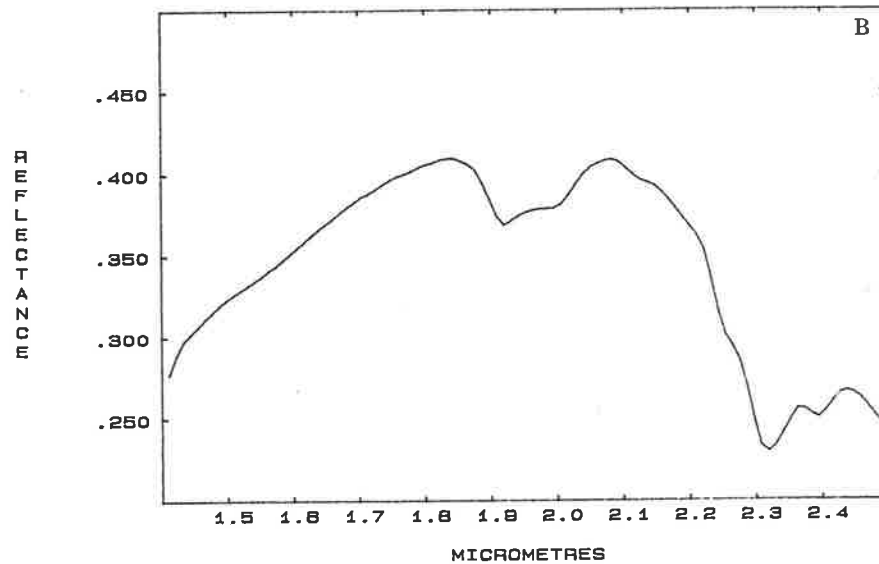
Figure 6.10c is the spectrum of an amphibole-chlorite rock powder. Most rocks containing almost equal amounts of amphibole and chlorite tend to have spectra dominated by chlorite, if their natural surfaces are measured. In some cases, powdered samples of the same material will produce spectra with slightly different characteristics. This spectrum appears to be dominated by effects due to amphibole.

Figure 6.10d is the spectrum of an amphibole-chlorite-epidote rock. The shape of this spectrum's major absorption feature is similar to those of both amphibole and epidote. The wavelength at which it occurs is $2.34\mu\text{m}$, which is longer than normal for amphibole, suggesting a contribution from epidote. The $2.0\mu\text{m}$ absorption feature is due to chlorite, which may also be influencing the position of the major feature.

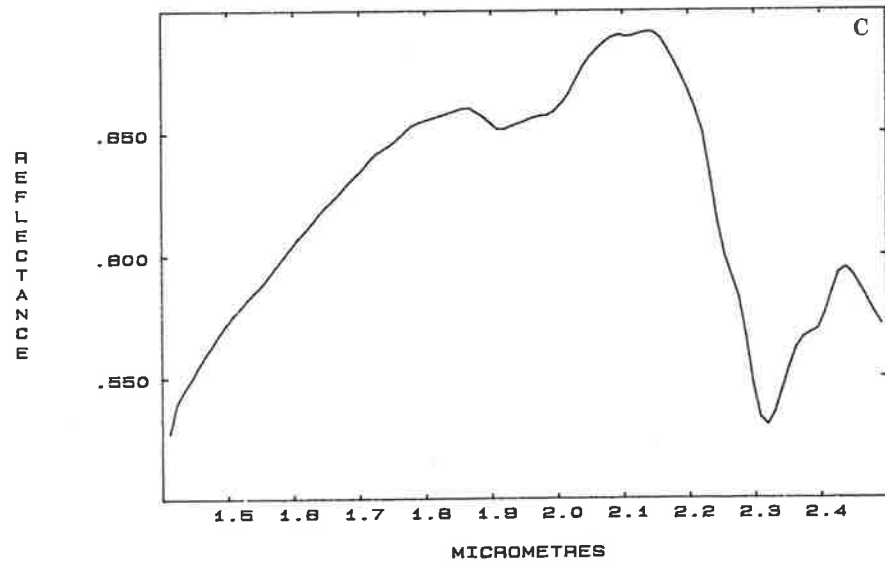
K7/288/F/W AMP44 CHL32 TC23



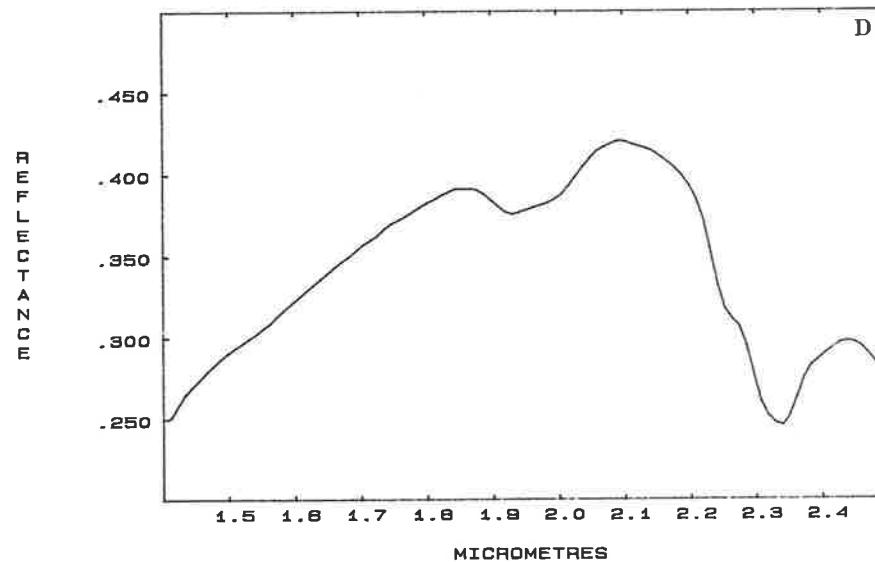
K7/180/F/W TC81 CHL33 PL4 AMP1



K1/31/F/P AMP50 CHL43 Q4 TC3



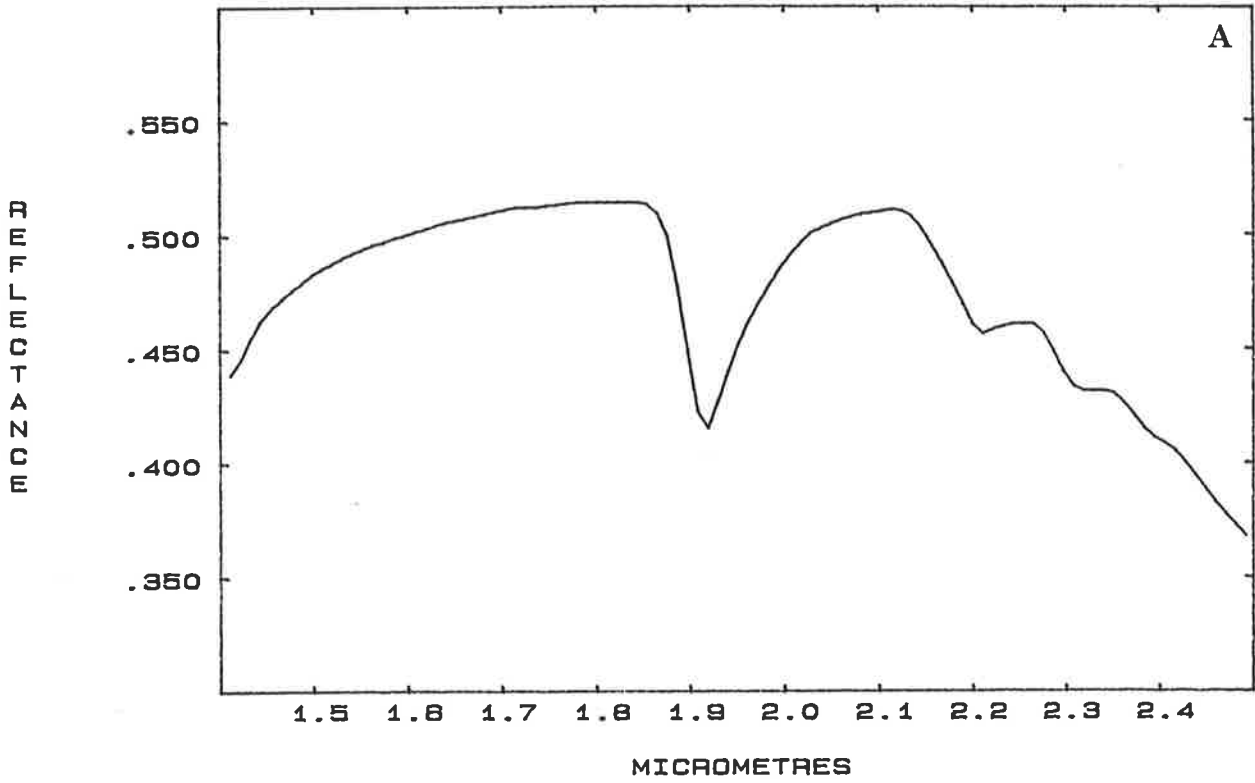
K8/448/F2/W2 AMP42 CHL25 Q20 AB14/EP10



Figures 6.11a&b SWIR laboratory reflectance spectra of materials exhibiting weak absorption features from Kambalda.

There are numerous examples of weak absorption features from Kambalda. Most of these cannot be interpreted confidently. Figure 6.11a is one example that shows weak features in both the 2.2 and 2.3 μ m regions. Figure 6.11b, on the other hand, exhibits a broad, weak absorption feature only in the 2.3 μ m region. This could be attributed to the 'mafic' mineralogy of Archaean basalts, except that XRD shows that diopside is present and the most likely cause of the absorption. Diopside is found only in Proterozoic dykes at Kambalda (D. Quick, pers. comm., 1984). This example underlines the importance of ground-truth work and supporting laboratory data.

K10/378/S Q63AB11CAL9AMP8TC4/KA0, HEM, M05



K10/273/F/W PL90 Q8 CHL2/DI40-50

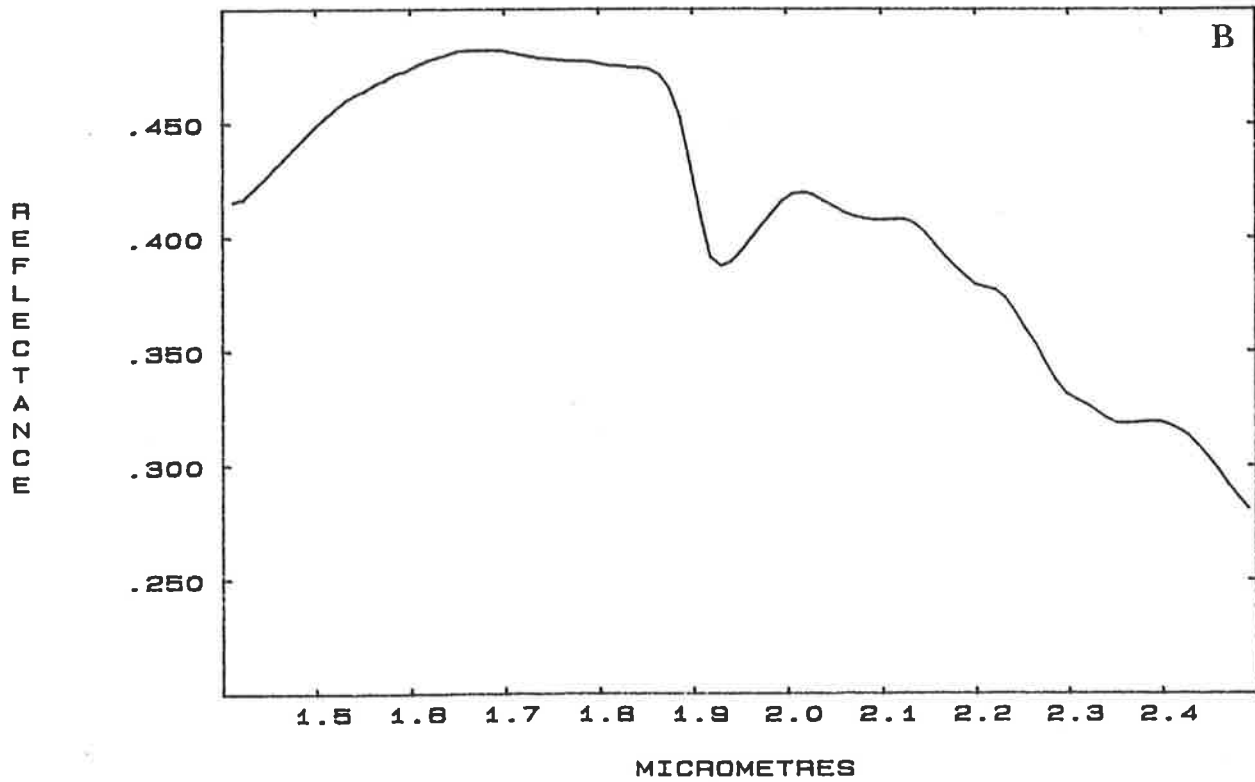
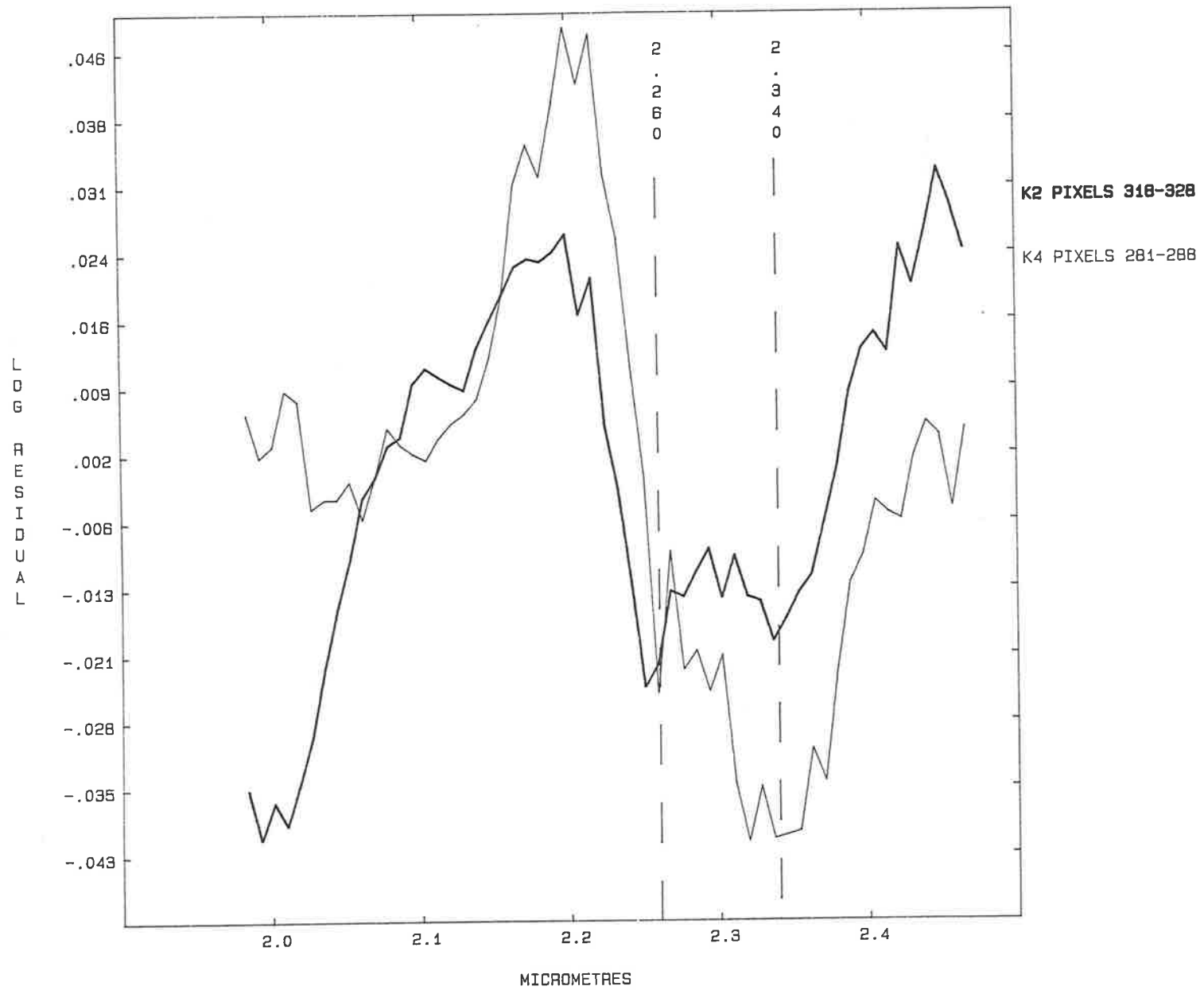


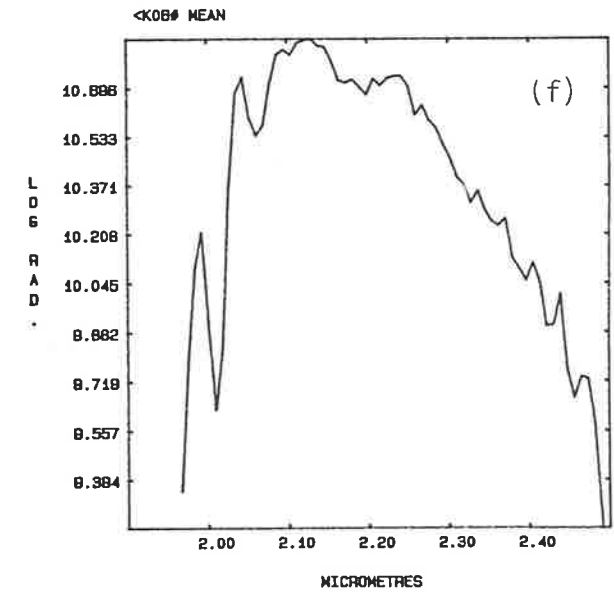
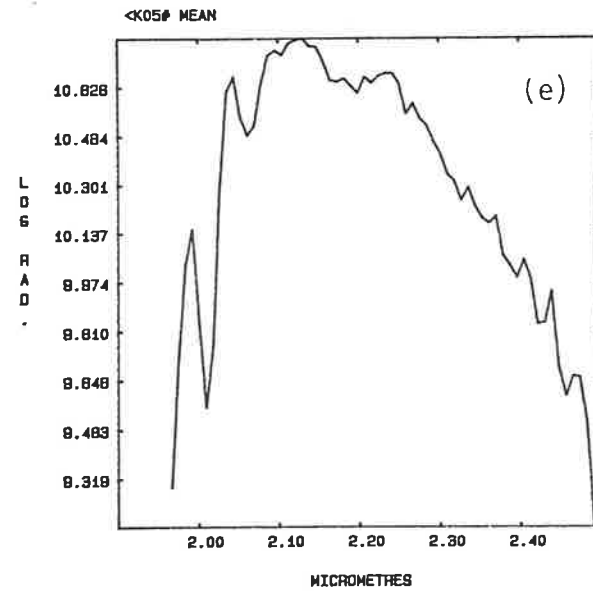
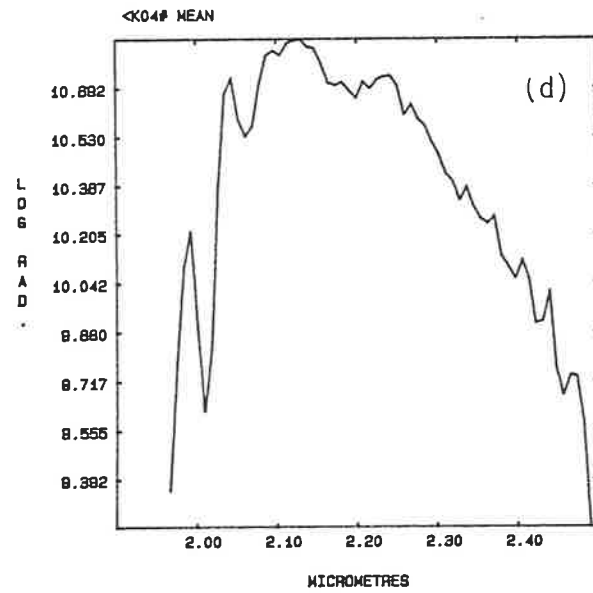
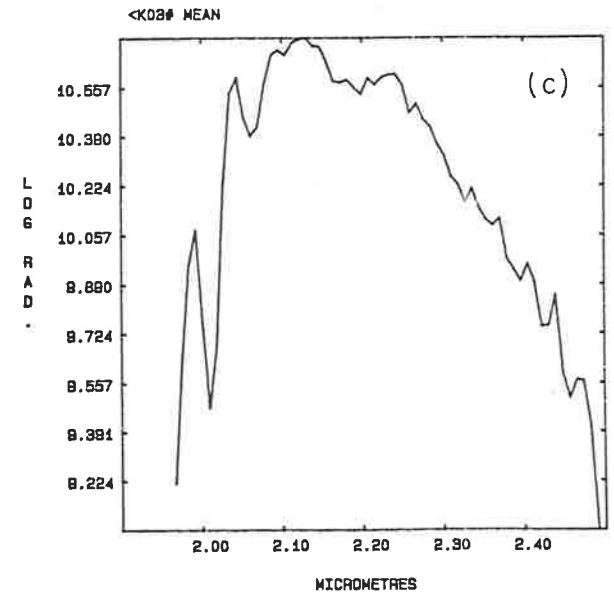
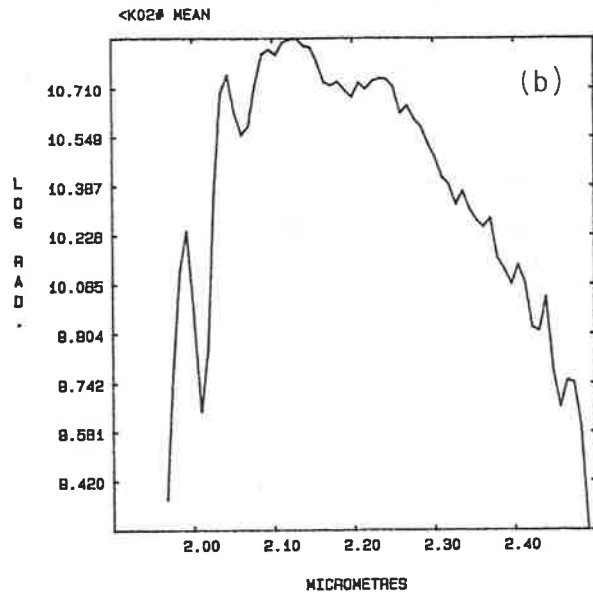
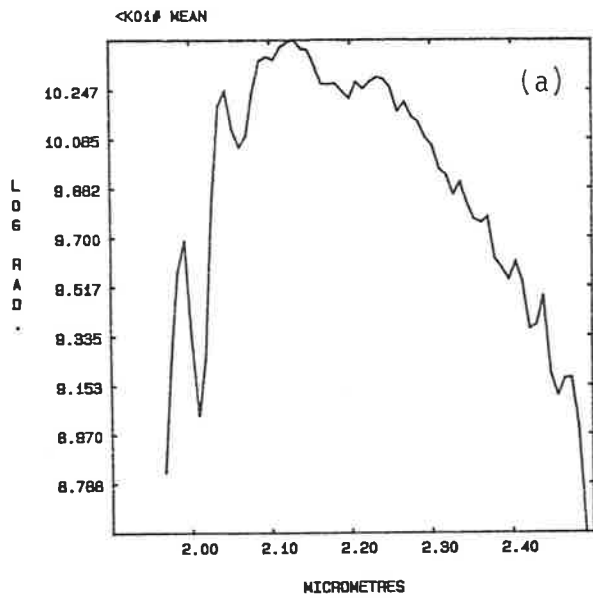
Figure 6.12 SWIR airborne log residual spectra of two chlorite-bearing areas at Kambalda.

Note particularly that the positions of the 2.26 μ m absorption features vary. This is difficult to explain other than by variation of the chlorites' composition. However, any variation in the position of the 2.34 μ m minimum could be explained by the presence of other minerals absorbing at slightly different wavelengths (e.g. talc, amphibole, or epidote).

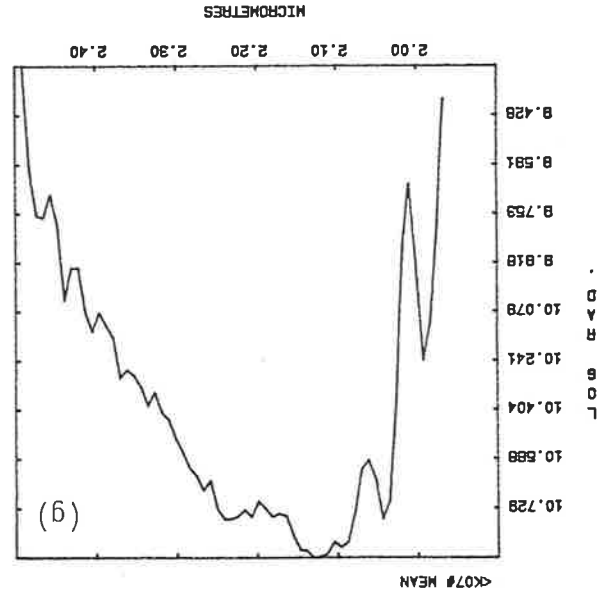
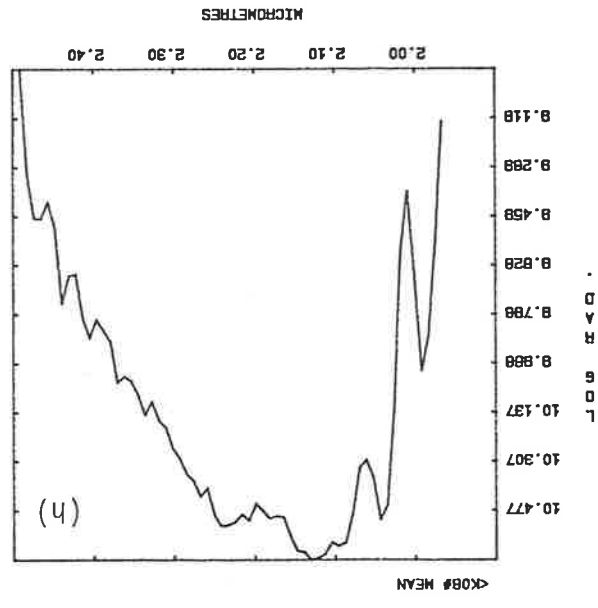
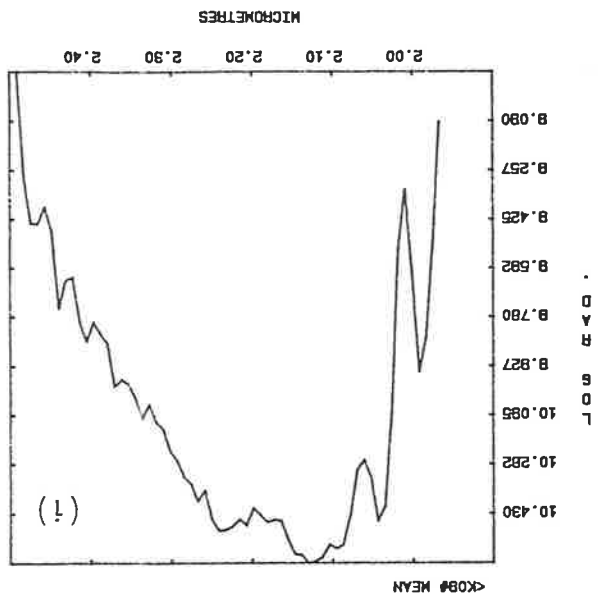
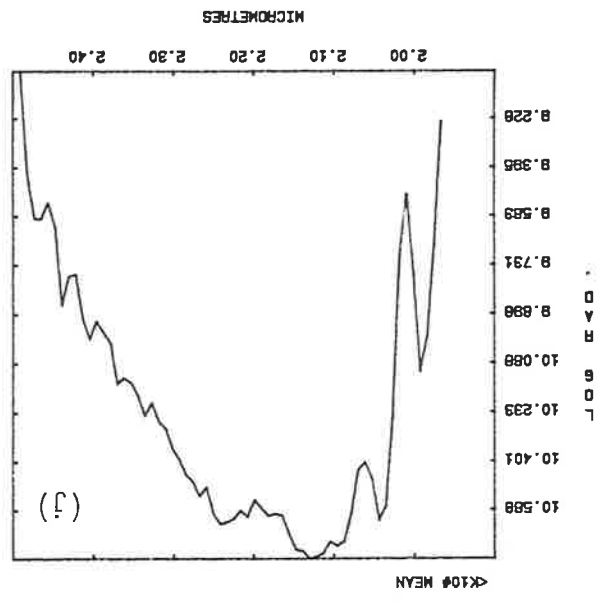
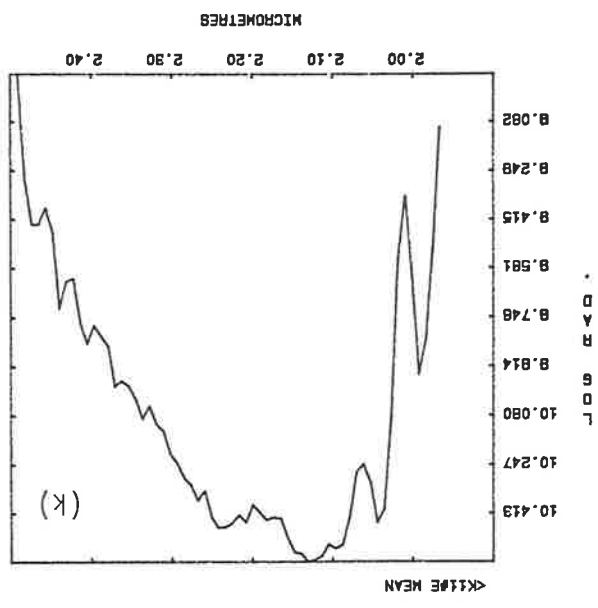
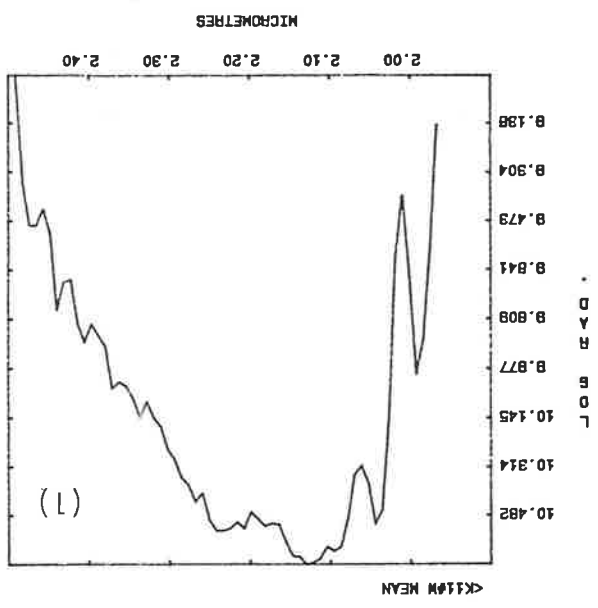


Figures 6.13a-f SWIR airborne mean radiance spectra of Kambalda flight lines.

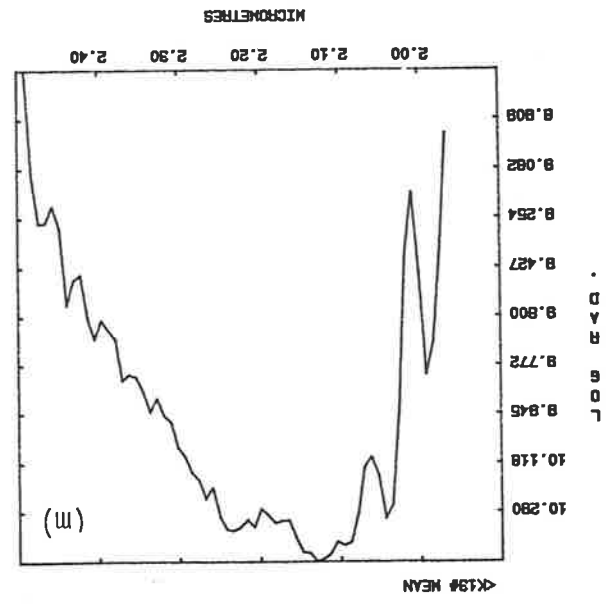
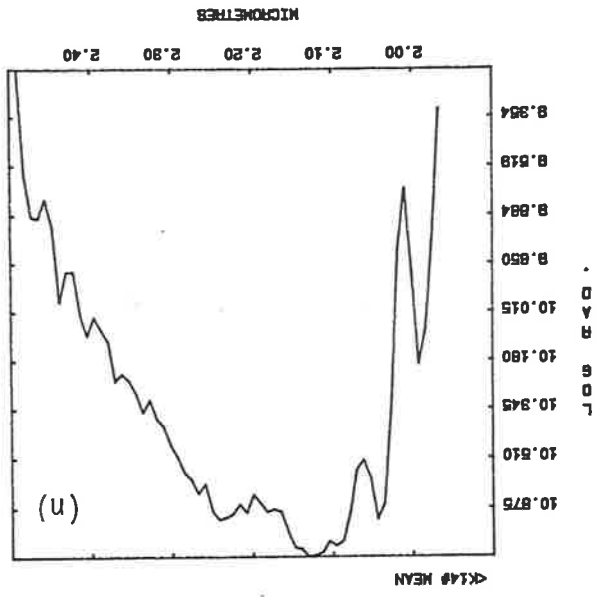
Note the similarity of all flight line means, and the absence of any recognisable mineralogical absorption features (in particular at $2.3\mu\text{m}$).



Figures 6.13g-1 SWIR airborne mean radiance spectra of Kambalda flight lines.



Figures 6.13m&n SWIR airborne mean radiance spectra of Kambalda flight lines.

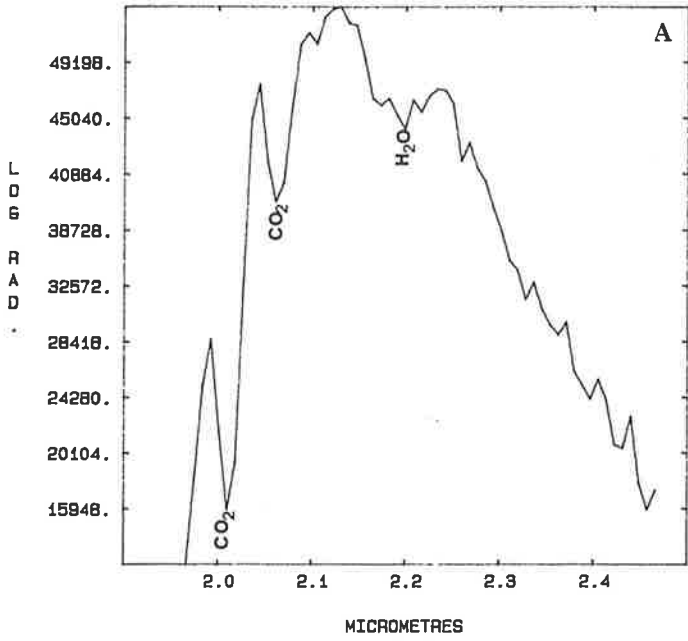


Figures 6.14a-d Two SWIR airborne radiance spectra and their corresponding log residuals from Kambalda.

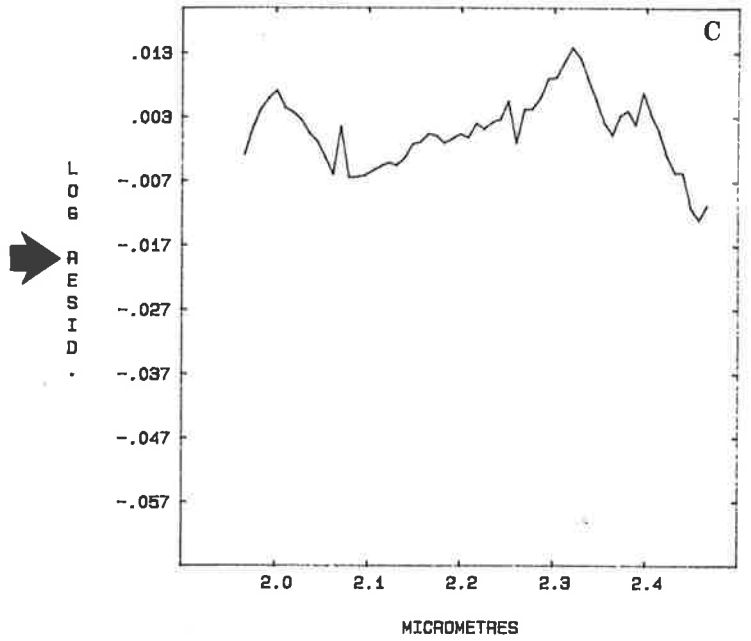
Figures 6.14a&b are averaged airborne radiance data over a spectrally flat area, and a talc-bearing area, respectively. Note the apparent absence of mineralogical features.

Figures 6.14c&d are the equivalent averaged log residual spectra. The spectrally flat area has no obvious absorption features, but the talc-bearing zone exhibits a strong talc log-residual signature.

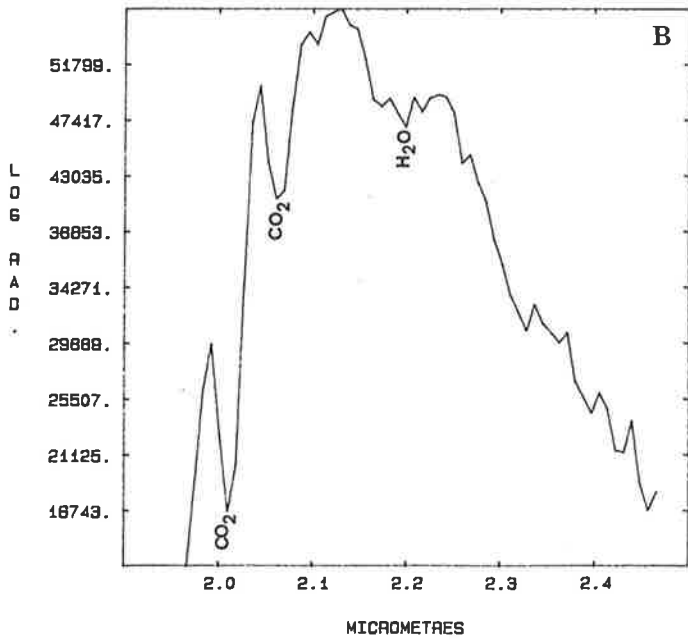
K2 108-225 RADIANCE



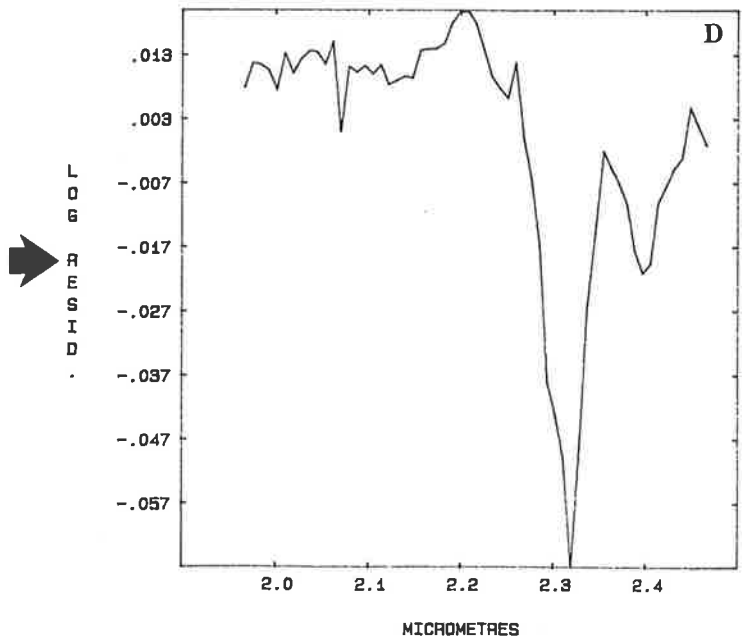
K2 108-225 LOG RESIDUAL



K2 458-495 RADIANCE



K2 458-495 LOG RESIDUAL



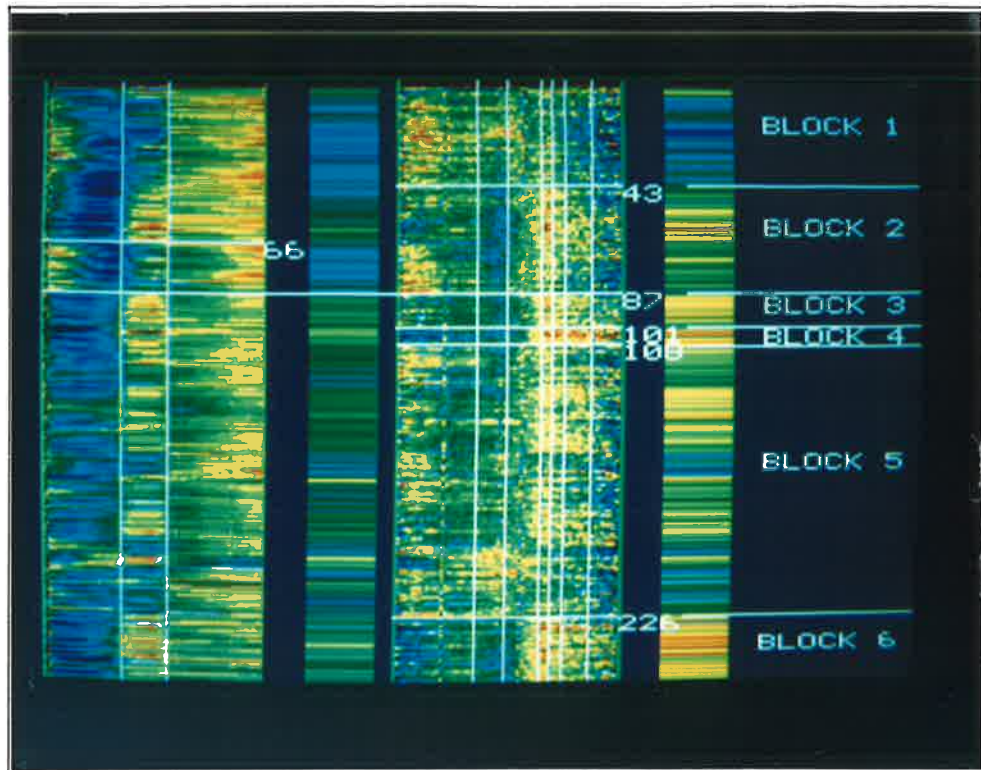
Figures 6.15a&b Airborne log residual spectra (line K2) - image format.

The log residual data have been put into an image format and density sliced. Dark blue colours represent the lowest log residual values (deepest absorption features), the mid-green colours (see vertical strips at extremities of some of the columns of data) represent zero values, and deep red colours represent the highest log residual values. The four columns of data are, from left to right, VNIR log residuals, VNIR albedo, SWIR log residuals and SWIR albedo.

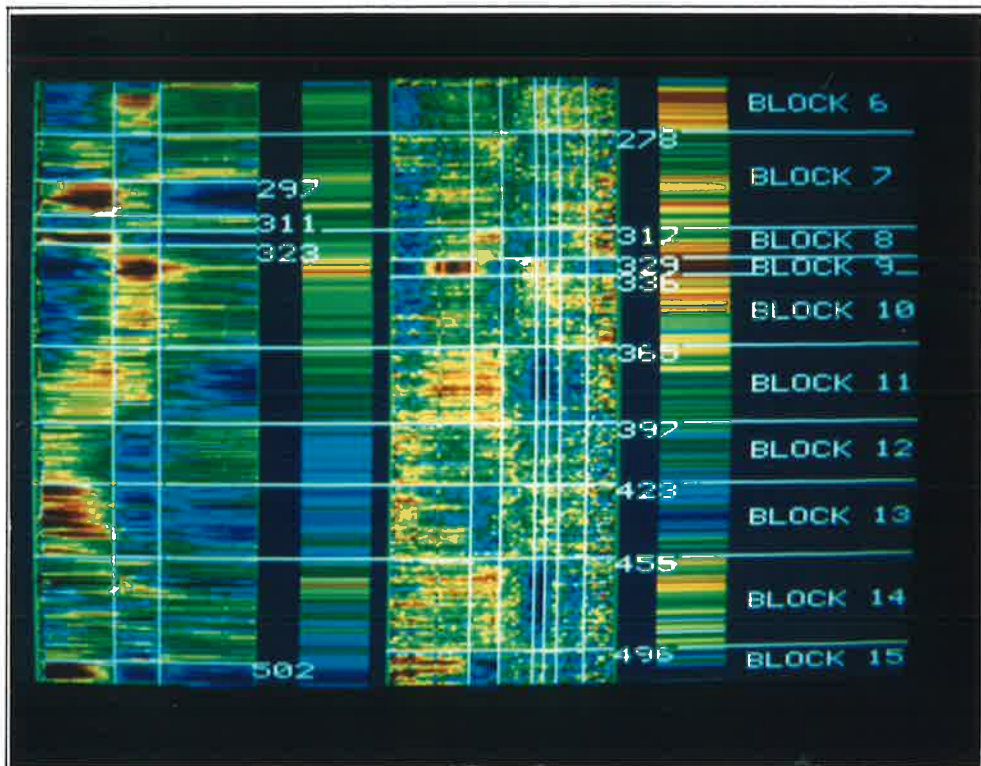
The block boundaries (horizontal lines) have been drawn after visual examination of the image. Each block has been numbered, and average log residual spectra for each block have been plotted and interpreted (see Appendix IX).

The data fall naturally into a number of 'bands' defined by the wavelength limits of the absorption features. The vertical lines outline these bands. Wavelengths of the band boundaries are ;

VNIR	1	0.327 μ m (start)
	2	0.635 μ m
	3	0.784 μ m
	4	1.012 μ m (end)
SWIR	5	1.954 μ m (start)
	6	2.147 μ m
	7	2.216 μ m
	8	2.302 μ m
	9	2.328 μ m
	10	2.353 μ m
	11	2.422 μ m
	12	2.500 μ m (end)



A

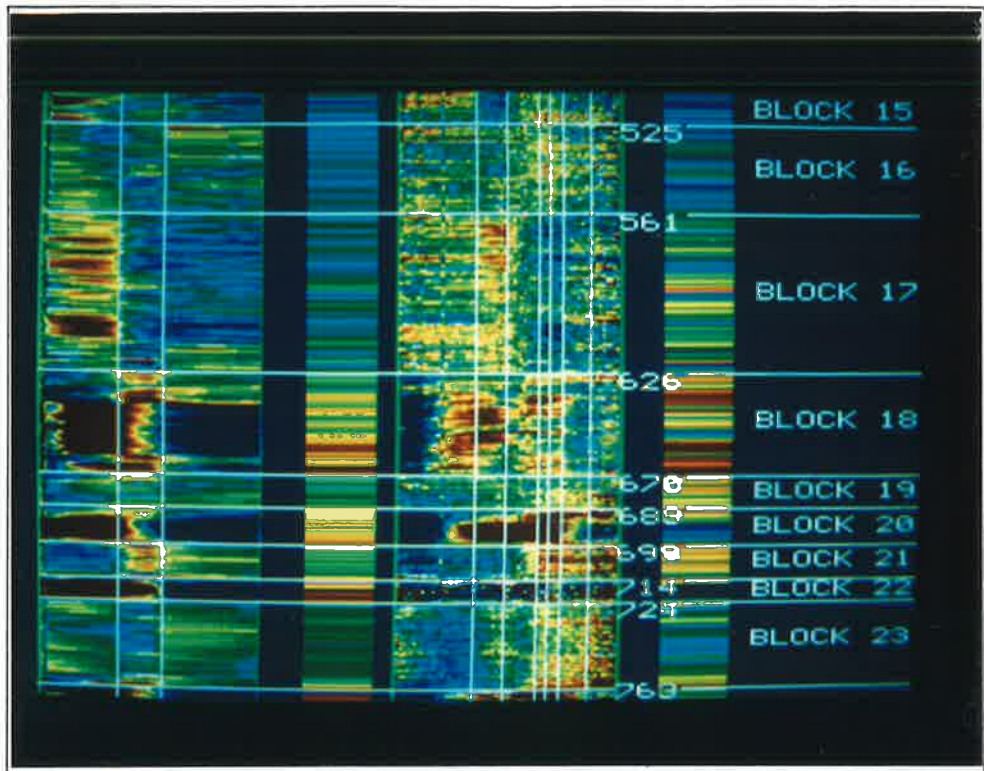


B

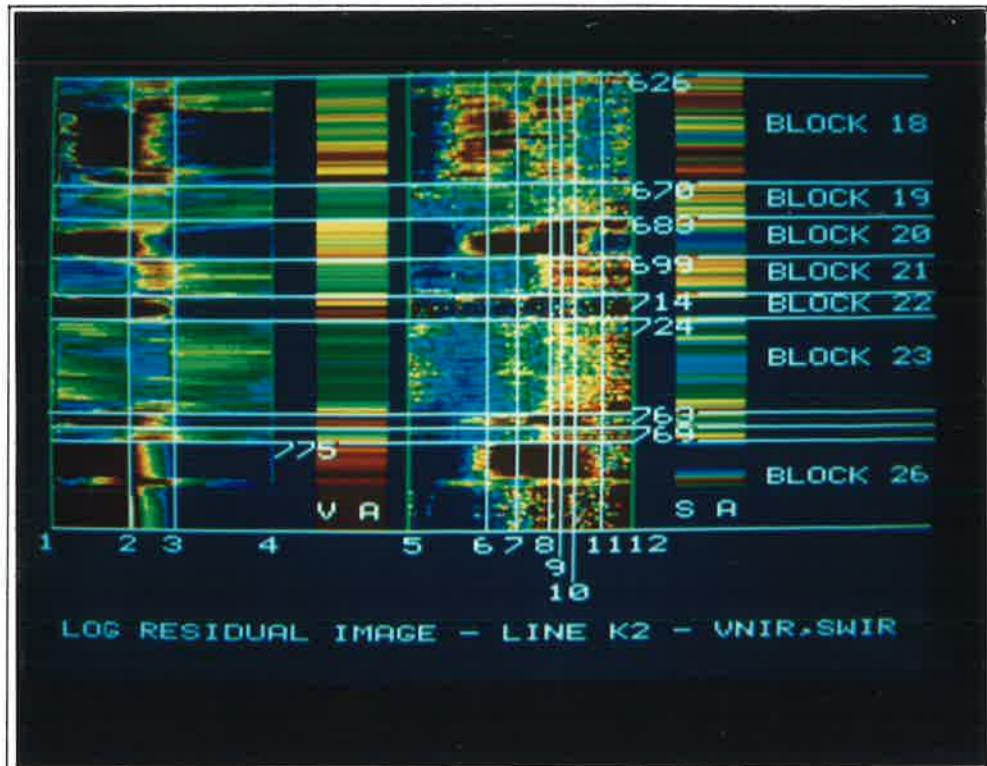
Figures 6.15c&d Airborne log residual spectra (line K2) - image format.

Blocks 15 to 26.

C



D

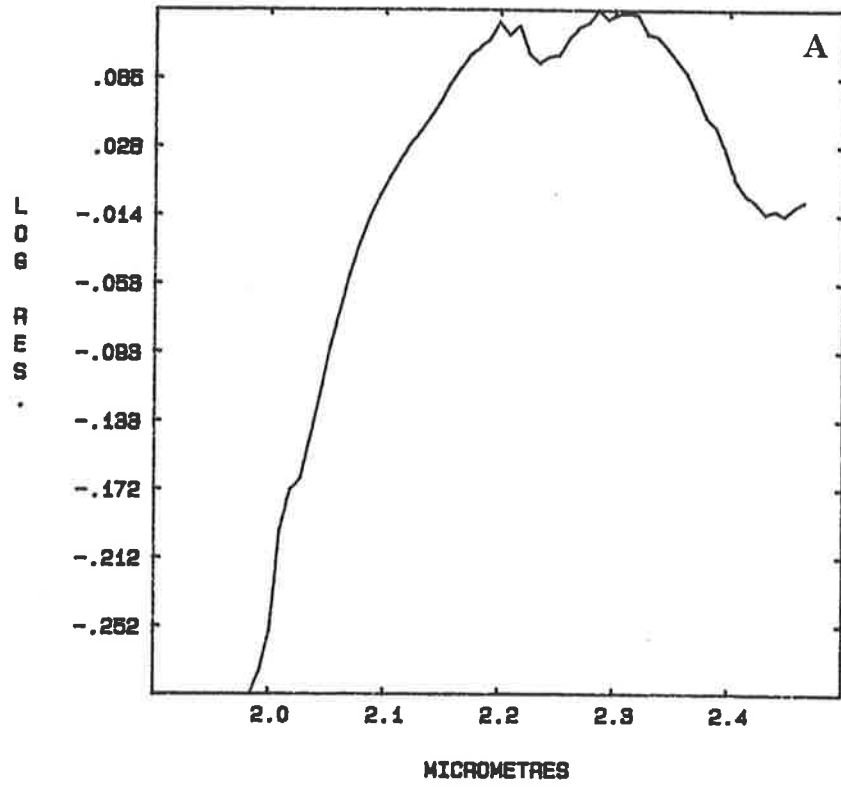


Figures 6.16a&b SWIR laboratory reflectance spectrum and airborne log residual spectrum of halite from Kambalda.

Figure 6.16a is the average airborne log residual spectrum over a portion of Lake Lefroy. It exhibits a weak local minimum at $2.25\mu\text{m}$ (similar to a feature documented by Hunt et al., 1972), which may be related to the moisture content of the halite.

Figure 6.16b is the laboratory spectrum of a (dry) halite sample from Lake Lefroy. It has no absorption features near $2.25\mu\text{m}$.

SA. 884-888



SWIR BIDIRECTIONAL REFLECTANCE

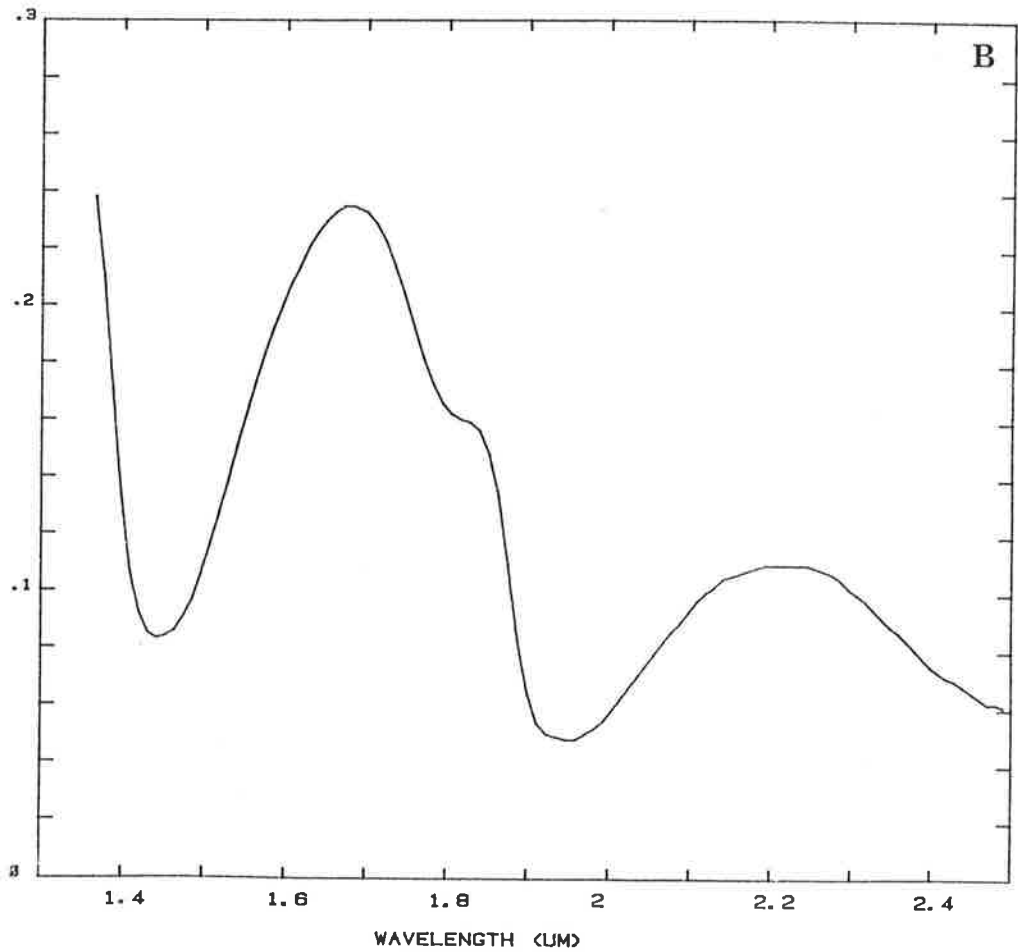


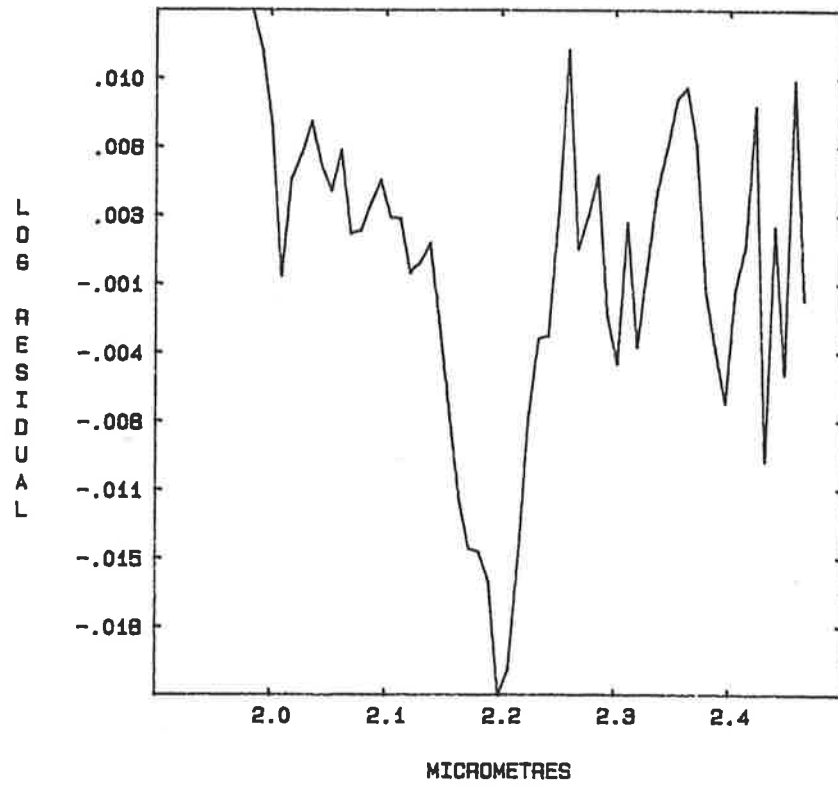
Figure 6.17 SWIR airborne log residual spectrum of a kaolinitic laterite at Kambalda.

Although somewhat noisy, the characteristic 2.17 and 2.208 μm absorption minima of kaolin are well developed in this spectrum.

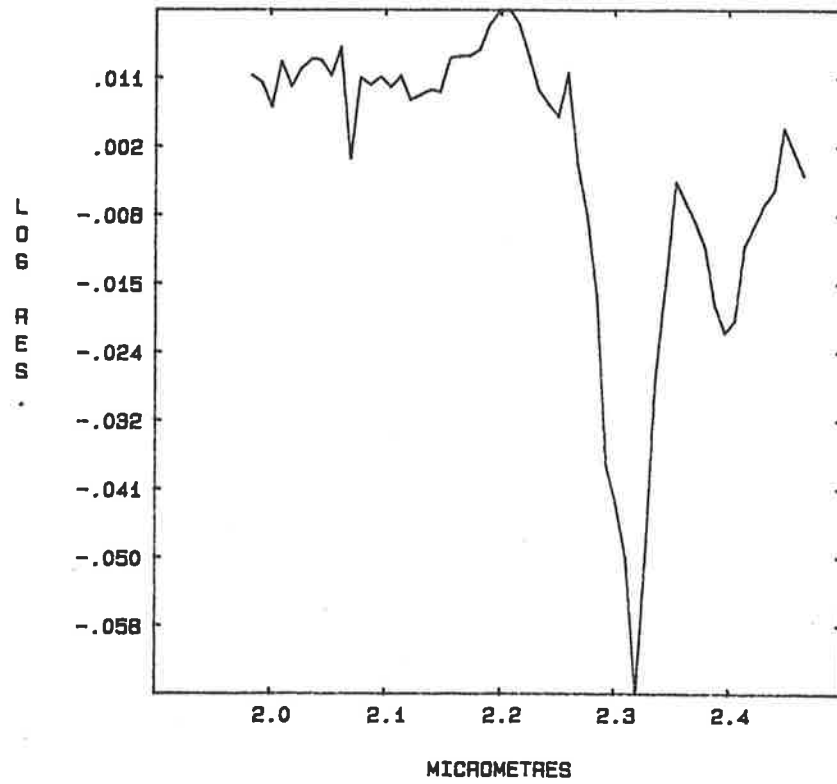
Figure 6.18 Airborne SWIR log residual spectrum of a talc-bearing soil at Kambalda.

This spectrum exhibits all of the characteristic features of talc, with minor features at 2.26, 2.30 (shoulder), 2.39 and 2.47 μm , and with the sharp and deep major absorption feature at 2.32 μm . Compare to the 2.2 μm absorption of kaolin in Fig. 6.17 above.

K5 PIXELS 283-298



K2 PIXELS 458-485



Figures 6.19a&b Photograph and airborne log residual spectrum of a talc-bearing portion of line K9.

Figure 6.19a shows the absence of outcrop or substantial float. Note also that soil lichen (right side of photo) and saltbush (just visible top and left) will contribute to the airborne spectrum. The author estimates that no more than 50% of each pixel in this area is composed of bare, talc-rich soil. XRD analysis shows that the talc content of the bare areas is approximately 20%. Therefore, it can be inferred that an effective talc concentration of only 10% has produced the log residual spectrum shown in Fig. 6.19b.

A

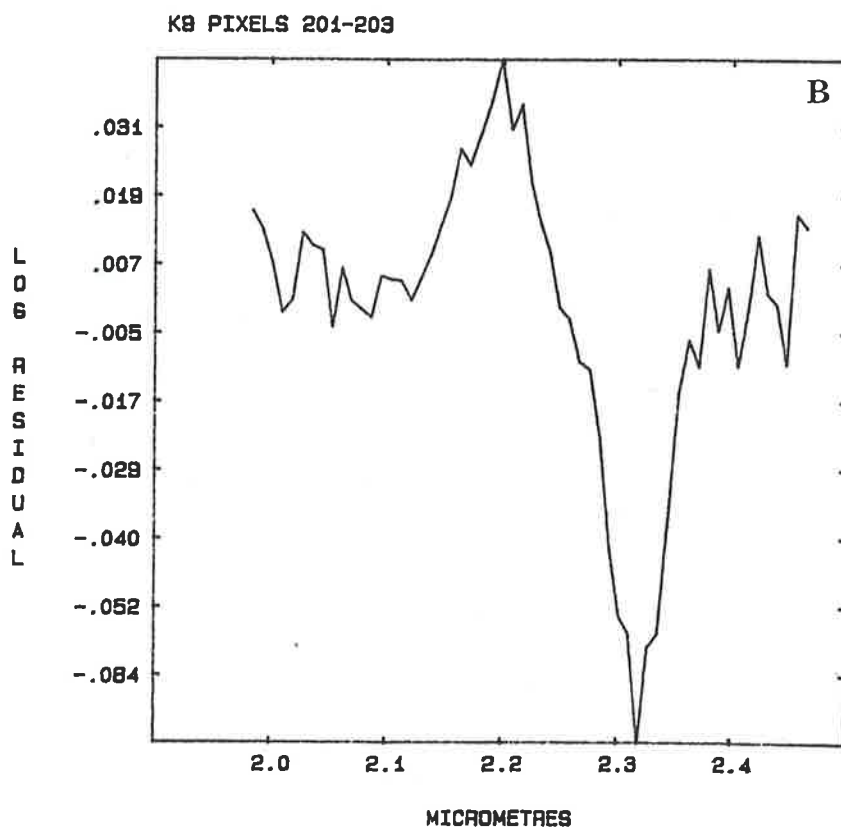


Figure 6.20 Interpreted and factual geology of the St Ives area,
with overlay showing mineralogy interpreted from
airborne spectral data.

KEY

- Kaolin
- Chlorite
- Chlorite/amphibole
- ▲ Amphibole/epidote
- △ Amphibole/chlorite
- ☆ Talc
- ★ Talc/chlorite
- ? Uncertain

35mm photo centres are marked for each flight line, and every hundredth pixel is numbered.

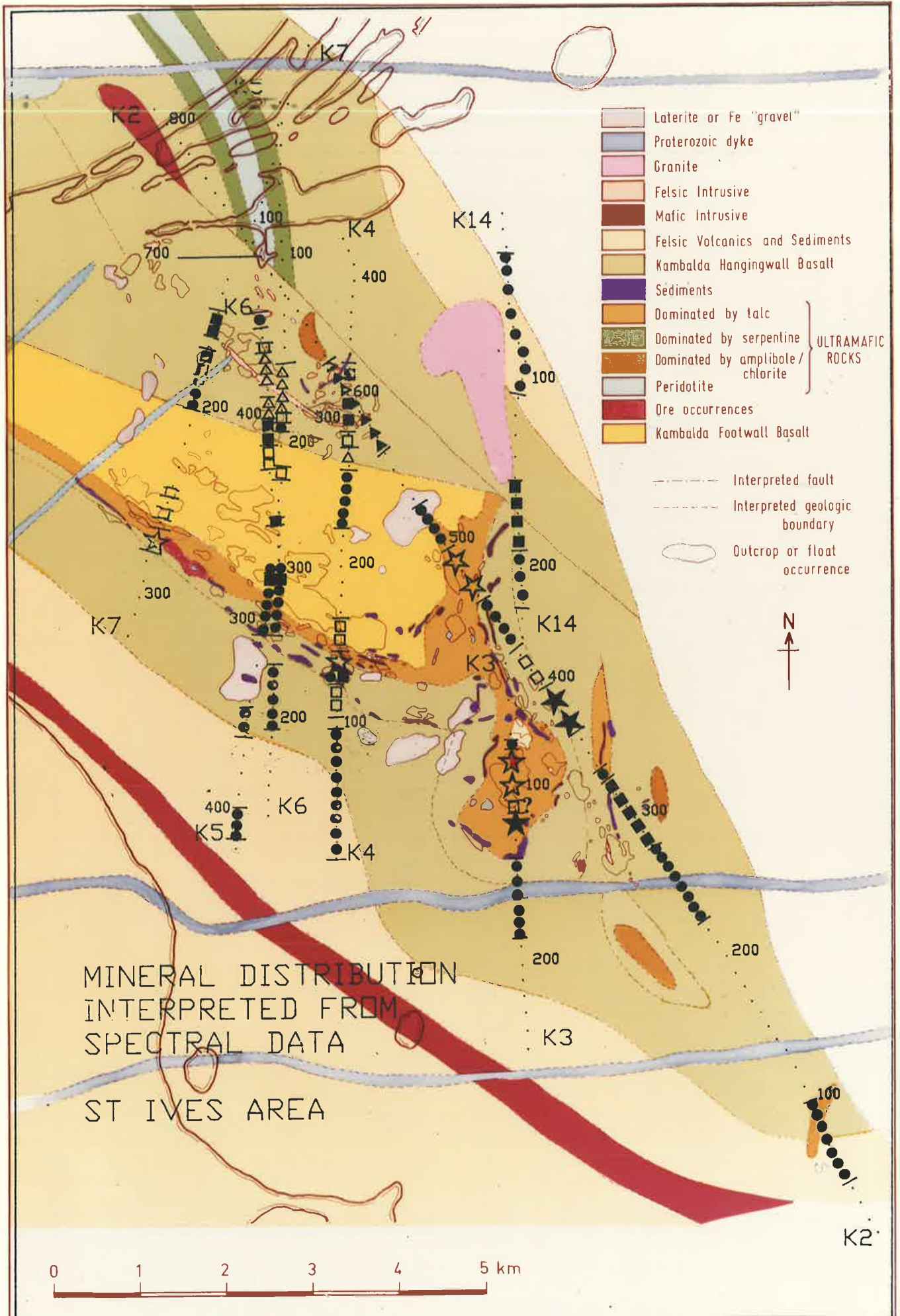
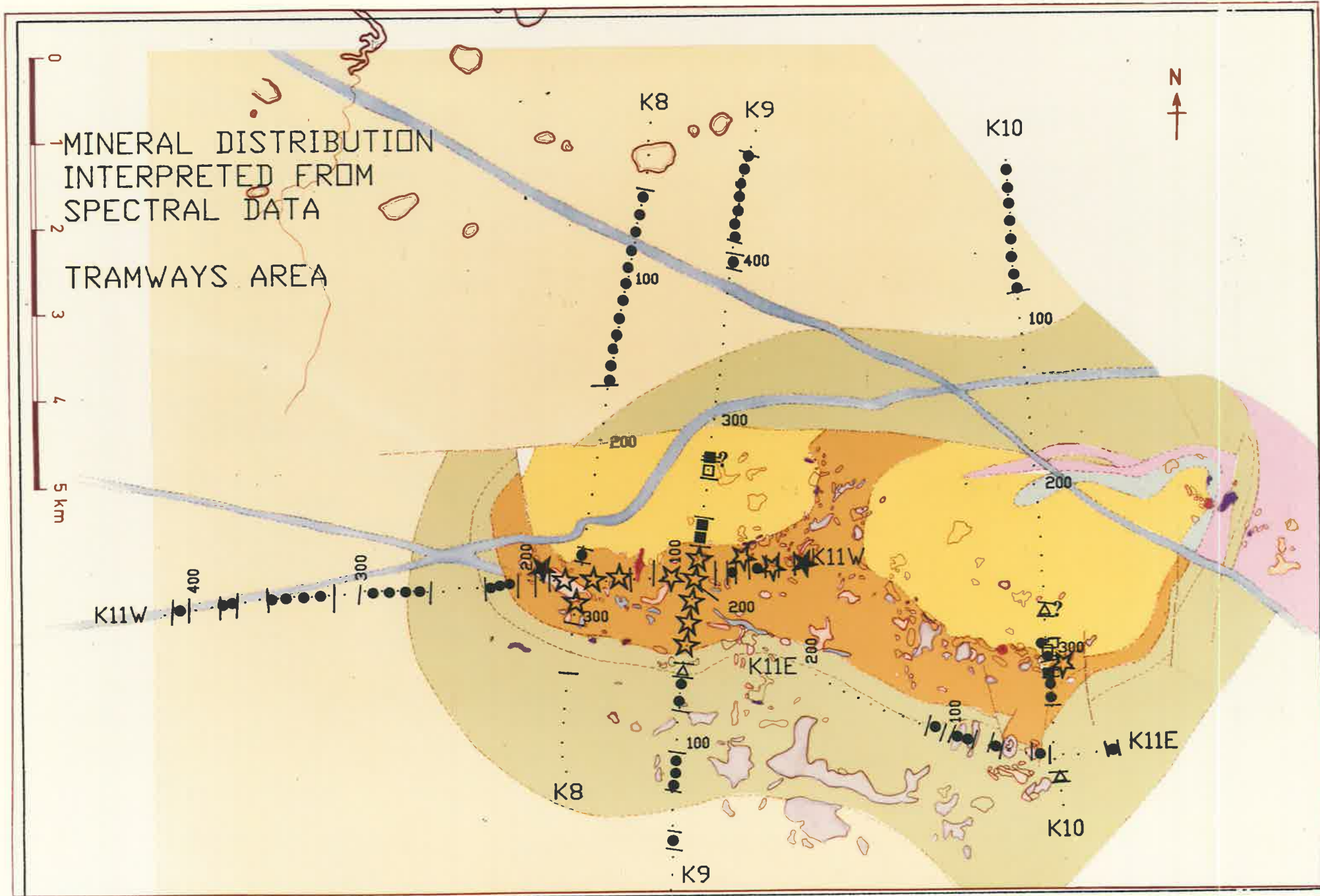


Figure 6.21 Interpreted and factual geology of the Tramways area,
with overlay showing mineralogy interpreted from
airborne spectral data.

KEY

- Kaolin
- Chlorite
- Chlorite/amphibole
- ▲ Amphibole/epidote
- △ Amphibole/chlorite
- ☆ Talc
- ★ Talc/chlorite
- ? Uncertain

35mm photo centres are marked for each flight line, and every hundredth pixel is numbered.



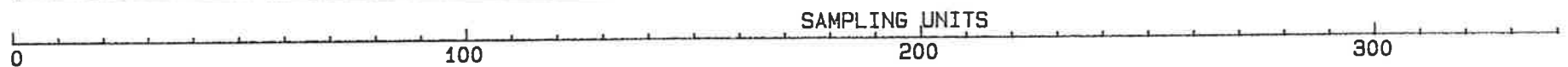
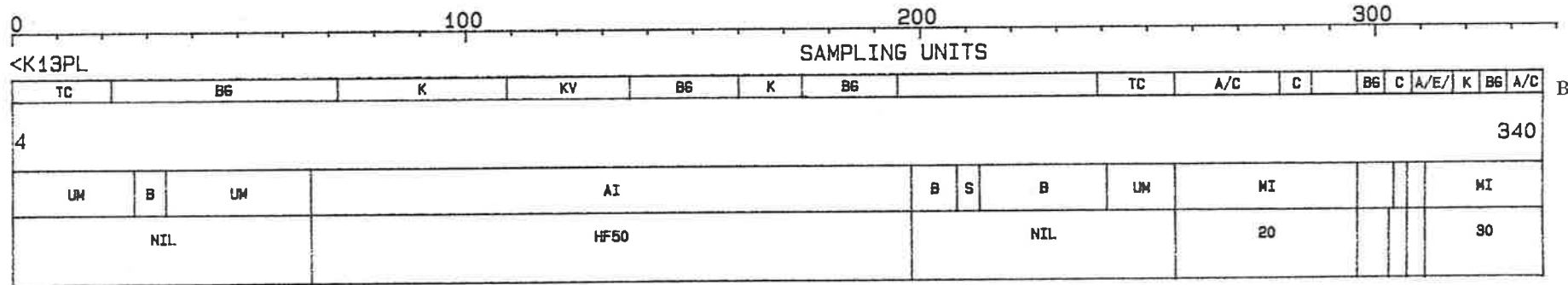
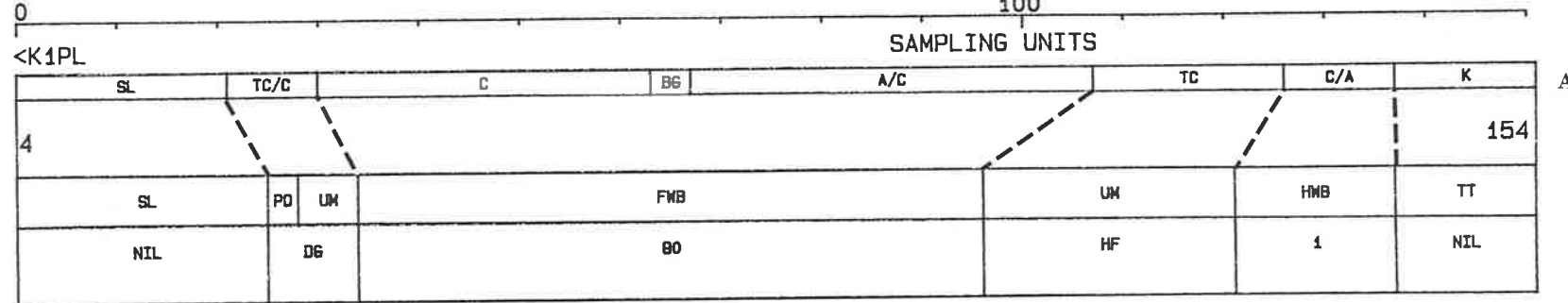
Figures 6.22a&b Geology and interpretation of mineralogy from airborne data for the Kambalda Dome and Democrat areas.

Figure 6.22a is the Kambalda Dome interpretation, and Fig. 6.22b that for Democrat. For both diagrams the spectral interpretation appears in the top line, mapped geology in the middle, and an estimate of the amount of outcrop or float in the bottom line.

KEY		
SPECTRA	SL	Salt Lake
	Tc	Talc
	C	Chlorite
	A	Amphibole
	E	Epidote
	K	Kaolin
	KV	Combined Kaolin/Vegetation
GEOLOGY	BG	Background
	PO	Porphyry (felsic)
	UM	Ultramafic
	FWB	Footwall Basalt
	HWB	Hangingwall Basalt
	AI	Acid Intrusive
	B	Basalt
	MI	Mafic Intrusive
OUTCROP	TT	Trotting Track
	DG	Disturbed Ground
	HF	Heavy Float
	90	% float

EAST

WEST



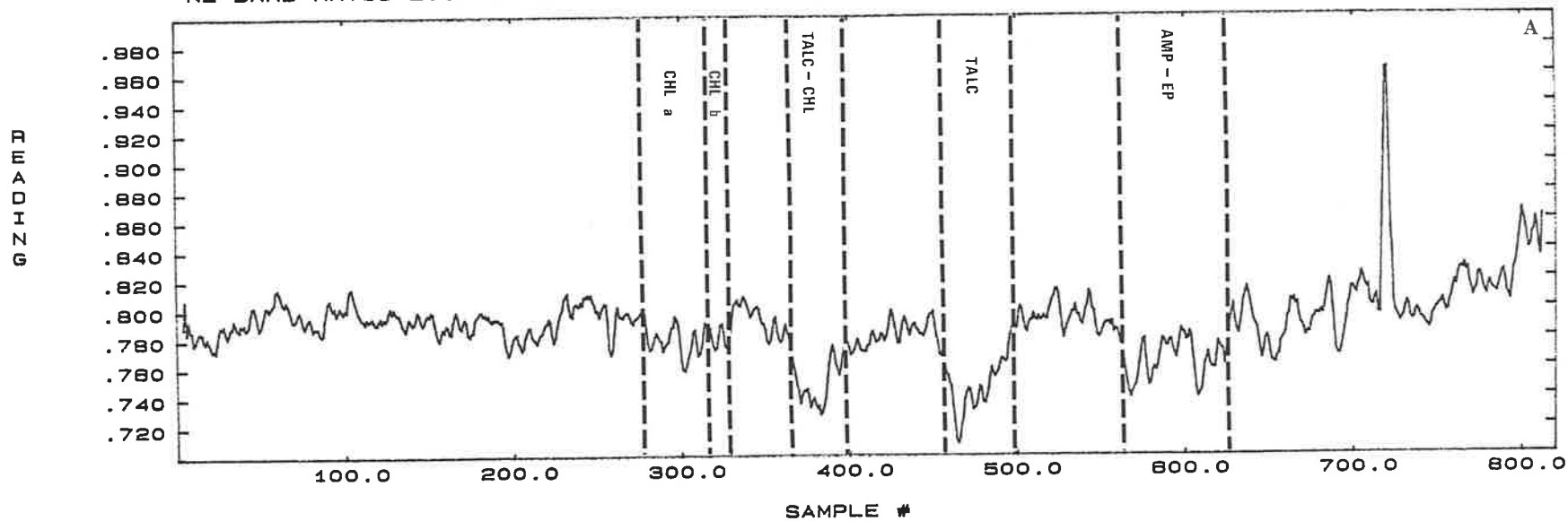
Figures 6.23a&b Variation of simulated narrow-band ratios derived from radiance data along flight line K2.

Figure 6.23a depicts the variation of a simulated narrow-band ratio (2.3018 to $2.3276\mu\text{m}/2.2158$ to $2.3018\mu\text{m}$) derived from radiance data.

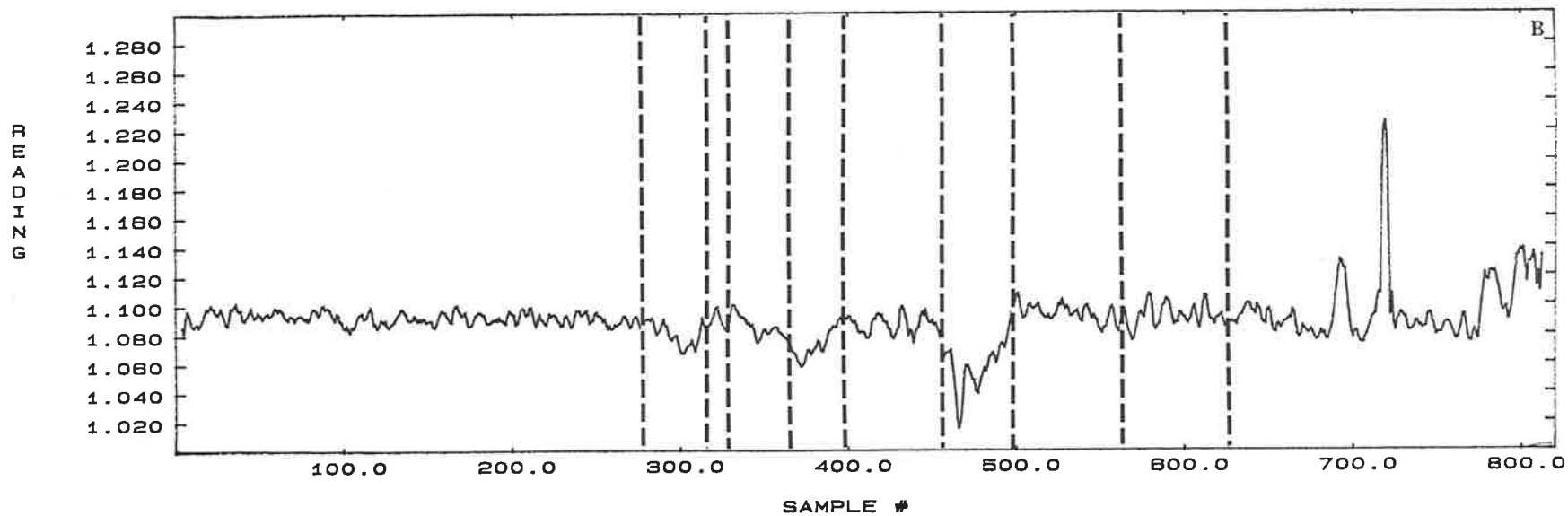
Figure 6.23b depicts a similar, but narrower ratio (2.3018 to $2.3276\mu\text{m}/2.3276$ to $2.3534\mu\text{m}$).

The areas within vertical dashed lines have been interpreted (from second-order effects) as containing the minerals shown.

K2 BAND RATIO 2.3018-2.3276UM / 2.2158-2.3018UM



K2 BAND RATIO 2.3018-2.3276UM / 2.3276-2.3534UM SMX7

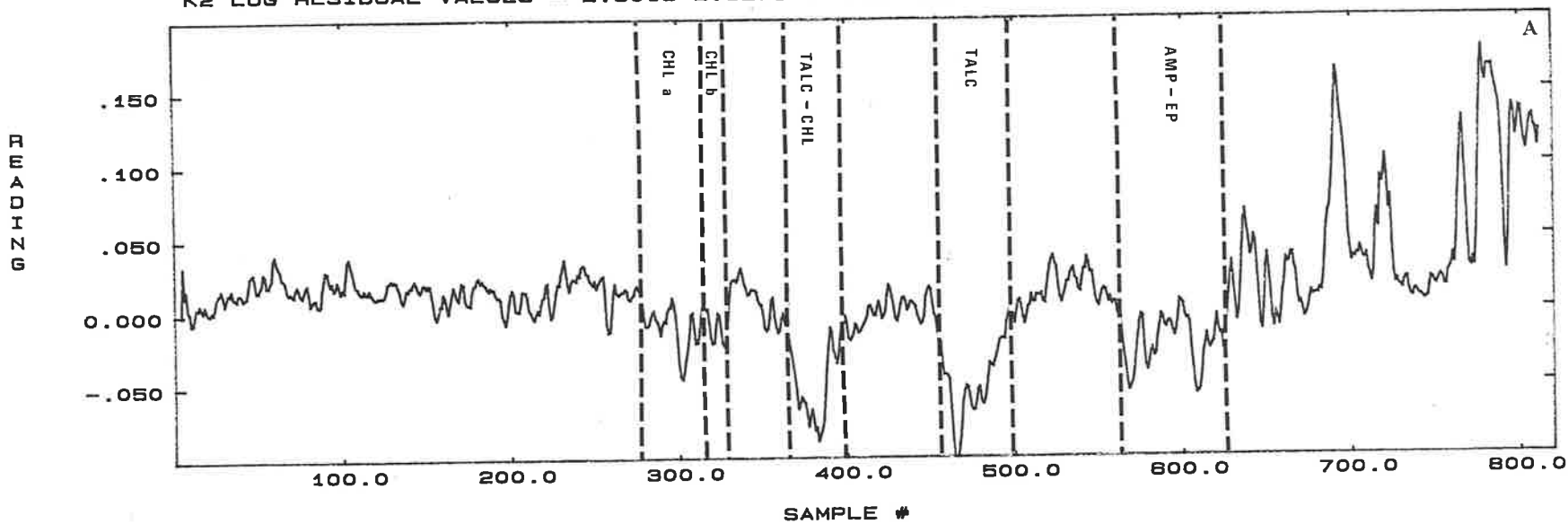


Figures 6.24a&b Variation of a simulated narrow-band derived from log residual data, and of a simulated broad-band (TM) ratio, along flight line K2.

Figure 6.24a plots the value of simulated narrow-band reflectance data (2.3018 to 2.3276 μm) derived from log residual values.

Figure 6.24b has been created from radiance data using two broad bands - 2.08 to 2.35 μm (equivalent to TM band) and 1.958 to 2.08 μm . The bands have then been ratioed to produce a result similar to the 1.6 μm /2.2 μm ratio commonly used in analysis of TM data. Note the lack of well-defined minima in the critical areas.

K2 LOG RESIDUAL VALUES - 2.3018-2.3278UM BAND SMX7



BROAD-BAND (TM) RATIO SIMULATION K2

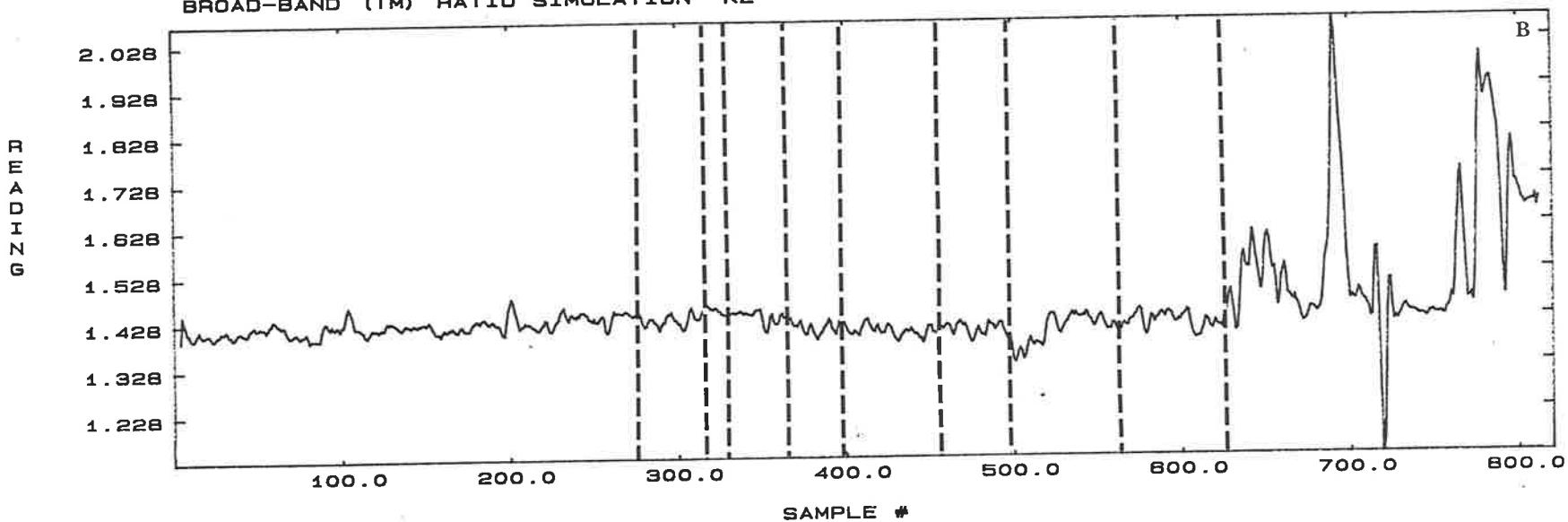


Figure 7.1 Geologic map of the Hamersley Basin showing the distribution of its major components (modified after Smith et al., 1982).

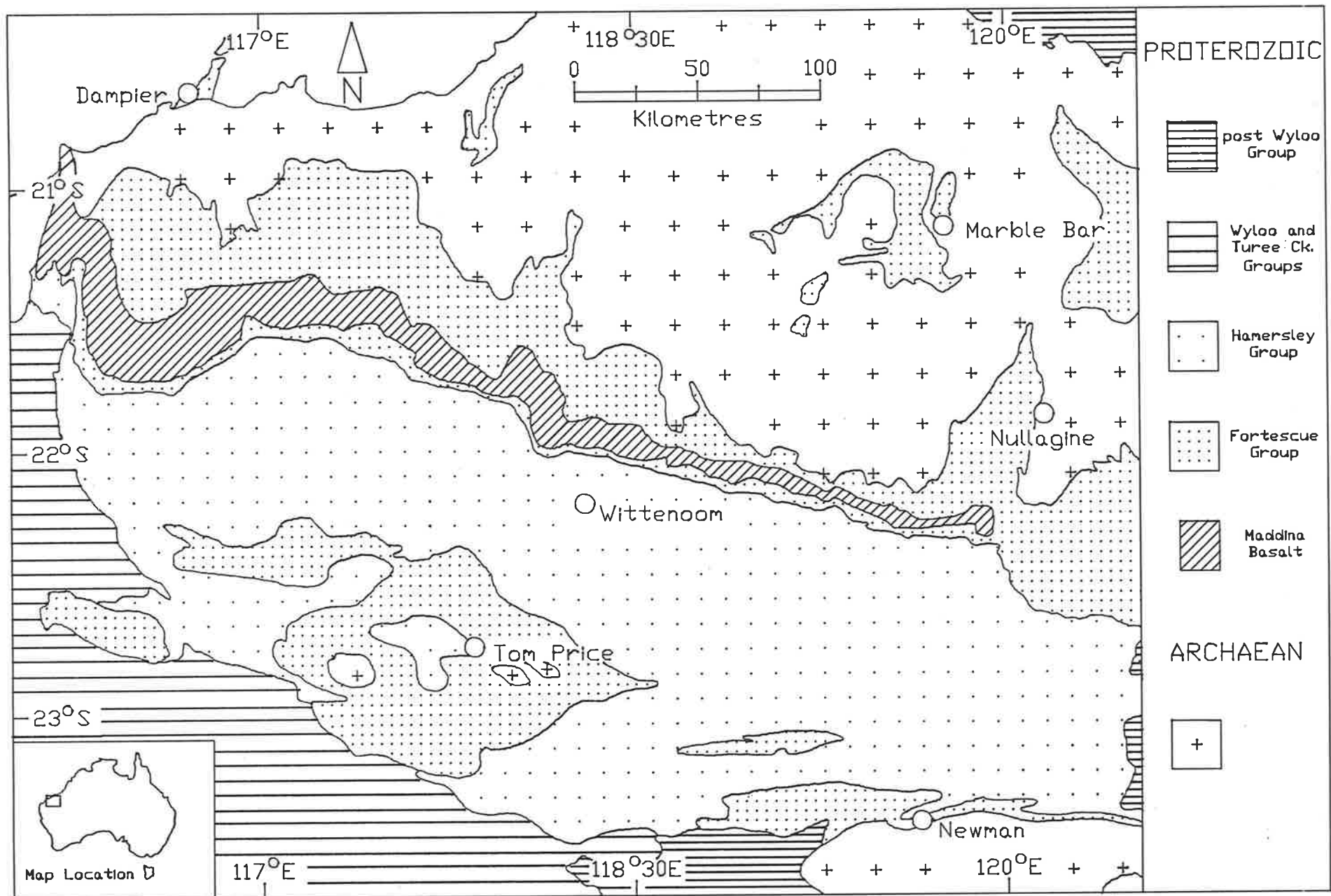


Figure 7.2 Typical profile through hydrothermally altered basic lava (after Smith et al., 1978).

Flows average about 50m in thickness. The altered flow-top lithologies are dominantly amygdaloidal, greenish-grey, and heterogeneous on an outcrop scale. Fresh material from the central layers is grey, fine-grained and uniform.

Figure 7.3 Plot of total Fe, as FeO, for two metasomatically altered lava flows in the Hamersley Basin (after Smith et al., 1982).

Analyses of samples from massive portions of flows C and E form a cluster of points of relatively uniform composition. Flow-top compositions vary widely, interpreted as being due to metasomatic alteration.

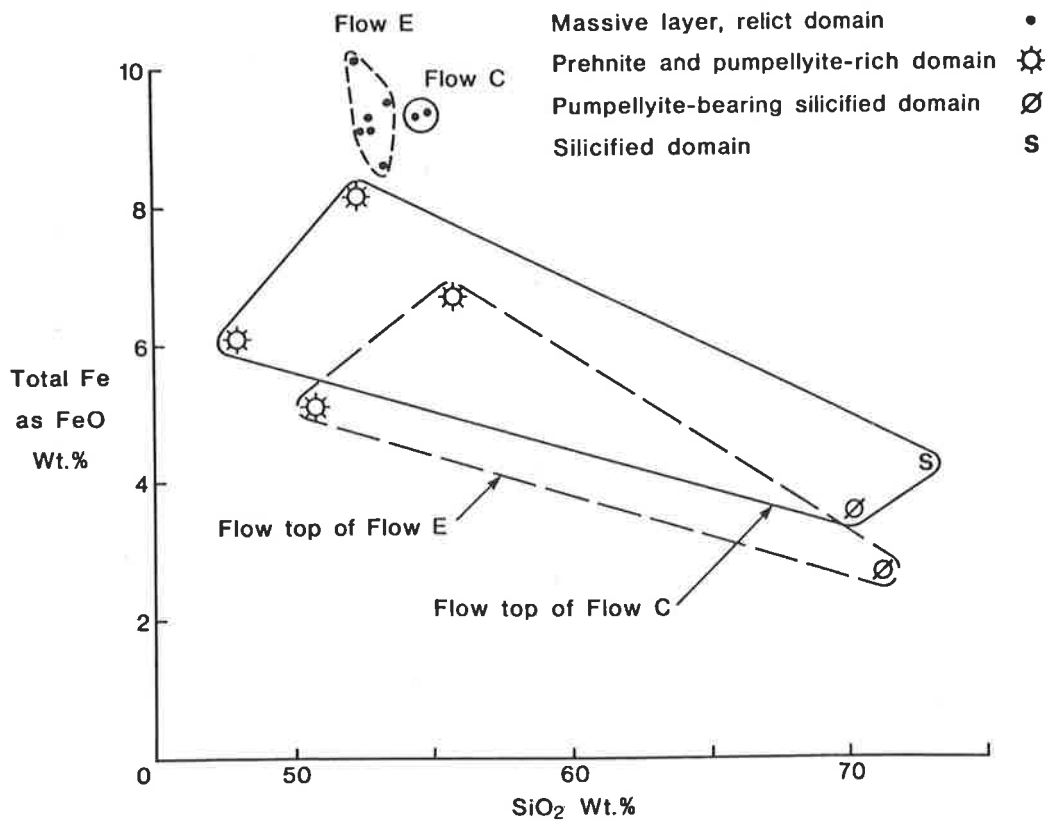
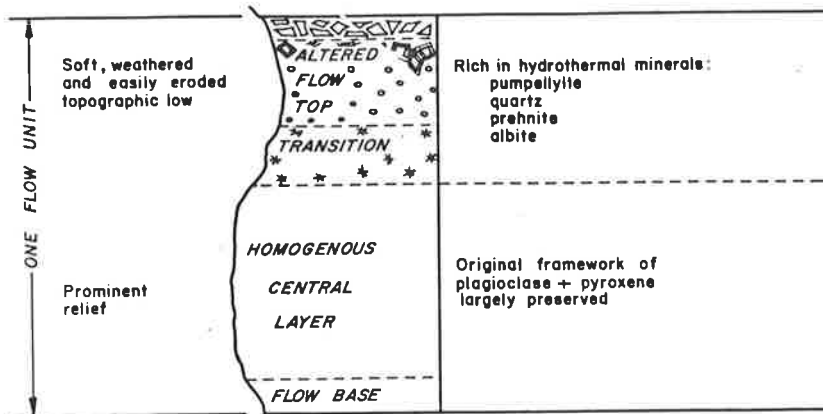
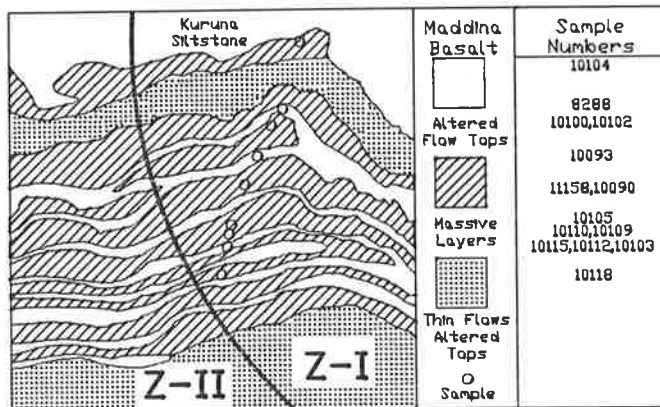
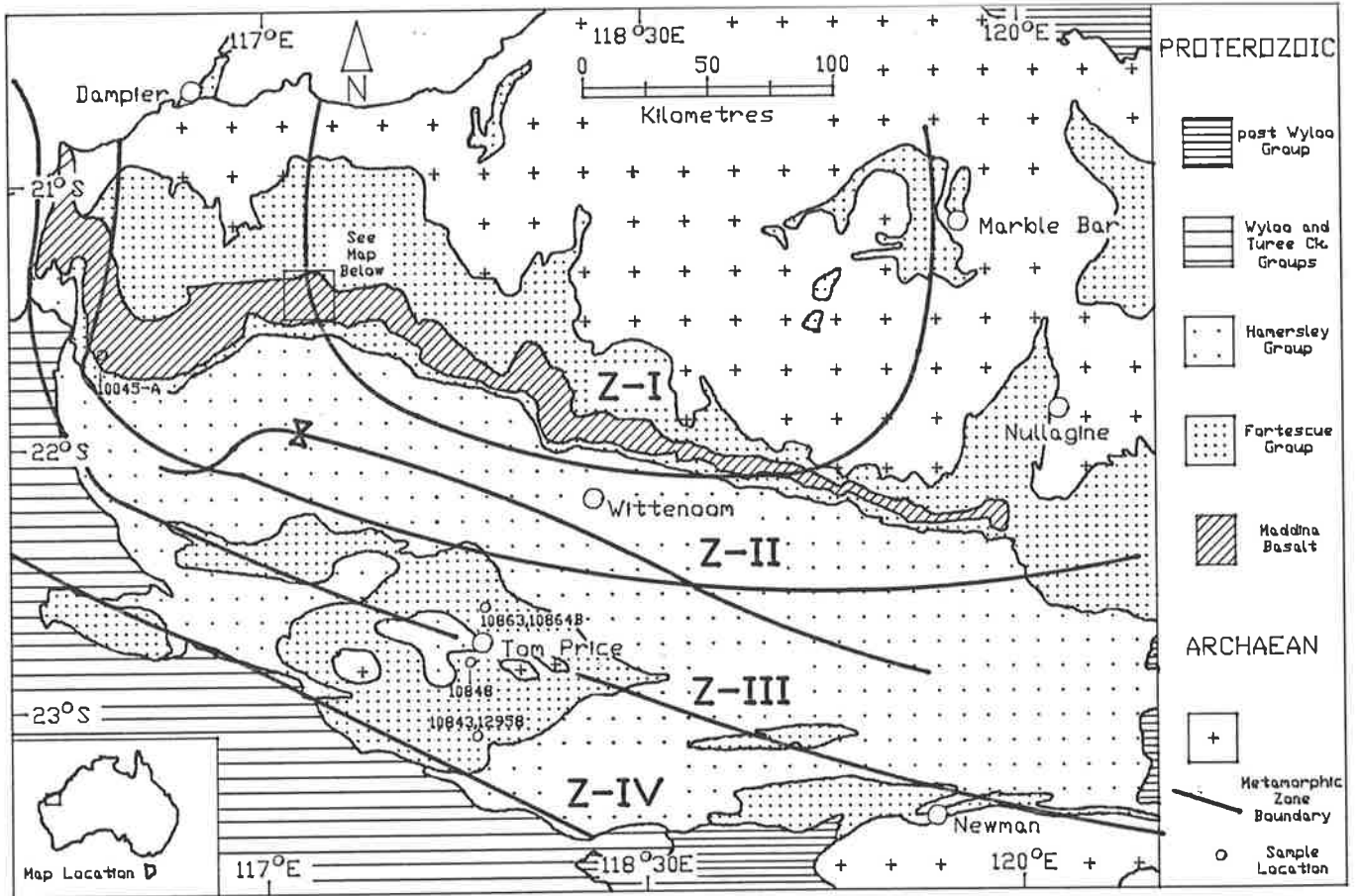


Figure 7.4 Metamorphic zone boundaries and sample locations
(modified after Smith et al., 1982).

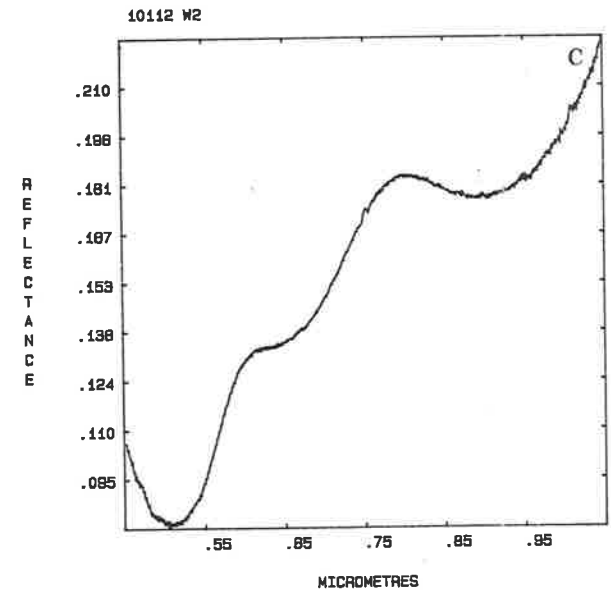
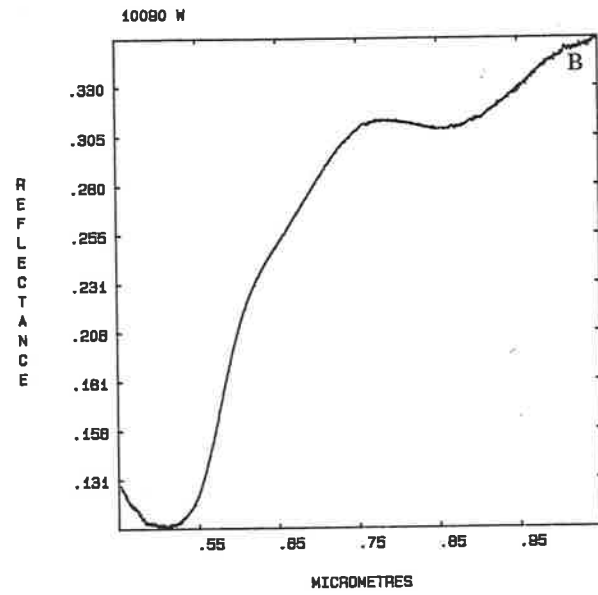
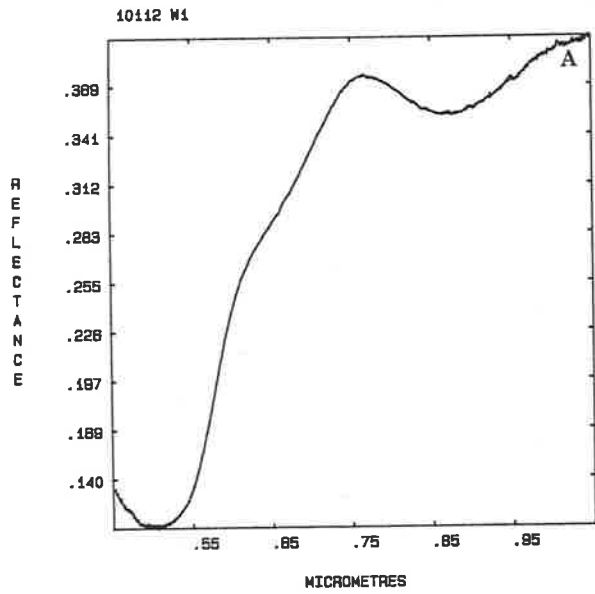
Z-I samples come from the Maddina Basalt (see magnified inset).
Other samples are from scattered locations, but mainly in the vicinity
of Tom Price.

METAMORPHIC ZONES AND SAMPLE LOCATIONS



Figures 7.5a-c VNIR laboratory reflectance spectra of typical samples from Z-I of the Fortescue Group - weathered surfaces.

Most weathered surface spectra of Z-I rocks have the characteristics of haematite, with a single, broad absorption feature near $.86\mu\text{m}$. Note that sample 10112 is from an altered flow top, and sample 10090 is from the massive, central portion of a flow. The single goethitic spectrum (10112W2, Fig. 7.5c) was measured on a different part of the same sample as Fig. 7.5a.

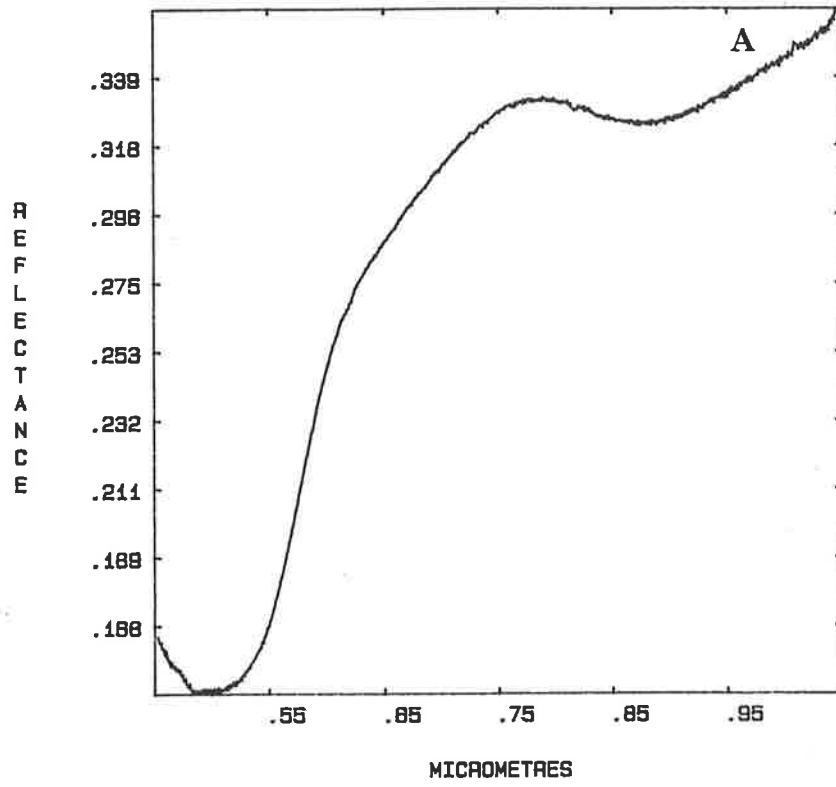


Figures 7.6a&b VNIR laboratory reflectance spectra of typical samples from higher-grade areas of the Fortescue Group - weathered surfaces.

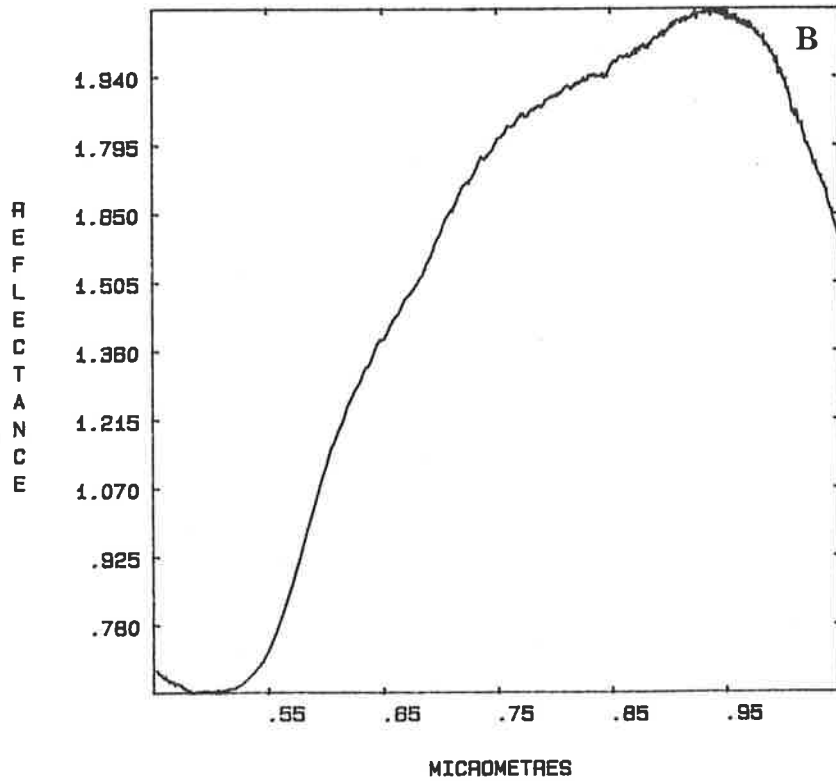
Figure 7.6a shows the spectrum of a sample from the massive part of a Z-IV flow. Note the similarity to Figs. 7.5a&b.

Figure 7.6b is the spectrum of another Z-IV sample. The lack of absorption features in this case is attributed to lower surficial iron, probably because of less severe weathering.

12958 W



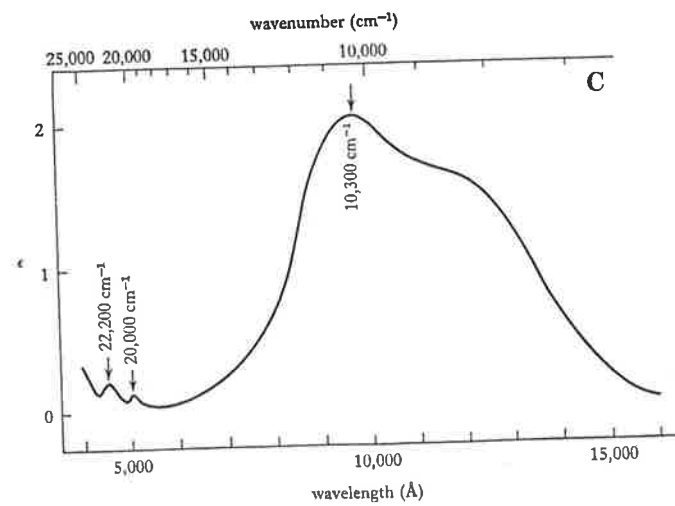
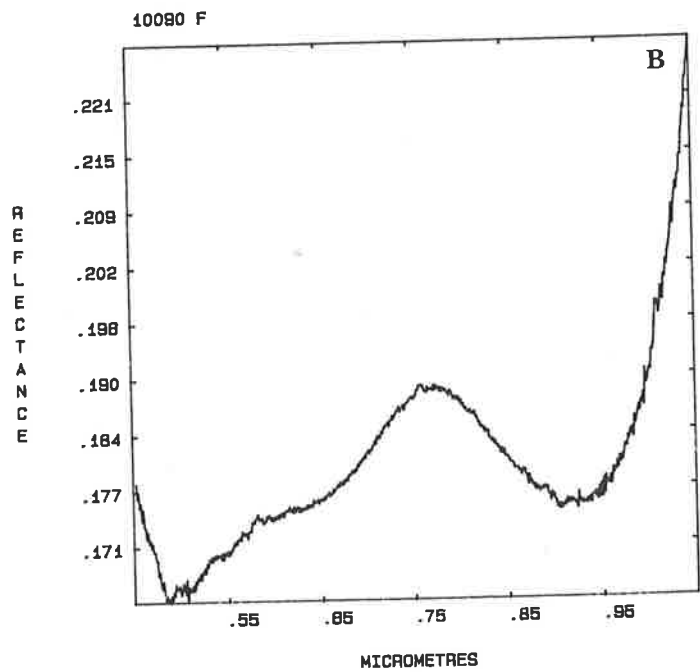
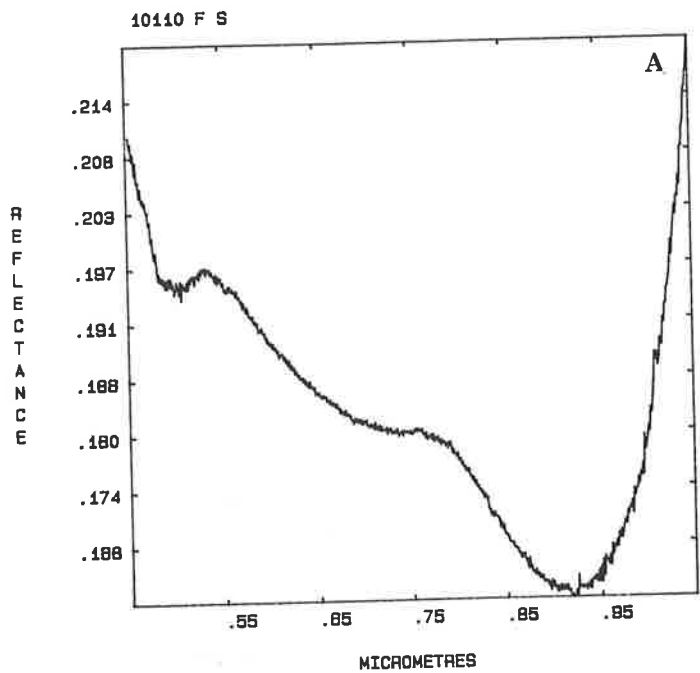
10848 W



Figures 7.7a-c VNIR laboratory reflectance spectra of typical samples from the Fortescue Group - fresh surfaces, and the spectrum of Fe^{2+} (after Burns, 1970).

Figures 7.7a&b are spectra of Z-I samples from the transition and from a massive, unaltered layer, respectively.

The absorption spectrum shown in Fig. 7.7c is of the aqueous Fe^{2+} ion in a solution of iron (II) ammonium sulphate, and is inverted relative to the reflectance spectra. Nevertheless, the same overall shape can be seen in all plots. Differences can be attributed to the crystal fields influencing absorption frequencies in the reflectance spectra.



Figures 7.8a-c VNIR laboratory reflectance spectra of pumpellyite- and epidote-bearing samples from the Fortescue Group - fresh surfaces, and the polarised absorption spectrum of epidote (after Burns, 1970).

Figures 7.8a&b are spectra of an epidote-bearing rock and a pumpellyite-bearing rock, respectively.

Figure 7.8c comprises polarised absorption spectra of epidote (dots, α spectrum, dashes β spectrum, solid line γ spectrum). Inset is the absorption spectrum of Fe^{3+} in ferric ammonium sulphate solution. Although inverted relative to the reflectance spectra, a number of similarities are evident.

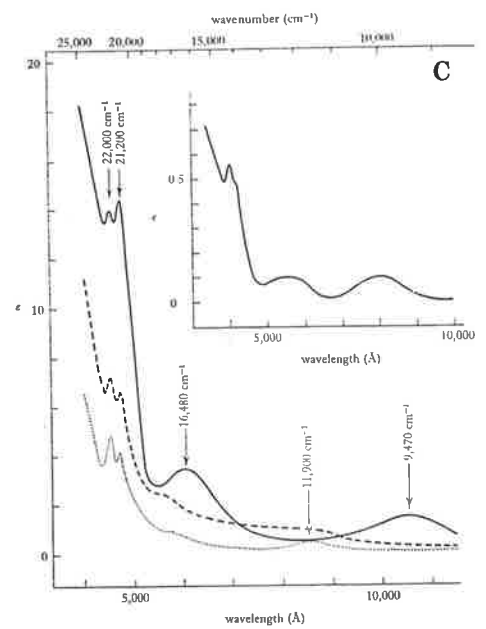
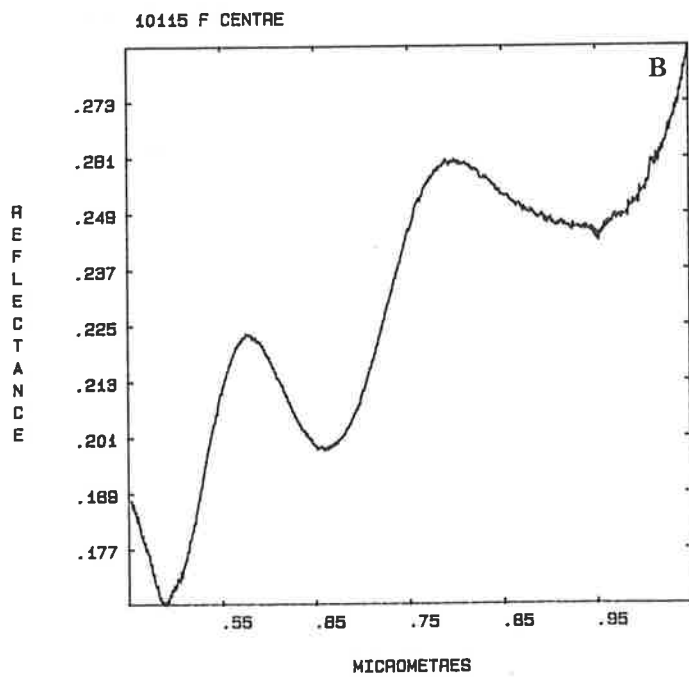
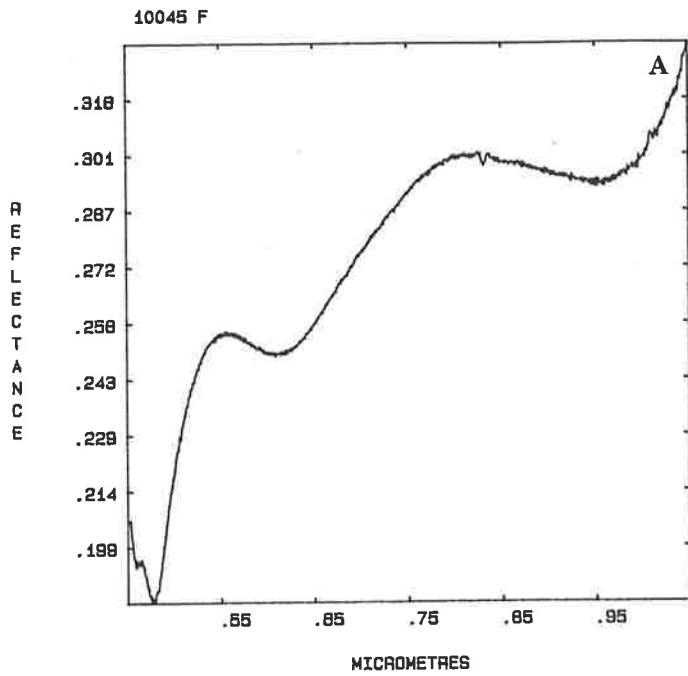
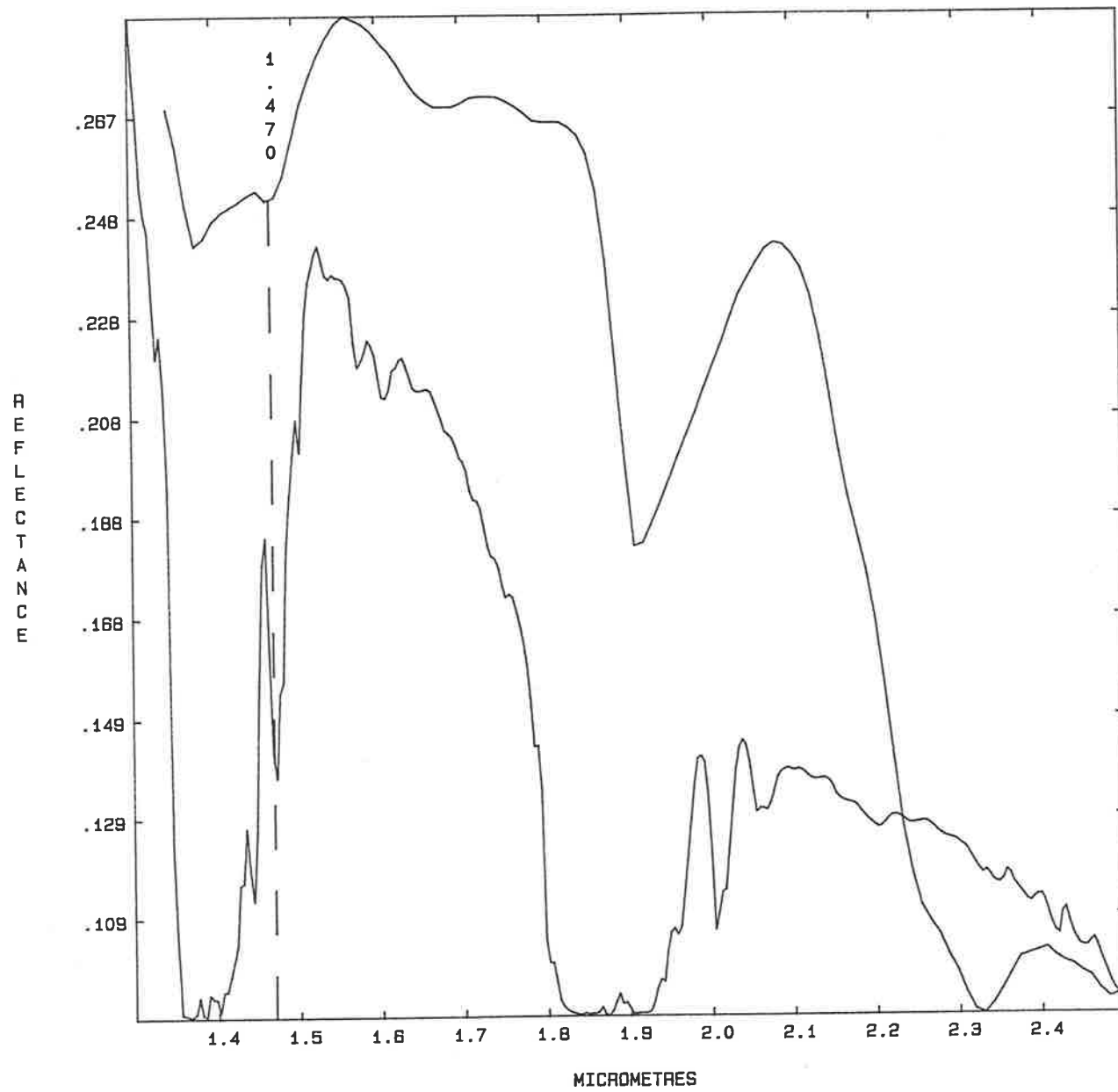


Figure 7.9 SWIR laboratory reflectance spectrum of pumpellyite compared to radiance of a 100% reflector at sea level.

The pumpellyite spectrum is shown as the dark line, while the radiance spectrum is the faint line. The radiance plot contains the theoretical atmospheric absorption features present in a spectrum measured at sea-level.

The comparison shows that all except the 1.8 μ m absorption feature should be detectable in a remote sensing mode.



PU+QTZ

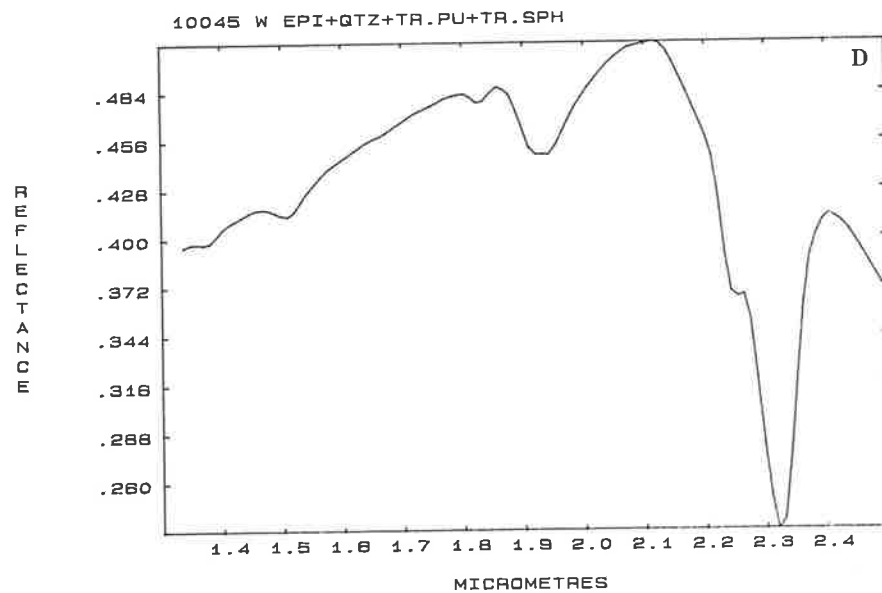
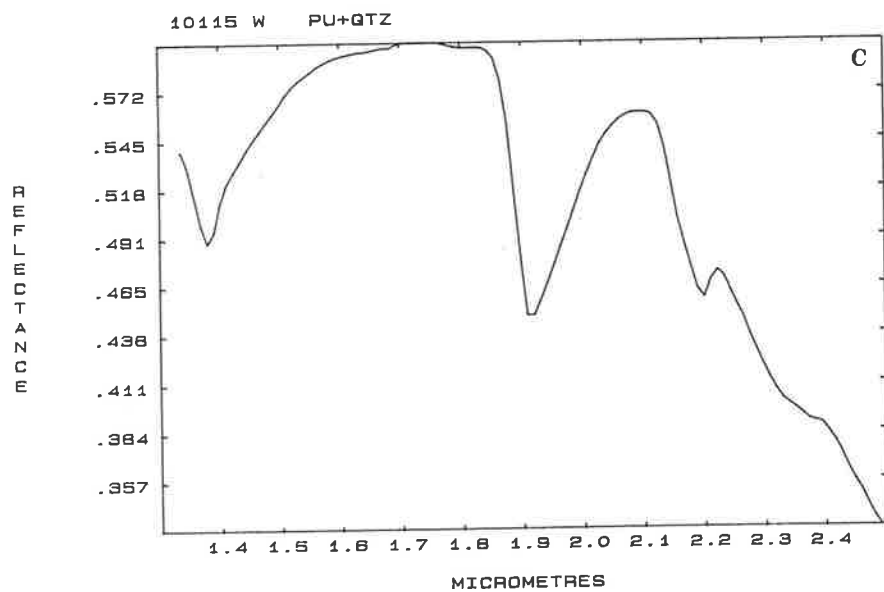
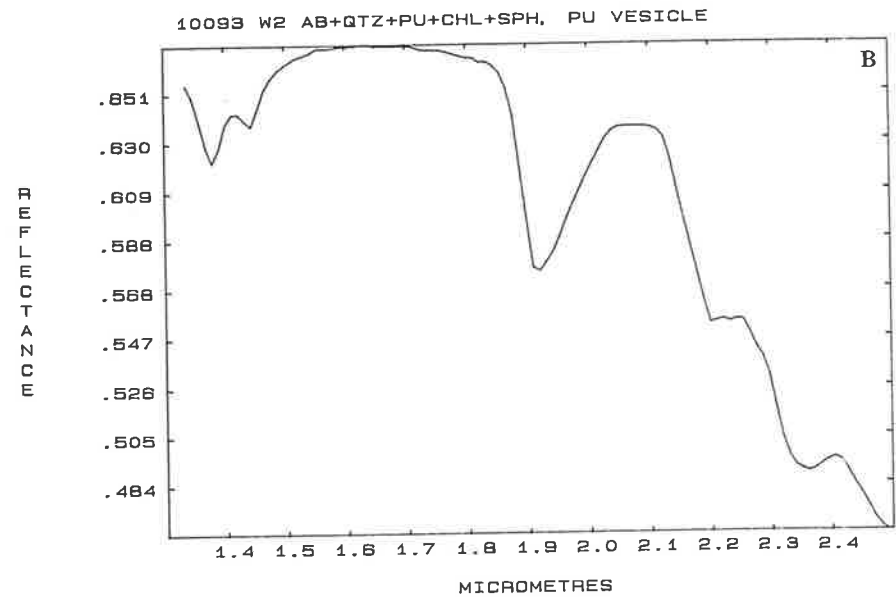
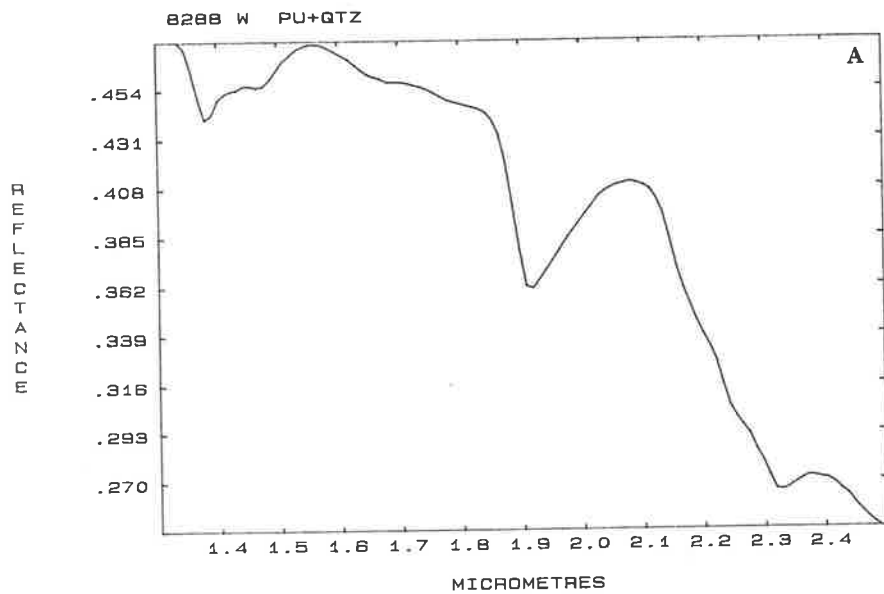
Figures 7.10a-d SWIR laboratory reflectance spectra of typical altered samples from the Fortescue Group - weathered surfaces.

Samples depicted in Figs. 7.10a-c are from Z-I, that in Fig 7.10d is from Z-II. Fig. 7.10a depicts a pumpellyite spectrum. Almost all of the features in Fig. 7.9 can be seen here.

Figure 7.10b exhibits only some of the pumpellyite features. In particular, the $1.47\mu\text{m}$ absorption feature is diagnostic. The $2.3\mu\text{m}$ absorption is due to a mixture of pumpellyite and chlorite, and is not recognisable as either.

Figure 7.10c shows a weak $2.2\mu\text{m}$ absorption feature (probably kaolin). Although the mineralogy of this sample is pumpellyite + quartz, in this case the surface weathering rind is too thick to allow diagnostic spectral features to be seen.

Despite being the spectrum of a weathered surface, Fig. 7.10d shows all of the characteristics of epidote (see Hunt et al., 1973a, p.91). The feature at $2.32\mu\text{m}$ is particularly strong. Epidote is a resistant mineral, and presumably persists in the weathered surface rind.

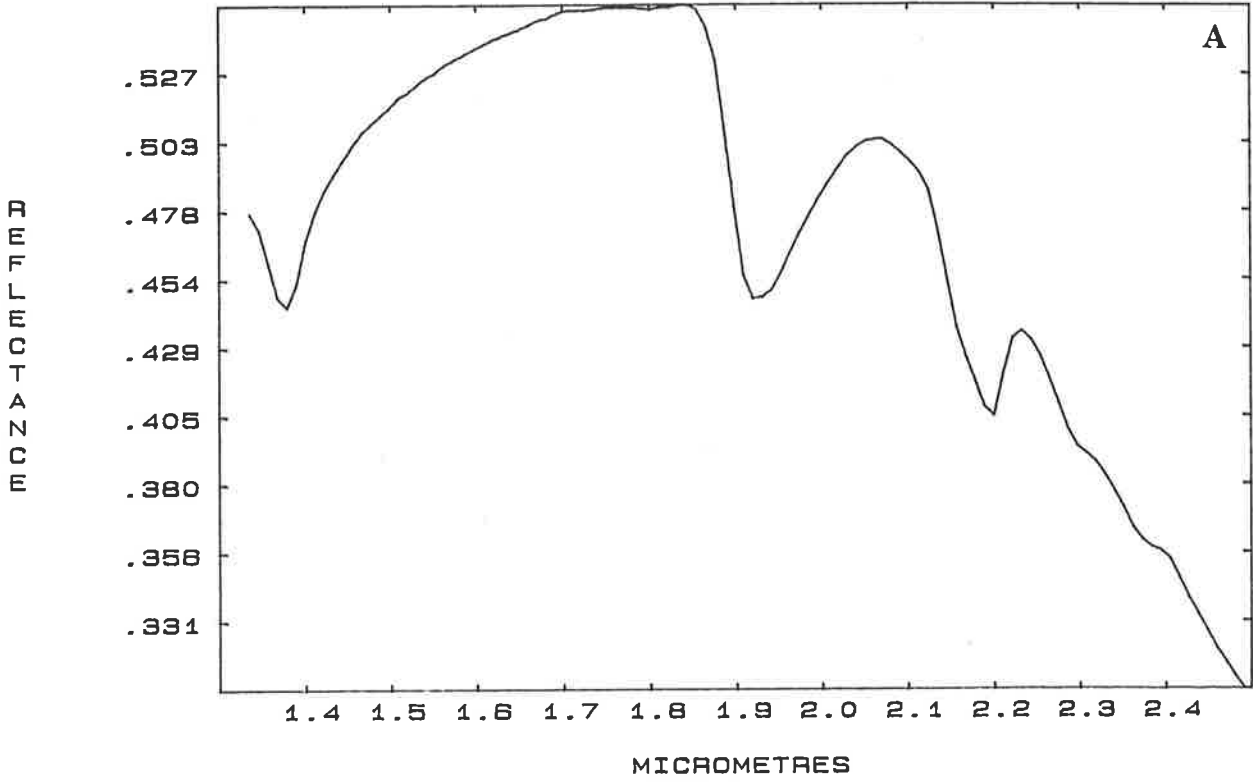


Figures 7.11a&b SWIR laboratory reflectance spectra of typical metadomain samples from Z-I of the Fortescue Group - weathered surfaces.

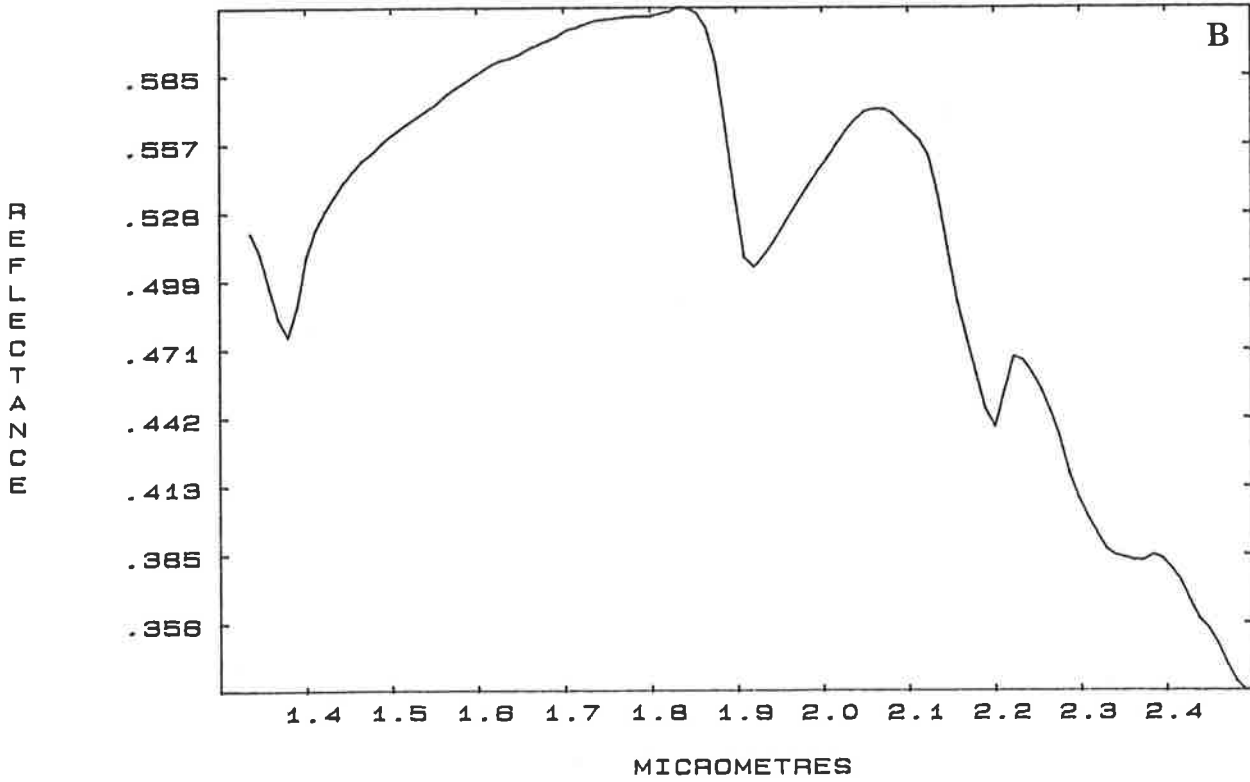
Both spectra have 2.2 μ m absorption features, probably due to kaolin. Figure 7.11b also has a 2.3 μ m feature due to chlorite, although this identification cannot be made from the spectrum alone.

The similarity of these plots to Fig. 7.10c means that there will be cases where SWIR reflectance spectra will not discriminate altered and unaltered samples.

10118 W CA PLAG+PYX+CHL+SPH+CC



10105 W CA PLAG+PYX+CHL+SPH

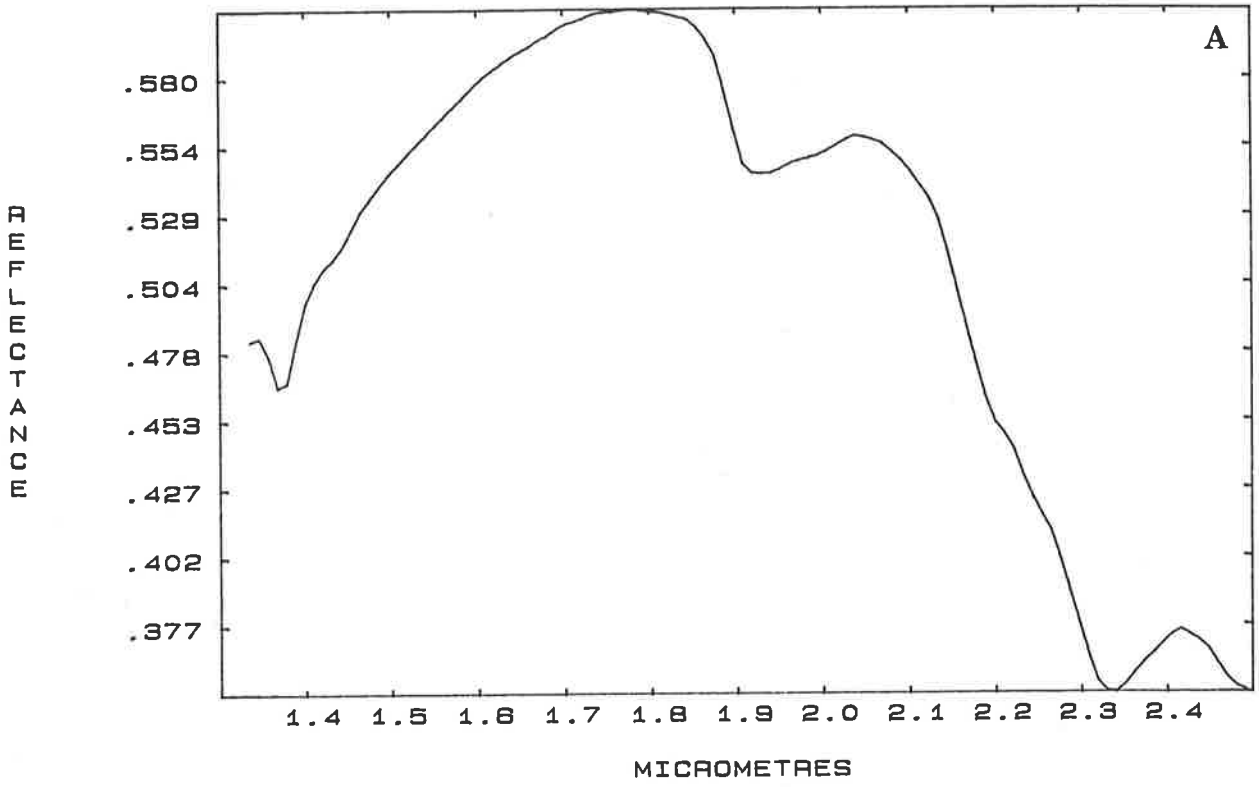


Figures 7.12a&b SWIR laboratory reflectance spectra of chlorite- and actinolite-bearing samples from higher-grade areas of the Fortescue Group - weathered surfaces.

Figure 7.12a is the chlorite spectrum. Note the similarity in the 2.3 μm region to pumpellyite (Fig. 7.9). However, the lack of an absorption feature at 1.47 μm , and the presence of the 2.0 μm absorption feature are sufficient to differentiate these two minerals.

Figure 7.12b is an actinolite spectrum. The shape of the 2.3 μm feature, and the shoulder at 2.4 μm indicate this. A larger number of spectra need to be measured before the feasibility of accurately mapping actinolite in this environment can be determined.

10848 W CA PLAG+CHL+ACT+PR



12958 W CA PLAG+CHL+ACT+EPI+SPH

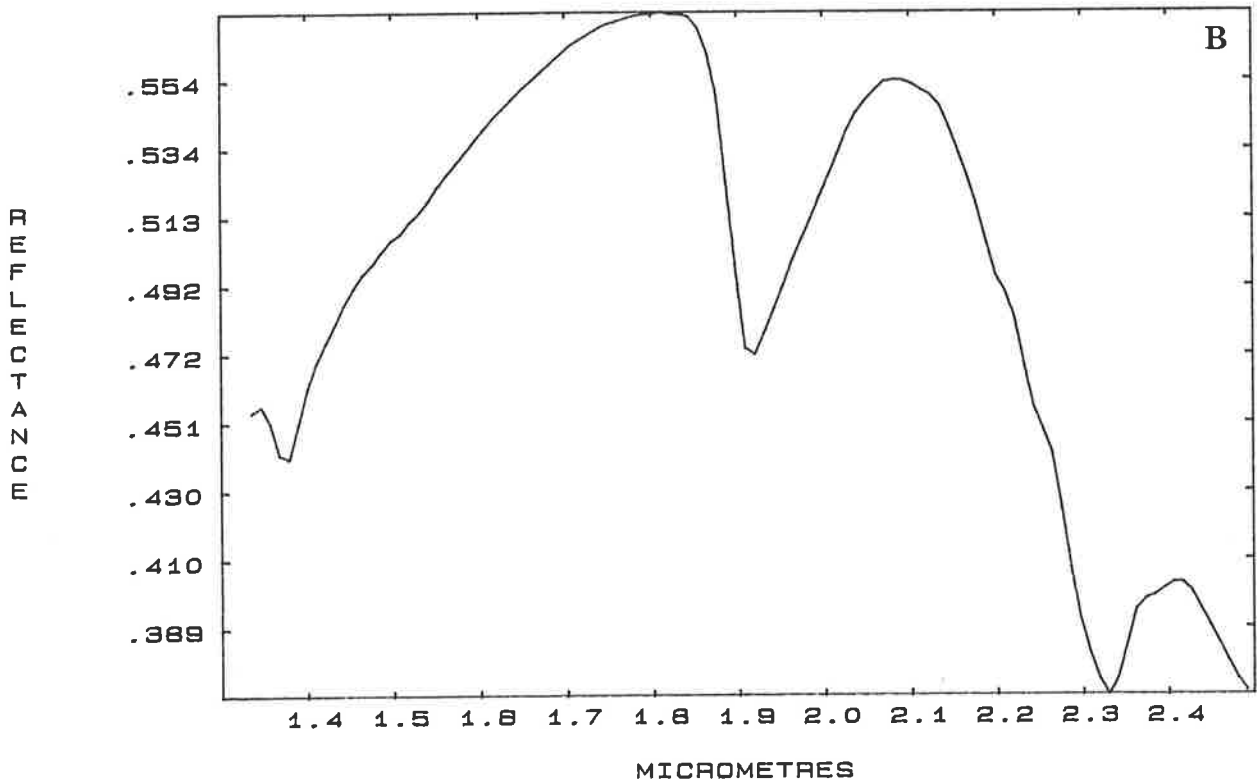
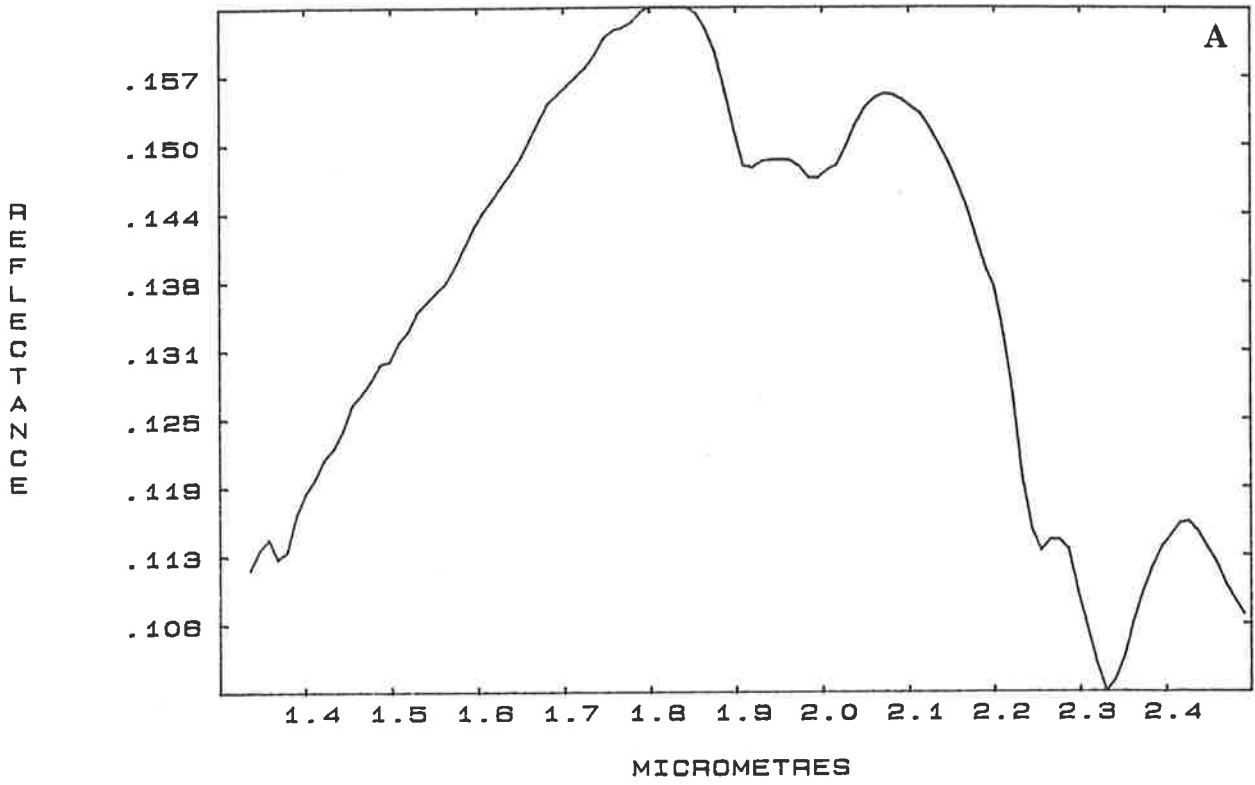


Figure 7.13a&b SWIR laboratory reflectance spectra of chlorite-bearing samples from Z-I of the Fortescue Group - fresh surfaces.

Both spectra have the characteristic chlorite absorption features at 2.0 and 2.26 μm .

10118 F CA PLAG+PYX+CHL+SPH+CC



10103 F AB+QTZ+CHL

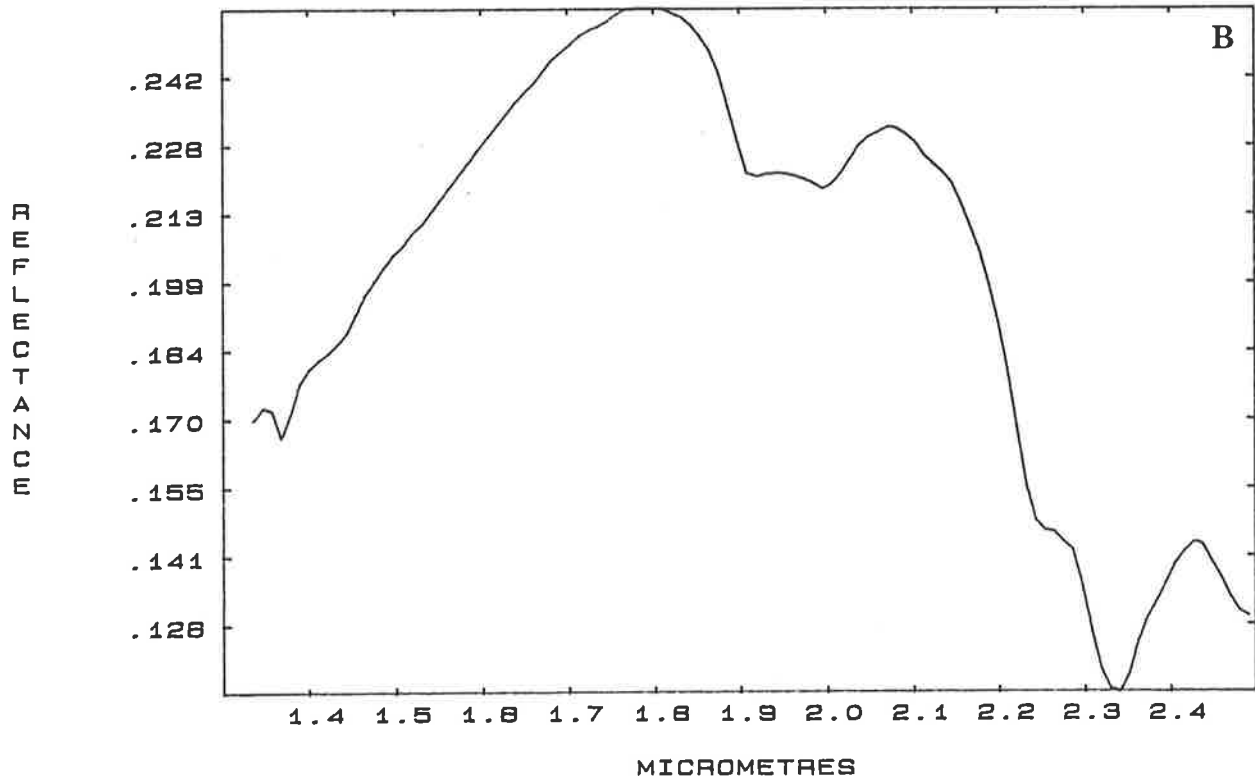
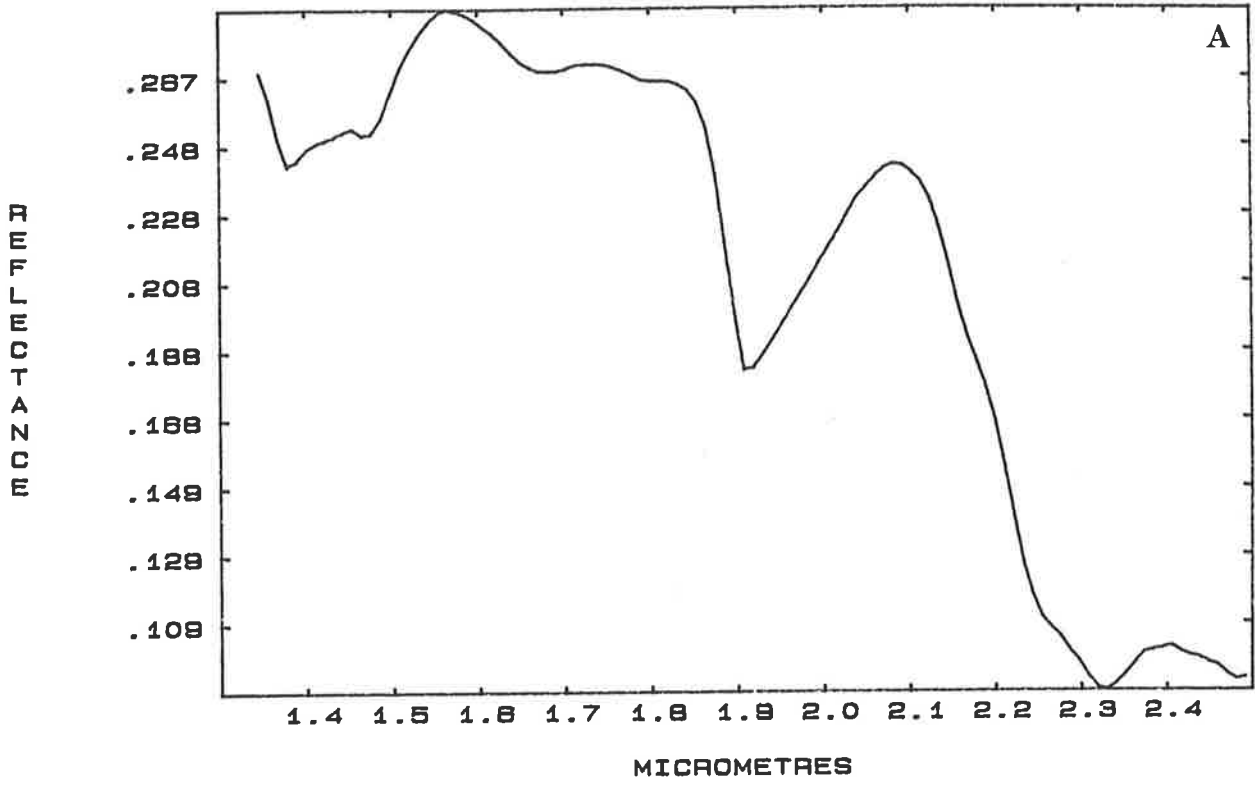
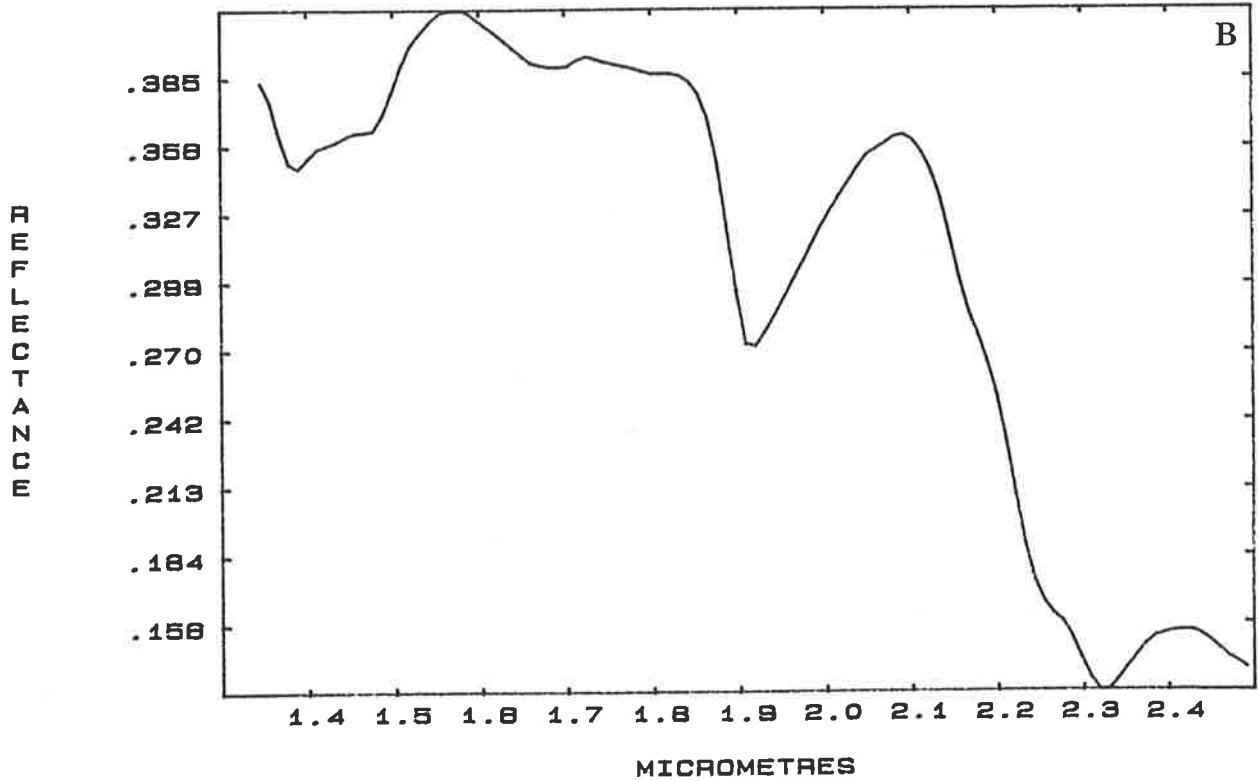


Figure 7.14a&b SWIR laboratory reflectance spectra of pumpellyite-bearing samples from Z-I of the Fortescue Group - fresh surfaces.

25/1/84 0208 F

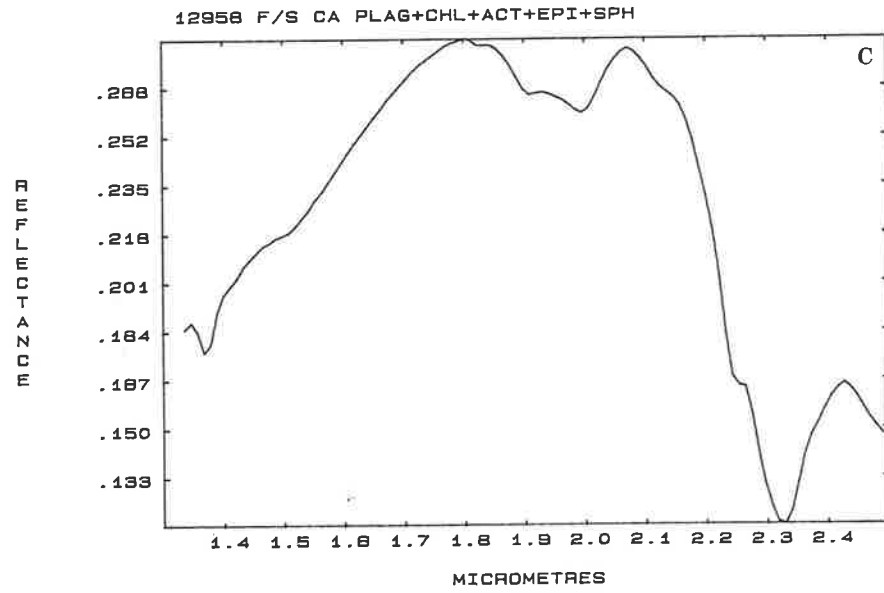
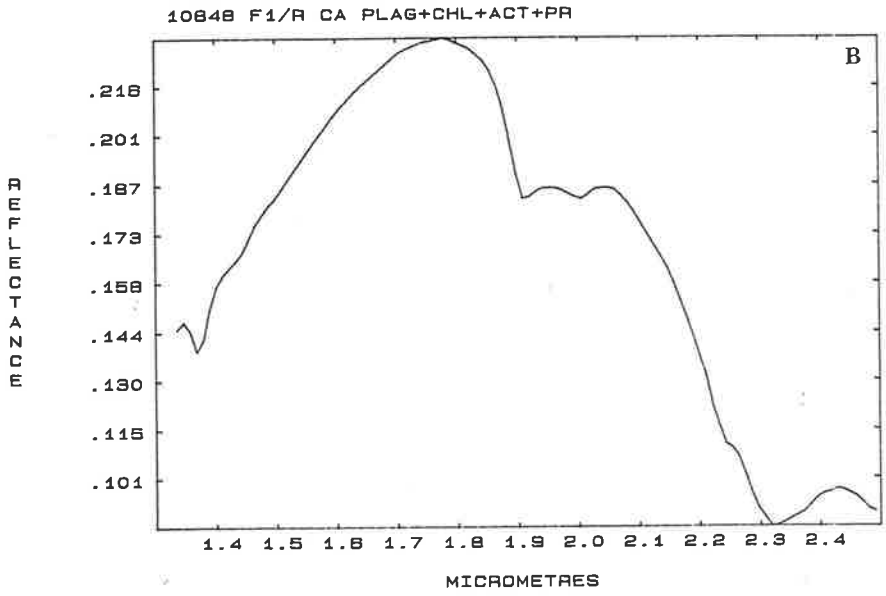
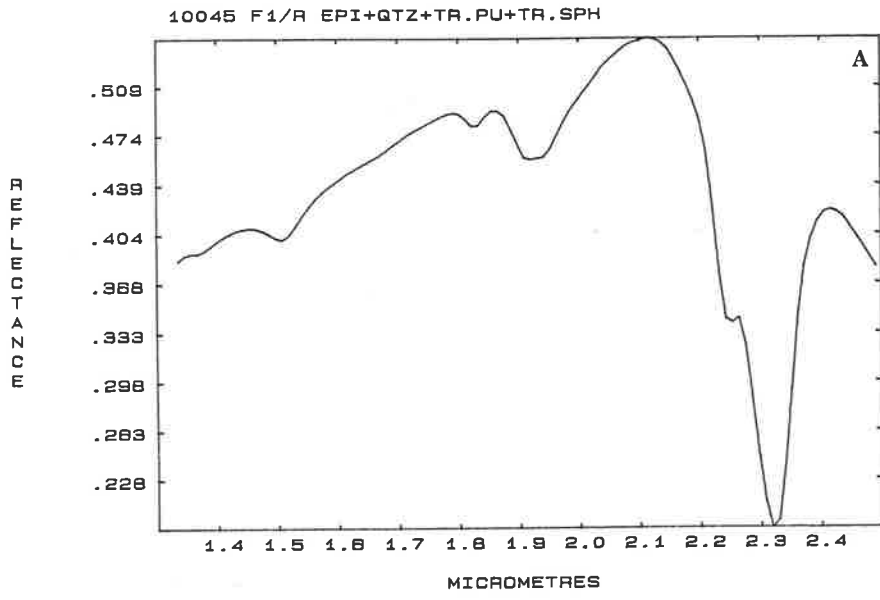


25/1/84 10115 F



Figures 7.15a-c SWIR laboratory reflectance spectra of typical samples from higher-grade areas of the Fortescue Group - fresh surfaces.

The spectra shown in Figs. 7.15a-c are dominated by effects due to epidote, chlorite, and a chlorite/epidote mixture, respectively. Figures 7.15b&c also appear to have a minor influence due to actinolite. Prehnite (which has a $2.3\mu\text{m}$ feature similar to pumpellyite, but different structure in the 1.4 to $1.6\mu\text{m}$ region - V. Gardavsky, pers. comm., 1986) may also be influencing the shape of the spectrum in Fig. 7.15b.



Figures 7.16a&b MIR laboratory laser reflectance spectra of typical unaltered and altered samples from the Fortescue Group - weathered surfaces.

Figures 7.16a&b are spectra of Z-I samples, while Figs. 7.16c&d are spectra of Z-II samples. All spectra exhibit the characteristics of haematite (J. Eberhardt, pers. comm., 1984). However, Figs. 7.16b&d (from altered flow tops in Z-I and Z-II, respectively) are both quartz-rich, and show an additional maximum near $9.3\mu\text{m}$. The wavelengths used for the ratio to discriminate altered and unaltered samples are marked by the vertical lines.

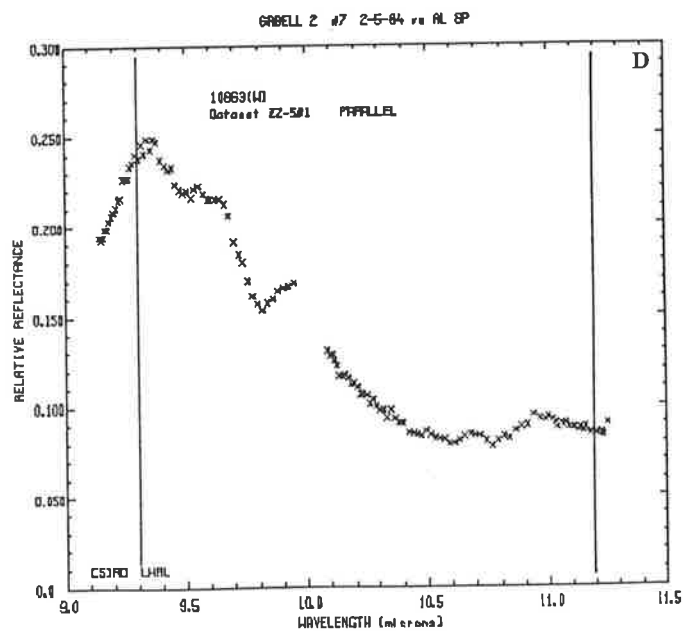
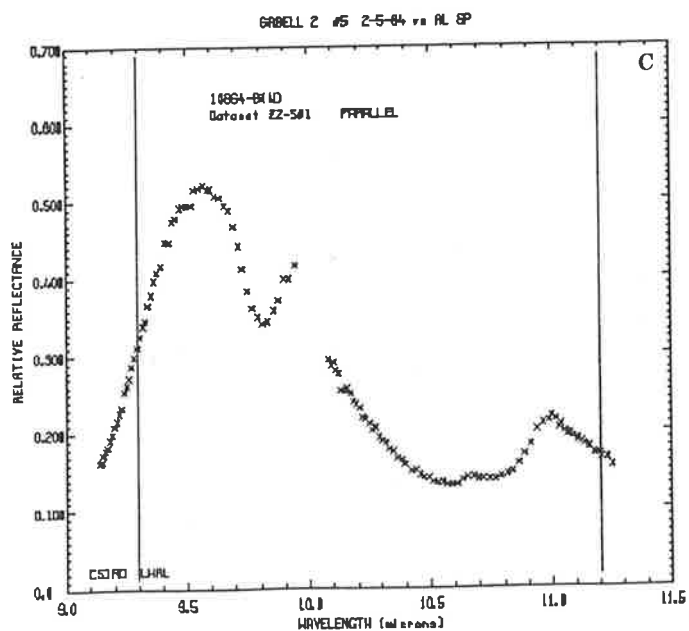
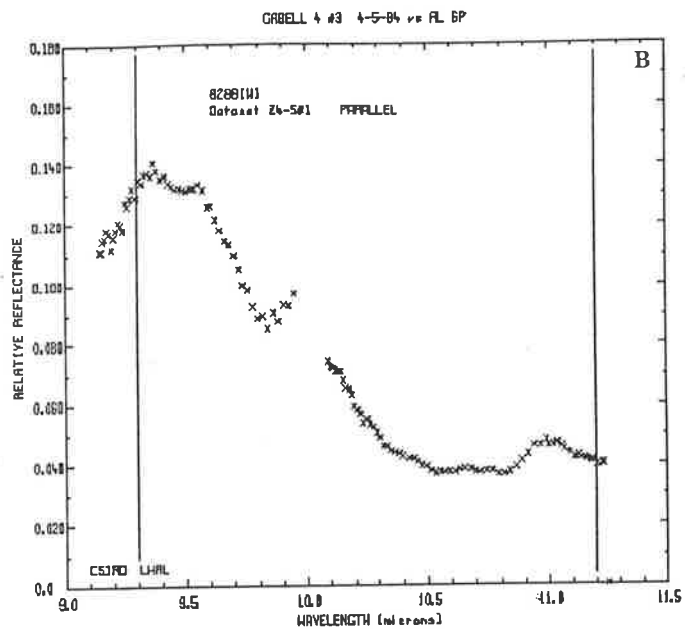
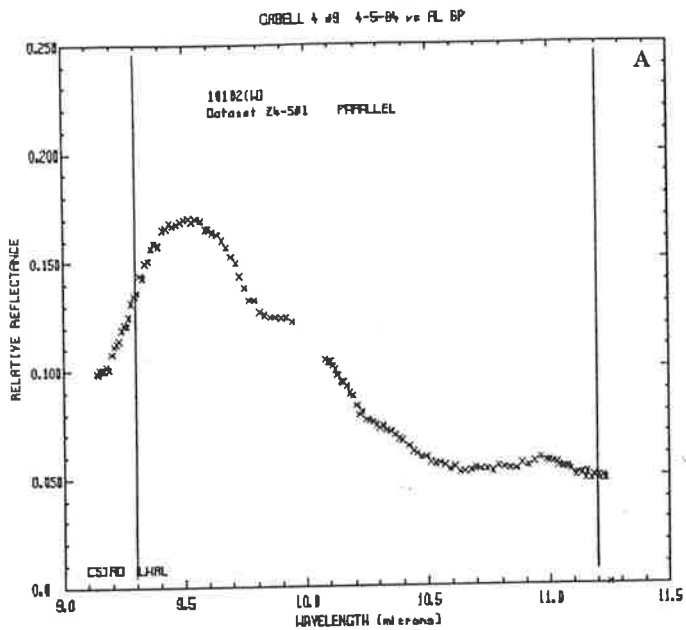


Figure 7.17a&b MIR laboratory laser reflectance spectra of typical unaltered and altered samples from the Fortescue Group - fresh surfaces.

These spectra were measured from the same samples as those in Figs. 7.16a-d. Compare Figs. 7.17a&c to spectra in Figs. 2.12 and 2.13. Most of the unaltered samples from the Fortescue Group have fresh surface spectra of this type. Of the altered samples, Fig. 7.17b exhibits the spectral features of quartz (its mineralogy is pumpellyite + quartz), while Fig. 7.17c has characteristics that can be attributed to both of its major components - quartz and epidote (see Farmer, 1974 for epidote). Note that the reflectance maximum of quartz in Fig. 7.17b is at a shorter, and more commonly observed, wavelength ($9.2\mu\text{m}$) than in the weathered surface spectra.

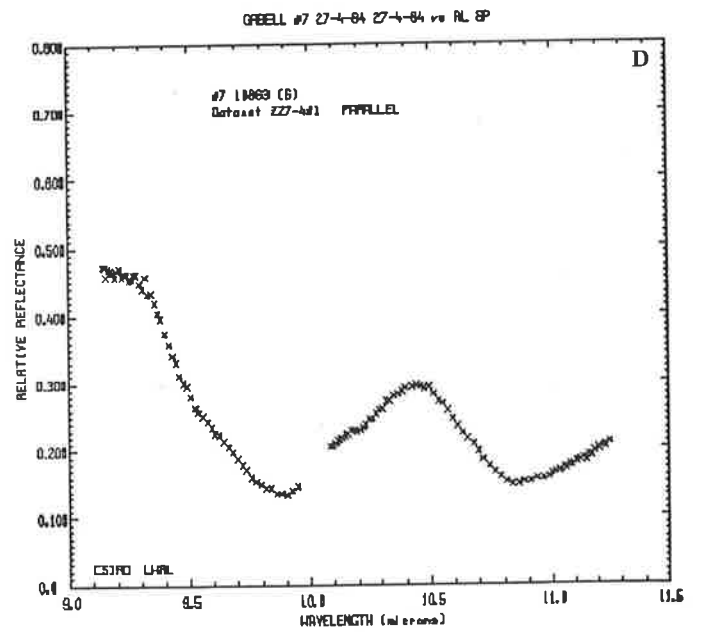
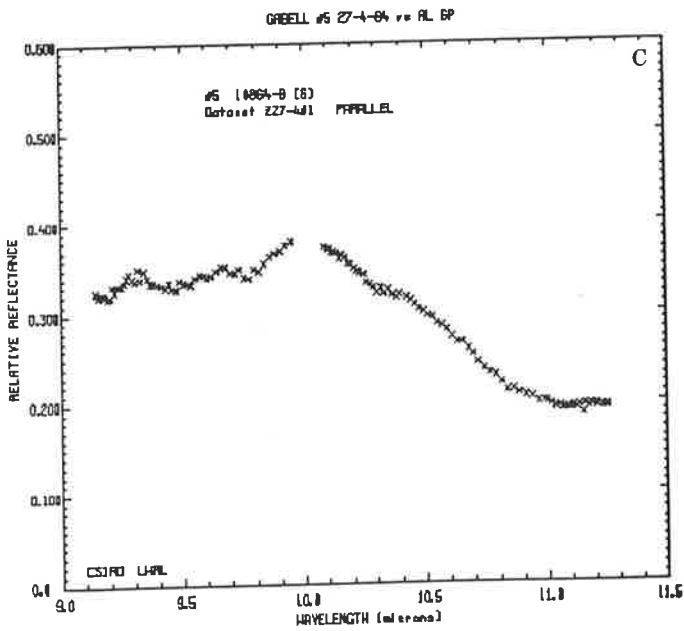
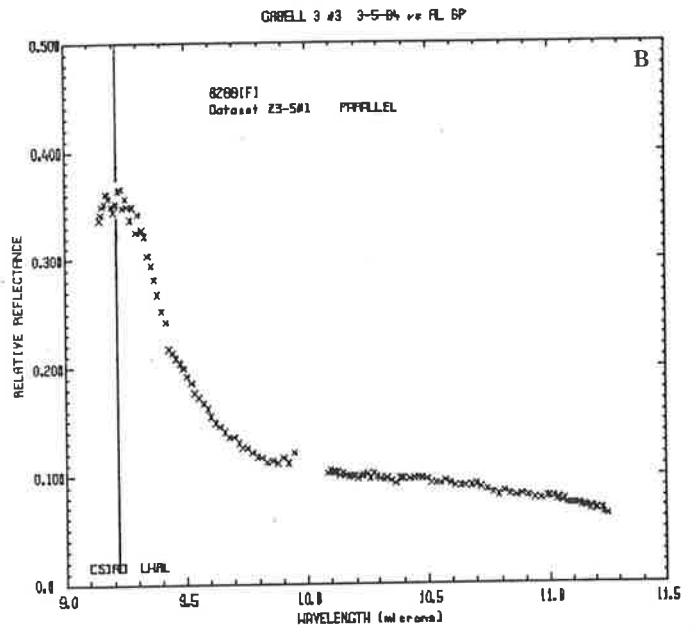
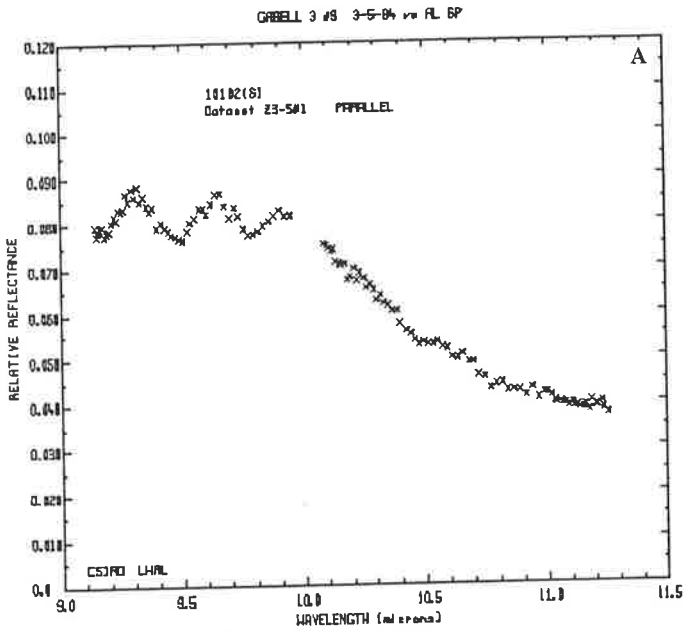


Figure 8.1 Photocopy of a colour aerial photograph of Peak Hill.

Peak Hill township lies immediately to the west of Peak Hill. Scale of the photograph is 1:25,000. Position of Gold Fields' grid (see Figs. 8.12 to 8.18) is shown.

05

27-2-82

3061-3096

PARKES

30R6

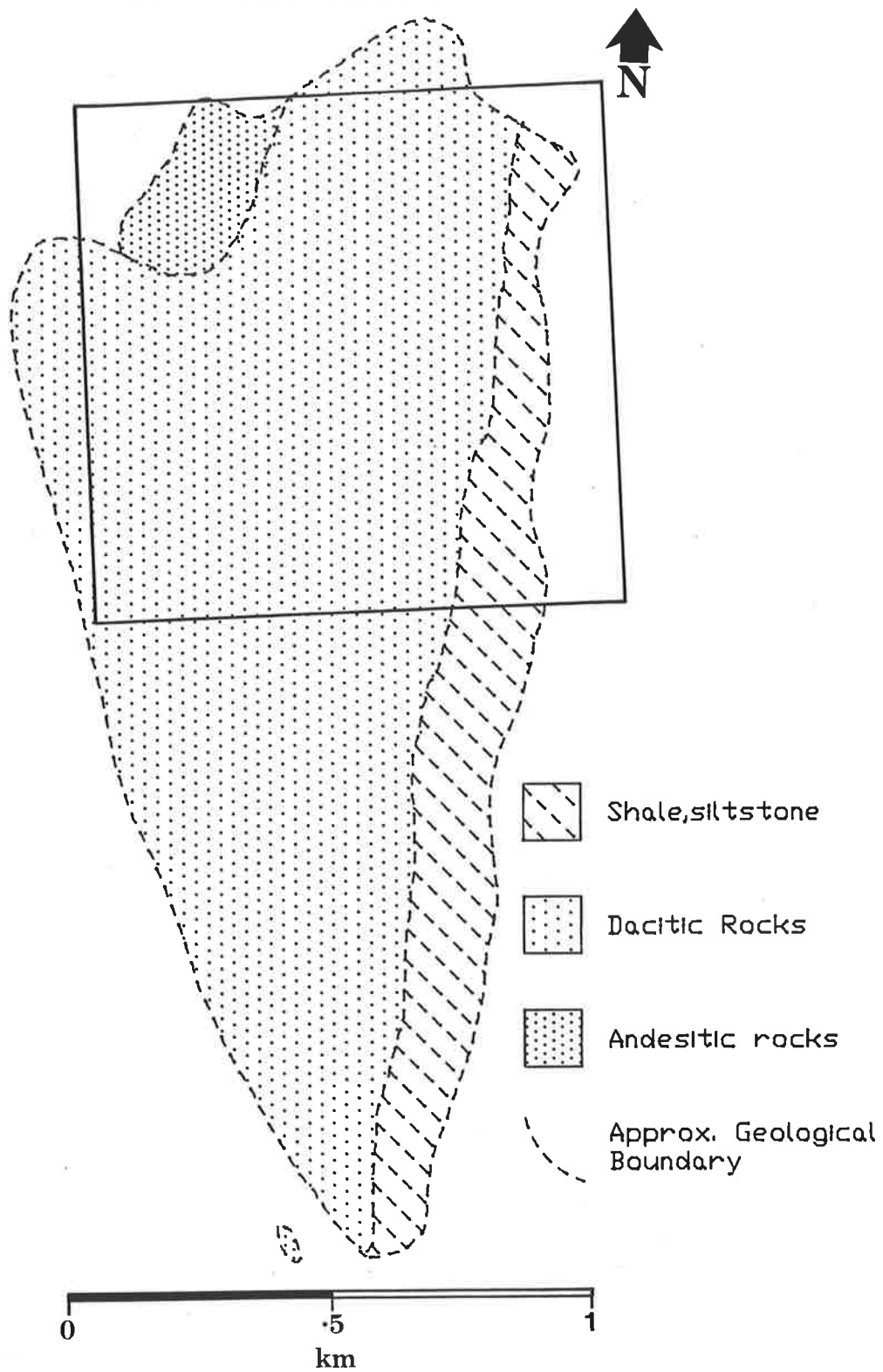
AAM
1550c



Figure 8.2 Geology of Peak Hill (after Bowman and Richardson,
1978).

Position of Gold Fields' grid is shown.

PEAK HILL GEOLOGY

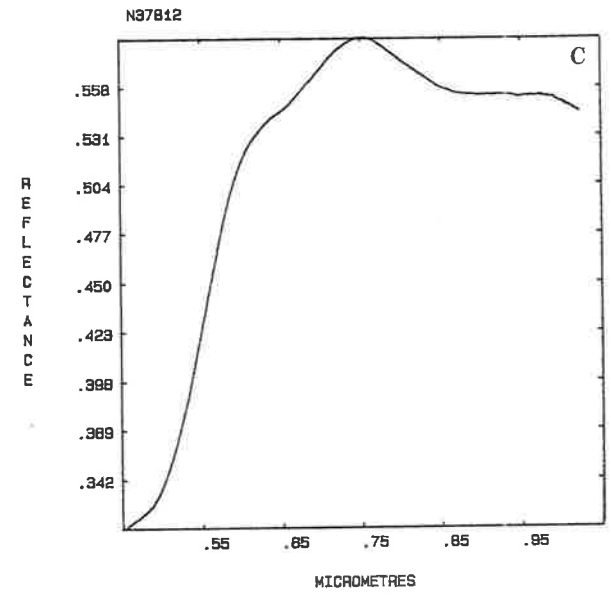
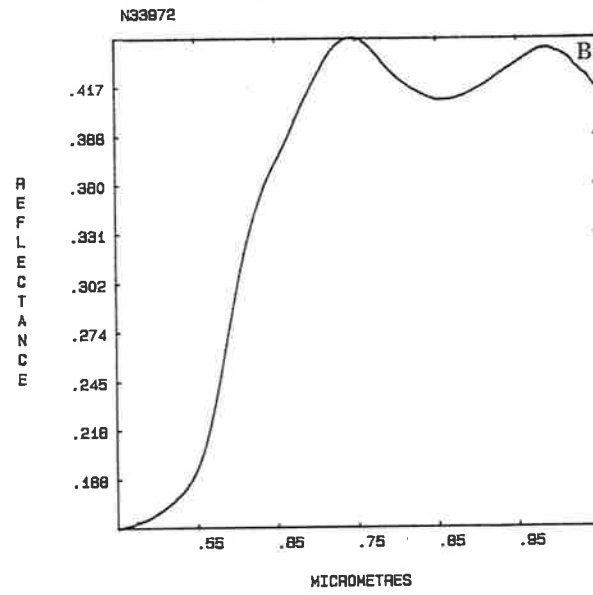
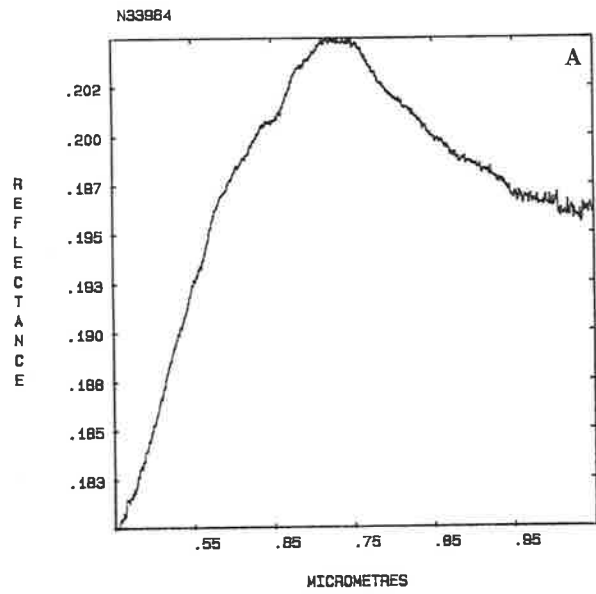


Figures 8.3a-c Representative VNIR laboratory reflectance spectra of core samples from Peak Hill.

The absorption feature in Fig. 8.3a is weak (0.08% depth), and centred at a longer wavelength than is usual for haematite (approx. $.87\mu\text{m}$) or goethite (approx. $.92\mu\text{m}$). The sample is from below the base of oxidation, and its spectral features can therefore be attributed to Fe^{2+} .

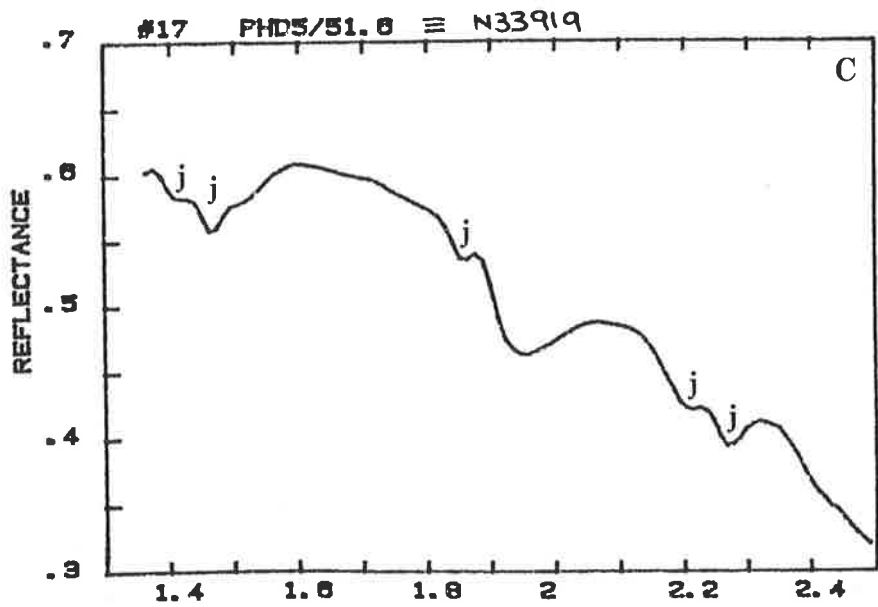
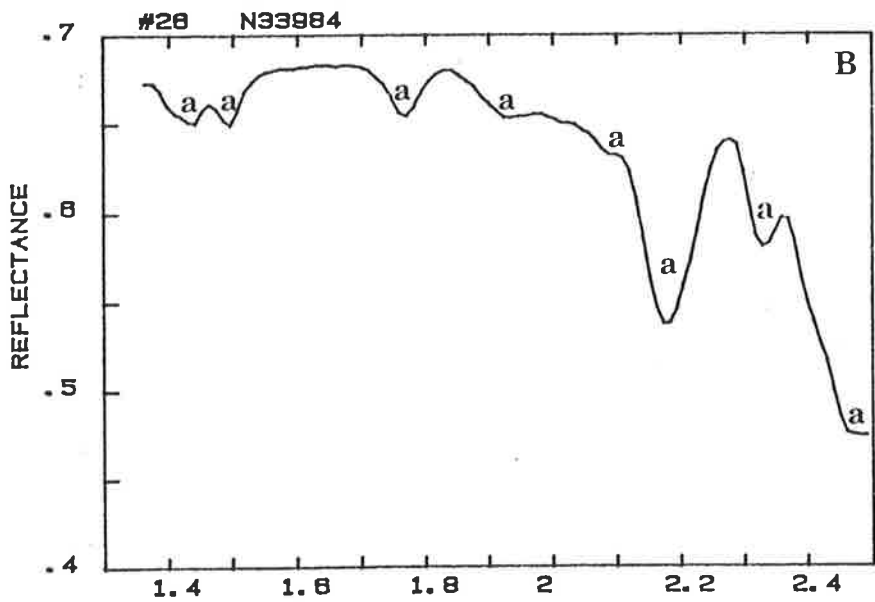
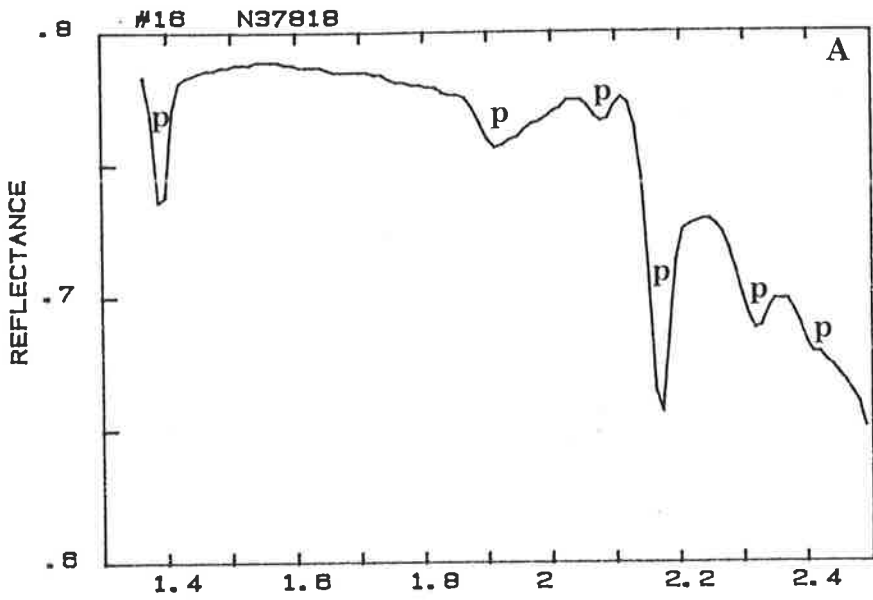
The spectrum shown in Fig. 8.3b is representative of the majority of core samples collected above the base of oxidation. The position of the major absorption feature ($.86\mu\text{m}$), and the absence of a shoulder near $.65\mu\text{m}$, indicate a haematite mineralogy.

Figure 8.3c does not have strong spectral features, but there is some evidence of absorption in the $.65\mu\text{m}$ region, and a feature longer than $.9\mu\text{m}$. These are characteristics of goethite, although jarosite can also have a similar spectral shape in the VNIR.



Figures 8.4a-c Representative SWIR laboratory reflectance spectra, with strong absorption features, of core samples from Peak Hill.

These plots are good examples of the spectral properties of pyrophyllite, alunite and jarosite. The strong, sharp $2.17\mu\text{m}$ feature of pyrophyllite (Fig. 8.4a) is diagnostic (see Fig. 2.5), as is the shape of the jarosite spectrum between 2.2 and $2.3\mu\text{m}$ (compare Figs. 8.4c and Fig. 2.9). Alunite has a doublet near $1.4\mu\text{m}$, a unique feature near $1.76\mu\text{m}$, and a characteristic shape between 2.17 and $2.2\mu\text{m}$ (compare Fig. 8.4b to Fig. 2.9).

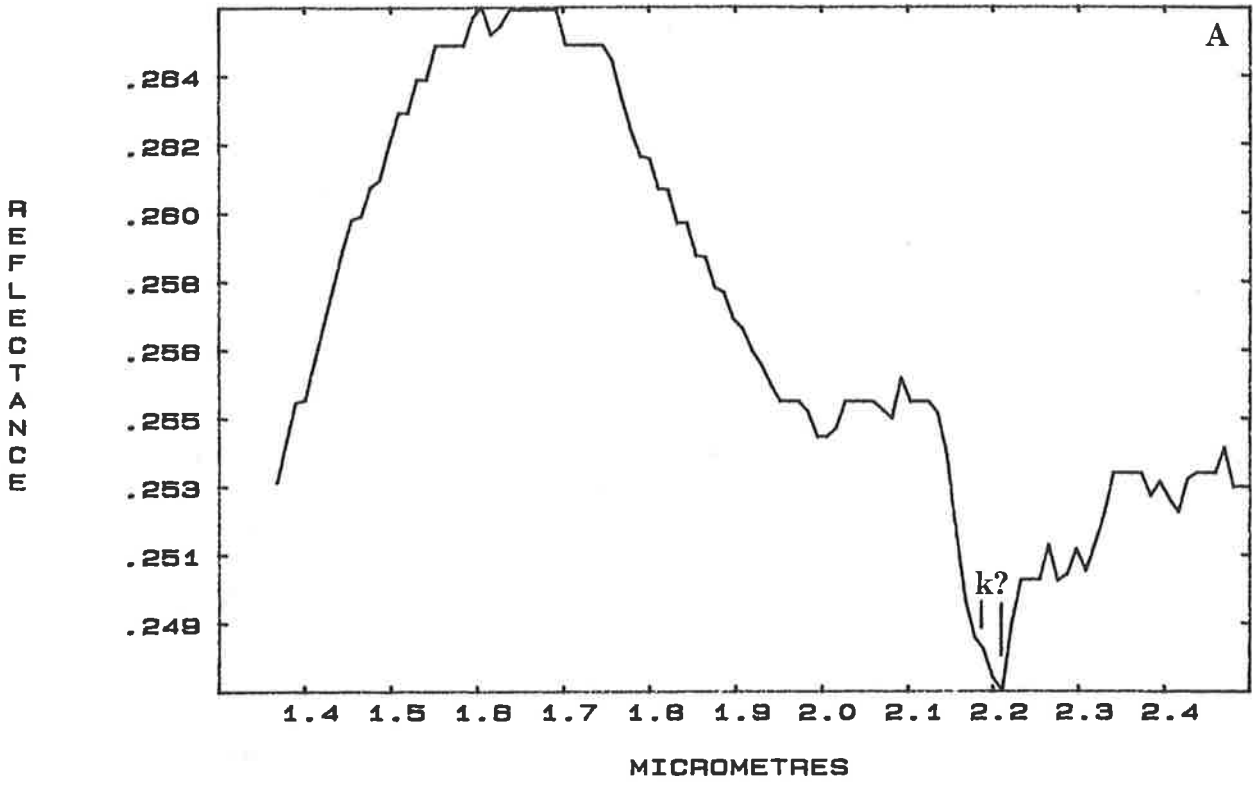


Figures 8.5a&b Representative SWIR laboratory reflectance spectra, with weak absorption features, of core samples from Peak Hill.

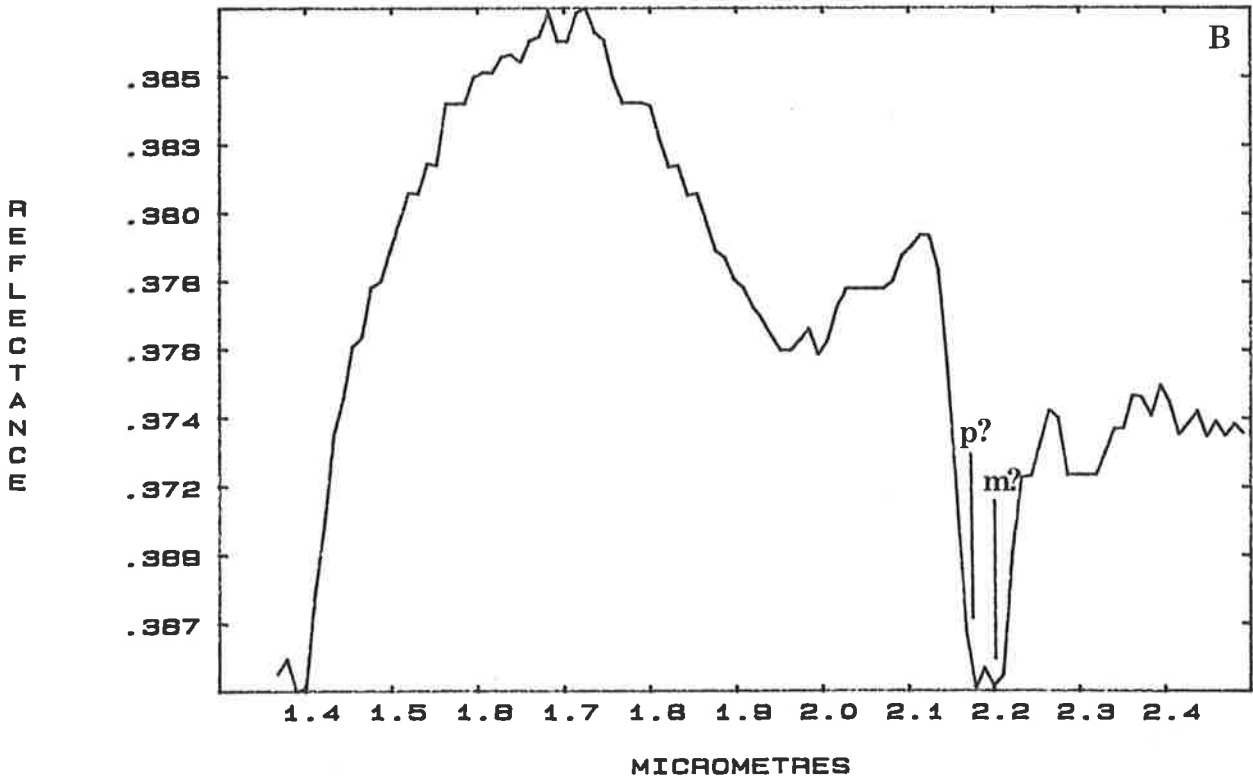
Figure 8.5a is a noisy spectrum, with weak absorption features. However, the feature near $2.2\mu\text{m}$, with a shoulder at $2.17\mu\text{m}$, is reminiscent of kaolinite.

Figure 8.5b appears to have a poorly-developed doublet at 2.17 and $2.2\mu\text{m}$. This shape was initially attributed to random noise in the spectrum (note the noise level at other wavelengths), but repeated measurement of the sample resulted in identical structure. These features may therefore represent true spectral absorption. If this is the case, a mixture of pyrophyllite (sharp $2.17\mu\text{m}$ absorption) and sericite (sharp $2.2\mu\text{m}$ absorption) could be responsible.

19-6-84 N33884

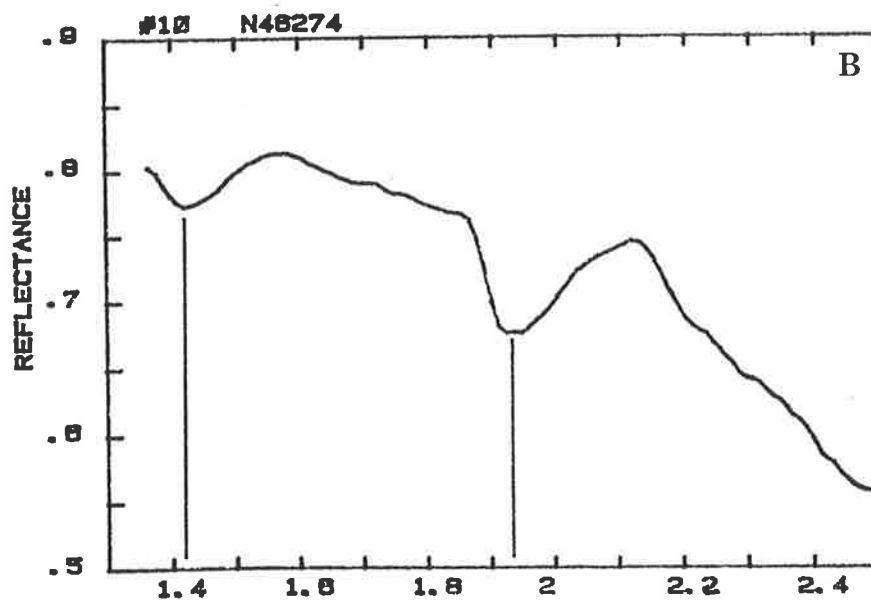
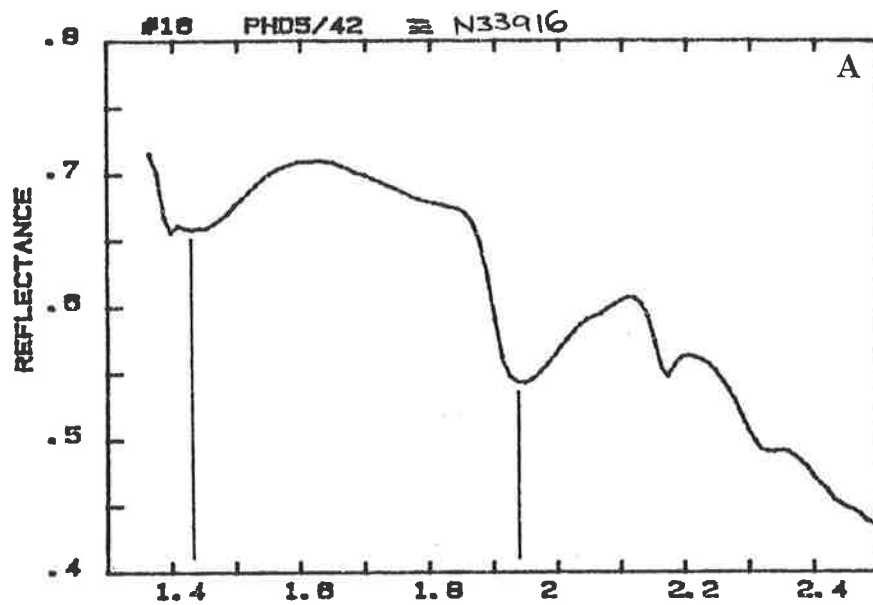


19-6-84 N33833



Figures 8.6a&b SWIR laboratory reflectance spectra of quartz-rich samples from Peak Hill.

Note the unusual, broad shapes of the 1.4 and 1.9 μ m water absorption features. Similar structure has been noted in a number of quartz samples containing fluid inclusions (J. Huntington, pers. comm., 1985). This information is of little use in a remote-sensing mode, however, because of atmospheric absorption in these regions.

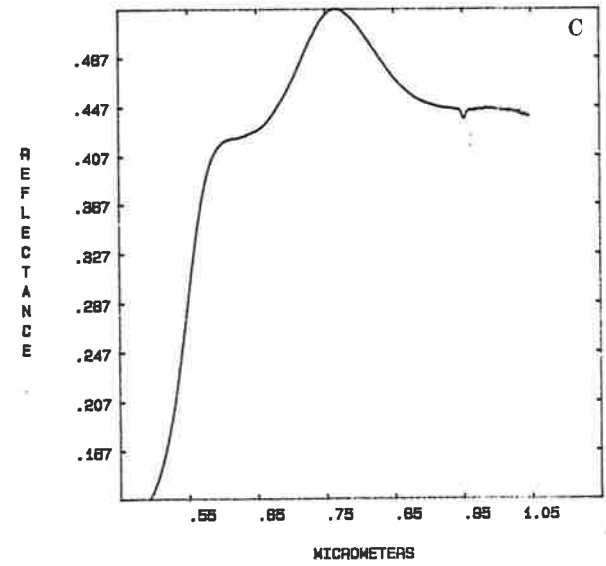
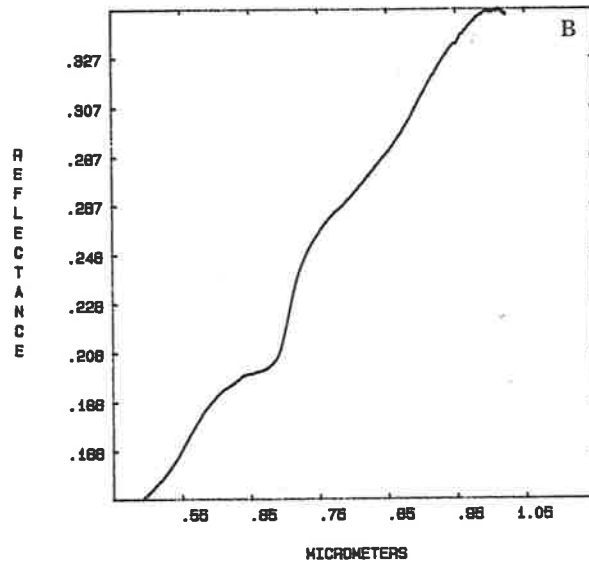
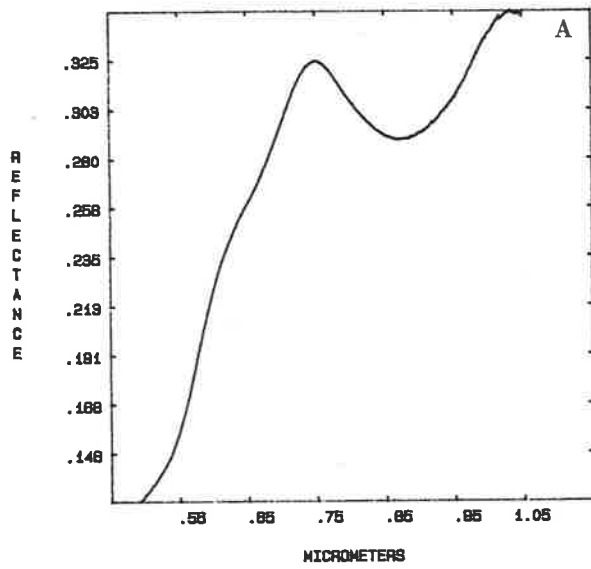


Figures 8.7a-c Representative VNIR laboratory reflectance spectra of surface samples from Peak Hill.

Figure 8.7a depicts the most common VNIR spectral reflectance curve from Peak Hill surface samples. Like the core samples, the features can be attributed to haematite.

Figure 8.7b has an unusual shape, caused by a significant coating of lichen on the surface of the rock.

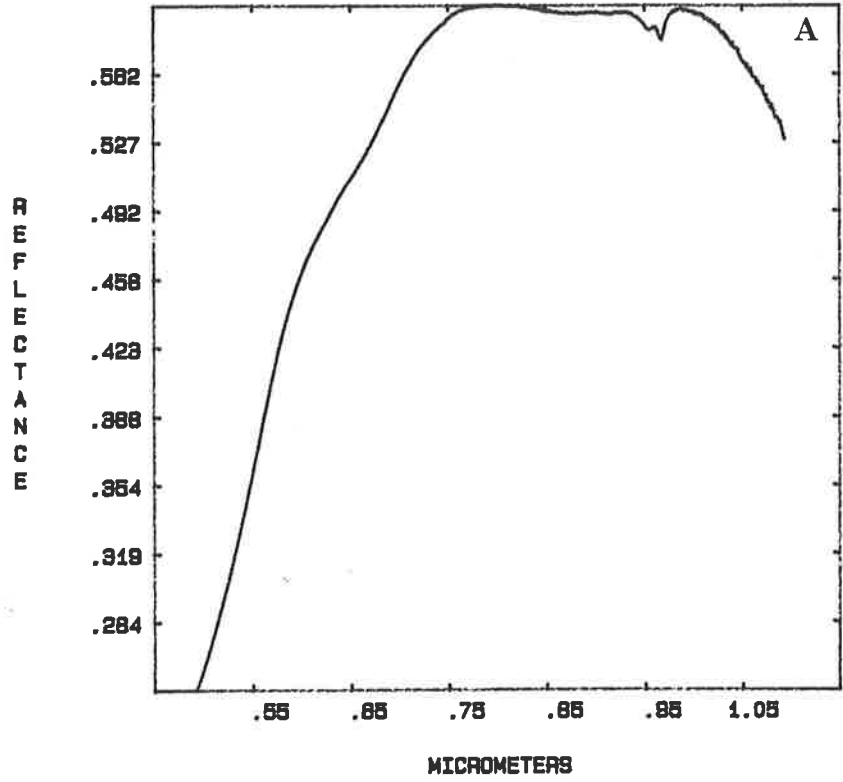
Figure 8.7c shows some subtle differences to the majority of the surface samples. Like Fig. 8.3c, absorption at a longer wavelength than haematite, coupled with a $.65\mu\text{m}$ absorption feature, is suggestive of goethite, or perhaps jarosite.



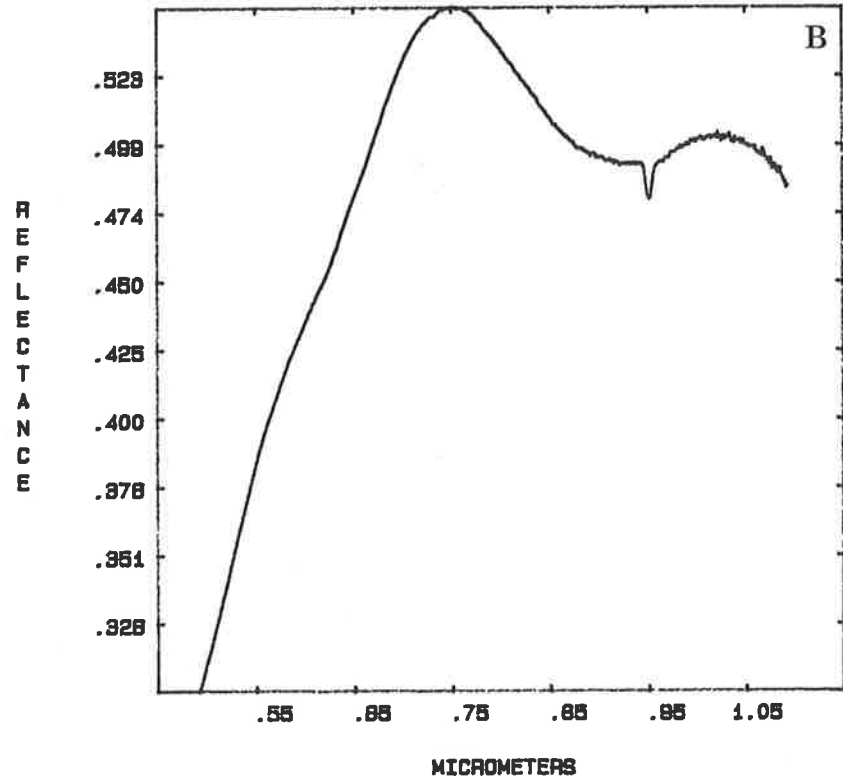
Figures 8.8a&b Representative VNIR laboratory reflectance spectra of surface samples from Peak Hill exhibiting .95 μ m clay absorption features.

Both plots show moderately well-developed .95 μ m absorption features. Figure 8.8a exhibits the doublet structure (.956 and .965 μ m) of kaolinite, whilst Fig. 8.8b has a single, sharp feature at .951 μ m - shorter than most $3\nu_{\text{OH}}$ features, but characteristic of pyrophyllite.

1 W2

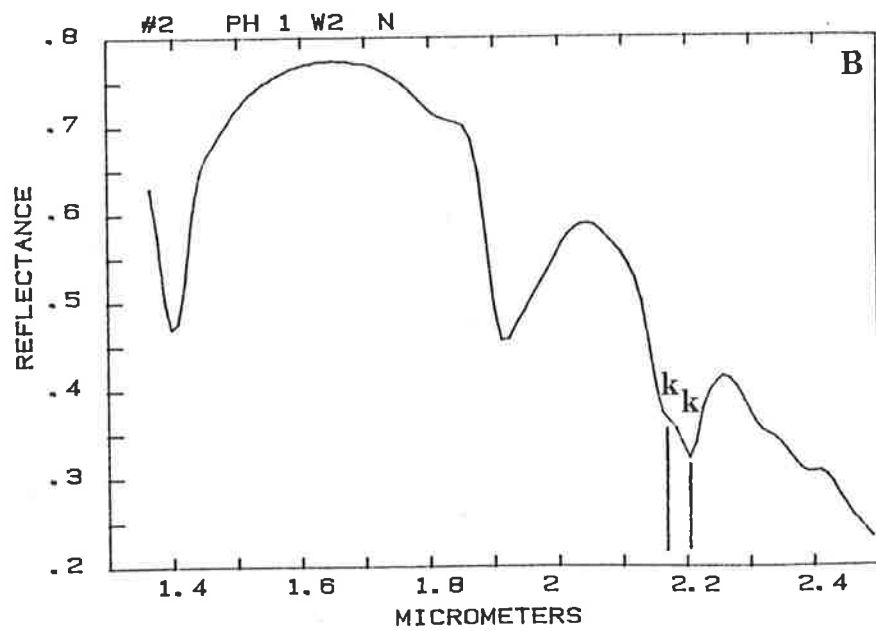
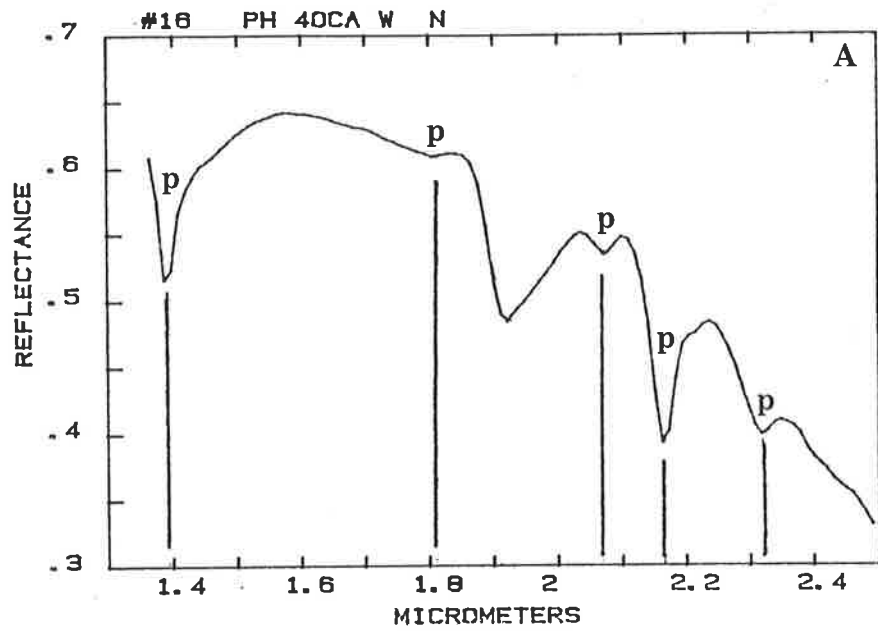


3 F



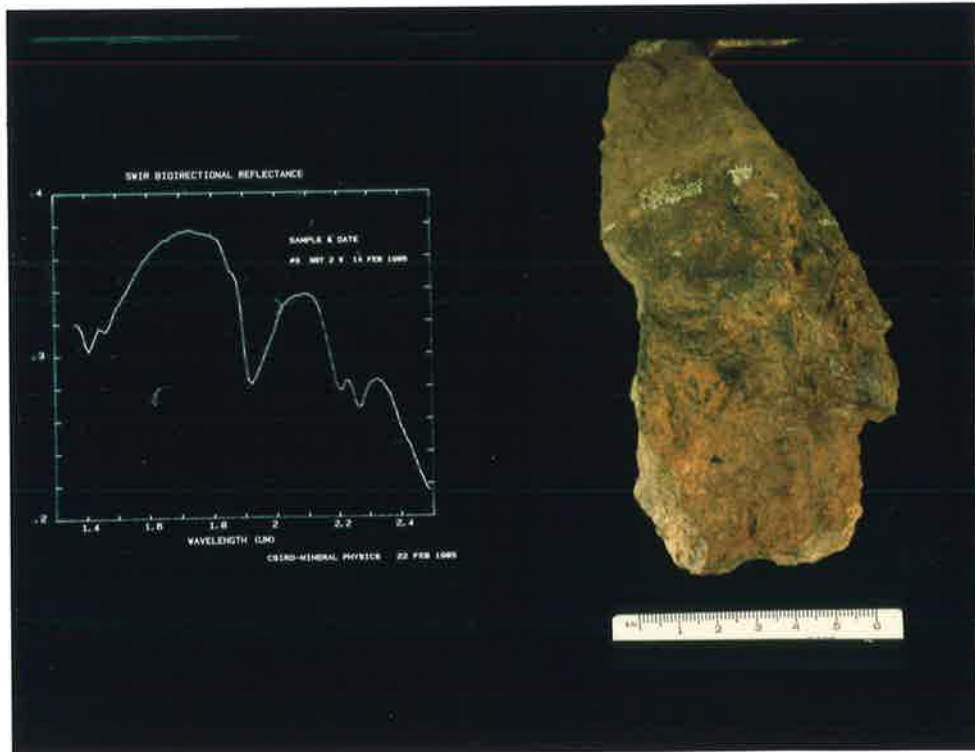
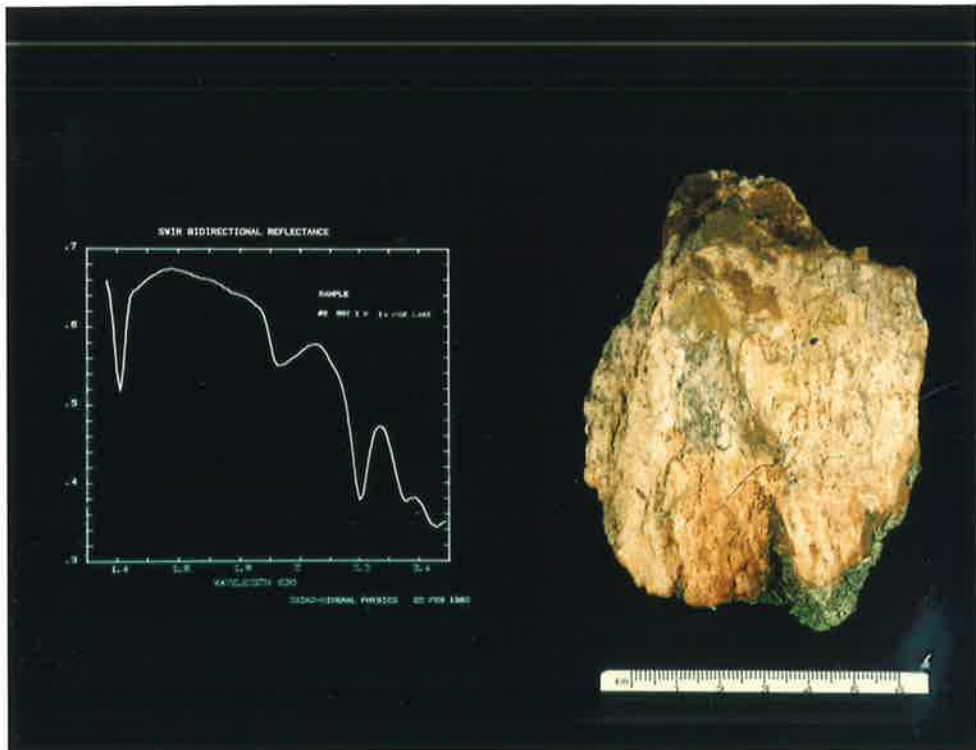
Figures 8.9a&b SWIR laboratory reflectance spectra of pyrophyllite- and kaolinite-bearing samples from Peak Hill.

Both spectra exhibit well-developed spectral features. The pyrophyllite spectrum (Fig. 8.9a) has the unmistakable characteristics also seen in Figs. 2.5 and 8.4a. The kaolinite spectrum (Fig. 8.9b) probably has the strongest features of any Peak Hill sample, yet it is the only one collected outside the alteration zone delineated by Gold Fields. The spectral features may, therefore, be due to intense weathering, rather than to alteration.



Figures 8.10a&b Photographs and SWIR laboratory reflectance spectra of muscovite- and jarosite-bearing rocks from Peak Hill.

Figures 8.10a&b show the rock surfaces measured (the black dot indicates the precise centre of measurement), and the resultant spectra of, muscovite- (sericitic) and jarosite-bearing samples, respectively. Both spectra are characteristic (see Fig. 2.5 for muscovite and Fig. 2.9 for jarosite).



Figures 8.11a&b Photographs and spectra of weathered and fresh surfaces of rock from Peak Hill with unusual spectral characteristics.

The spectral features of this sample occur in weathered and fresh surfaces, and are both moderately sharp. Both facts point to a mineralogical origin, rather than to lichen. The feature at the shorter wavelength of $2.08\mu\text{m}$ appears at the same wavelength as that of topaz (see Figs. 5.29a&b), whilst the longer wavelength feature appears at $2.17\mu\text{m}$ (the same wavelength as pyrophyllite). Neither mineral was reported present from XRD analysis.

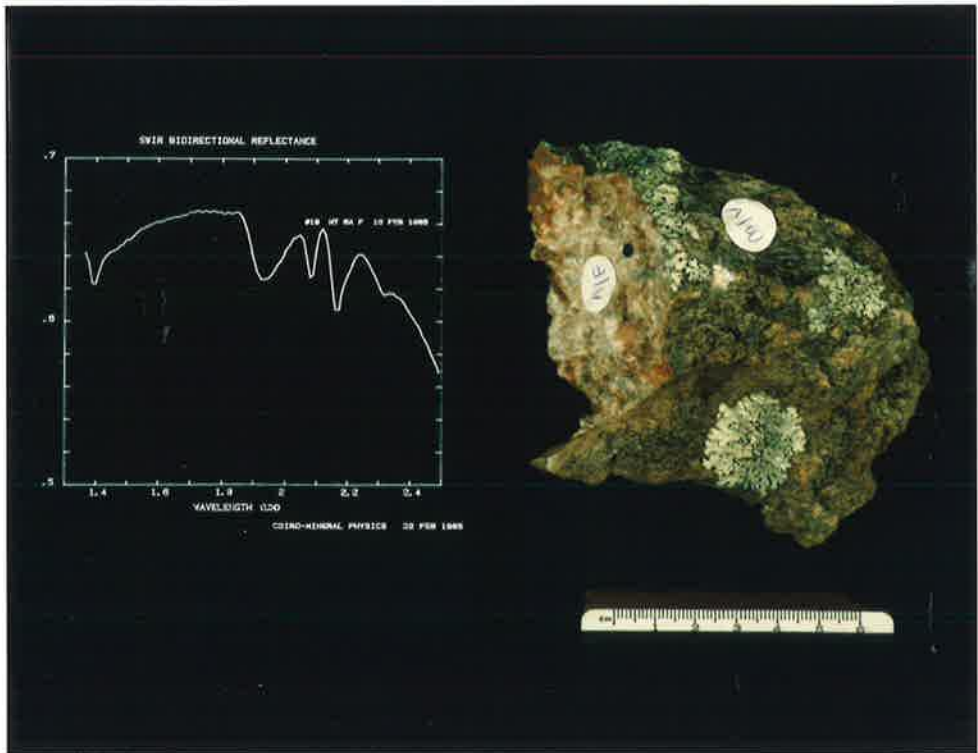
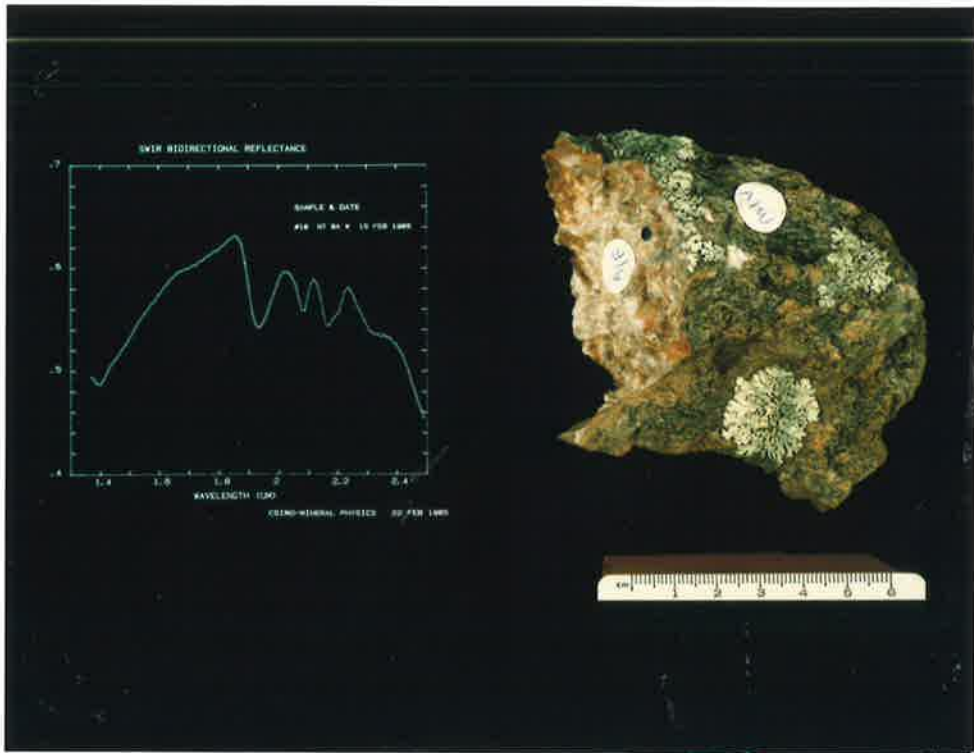


Figure 8.12 Location of samples used in the Peak Hill study.

See Figs. 8.1 and 8.2 for location of the grid. Gold Fields' core samples are shown as circles, surface samples collected by the author are shown as crosses.

PEAK HILL MINERALOGY SAMPLE LOCATIONS

KEY

O = CORE SAMPLE (XRD)
X = SURFACE SAMPLE (SPECTRA)

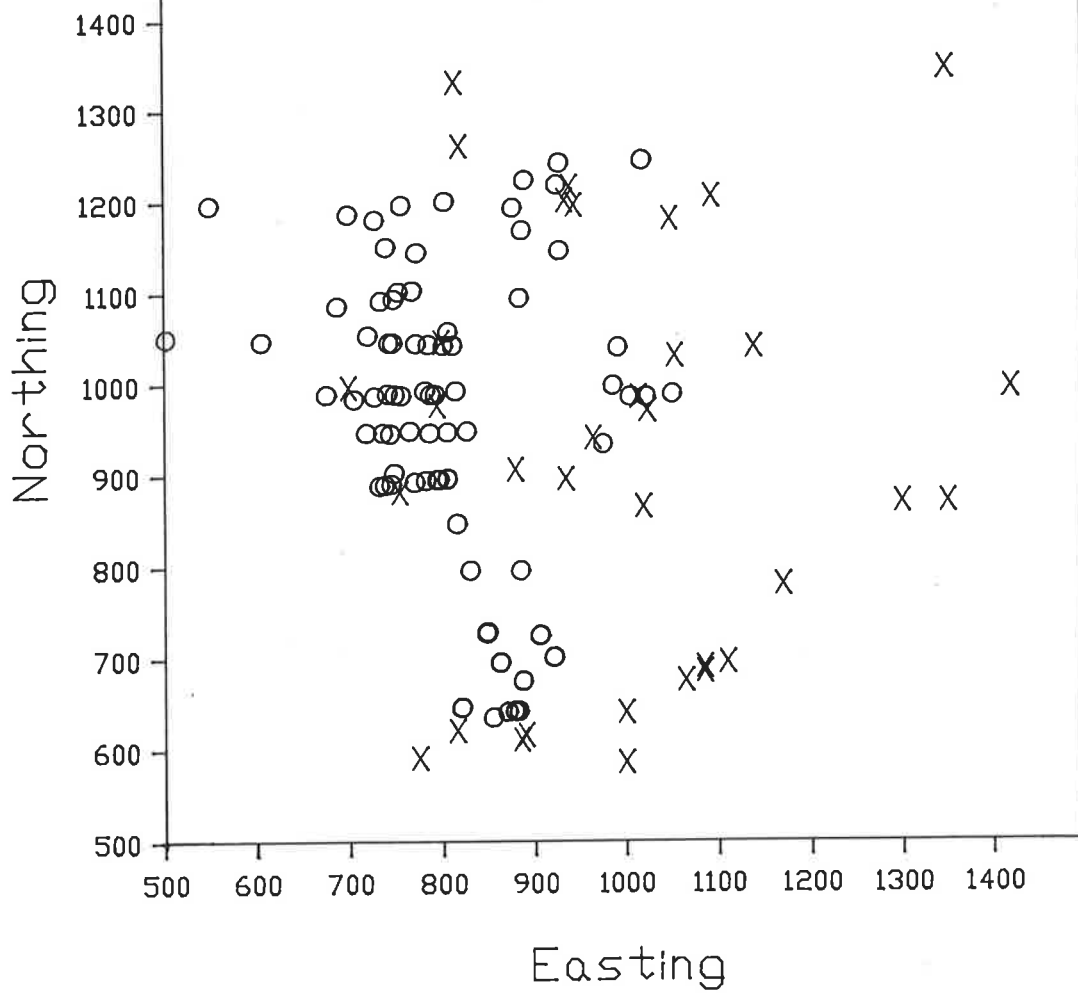


Figure 8.13 Distribution of pyrophyllite at Peak Hill determined from XRD of core samples and interpretation of SWIR laboratory reflectance spectra of surface samples (overlay).

The hatched areas on Figs. 8.14 to 8.18 outline the core samples containing 20% or more of the minerals indicated (as determined by XRD analysis).

Note the good correspondence between the location of the pyrophyllite zone, as determined from XRD of core, with that determined from the spectra of surface samples (overlay).

PEAK HILL MINERALOGY
SPECTRA FROM SURFACE SAMPLES

KEY

Pyrophyllite

- <20%
- 20-50%
- >50%

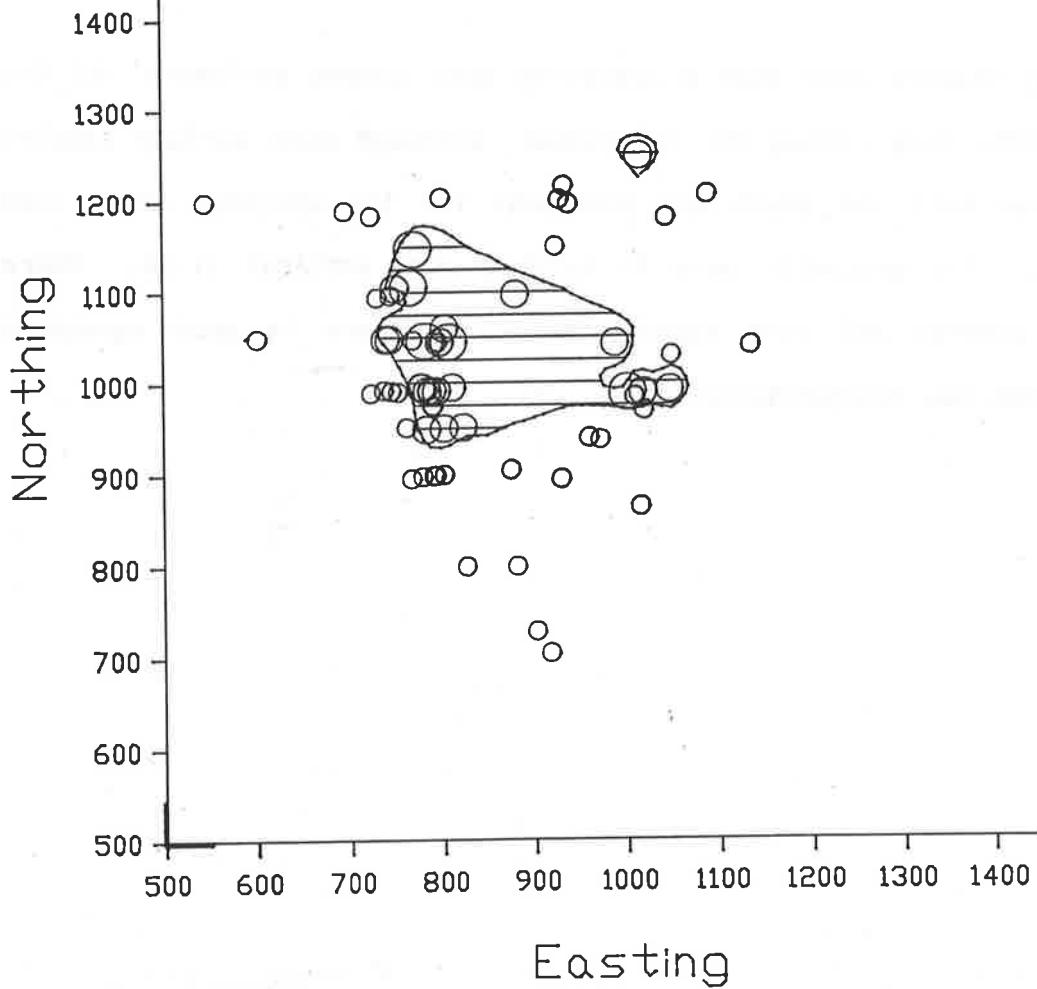


Figure 8.14 Distribution of muscovite at Peak Hill determined from XRD of core samples and interpretation of SWIR laboratory reflectance spectra of surface samples (overlay).

Both studies show that a muscovite zone occurs peripheral to the pyrophyllite zone (shown for reference), although more surface samples were taken from the south and southeast for the spectral study (see overlay). The muscovite zone is hatched with vertical lines. Where surface samples and core samples coincide, there is good agreement between the two interpretations.

PEAK HILL MINERALOGY EXTRAFILM SURFACE SAMPLES

KEY

Muscovite

- <20%
- 20-50%
- >50%

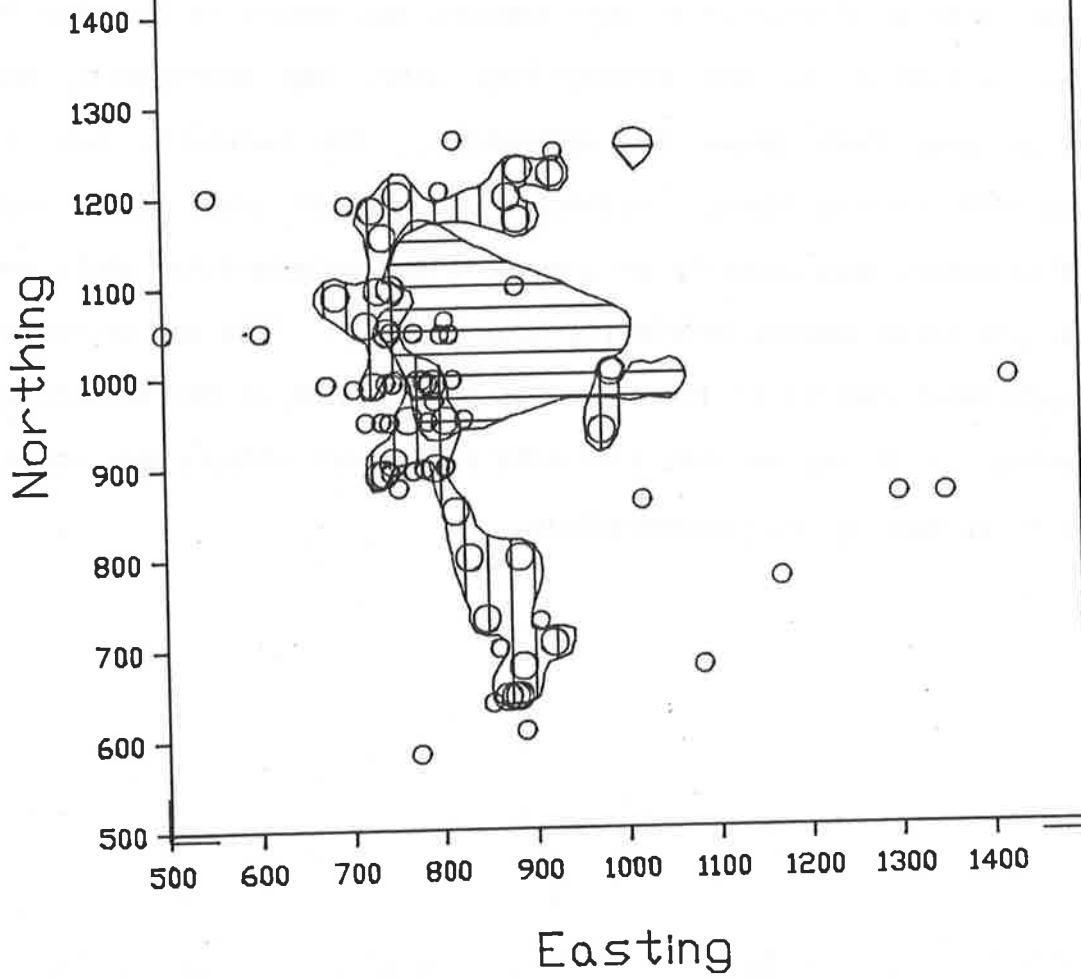


Figure 8.15 Distribution of kaolinite at Peak Hill determined from XRD of core samples and interpretation of SWIR laboratory reflectance spectra of surface samples (overlay).

Kaolinite is widespread in core samples, but occurs in quantity in a zone peripheral to the pyrophyllite zone, and overlapping the muscovite zone (both shown for reference). The kaolinite zone is hatched with sloping lines. Although the spectral study shows that kaolinite occurs most commonly peripheral to the pyrophyllite, there are several kaolinite samples within its zone boundary. This may be due to some additional control of the kaolinite distribution at the surface by weathering, or it may be that kaolinite's spectral effects are strong enough to be seen at low concentrations.

PEAK HILL MINERALOGY SPECTRA FROM SURFACE SAMPLES

KEY

Kaolinite

- <20%
- 20-50%
- >50%

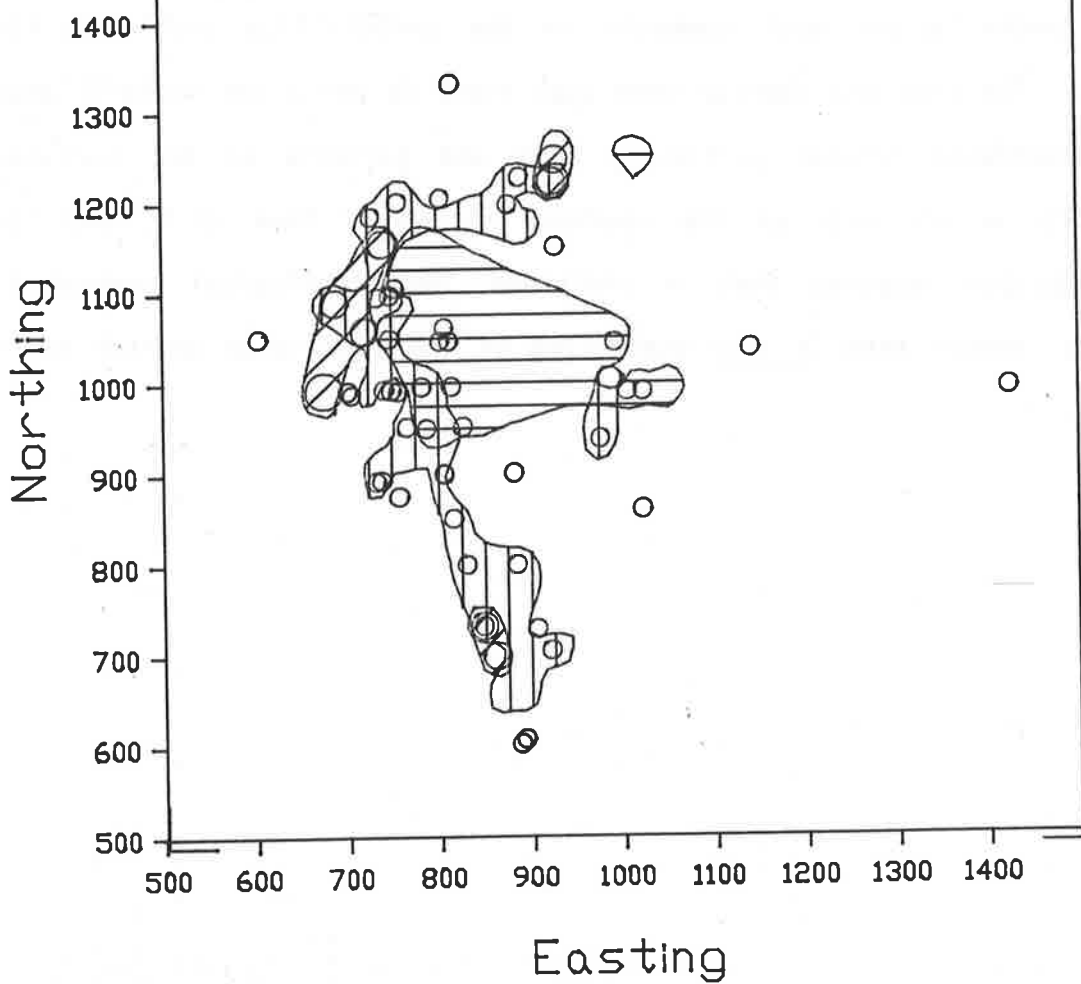


Figure 8.16 Distribution of pyrite at Peak Hill determined from XRD of core samples and distribution of jarosite determined from interpretation of SWIR laboratory reflectance spectra of surface samples (overlay).

Pyrite occurs most commonly in the pyrophyllite and muscovite zones. The plot and overlay show that there is not a particularly good correspondence between pyrite in core and jarosite at the surface. Jarosite occurs only on the southern slopes of Peak Hill, and its distribution suggests that an additional geomorphological control is likely, rather than in situ weathering of the most sulphide-rich rocks alone.

PEAK HILL MINERALOGY
SPECTRA FROM SURFACE SAMPLES

KEY

Pyrite

Jarosite

- <20%
- 20-50%
- >50%

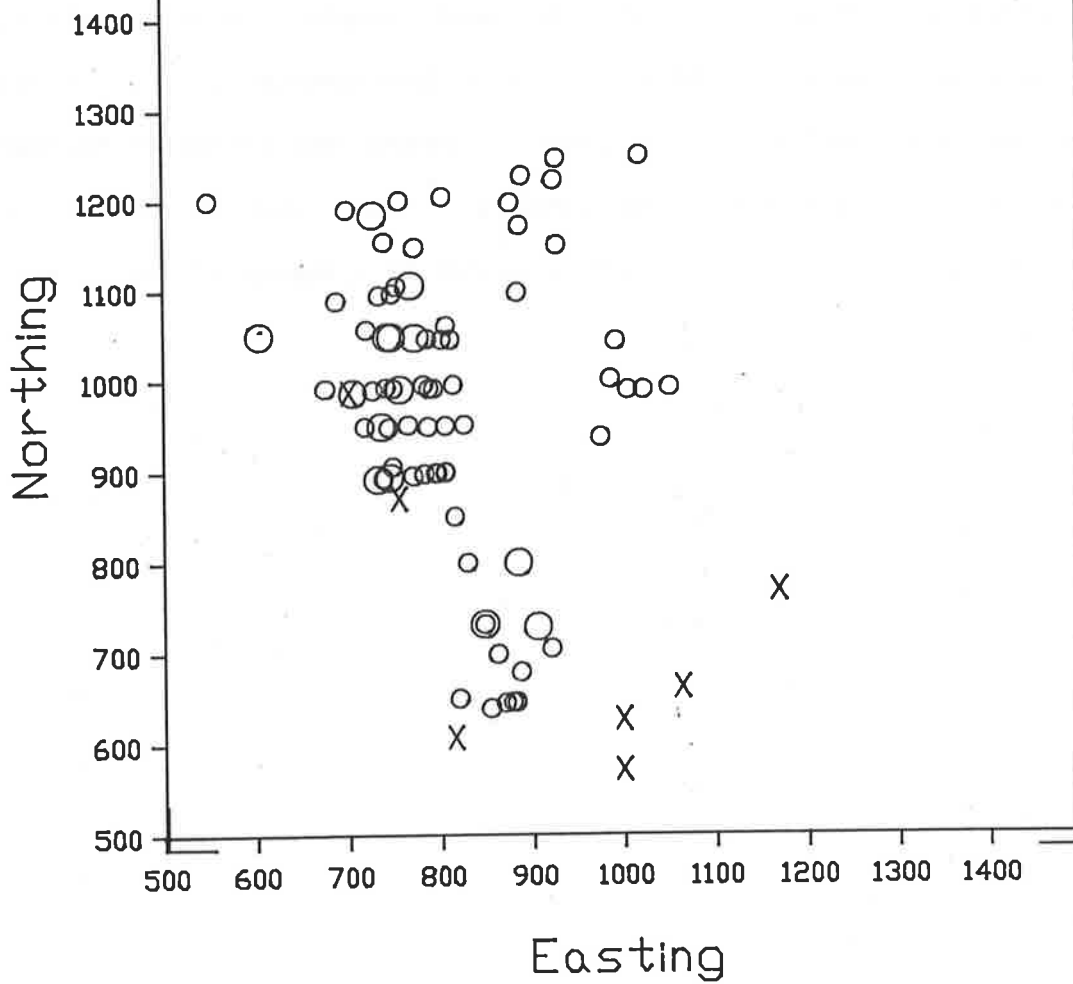


Figure 8.17 Distribution of alunite at Peak Hill determined from XRD of core samples.

Alunite determined from XRD of core samples shows a strong coincidence with the muscovite distribution (see overlay). Alunite was interpreted from spectra of core samples (despite the estimated maximum content of 10% shown for these samples). The absence of alunite features in spectra of surface samples is due to a dearth of the mineral in outcrop.

PEAK HILL MINERALOGY

XRD FROM CORE SAMPLES

KEY

Alunite

Muscovite

All 2-10%

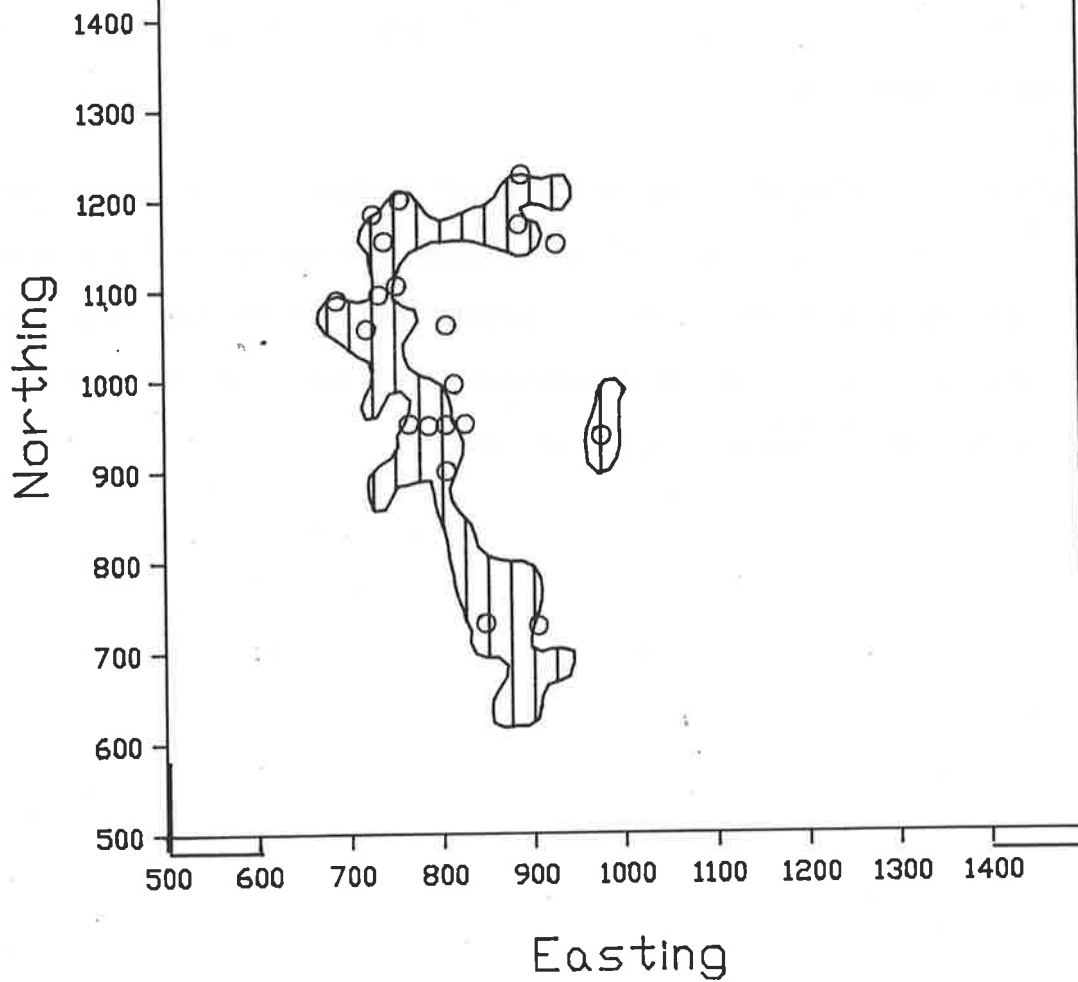


Figure 8.18 Summary diagram showing pyrophyllite, muscovite and kaolinite zones (determined from XRD analyses of core samples) compared to mineralogy interpreted from SWIR spectral features (overlay).

The results from Figs.8.14 to 8.17 have been summarised and presented together here.

Note the location of anomalous spectral features in the north and northeast of the area. Three of these samples have tentatively been interpreted as topaz- and pyrophyllite-bearing, whilst the last has tentatively been interpreted as diaspore-bearing (although XRD analysis has failed to confirm these interpretations).

PEAK HILL MINERALOGY SPECTRA OF SURFACE SAMPLES

KEY

Pyrophyllite	☆
Muscovite	○
Kaolinite	●
Jarosite	x
Topaz + Pyrophyllite ?	★
Diaspore + Pyrophyllite ?	⊙

